



# Determining the reuse of metal mine wastes based on leaching test and human health risk assessment

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## ABSTRACT

Meeting the regulations based on the short-term leaching tests may not necessarily assure the environmental and human health safety of reusing mine wastes. This study investigated heavy metal leachability of four metal mine waste samples (e.g., Z, Y, H, and M) and human health risk of reusing them as construction materials. The heavy metal leachability did not depend on the total heavy metal contents. For example, the Z sample contained greater amounts of As and Fe than Zn, but the leachates contained only Zn at a detectable level. This can be attributed to the crystalline structure and heavy metal fractions of the mine wastes. The leaching test results suggested that the four mine waste samples are potentially reusable. But the Z and M samples reused in industrial areas imposed carcinogenic risks. This was largely attributed to As that is exposed via dermal contact. The Y and H samples reused in residential areas imposed carcinogenic risk. The major exposure route was the ingestion of crops grown on the mine wastes and Cr was the major concern. The two-stage assessment involving leaching tests and risk assessment can be used to promote safe reuse of mine wastes.

**Keywords:** Heavy metal, Human health risk, Leachability, Mine waste, Reuse

## 1. Introduction

Mine wastes generated during beneficiation of the extracted ore are often heaped and stored without any treatment near mining sites. These mine wastes are prone to metal leaching, which may often lead to acid mine drainage (AMD) generation, and this can cause environmental contamination near mining sites [1, 2]. The leachate generated from mine wastes can have impacts on surrounding environments and nearby residents due to their low pH and/or harmful heavy metal contents [3, 4]. Conventional treatment methods often involve post-treatment of the generated AMD from mine wastes or the contaminated water and/or soil by AMD [5]. Also, potentials for exposure of mine wastes to air or water and consequential oxidation of sulfide minerals such as pyrite (FeS<sub>2</sub>) contained in mine wastes and AMD generation can be removed by covering mine wastes or by removing potential contaminants in mine wastes [6].

The growing movement towards reusing and recycling of solid wastes is promoting research on beneficial reuses of mine wastes

as well. One of the major environmental issues in mine waste reuse is leaching of various heavy metals contained in mine wastes, in particular, metal mine wastes. Heavy metal concentrations in the leachates from solid wastes in Republic of Korea are regulated based on the short-term leaching tests. However, meeting the regulations based on short-term leaching tests does not necessarily mean that it is safe to reuse mine wastes. The continuous release of leachates from mine wastes with higher metal contents may result in contamination of the surrounding soil environment. The soil impacted by mine wastes can have phytotoxicity [7, 8] and human health risk [9, 10]. Therefore, in order to promote the reuse of mine wastes, it is necessary to prepare a new framework that can assess the environmental and human health safety.

Heavy metal leaching from mine wastes is highly dependent on the proportions of silicate, acidic minerals, and organic matter that constitute mine wastes and the leachability of biogenic elements (K, Na, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>) [11]. For example, leaching of heavy metals is likely to be lower for the mine wastes with lower sulfide mineral contents as the generation of acid that will dissolve



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Received December 15, 2017 Accepted June 20, 2018

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heavy metals will be less [12]. The current regulation determines the reusability of solid wastes (e.g., mine waste) based on the contaminant concentrations in the leachates; however, mine wastes are not readily reused even though they satisfy the regulations. This study used the metal mine waste samples collected in different places of Republic of Korea to investigate the leachability of heavy metals, the ecological toxic effects of the leachates, and the human health risk of reusing the mine waste samples.

This study aims at providing a new framework that can promote the reuse of mine wastes, and a two-stage assessment method involving short-term leaching tests and human health risk assessment is suggested. This method can be used to determine the reusability of mine wastes considering both the current regulations on leachates and the information on human health risk, and promote the safe reuse of mine wastes.

