Research Article

pISSN 2799-8525 · eISSN 2799-8681 J Powder Mater 2024;31(2):119-125 https://doi.org/10.4150/jpm.2024.00045

Journal of Powder Materials

Exploring Thermoelectric Transport Properties and Band Parameters of *n*-Type Bi_{2-x}Sb_xTe₃ Compounds Using the Single Parabolic Band Model

Linh Ba Vu^{1,2}, Soo-ho Jung¹, Jinhee Bae¹, Jong Min Park¹, Kyung Tae Kim¹, Injoon Son², Seungki Jo^{1,*}

¹Department of 3D Printing Materials, Korea Institute of Materials Science, Changwon, Gyeongnam 51508, Republic of Korea ²Department of Materials Science and Metallurgical Engineering, Kyungpook National University, Daegu 41566, Republic of Korea

The *n*-type $Bi_{2-x}Sb_xTe_3$ compounds have been of great interest due to its potential to achieve a high thermoelectric performance, comparable to that of *p*-type $Bi_{2-x}Sb_xTe_3$. However, a comprehensive understanding on the thermoelectric properties remains lacking. Here, we investigate the thermoelectric transport properties and band characteristics of *n*-type $Bi_{2-x}Sb_xTe_3$ (x = 0.1 - 1.1) based on experimental and theoretical considerations. We find that the higher power factor at lower Sb content results from the optimized balance between the density of state effective mass and nondegenerate mobility. Additionally, a higher carrier concentration at lower *x* suppresses bipolar conduction, thereby reducing thermal conductivity at elevated temperatures. Consequently, the highest *zT* of ~ 0.5 is observed at 450 K for x = 0.1 and, according to the single parabolic band model, it could be further improved by ~70 % through carrier concentration tuning.

Keywords: Bi_{2-x}Sb_xTe₃; thermoelectrics; SPB modeling; Effective mass; Nondegenerate mobility

1. Introduction

Thermoelectric (TE) technology directly converting waste heat into electricity holds significant promise in environmental protection [1, 2]. As the efficiency of thermal-to-electrical energy conversion relies heavily on TE materials, extensive research efforts have been devoted to enhancing TE performance and developing new materials over the past 60 years [3-6]. Bismuth telluride (Bi₂Te₃)-based materials, including (Bi,Sb)₂Te₃ for *p*-type and Bi₂(Te,Se)₃ for *n*-type, have long been recognized for their significant potential in commercial thermoelectric applications, owing to high TE conversion efficiency [7-9]. The materials' TE performance is evaluated by a dimensionless figure of merit $zT = S^2 \sigma T/\kappa_{tot}$ ($\kappa_e + \kappa_L$), where *S*, σ , *T*, κ_{tot} , κ_e and κ_L are the Seebeck coefficient, electrical conductivity, absolute temperature, total thermal conductivity, respectively [10]

LB.Vu, J.M.Park : researcher, SJung, J.Bae, SJo: Senior researcher, K.T.Kim : Principal researcher, I.Son : Professor

© 2024 The Korean Powder Metallurgy & Materials Institute

While *zT* value serves as a primary parameter, balancing between the p- and *n*-type materials also plays a pivotal role in determining the overall performance of a TE device [11]. Over the decades, efforts to improve *zT* of TE materials have led to notable advancements achieved through a diverse array of strategies including nanostructuring [12-14], lattice planification [15], carrier filtering [16, 17], resonant doping [18, 19], band convergence [20, 21] and so forth. However, the TE performance of *n*-type Bi₂(Te,Se)₃ compounds (with *zT* ~1.22 at 323K) [22] lags somewhat behind that of their counterpart *p*-type (Bi,Sb)₂Te₃ (with *zT* ~1.86 at 320K) [23], posing challenges in the development of high-performance TE devices and thereby limiting their practical applications.

The inferior TE performance of *n*-type $Bi_2(Te,Se)_3$ originates from intrinsically unfavorable electronic transport properties, such as high carrier concentration [24], strong dependence of carrier mobility on structural texturing [25], and single valley contribution to electrical properties of Bi_2Se_3 [26]. Recently, Zhou et al. developed a new *n*-type $Bi_{2-x}Sb_xTe_3$, tackling the aforementioned issues in conventional *n*-type $Bi_2(Te,Se)_3$ [11].

