

국내 리튬이온전지 재활용 산업현황

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Lithium Ion Battery Recycling Industry in South Korea

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요 약

이 글은 현재 국내 리튬이온전지 상용 재활용 공정을 정리하고, 리튬이온전지 재활용 공정의 새로운 방향을 제시한다. 대표적인 리튬이온전지 재활용 업체인 (주)성일하이텍은 10년 이상 리튬이온전지 재활용 공정을 성공적으로 운영해 왔으며 최근 많은 재활용 업체 및 배터리 제조업체들이 새로운 재활용 공정을 제안하고 개발하고 있다. 새로운 재활용 공정에서는 리튬 가격의 급격한 상승으로 니켈과 코발트보다 먼저 리튬이 회수되고, 금속 황산염 용액을 최종 제품으로 배터리 제조업체에 공급하는 특징이 있다. 향후 대량으로 발생할 폐전지 처리를 위해 기존 공정이 개선될 필요가 있으며, 폐기된 자동차와 함께 유입되는 성분들과 리튬이온전지의 새로운 첨가제는 향후 리튬이온전지 재활용 공정에서 주요 공정효율 저감 요인이 될 수 있다.

주제어 : 리튬이온전지, 재활용, 니켈, 코발트

Abstract

The objective of this article is to summarize the commercial lithium ion battery (LIB) recycling processes in Korea and to suggest new direction for LIB recycling. A representative LIB recycler, SungEel Hitech Co. has successfully operated the LIB recycling process for over 10 years, and new recycling processes were recently proposed or developed by many recycling companies and battery manufacturers. In the new recycling processes, lithium is recovered before nickel and cobalt due to the rapid rise in lithium prices, and metal sulfate solution as final product of recycling process can be supplied to manufacturers. The main problem that the new recycling process will face is impurities, which will mainly come from end-of-life electric vehicles or new additives in LIB, although the conventional processes must be improved for mass processing.

Key words : lithium ion battery, recycling, nickel, cobalt

1. Introduction

In the 21th United Nations Framework Convention

on Climate Change Conference of the Parties (UNFCCC COP21), the Paris Agreement was adopted on 12 December 2015 and entered into force on 4 November 2016, and its

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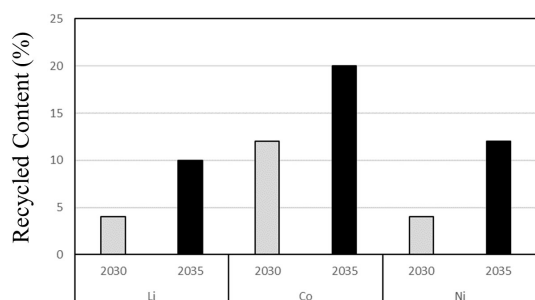


Fig. 1. Recycled contents of Li, Co, and Ni in EV battery by EU in 2030 and 2035.

goal is to limit global warming to 1.5 °C¹⁾. The Korean government has declared Carbon Neutrality in 2020, and announced ‘Carbon Neutrality 2050 Scenario’ prepared by the 2050 Carbon Neutral Green Growth Committee²⁾, which was established based on the ‘Act On Carbon Neutrality And Green Growth For Coping With Climate Crisis³⁾’. In this scenario, the use of electric vehicles has been encouraged by the Korean government because one of main goals is to reduce the carbon footprint in a transport field.

The Korean government announced the development strategy for future vehicle industry in 2019, where the government set the share of fuel cell and battery electric vehicles at 30% until 2030. Another development strategy entitled ‘2030 secondary battery industry development strategy’ (K-battery strategy) was prepared by the Korean government. Three detailed strategies are securing unrivaled 1st-technology, building a global leading base, and expanding the secondary battery market, where the market also contains battery recycling industries. The recycling industries can be classified into remanufacturing and recycling industries, and the representative products are the batteries for energy storage system (ESS) from remanufacturing processes and metal sulfates such as CoSO₄ and NiSO₄ from recycling processes, respectively.

A new EU regulatory framework was announced in 2021, and it includes a recycled-content declaration requirement⁴⁾. This regulation would apply to various batteries such as industrial batteries and electric vehicle

batteries, which contain lithium, cobalt, nickel, or manganese in active materials. As shown in Fig. 1, mandatory minimum levels of recycled content would be set to 12% cobalt, 4% lithium, and 4% nickel in 2030, increasing to 20% cobalt, 10% lithium, and 12% nickel in 2035⁴⁾. These regulation makes vehicle and battery manufacturers aware of the importance of battery recycling.