## 2. Materials and Methods

### 2.1. Preparation of Mine Waste Samples

Mine waste samples were collected from an operating lead (Pb) and zinc (Zn) mine (hereafter referred to as Z mine, located in Gyeongsangbuk-do province, South Korea), two operating magnetite mines (hereafter referred to as Y mine and H mine, located in Gangwon-do province, South Korea), and an operating molybdenum mine (hereafter referred to as M mine, located in Chungcheongbuk-do province, South Korea). Mine wastes were sampled after the mineral dressing step, dried at room temperature, and sealed to reduce contact with air until the use.

### 2.2. Mine Waste Characteristics

The crystal structures of the mine waste samples were analyzed using the X-ray powder diffraction (XRD) (D8 Advance, Bruker, Germany). The United States Environmental Protection Agency (USEPA) 3052 method, a microwave-assisted acid digestion method for extracting residual elements in soil, was used to analyze total heavy metal contents in the mine waste samples [13]. Briefly, 0.5 g of each mine waste sample was placed in a mixture of 9 mL HNO<sub>3</sub>, 3 mL HF, 2 mL HCl, and 1 mL H<sub>2</sub>O<sub>2</sub> in a Teflon tube. The mine waste samples were dissolved by using a microwave (MARS 6 230/60, CEM, US), which was programmed to raise its temperature to 180°C within 5.5 min and then maintained at 180°C for 9.5 min. The samples were cooled down to room temperature, and the digested samples were filtered through 0.45 μm GHP syringe filters for further analysis. The sequential extraction procedure suggested by Tessier et al. [14] was used to identify metals bound to exchangeable fraction, bound to carbonate, and bound to Fe and Mn oxides. Briefly, 1 M MgCl<sub>2</sub> (pH 7), 1 M sodium acetate (pH 5.0 adjusted with acetic acid), and 0.04 M NH<sub>2</sub>OH·HCl in 25% acetic acid (v:v) were used to extract metals bound to exchangeable fraction, carbonate, and Fe and Mn oxides, respectively. The extracts obtained were filtered through 0.45 μm GHP syringe filters, and analyzed for arsenic (As), Pb, cadmium (Cd), Zn, copper (Cu),

nickel (Ni), and chromium (Cr). The pH of the mine waste samples was measured using the 5-star portable meter (Thermo Fisher Scientific) equipped with a pH probe (8102BNUWP).

### 2.3. Heavy Metal Leaching Tests from the Mine Waste Samples

The heavy metal leaching potentials from the four mine waste samples were tested using the Korea Standard Test Methods for Solid Wastes (hereafter, referred to as STM) provided by the Korea Ministry of Environment (KMOE) [15]. Briefly, 100 g of the mine waste sample, passed through a 2 mm sieve, was mixed with pH-adjusted distilled water (pH 5.8-6.3 with HCl) at a ratio of 1:10 (w:v). The mixture was shaken for 6 h at 200 rpm with 4-5 cm of amplitude (at room temperature), and then centrifuged at 12,000 g for 15 min. The heavy metal content of the filtered supernatant (0.45 μm GHP) was analyzed.

To determine the heavy metal leachability of the mine waste samples when they are exposed to acidic rainfall, the USEPA 1312 method (Synthetic Precipitation Leaching Procedure; hereafter referred to as SPLP), which was designed to simulate in-situ exposure of materials to slightly acidic rainfall, was used [16]. The mine waste samples, passed through a 9.5 mm sieve, were mixed with the leaching solution (i.e., distilled water with pH adjusted to 4.2 ± 0.05 using a mixture of H<sub>2</sub>SO<sub>4</sub>:HNO<sub>3</sub> at 60:40 (w:w)) at a ratio of 1:20 (w:v). The mixture was shaken on a rotary agitator (30 rpm, 23 ± 2°C) for 18 h, and then centrifuged at 12,000 g for 15 min. The supernatant was filtered through a 0.45 μm GHP syringe filter for analysis of its heavy metal content (As, Pb, Cd, Zn, Cu, Ni, iron (Fe), Cr).