Received: April 8, 2024 **Revised:** April 21, 2024 **Accepted:** April 21, 2024

*Corresponding author: Seungki Jo

TEL: +82-55-280-3474 FAX: +82-55-280-3289 E-mail: seungkijo@kims.re.kr It offers several advantages including optimized electron concentration, multi-valley contribution for high *S*, and larger mass differences for low κ , resulting in a high *zT* value of ~1.0 at 330 K for Bi_{1.5}Sb_{0.5}Te_{3.0} even without applying strategies to enhance TE properties [11]. This highlights the possibility of further improving the *zT* of *n*-type Bi₂Te₃-based materials. In this regard, a comprehensive investigation into the TE transport properties and understanding of the electronic band structure of *n*-type Bi_{2.x}Sb_xTe₃ is necessary as a preliminary study.

In this work, we investigated the temperature-dependent TE transport properties of *n*-type $Bi_{2-x}Sb_xTe_3$ (x = 0.1 - 1.1) and characterized the band parameters using the single parabolic band (SPB) model. The calculated weighted mobility (μ_w) was found to be highest at x = 0.1 and decreased with increasing Sb content, indicating that lower Sb content led to more favorable electronic transport properties. Consequently, a maximum power factor $(S^2\sigma)$ of ~2.2 mW m⁻¹ K⁻² was observed at 300 K for x = 0.1. Meanwhile, the higher carrier concentration at lower *x* suppressed the bipolar contribution to the total thermal conductivity (κ_{tot}). As a result, the peak *zT* value for x = 0.1 was the highest, reaching ~ 0.5 at 450 K. Furthermore, SPB model-based estimation of Hall carrier concentration $(n_{\rm H})$ dependent *zT* values suggests that the *zT* value for x = 0.1 could be improved by ~70% (from 0.3 to 0.5) through optimizing $n_{\rm H}$ at room temperature.

2. Experimental Procedure

 $Bi_{2-x}Sb_xTe_3$ (x = 0.1, 0.3, 0.5, 0.7, 0.9, 1.1) powders were synthesized via a ball-milling process. High-purity (~99.999%) Bi, Sb, and Te shots were used as raw materials. They were weighed according to the desired ratio to achieve stoichiometry and ball-milled using planetary mill (Pulverisette 5, Fritsch, Germa-

ny). The ball-milling process consisted of two steps: the first step operated at 360 rpm for 48 cycles, and the second step operated at the same speed for 24 cycles. The obtained powder was reduced in the H2 atmosphere at 573K in 90 min. Cylinder-type bulks (15 mm in diameter and 12 mm in thickness) were fabricated by using spark plasma sintering (SPS) at 753K for 5 min under 30, 40, and 50MPa. The crystalline phases of the ball-milled powders were identified using X-ray diffraction (XRD) with a Cu K_{a1} X-ray source (Malvern Panalytical, United Kingdom). The temperature-dependent electrical conductivity and Seebeck coefficient were measured by SBA458 Nemesis (Netzsch, Germany) from 298K to 473K in an argon atmosphere. The Hall carrier concentration and Hall mobility were measured using the van der Pauw configuration with an HMS3500 (Ecopia, South Korea) at room temperature. The temperature-dependent thermal conductivity was determined using the equation of $\kappa = D\rho C_{\nu}$, where D, ρ and C_{ν} are thermal diffusivity, density and specific heat. Thermal diffusivity was measured by laser flash method (NETZSCH, Germany) and the density was measured using Archimedes method. The specific heat was estimated by the corrected Dulong-Petit law [27, 28]. The electrical and thermal properties were measured along the direction perpendicular to the pressing direction of SPS.