The Korea has been found as a representative resources-poor country^{5,6)}. Furthermore, battery minerals such as cobalt and nickel are limited to a few countries, e.g., 71% of cobalt is produced in the Democratic Republic of the Congo, and nickel is mainly produced in Indonesia⁷⁾. Therefore, the Korean government has recognized the importance of establishing a secure supply-chain of critical mineral for the domestic battery industries by securing critical minerals such as Li, Co, and Ni through recycling, and the government has supported small business recycling companies by R&D funding⁸⁾. In the present article, the commercial lithium ion battery (LIB) recycling processes in Korea were summarized, and new technological and industrial challenges were discussed to suggest new direction for LIB recycling.

2. Commercial LIB recycling processes in Korea

The first LIB recycling company in Korea is Sungil Hitech Co., Ltd. and after establishing a LIB recycling factory in 2008 and operating a pretreatment process called ‘Recycling Park’, it has operated a hydrometallurgical process called ‘Hydro Center’ since 2011 to produce metal sulfate supplied to battery manufacturers. This recycling process was developed together with Korea Institute of Geoscience and Mineral Resources (KIGAM) funded by ‘Resources Recycling R&D Center’ project of Korean Government⁸⁾. Therefore, the brief summary of the research conducted by KIGAM is as follows.

Early lithium-ion battery recycling studies were conducted on lithium cobalt oxide (LCO) cathode materials. Lee and Rhee proposed the LCO anode recycling process,

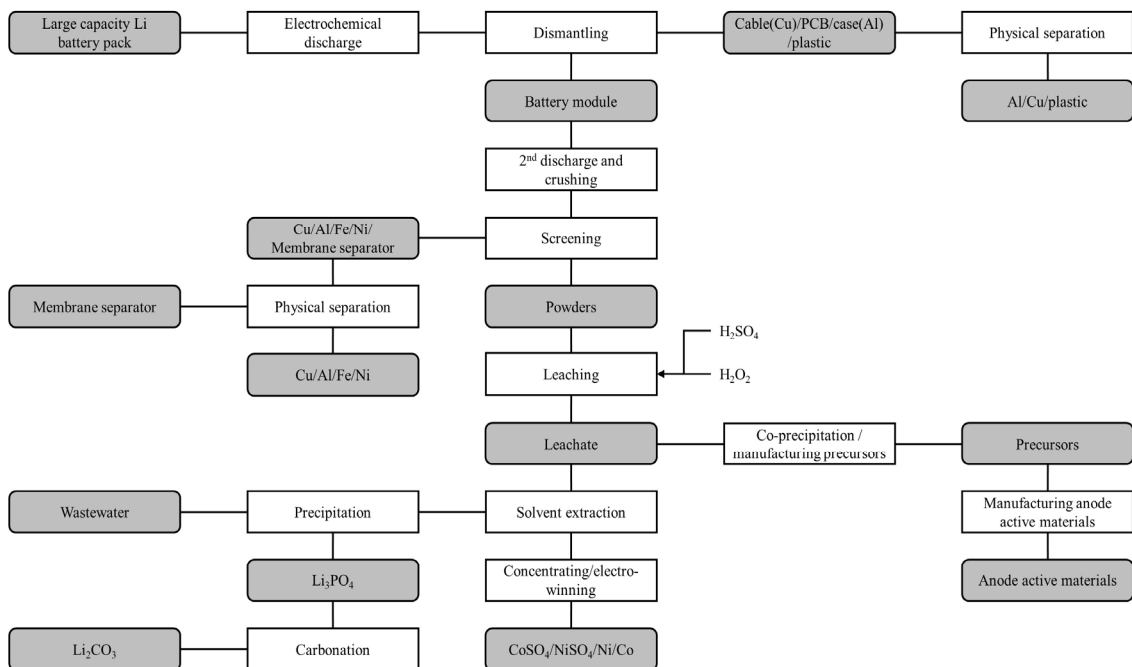


Fig. 2. The recycling process of a spent large capacity Li battery pack proposed by KIGAM.

in which, after a two-step thermal and mechanical treatment, the anode concentrate was leached in 1 M HNO_3 solution with H_2O_2 and then lithium and cobalt were recovered by the amorphous citrate precursor process^{9,10}. The H_2O_2 in this leaching process was used as a reducing agent, which could reduce the oxidation state of metals from +3 to +2⁴. The reductive leaching system was changed from HNO_3 with H_2O_2 to H_2SO_4 with H_2O_2 ^{11,12}.