### 2.4. Toxic Effects of Mine Waste Leachates

The acute toxicities of the leachates generated using the STM and SPLP were determined by using *Daphnia magna* (*D. magna*) as a test species. *D. magna* was exposed to the leachates for 24 h and 48 h in accordance with the acute toxicity test method described in the Korea Standard Test Method for Water Quality [17]. The toxic unit (TU) was evaluated using the TOXCAL program (Tidepool Scientific Software, USA). Assuming that the leachates get to a nearby water system without considering dilution effect, the determined TU values were compared with the Korea Effluent Water Quality Standards [18].

### 2.5. Human Health Risk of Reusing Mine Wastes

The *in vitro* bioavailability of heavy metals contained in the mine waste samples was determined by using the stomach-phase extraction test developed by Solubility/Bioavailability Consortium (SBRC) [19]. Briefly, 1 g of the mine waste sample, passed through a 150 μm sieve, was mixed with 100 mL of synthetic gastric juice, which is 0.4 M glycine adjusted to pH 1.5 with HCl. The mixture was agitated (38°C, 200 rpm) for 1 h, and then centrifuged at 12,000 g for 15 min. The supernatant was filtered through a 0.45 μm GHP syringe filter for its heavy metal content analysis. The total heavy metal contents and the SBRC extracted heavy metal concentrations were then used to estimate the carcinogenic and non-carcinogenic human health risks, following the procedure described in the Soil Environment Conservation Act of

Republic of Korea with modification [20]. Risk assessment was carried out assuming that the mine waste samples were reused in industrial and residential areas.

### 2.6. Heavy Metal Concentration Analyses

The concentrations of heavy metals/metalloids (i.e., Cr, Fe, Ni, Cu, Zn, As, Cd, Pb) in the extracts obtained from the total heavy metal content analysis and SBRC test and in the leachates obtained from the leaching tests were analyzed by using Inductively Coupled Plasma-Optical Emission Spectrometer (ICP-OES) (ICAP 7400 DUO, Thermo Scientific, US). The detection limit of each element analyzed was as follows: As 0.089 mg L<sup>-1</sup>, Cd 0.015 mg L<sup>-1</sup>, Cr 0.010 mg L<sup>-1</sup>, Cu 0.005 mg L<sup>-1</sup>, Fe 0.007 mg L<sup>-1</sup>, Ni 0.0134 mg L<sup>-1</sup>, Pb 0.065 mg L<sup>-1</sup>, and Zn 0.004 mg L<sup>-1</sup>.

## 3. Results and Discussion

### 3.1. Potential Contaminants in the Mine Waste Samples

The total heavy metal contents of the four metal mine waste samples are shown in Table 1. The types and amounts of heavy metals contained in the different metal mine waste samples varied significantly depending on the types of mines (Table 1). For example, the Pb and Zn contents of the Z mine waste sample were relatively higher than the other mine waste samples since the Z mine produces Pb and Zn (Table 1). The M and H mine waste samples contained relatively higher concentrations of Fe than other heavy metals (Table 1). Also, the Fe content was the highest in the Y mine waste as the Y mine produces magnetite (Table 1). The major heavy metals contained in the mine waste samples were dependent on the metal that each mine produces. The high heavy metal contents of the studied mine waste samples may limit the reuse of mine wastes, if they are released as leachates.

### 3.2. Heavy Metal Leaching from the Mine Waste and Leachate Toxicity

The high heavy metal contents of the mine waste samples do not necessarily mean higher leachability of heavy metals from the mine waste samples. The heavy metal concentrations of the STM and SPLP leachates generated from the mine waste samples and their toxic effects on *D. magna* are shown in Table 2. Generally, the heavy metal concentrations of the STM and SPLP leachates were lower than the detection limits (Table 2). With

the detected heavy metals, their concentrations were lower than the regulation levels given in the Korea Waste Control Act that need to be met in order to reuse any waste materials (Table 2). Only the STM and SPLP leachates from the Z mine waste sample contained a detectable level of Zn (Table 2), although the amounts of the other constituents such as As and Fe were greater than that of Zn (Table 1). Also, the SPLP leachate of the Y mine waste contained a detectable level of Fe (Table 2).