3. Results and Discussion

Fig. 1(a) and Fig. 1(b) shows the X-ray diffraction (XRD) patterns of *n*-type $Bi_{2-x}Sb_xTe_3$ (x = 0.1 - 1.1) ball-milled powders. The peaks are clearly indexed to Bi_2Te_3 (JCPDS #15-0863), indicating the synthesis of single phases of *n*-type $Bi_{2-x}Sb_xTe_3$ (x = 0.1 - 1.1) across all compositions. The peaks shifted to higher angles with increasing *x*, suggesting the change in lattice parameters upon the substitution of Sb at Bi sites. As illustrated in



Fig. 1. (a) X-ray diffraction (XRD) patterns and (b) three close-up views of (a) with the 2θ ranging between 27–29°, 36–43° and 44–46° (c) lattice parameters (a and c) and unit cell volume of n-type Bi_{2-x}Sb_xTe₃ (x = 0.1 - 1.1).

Fig. 1(c), the calculated lattice parameter *a* gradually decreased with increasing Sb content, which can be attributed to the smaller atomic size of Sb compared to Bi ($r_{\rm sb} = 1.45$ Å, $r_{\rm Bi} = 1.60$ Å). Meanwhile, it was observed that the lattice parameter c increased, likely due to the differences in electronegativity between Bi and Sb [11] These results suggest that the successful substitution of Sb atoms at the Bi site without the formation of any second phases.

Fig. 2 shows the electrical conductivity (σ), Seebeck coefficient (S), and power factor $(S^2\sigma)$ as a function of temperature for spark plasma sintered (SPSed) $Bi_{2,x}Sb_xTe_3$ (x = 0.1 - 1.1). As shown in Fig. 2(a), Bi_{1.9}Sb_{0.1}Te₃ exhibited the highest σ at room temperature, ~1895 S cm⁻¹, and decreased with increasing temperature to ~1093 S cm⁻¹ at 473K. The σ of the Bi_{2,x}Sb_xTe₃ compound gradually decreased as the x decreases due to the suppression of donor-like effect by excess Bi_{Te} antisites defect, with the lowest σ of ~237 S cm⁻¹ observed at room temperature for x = 1.1. The measured S values are negative, indicating n-type carrier transport behavior. The absolute value of S increased with increasing x from 108 μ V K⁻¹ (x = 0.1) to 148 μ V K⁻¹ (x = 1.1) at 300 K (Fig. 2b). Also, it is observed that the temperature at which the peak S value occurs shifted to higher temperatures with decreasing Sb content. The changes in σ and S can be ascribed to the trade-off relationship between σ and S, and the differences in temperature dependence of S with Sb content are due to the increased or reduced bipolar contribution at different x [29-33]. The power factor ($S^2 \sigma$) calculated from the measured σ and S exhibited the highest value of ~2.2 mW m⁻¹ K⁻² for x = 0.1 and gradually decreased with increasing x, and reaching ~0.5 mW m⁻¹ K⁻² for x = 1.1 at 300K (Fig. 2c).

To understand the change in electrical properties, we mea-

sured the Hall carrier concentration $(n_{\rm H})$ and Hall mobility $(\mu_{\rm H})$, as shown in Table 1. As expected, both $n_{\rm H}$ and $\mu_{\rm H}$ gradually decreased with increasing x, which accounts for the decrease in σ and the increase in S with increasing Sb content. To gain a deeper understanding on the origin of differences in electronic transport properties of Bi_{2-x}Sb_xTe₃ (x = 0.1 - 1.1), we estimated the density-of-state effective mass ($m_{\rm d}^{*}$) and nondegenerate mobility (μ_0) using the single parabolic band (SPB) model. We obtained $n_{\rm H}$ dependent S ($n_{\rm H}$ -S) curve for a given $m_{\rm d}^{*}$ using the following equations and fitted them to the experimentally measured $n_{\rm H}$ and S (Fig 3a),

$$S = \frac{k_B}{e} \left(\frac{2F_1}{F_0} - \eta \right) \tag{1}$$

$$n_{H} = \frac{16\pi}{3} \left(\frac{2m_{d}^{*}k_{B}T}{h^{2}}\right)^{3/2} \frac{\left(F_{0}(\eta)\right)^{2}}{F_{-1/2}(\eta)}$$
(2)