As demand for lithium-ion batteries increased, it was expected that the use of NCM (nickel-cobalt-manganese) type cathode material increases. The reductive leaching process can be also applied to NCM cathode recycling process, but cobalt, nickel, and manganese must be separated into each metal component¹³⁻¹⁷. The separation can be achieved by solvent extraction processes using Cyanex 272¹³, PC88A and Versatic 10 acid¹⁵, or Versatic 10 acid and LIX 84-I¹⁶. Fig. 2 shows a recycling process based on the previous studies summarized above⁸.

The SungEel Hitech Co. Ltd. has continuously im-

proved the recycling process for last ten years, and the process was summarized as shown in Fig. 3¹⁸. The process is divided into recycling park and hydro center. The recycling park consists of discharging, dismantling, heat treatment, and crushing/grinding for producing battery powder, which is the mixture of anode and cathode powders. In the hydro center, the battery powder is dissolved in sulfuric acid with hydrogen peroxide, and then separated by solvent extraction into each metal component. SungEel Hitech Co. has the plan to construct 30 recycling parks and 5 hydro centers around the world as shown in Fig. 4. The hydro center in Gunsan head-quarter in Korea produces nickel and cobalt compounds by receiving the battery powders from 8 recycling parks. The capacity of recycling park and hydro center is plan to be 173,000 ton and 14,320 ton, respectively, as shown in Fig. 5. Two hydro centers will be constructed in North America and Europe, and the capacity will reach 20,080 ton/year.

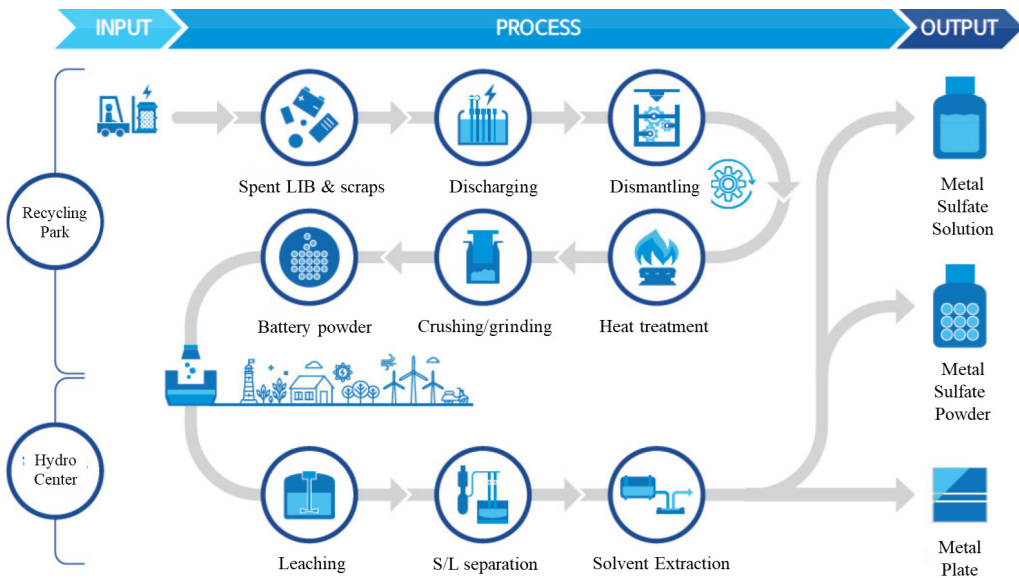


Fig. 3. The LIB recycling process operated by SungEel Hitech Co.