The low heavy metal concentrations in the leachates can be attributed to the pH conditions of the leaching solutions. Heavy metal leachability at neutral or alkaline pH conditions is generally lower than that at acidic pH conditions [21]. The average initial pH of the leaching solution was 5.8 for the STM, but the final pH of the leaching solution was increased to 7.0 for the Z mine waste, 8.7 for the M mine waste, 8.9 for the Y mine waste, and 9.2 for the H mine waste. The increase in the pH of the leaching solution can be expected, as all the four mine waste samples contained CO<sub>3</sub><sup>-</sup> or OH-bearing minerals, which can potentially contribute to the pH increases (Table 3). Among the four mine waste samples, only the Z mine waste contained a sulfur-bearing mineral, pyrite, and the final pH of the Z mine waste leachate showed the smallest increase in the pH. This may be attributed to the generation of acid due to sulfur-bearing minerals such as pyrite [12].

The heavy metal concentrations in the Z mine waste leachate did not depend on the total heavy metal contents of the Z mine waste (Table 1 and Table 2), and this can be attributed to the crystal structures of the Z mine waste (Table 3). The Z mine waste contained higher amounts of As and Fe than Zn, but the leachate only contained a detectable level of Zn. Based on the crystal structure of the Z mine waste sample as shown in Table 3, As and Fe are found in minerals having crystalline structures such as pyrite, merrihueite ((K, Na)<sub>2</sub>(Fe, Mg)<sub>5</sub>Si<sub>12</sub>O<sub>30</sub>), butlerite (FeSO<sub>4</sub>(OH)·2H<sub>2</sub>O), and tennantite (Cu<sub>6</sub>(Cu<sub>0.48</sub>Fe<sub>5.52</sub>)As<sub>4</sub>S<sub>13</sub>), while Zn is not observed in minerals having crystalline structures. This suggests that the Zn leachability may be greater than the As or Fe leachability, because the crystalline form is generally more stable than the amorphous counterpart, indicating that it requires a higher energy to break its chemical structure. In addition, based on the Tessier's sequential extraction, the readily soluble amounts of Zn and As (i.e., the sum of the exchangeable and carbonate-bound fractions) were 1.6% and 0.29% of the total content (Table 4). This indicates that Zn is more readily dissolved than As. Overall, the crystalline structures and the heavy metal fractions are more likely to influence the

**Table 1.** Heavy Metals Contained in the Mine Wastes Used in This Study (Unit: mg kg<sup>-1</sup>)

Mine wastes	As	Cd	Cr	Cu	Fe	Ni	Pb	Zn
Z	45,424 ± 5,620	2,293 ± 246	118 ± 11	651 ± 47	88,943 ± 6,539	ND <sup>1)</sup>	2,671 ± 15	14,851 ± 541
M	401 ± 259	0.02	636 ± 175	158 ± 78.3	177,000 ± 34,400	258 ± 117	1,171 ± 613	2,330 ± 111
Y	ND	ND	31.5 ± 2.60	26.6 ± 4.69	6,360 ± 1,910	ND	86.1 ± 1.57	390 ± 4.98
H	ND	12.4 ± 6.84	141 ± 5.48	ND	103,000 ± 7,540	185 ± 35.8	1,180 ± 88.0	1,790 ± 83.6

1) ND: Less than the ICP-OES detection limits (As 0.089 mg L<sup>-1</sup>, Cd 0.015 mg L<sup>-1</sup>, Cr 0.010 mg L<sup>-1</sup>, Cu 0.005 mg L<sup>-1</sup>, Fe 0.007 mg L<sup>-1</sup>, Ni 0.0134 mg L<sup>-1</sup>, Pb 0.065 mg L<sup>-1</sup>, Zn 0.004 mg L<sup>-1</sup>)

**Table 2.** Heavy Metal Concentrations in the Leachates from the Mine Wastes (Unit: mg L<sup>-1</sup> except for TU) and Toxic Units (TU) of the Leachates Determined by Using *Daphnia magna*