$$F_j(\eta) = \int_0^\infty \frac{\varepsilon^j d\varepsilon}{1 + \exp[\varepsilon - \eta]}$$
(3)

where $k_{\rm B}$, e, F_{j} , η , and h are the Boltzmann constant, elemental electric charge, Fermi integral of order j, reduced Fermi energy, and Planck's constant, respectively. The estimated values of $m_{\rm d}^*$ of Bi_{2-x}Sb_xTe₃ (x = 0.1 - 1.1) range from ~0.9 for x = 1.1to 1.15 m_0 for x = 0.5 where m_0 is the free electron mass (Fig. 3d). These variations in $m_{\rm d}^*$ could be attributed to changes in band curvature that occur with compositional changes [34]. In addition, we obtained $n_{\rm H}$ dependent $\mu_{\rm H}$ ($n_{\rm H}$ - $\mu_{\rm H}$) curves for a given μ_0 using the following equations (Fig. 3b),



Fig. 2. Temperature-dependent (a) electrical conductivity (σ), (b) Seebeck coefficient (*S*), and (c) power factor ($S^2\sigma$) of spark plasma sintered (SPSed) Bi_{2-x}Sb_xTe₃ (x = 0.1 - 1.1).

Table 1. Hal	l carrier	concentration ($(n_{\rm H})$) and H	Hall	mob	ility	$(\mu_{\rm H})$	of	Bi _{2-x}	Sb _x 7	Г е 3 (x =	0.1	- 1	.1)	measured	l at room	tem	peratu	re
--------------	-----------	-----------------	---------------	---------	------	-----	-------	-----------------	----	-------------------	-------------------	----------------	-----	-----	-----	-----	----------	-----------	-----	--------	----

	x = 0.1	x = 0.3	x = 0.5	x = 0.7	x = 0.9	x = 1.1
$n_{\rm H} (10^{19}{\rm cm}^{-3})$	7.74	6.94	5.86	4.40	4.10	3.25
$\mu_{\rm H} ({\rm cm}^2{\rm V}^1{\rm s}^{-1})$	151.4	127.0	110.6	85.0	60.0	44.9



Fig. 3. Hall carrier concentration ($n_{\rm H}$) dependent (a) Seebeck coefficient (*S*), (b) Hall carrier mobility ($\mu_{\rm H}$), and (c) power factor ($S^2\sigma$) of Bi₂₋ _xSb_xTe₃ (x = 0.1 - 1.1). (d) density-of-state effective mass ($m_{\rm d}^*$), (e) nondegenerate mobility (μ_0), and weighted mobility (μ_w) as a function of x. The single parabolic band (SPB) modeling was conducted using experimentally measured values at room temperature.

$$\mu_H = \mu_0 \frac{F_{-1/2}(\eta)}{2F_0(\eta)} \tag{4}$$

$$\mu_0 = \frac{2^{3/2} \pi^{1/2} \hbar^4 e C_l}{3(m_b^*)^{5/2} (k_B T)^{3/2} E_{def}^2} \tag{5}$$

where $C_b m_b^*$ and E_{def} are the elastic constant, single band effective mass and deformation potential, respectively. The estimated μ_0 values range from ~245 cm² V¹ s⁻¹ for x = 0.1 to 63 cm² V⁻¹ s⁻¹ for x = 1.1, showing a decreasing trend with increasing x (Fig. 3e). Given that μ_0 is inversely proportional to the m_b^* , it is reasonable that μ_0 decreased with increasing x up to x = 0.5. However, for x > 0.5, the m_d^* and μ_0 decreased simultaneously, which might be a result of the intensified carrier scattering induced by enhanced atomic disorder [35]. Fig. 3c shows the $n_{\rm H}$ -dependent power factor ($n_{\rm H}$ -S² σ) curve predicted by the SPB