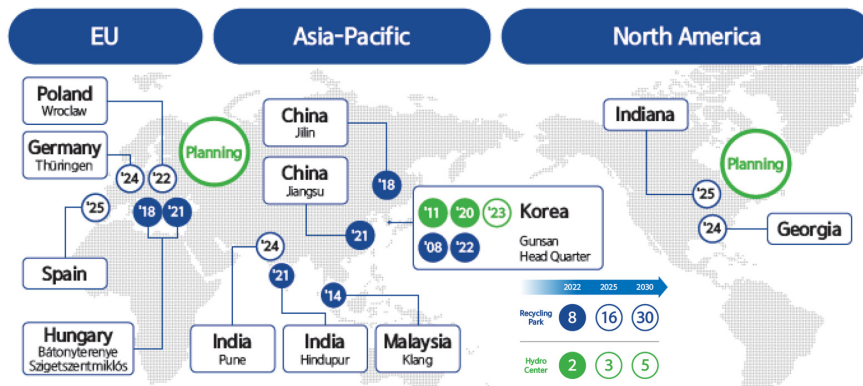


Fig. 4. Recycling parks and hydro centers around the world operated by SungEel Hitech Co. Ltd.

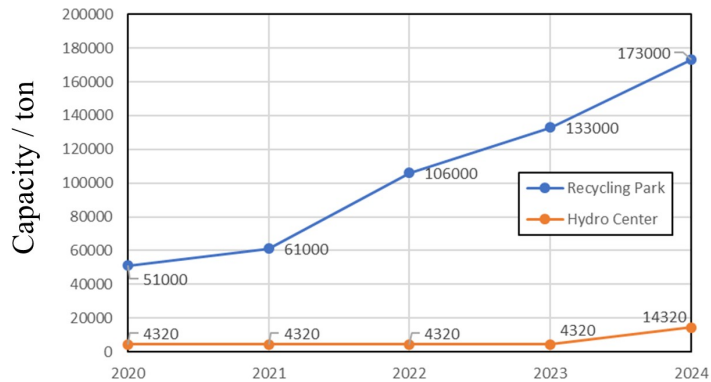


Fig. 5. The capacity of recycling park and hydro center by SungEel Hitech, Co.

3. New challengers and their strategies

As the importance of the battery mineral supply chain has been recognized, many companies in Korea have increased their interest in the battery recycling industry. The companies and their location are summarized in Fig. 6. The representative challenger is EcoPro C&G (clean and green) Co., a subsidiary of EcoPro Group. As shown in Fig. 7, EcoPro group built the closed loop system for

lithium ion battery manufacturers, in which EcoPro CNG provides metals, obtained from spent LIB and scraps, to EcoPro Materials and EcoPro innovation, respectively¹⁹⁾. The EcoPro Materials and EcoPro innovation received the metal sources from mineral resources as well as EcoPro CNG.

In the conventional recycling processes, main final products are metal sulfate such as NiSO_4 or CoSO_4 . However, since EcoPro CNG can supply their product

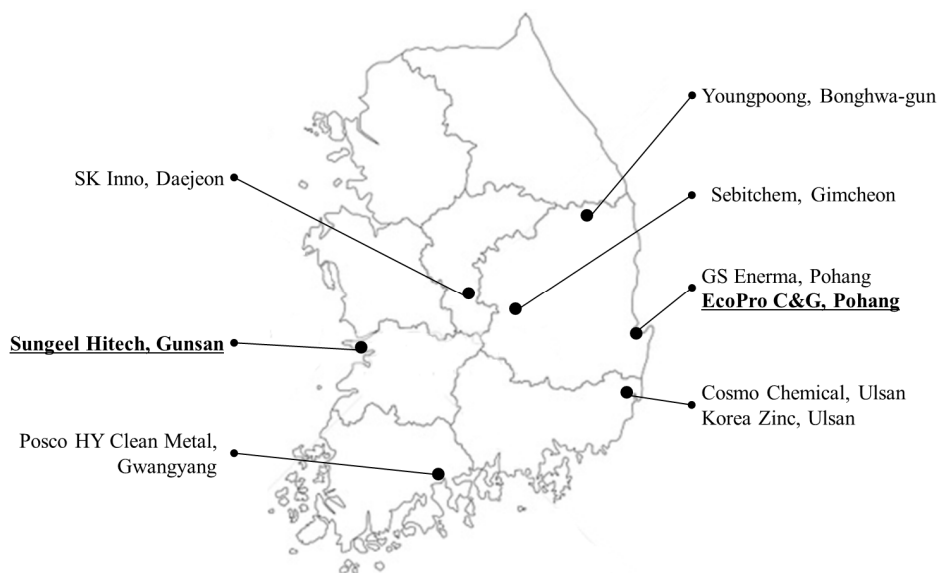


Fig. 6. The name and location of the companies operating or planning a LIB recycling business.