Mine wastes	Leaching test	As	Cd	Cr	Cu	Fe	Ni	Pb	Zn	TU	
										24 h	48 h
Z	STM <sup>1)</sup>	ND <sup>2)</sup>	ND	ND	ND	ND	ND	ND	0.123 ± 0.010	0	0.6
	SPLP <sup>3)</sup>	ND	ND	ND	ND	ND	ND	ND	0.052 ± 0.016	0.6	0.6
M	STM	ND	ND	ND	ND	ND	ND	ND	ND	0	0.2
	SPLP	ND	ND	ND	ND	ND	ND	ND	ND	0	0
Y	STM	ND	ND	ND	ND	ND	ND	ND	ND	0	0.4
	SPLP	ND	ND	ND	ND	0.015 ± 0.004	ND	ND	ND	0	0
H	STM	ND	ND	ND	ND	ND	ND	ND	ND	0	0.2
	SPLP	ND	ND	ND	ND	ND	ND	ND	ND	0	0.2
Korea Waste Control Act		1.5	0.3	Cr(VI) 1.5	3.0	-	-	3.0	-	-	-
Korea Allowable Effluent Water Quality Standards (Clean area)		< 0.05	< 0.02	< 0.5	< 1	< 2	-	< 0.1	< 1	< 1	< 1
Korea Allowable Effluent Water Quality Standards <sup>4)</sup> (Area 1)		< 0.25	< 0.1	< 2	< 3	< 10	-	< 0.5	< 5	< 2	< 2
Korea Allowable Effluent Water Quality Standards (Area 2)		< 0.25	< 0.1	< 2	< 3	< 10	-	< 0.5	< 5	< 2	< 2

1) STM: Standard Test Method for Solid Wastes

2) ND: Less than the ICP-OES detection limit (As 0.089 mg L<sup>-1</sup>, Cd 0.015 mg L<sup>-1</sup>, Cr 0.010 mg L<sup>-1</sup>, Cu 0.005 mg L<sup>-1</sup>, Fe 0.007 mg L<sup>-1</sup>, Ni 0.0134 mg L<sup>-1</sup>, Pb 0.065 mg L<sup>-1</sup>, Zn 0.004 mg L<sup>-1</sup>)

3) SPLP: Synthetic Precipitation Leaching Procedure

4) Area 1 and Area 2 of the Korea Allowable Effluent Water Quality Standards are classified according to the administrative district of South Korea.