model, which agrees well with experimentally obtained $S^2\sigma$ at all compositions. Also, it is noted that with smaller x, $n_{\rm H}$ deviates more from the optimized value for maximum $S^2\sigma$, suggesting that the electrical properties of ${\rm Bi}_{2-x}{\rm Sb}_x{\rm Te}_3$ (x = 0.1 - 1.1) can be improved by adjusting $n_{\rm H}$. The weighted mobility ($\mu_w = \mu_0 (m_{\rm d}/m_0)^{3/2}$) as a function of x was presented in Fig. 3(f), calculated using theoretically estimated $m_{\rm d}^*$ and μ_0 (Fig. 3f). The μ_w is the primary parameter for the electrical properties of a given material, as the maximum $S^2\sigma$ is in direct proportion to the μ_w [36]. As shown in Fig. 3(f), the calculated μ_w gradually decreases as x increases. Therefore, it is concluded that ${\rm Bi}_{2-x}{\rm Sb}_x$. Te₃ exhibits the most favorable electronic transport properties at x = 0.1 thanks to the optimized balance between $m_{\rm d}^*$ and μ_0 .

The temperature-dependent thermal properties of $\text{Bi}_{2-x}\text{Sb}_x\text{Te}_3$ (x = 0.1 - 1.1) are shown in Fig. 4. As *x* increases, the total thermal conductivity (κ_{tot}) decreases and the lowest κ_{tot} of ~1.02 W m⁻¹ K⁻¹ was observed at x = 1.1, which should be attributed to the reduced electronic contribution as we observed in Fig. 2(a). To further understand the changes in thermal properties, we calculated the lattice thermal conductivity ($\kappa_{\rm L} = \kappa_{\rm tot} - \kappa_{\rm e}$) by subtracting the $\kappa_{\rm e}$ from the $\kappa_{\rm tot}$. The $\kappa_{\rm e}$ was determined using the Wiedemann-Franz law ($\kappa_{\rm e} = L_0 \sigma T$), where L_0 is the Lorenz number (in 10⁻⁸ W Ω K⁻²). We determined L_0 using the following equation proposed by Kim et al [37].

$$L_0 = 1.5 + \exp\left(-\frac{|S|}{116}\right)$$
(6)

The $\kappa_{\rm L}$ exhibited a decreasing trend as the Sb content increased with the lowest $\kappa_{\rm L}$ of ~0.9 W m⁻¹ K⁻¹ observed at x = 1.1 (Fig. 4b). The reduction of $\kappa_{\rm L}$ with increasing x could be attributed to the intensified point defect scattering as discussed in Fig. 3 and indicated by the following equation [29],

$$\Gamma = x(1-x) \left[\left(\frac{\Delta M}{M} \right)^2 + \varepsilon \left(\frac{a_{\text{disorder}} - a_{\text{pure}}}{a_{\text{pure}}} \right)^2 \right] \quad (7)$$

where *x* is the doping fraction, $\Delta M/M$ is the rate of change of relative atomic mass, a_{disorder} and a_{pure} represent the lattice constants of disordered and pure alloys, and ε is an elasticity. It is also worth noting that κ_{L} decreased at elevated temperature with increasing *x*, suggesting the suppressed contribution of bipolar carriers due to increased n_{H} [33].

The temperature-dependent figure of merit zT was evaluated from the measured σ , S, and κ_{tot} (Fig. 4c). The maximum zTvalue of ~0.5 was observed at 450 K for the x = 0.1 sample and the zT values decreased with increasing x within the measured temperature ranges. Furthermore, we obtained the n_{H} -dependent zT (n_{H} -zT) curve using the SPB model (Fig. 4d), and it was found that the zT values predicted based on the SPB model agree well with the experimentally evaluated ones. It is note-



Fig. 4. Temperature-dependent (a) total thermal conductivity (κ_{tot}), (b) lattice thermal conductivity (κ_L), and (c) figure of merit *zT* of Bi₂₋ _xSb_xTe₃ (x = 0.1 - 1.1). (d) Hall carrier concentration (n_H) dependent zT (n_H -*zT*) curves were estimated at room temperature using the single parabolic band (SPB) model.

worthy that $n_{\rm H}$ deviates more from the optimized value with lower Sb content (smaller *x*), indicating the need to optimize $n_{\rm H}$ at lower *x* to further improve *zT*. According to the SPB model, *zT* can be improved by up to ~70% (e.g., from ~0.3 to ~0.5 for x = 0.1) by tuning $n_{\rm H}$ at room temperature (Fig. 4d).