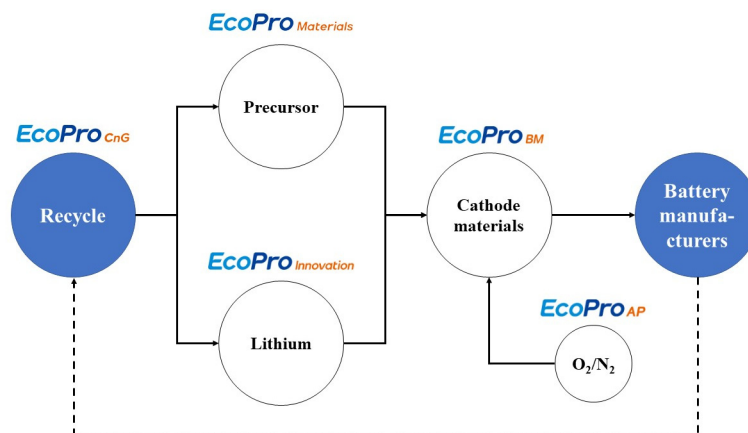


Fig. 7. The closed loop system of EcoPro group.



Fig. 8. Lithium price trend from 2012 to present.

directly to EcoPro Materials, a precursor manufacturer, EcoPro CNG can deliver the nickel and cobalt as metal solution. Therefore, as shown in Fig. 3, SungEel Hitech now supplies the metal sulfate solution to manufacturers.

Many new challengers developed their LIB recycling process considering the recycling process developed by KIGAM and SungEel Hitech Co. However, when the recycling process of SungEel Hitech Co. was developed, the recovery of lithium is not an important issue because lithium prices are relatively low as shown in Fig. 8. The price is only 41 RMB/kg in November, 2012, but it increased rapidly after 2020, rising to 575.5 RMB/kg in November, 2022. Therefore, SungEel Hitech Co. added a Li recovery process after the recovery of Ni and Co by solvent extraction process. Although the new challengers didn't disclose their new LIB recycling processes, it has been found that lithium is recovered before nickel and cobalt in a new recycling process.

Sebit Chem Co. Ltd has operated the recycling process after 2017 and supply the metal solution to battery manufacturers. Posco HY Clean Metal Co., GS Enerma, and Cosmo Chemical Co. are building the recycling facilities. The Posco HY Clean Metal Co. is a joint venture between Posco Co. and Huayou Co. SK Innovation Co. and Youngpoong Co. are now testing their pilot plants based on their own recycling processes. Most of LIB

recycling processes are based on the hydrometallurgical processes, but the new recycling process by Youngpoong Co. consists of pyrometallurgical processes, and the pilot plant is now operated with 2,000 ton/year.

Many joint ventures were announced for LIB recycling; LG Energy Solution and GM, POSCO and GS, SK Innovation and SungEel Hitech, LG Energy Solution and Huayou Cobalt, LG Chem and Korea Zinc, LG Energy Solution and KEMCO. Other battery manufacturers such as Samsung SDI and SK ON also develop their LIB recycling processes. All companies are building or have plans to construct recycling facilities with a capacity of 10,000 ton/year or more, and there is an increasing concern about the competition between LIB recycling companies.

4. Conclusions

It is important to establish a stable supply chain of metals such as Li, Ni, and Co for the continuous development of the domestic battery manufacturing industry. Therefore, the Korean government has supported small business recycling companies by R&D funding. SungEel Hitech Co. has run the LIB recycling facility successfully for last 10 years. New LIB recycling processes were proposed and constructed, where some improvement

was achieved; lithium is recovered before Ni and Co, and metal sulfate solution is supplied directly to manufactures. However, the new recycling processes should overcome the following issues.

1. The conventional recycling processes have treated the scraps obtained from battery manufactures. However, as the end of life EV will increase, the EV contaminated with impurities will be treated.
2. The conventional recycling processes have treated only LCO (lithium cobalt oxides) or NCM (nickel-cobalt-manganese) battery, but, as the use of LFP (lithium iron phosphate) in EV increases, the mixed batteries of LCO, NCM, and LFP should be treated by the recycling processes.
3. The manufacturing technologies and processes for LIB are continuously being improved. New additives are investigated and developed to improve the stability of LIB. These kinds of additives act as impurities in the recycling processes.