**Table 3.** Crystal Structures Contained in the Mine Wastes Used in This Study

Mine wastes	Crystal structures
Z	Quartz (SiO <sub>2</sub> ), Pyrite (FeS <sub>2</sub> ), Pyroxomangite (Mn <sub>0.97</sub> Mg <sub>0.03</sub> SiO <sub>3</sub> ), Dolomite (CaMg(CO <sub>3</sub> ) <sub>2</sub> ), Merrihueite ((K, Na) <sub>2</sub> (Fe, Mg) <sub>5</sub> Si <sub>12</sub> O <sub>30</sub> ), Chlorite-serpentine ((Mg, Al) <sub>6</sub> (Si, Al) <sub>4</sub> O <sub>10</sub> (OH) <sub>8</sub> ), Butlerite (FeSO <sub>4</sub> (OH) 2H <sub>2</sub> O), Tennantite (Cu <sub>6</sub> (Cu <sub>0.48</sub> Fe <sub>0.52</sub> ) As <sub>4</sub> S <sub>13</sub> ), Unnamed mineral ((K, H <sub>3</sub> O) Al <sub>2</sub> Si <sub>3</sub> AlO <sub>10</sub> (OH) <sub>2</sub> )
M	Grossular (Ca <sub>3</sub> Al <sub>2</sub> (SiO <sub>4</sub> ) <sub>3</sub> ), Chlorite-serpentine ((Mg, Al) <sub>6</sub> (Si, Al) <sub>4</sub> O <sub>10</sub> (OH) <sub>8</sub> ), Calcium magnesium (CaMg <sub>0.52</sub> Fe <sub>0.48</sub> (Si <sub>2</sub> O <sub>6</sub> )), Clinocllore Iib-4 (Mg <sub>9.8</sub> Al <sub>1.6</sub> Fe <sub>0.6</sub> (Si <sub>6.32</sub> Al <sub>1.68</sub> O <sub>20.24</sub> )(OH) <sub>15.72</sub> ), Meionite ((Ca, Na) <sub>2</sub> (Si, Al) <sub>6</sub> O <sub>12</sub> (CO <sub>3</sub> ) <sub>0.5</sub> ), Muscovite-3T ((K, Na)(Al, Mg, Fe) <sub>2</sub> (Si <sub>3.1</sub> Al <sub>0.9</sub> )O <sub>10</sub> (OH) <sub>2</sub> ), Pargasite (Na <sub>3.38</sub> Ca <sub>1.7</sub> Mg <sub>3.34</sub> Fe <sub>1.77</sub> (Al <sub>1.8</sub> Si <sub>6.2</sub> )O <sub>23</sub> (OH)), Heptairon tetrakis (Fe <sub>7</sub> (P <sub>2</sub> O <sub>7</sub> ) <sub>4</sub> )
Y	Biotite (KFeMg <sub>2</sub> (AlSi <sub>3</sub> O <sub>10</sub> )(OH) <sub>2</sub> ), Magnesioriebeckite ((Na, Ca) <sub>2</sub> (Mg, Fe) <sub>5</sub> Si <sub>8</sub> O <sub>22</sub> (OH) <sub>2</sub> ), Clinocllore-1M <sub>Ib</sub> , ferroan ((Mg, Fe) <sub>6</sub> (Si, Al) <sub>4</sub> O <sub>10</sub> (OH) <sub>8</sub> ), Laumontite (Ca <sub>4</sub> Al <sub>8</sub> Si <sub>16</sub> O <sub>48</sub> 14H <sub>2</sub> O), Bementite (Mn <sub>8</sub> Si <sub>6</sub> O <sub>15</sub> (OH) <sub>10</sub> ), Cummingtonite manganonaon (Na <sub>0.1</sub> Ca <sub>0.4</sub> Mg <sub>5.6</sub> Mn <sub>1.0</sub> Si <sub>8</sub> O <sub>22</sub> (OH) <sub>2</sub> ), Unnamed mineral ((Mn, Zn) <sub>3</sub> Zn <sub>2</sub> AsO <sub>4</sub> (OH, O) <sub>6</sub> )
H	Diopside (CaMg(Si <sub>2</sub> O <sub>6</sub> )), Actinolite (Ca <sub>2</sub> (Mg, Fe) <sub>5</sub> Si <sub>8</sub> O <sub>22</sub> (OH) <sub>2</sub> ), Antigorite-8.0M (Mg <sub>3</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>4</sub> ), Clinocllore ((Mg <sub>2.8</sub> Fe <sub>1.7</sub> Al <sub>1.2</sub> )(Si <sub>2.8</sub> Al <sub>1.2</sub> )O <sub>10</sub> (OH) <sub>8</sub> ), Mutinaite (Al <sub>2</sub> O <sub>3</sub> 54SiO <sub>2</sub> ), Cummingtonite manganonaon (Na <sub>0.1</sub> Ca <sub>0.4</sub> Mg <sub>5.6</sub> Mn <sub>1.0</sub> Si <sub>8</sub> O <sub>22</sub> (OH) <sub>2</sub> )

heavy metal leachability from the mine waste than the total metal contents. Even though the heavy metal concentrations of the leachate from the mine wastes met the regulations, the combined toxic effects of trace amounts of heavy metals in the leachates could still exhibit toxic effects on aquatic organisms

[22]. The toxic effects of the leachates on *D. magna* in TU were lower than the Korea Allowable Effluent Water Quality Standards for all the areas (TU < 1 or < 2) (Table 2). All the four mine waste samples tested in this study satisfied the regulations on the leachates, indicating that they can be reused.