4. Conclusion

In this study, we investigated the thermoelectric properties and related band parameters of Bi₂, Sb₂Te₃ (x = 0.1 - 1.1) using both experimental and theoretical approaches. Our findings, based on the estimation of weighted mobility (μ_w) , revealed that the electronic transport properties are most favorable for x =0.1, attributed to the balanced density-of-state effective mass (m_d^*) and nondegenerate mobility (μ_0) . Additionally, at lower Sb content, the bipolar contribution to thermal conductivity is reduced at elevated temperatures. These synergetic effects led to the highest zT value of ~0.5 at 450 K in Bi_{1.9}Sb_{0.1}Te₃. Furthermore, theoretical predictions by the single parabolic band (SPB) model suggest that the zT values can be further improved by up to ~70% through manipulating the carrier density. We believe that the current study not only contributes to the basic understanding of the *n*-type $Bi_{2,x}Sb_xTe_3$ but also expedites the development of high-performance *n*-type Bi₂Te₃-based thermoelectric materials.

Acknowledgments

This work was supported by KIMS' Principal R&D project (PNK9950) and the NRF' project (code no. 2021M3C1C30 97540)of the Republe of Korea.

References

- K. T. Kim, T. Min, and K. D. Won: J. Powder Mater., 23 (2016) 263.
- [2] J. Bae, S. Jo, and K. T. Kim: J. Powder Mater., **30** (2023) 318.
- [3] H. Jouhara, A. Żabnieńska-Góra, N. Khordehgah, Q. Doraghi, L. Ahmad, L. Norman, B. Axcell, L. Wrobel, and S. Dai: Int. J. Thermofluids., 9 (2021) 100063.
- [4] S. Twaha, J. Zhu, Y. Yan, and B. Li: Renew. Sustain. Energy Rev., 65 (2016) 698.
- [5] D. Enescu and E. O. Virjoghe: Renew. Sustain. Energy Rev., 38 (2014) 903.
- [6] M. H. Elsheikh, D. A. Shnawah, M. F. M. Sabri, S. B. M. Said, M. H Hassan, M. B. A. Bashir, and M. Mohamad: Re-

new. Sustain. Energy Rev., 30 (2016) 337.

- [7] K. T. Lee, D. S Lee, W. H. Chen, Y. L. Lin, D. Luo, Y. K Park, and A Bandala: iScience., 26 (2023) 107874.
- [8] G. Yang, D.R.G. Mitchell, F. F. Yun, K.W. See, A. J. Ahmed, S. Sayyar, A. Bake, P. Liu, L.i Chen, Z. Yue, D. Cortie, and X. Wang: J. Chem. Eng., 428 (2022) 131205.
- [9] W. Xie, X. Tang, Y. Yan, Q. Zhang, and T. M. Tritt: J. Appl. Phys., 105 (2009) 113713.
- [10] I. T. Witting, T. C Chasapis, F. Ricci, M. Peters, N. A Heinz, G. Hautier, and G. J Snyder: Adv. Electron. Mater., 5 (2019) 1800904.
- [11] Y. Zhou, F. Meng, J. He, A. Benton, L. Hu, F. Liu, J. Li, C. Zhang, W. Ao, and H. Xie: ACS Appl. Mater. Interfaces., 12 (2020) 31619.
- [12] H. Jung, Y. J. Woo, K. T. Kim, and S. Jo: J. Powder Mater., 30 (2023) 123.
- [13] S. R. Szczech, J. M. Higgins, and S. Jin: J. Mater. Chem., 21 (2011) 4037.
- [14] Y. Zheng, Q. Zhang, X. Su, H. Xie, S. Shu, T. Chen, G. Tan, Y. Yan, X. Tang, C. Uher, and G. J. Snyder: Adv. Energy Mater., 5 (2014) 1401391.
- [15] H. Pang, Y. Qin, B. Qin, L. Yu, X. Su, H. Liang, Z. H. Ge, Q. Cao, Q. Tan, and L. D. Zhao: Adv. Funct. Mater., (2024) 2401716.
- [16] C. Gayner and Y. Amouyal: Adv. Funct. Mater., 30 (2020) 1901789.
- [17] J. Choi, J. Y. Lee, S. S. Lee, C. R. Park, and H. Kim: Adv. Energy Mater., 6 (2016) 1502181.
- [18] S. Kim, J. Hwang, T. S. You, S. Yeon, J. Kim, B. K. Yu, M. K. Han, M. Lee, S. Acharya, J. Kim, W. Kim, and S. Kim: Phys. Rev. Appl., **19** (2023) 014034.
- [19] M. Zhou, Z. M. Gibbs, H. Wang, Y. Han, L. Li, and G. J. Snyder: Appl. Phys. Lett., 109 (2016) 042102.
- [20] S. Ahmad, A. Singh, S. Bhattacharya, M. Navaneethan, R. Basu, R. Bhatt, P. Sarkar, K. N. Meshram, A. K. Debnath, K. P. Muthe, and D. K. Aswal: ACS Appl. Energy Mater., 3 (2020) 8882.
- [21] K. H. Lee, S. Kim, H. S. Kim, and S. W. Kim: ACS Appl. Energy Mater., 3 (2020) 2214.
- [22] D. Li, J. M. Li, J. C. Li, Y. S. Wang, J. Zhang, X. Y. Qin, Y. Cao, Y. S. Li, and G. D. Tang: J. Mater. Chem., 6 (2018) 9642.
- [23] S. I. Kim, K. H. Lee, H. A. Mun, H. S. Kim, S. W. Hwang, J. W. Roh, D. J. Yang, W. H. Shin, X. S. Li, Y. H. Lee, G. J. Snyder, and S. W. Kim: Science., **348** (2015) 109.
- [24] T. Zhu, L. Hu, X. Zhao, and J. He: Adv. Sci., 3 (2016)