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References

1. Jung, E. and Kim, J., 2022 : Greenhouse Gas Emission Reduction and Economic Benefit Evaluation of Carbon Mineralization Technology using CFBC Ash, Resources Recycling, 31(3), pp.40-52.
2. The 2050 Carbon Neutral Green Growth Committee, Carbon Neutrality 2050 Scenario, <https://2050cnc.go.kr>, January 1st, 2023.
3. Korean Law Information Center, Act On Carbon Neutrality And Green Growth For Coping With Climate Crisis. <https://www.law.go.kr/LSW/eng/engLsSc.do?menuId=2§ion=lawNm&query=Act+On+Carbon+Neutrality+And+Green+Growth+For+Coping+With+Climate+Crisis&x=28&y=27#libGcolor0>, January 1st, 2023.
4. Ahn, Y., Koo, W., Yoo, K., et al., 2022 : Carbothermic Reduction Roasting of Cathode Active Materials Using Activated Carbon and Graphite to Enhance the Sulfuric-Acid-Leaching Efficiency of Nickel and Cobalt, Minerals, 12, pp. 1021.
5. Lee, S. and Yoo, K., 2021 : Waste and Recycling Status of Europe, Japan and USA, Resources Recycling, 30(1), pp. 92-101.
6. Kim, B., Kim, J., and Yoo, K., 2019 : Recycling Status of Gold, Silver, Platinum and Palladium, Journal of the Korean Society of Mineral and Energy Resources Engineers, 56(4), pp. 359-366.
7. Kim, R., Lee, J., Park, J., et al., 2022 : Current Status in the Mining Industry of Critical Minerals for Battery (Li, Ni, Co, and C) in the Energy Transition Era, Journal of the Korean Society of Mineral and Energy Resources Engineers, 59(2), pp. 218-232.
8. Cho, B., Cho, Y., Lee, J., et al., 2019 : Korea's metal resources recycling research project - valuable recycling, Geosystem Engineering, 22(1), pp. 48-58.
9. Lee, C.K. and Rhee, K., 2002 : Preparation of LiCoO₂ from spent lithium-ion batteries, Journal of Power Sources, 109, pp. 17-21.
10. Lee, C.K. and Rhee, K., 2003 : Reductive leaching of cathodic active materials from lithium ion battery wastes, Hydrometallurgy, 68, pp. 5-10.
11. Shin, S.M., Kim, N.H., Sohn, J.S., et al., 2005 : Development of a metal recovery process from Li-ion battery wastes, Hydrometallurgy, 79(3-4), pp. 172-181.
12. Kang, J., Sohn, J., Chang, H., et al., 2010 : Preparation of cobalt oxide from concentrated cathode material of spent lithium ion batteries by hydrometallurgical method, Advanced Powder Technology, 21(2), pp. 175-179.
13. Kang, J., Senanayake, G., Sohn, J., et al., 2010 : Recovery of cobalt sulfate from spent lithium ion batteries by reductive leaching and solvent extraction with Cyanex 272, Hydrometallurgy, 100(3-4), pp. 168-171.
14. Joo, S. H., Kang, J., Woong, K., et al., 2013 : Production of chemical manganese dioxide from lithium ion battery ternary cathodic material by selective oxidative precipitation of manganese. Materials Transactions, 54(5), pp. 844-849.
15. Joo, S.H., Shin, S.M., Shin, D., et al., 2015 : Extractive separation studies of manganese from spent lithium battery leachate using mixture of PC88A and Versatic 10 acid in kerosene, Hydrometallurgy, 156, pp. 136-141.
16. Joo, S.H., Shin, D., Oh, C., et al., 2016 : Selective extraction

- and separation of nickel from cobalt, manganese and lithium in pre-treated leach liquors of ternary cathode material of spent lithium-ion batteries using synergism caused by Versatic 10 acid and LIX 84-I, Hydrometallurgy, 159, pp. 65-74.
17. Joo, S.H., Shin, D., Oh, C., et al., 2016 : Extraction of manganese by alkyl monocarboxylic acid in a mixed extractant from a leaching solution of spent lithium-ion battery ternary cathodic material, Journal of Power Sources, 305, pp. 175-181.
18. SungEel Hitech Homepage, LIB Recycling Process, <https://www.sungeelht.com/html/12>, January 1st, 2023.
19. EcoPro, Business Model, <https://www.ecopro.co.kr/sub010301>, January 1st, 2023.

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