1600004.

- [25] N. W. Park, W. Y. Lee, Y. S. Yoon, G. S. Kim, Y. G. Yoon, and S. K. Lee: ACS Appl. Mater. Interfaces., 11 (2019) 38247.
- [26] G. R. Miller, C.-Y. Li, and C. W. Spencer: J. Appl. Phys., 34 (1963) 1398.
- [27] K.A. Borup, J. Boor, H. Wang, F. Drymiotis, F. Gascoin, X. Shi, L. Chen, and G.J. Snyder: Energy Environ. Sci., 8 (2015) 423.
- [28] H. Wang, W. D. Porter, J. Sharp, J. Lo, H. Kleinke, and L. Kiss: J. Electron. Mater., 42 (2013) 1073.
- [29] Y. Sun, Y. Liu, R. Li, Y. Li, and S. Bai: Front. Chem., 10 (2022) 865281.
- [30] X. Zhou, Y. Yan, X. Lu, H. Zhu, X. Han, G. Chen, and Z. Ren: Mater. Today., 21 (2018) 974.

- [31] Z. Chen, X. Zhang, J. Ren, Z. Zeng, Y. Chen, J. He, L. Chen, and Y. Pei: Nat. Commun., 12 (2021) 3837.
- [32] S. Foster and N. Neophytou: Comput. Mater. Sci., 164 (2019) 91.
- [33] J. J. Gong, A. J. Hong, J. Shuai, L. Li, Z. B. Yan, Z. F. Renb, and J. M. Liu: Phys. Chem. Chem. Phys., 18 (2016) 16566.
- [34] S. H. Namiki, M. Kobayashi, K. Nagata, Y. Saito, N. Tachibana, and Y. Ota: Mater. Today Sustain., 18 (2022) 100116.
- [35] F. Li, X. Liu, N. Ma, L. Chen, and L. M. Wu: Angew. Chem. Int. Ed., 134 (2022) e202208216.
- [36] G. J. Snyder, A. H. Snyder, M. Wood, R. Gurunathan, B. H. Snyder, and C. Niu: Adv. Mater., 32 (2020) 2001537.
- [37] H. S. Kim, Z. M. Gibbs, Y. Tang, H. Wang, and G. J. Snyder: APL Mater., 3 (2015) 041506.