**Table 4.** Exchangeable, Carbonate-bound, and Iron and Manganese Oxides-bound Fractions of Heavy Metals in the Z Mine Waste Determined by Using the Tessier's Sequential Extraction Procedure

Heavy metal	Exchangeable (F1) (mg kg <sup>-1</sup> )	Carbonate-bound (F2) (mg kg <sup>-1</sup> )	Iron and manganese oxides-bound (F3) (mg kg <sup>-1</sup> )	Ratio of the sum of F1 and F2 and the total content (%)
As	134	ND	454	0.29
Pb	764	ND	1,288	28
Cd	8	ND	27	0.35
Zn	226	7	403	1.6
Cu	6	ND	3	0.92
Ni	ND	ND	ND	ND
Cr	ND	ND	17	ND

ND: Not determined, Less than the detection limit

### 3.3. Human Health Risk Assessment for Workers in Industrial Areas Where the Mine Wastes Are Reused

The ingestion of soil is the major exposure pathway for heavy metal contaminated soils [23], so the intake of mine wastes via oral ingestion can be considered as the major exposure pathway for workers at mine sites. In addition to this, the exposure through inhalation of fugitive dust can be considered. Carcinogenic risk (CR) and non-carcinogenic risk (i.e., sum of Hazard Quotient (HQ) or Hazard Index (HI)) of each metal in mine wastes can be calculated using Eq. (1)-(7) considering an average daily dose (ADD) via ingestion and dermal contact or an average daily exposure (ADE) via inhalation and toxicity values [24, 25]. The input parameters and exposure factors used in risk calculation are shown in Table 5.

$$\begin{aligned} \text{Ingestion of mine wastes: ADD} & & (1) \\ = (Cs \times FI \times IR \times EF \times ED) / (BW \times AT) \end{aligned}$$

$$\begin{aligned} \text{Dermal contact with mine wastes: ADD} & & (2) \\ = (Cs \times AF \times ABS_D \times SA \times EF \times ED) / (BW \times AT) \end{aligned}$$

$$\begin{aligned} \text{Inhalation of fugitive dust: ADE} & & (3) \\ = (Cs \times TSP \times frs \times FR \times EF \times ED) / (AT) \end{aligned}$$

$$\begin{aligned} \text{CR for ingestion of and dermal contact} & & (4) \\ \text{with mine wastes: CR} = \text{ADD} \times \text{SF} \end{aligned}$$

$$\begin{aligned} \text{CR for inhalation of fugitive dust: CR} & & (5) \\ = \text{ADE} \times \text{IUR} \end{aligned}$$

$$\begin{aligned} \text{Non-CR for ingestion of and dermal contact} & & (6) \\ \text{with mine wastes: HQ} = \text{ADD} / \text{RfD}_o \end{aligned}$$

$$\begin{aligned} \text{Non-CR for inhalation of fugitive dust: HQ} & & (7) \\ = \text{ADE} / \text{RfC} \end{aligned}$$

where ADD: average daily dose (mg kg<sup>-1</sup> d<sup>-1</sup>); Cs: total content of metal (mg kg<sup>-1</sup>); FI: bioavailable fraction (%); IR: ingestion rate (mg d<sup>-1</sup>); EF: exposure frequency (day y<sup>-1</sup>); ED: exposure duration (y); BW: body weight (kg); AT: averaging time (d); AF: Mine waste-to-skin adherence factor (mg cm<sup>-2</sup>); ABS<sub>D</sub>: Dermal

absorption factor (dimensionless); SA: Skin surface area (cm<sup>2</sup> d<sup>-1</sup>); ADE: average daily exposure (mg kg<sup>-1</sup> d<sup>-1</sup>); TSP: Total suspended particle (mg m<sup>-3</sup>); frs: Fraction of mine waste in air (dimensionless); FR: Retention factor of mine waste in lungs (dimensionless); CR: Carcinogenic risk (dimensionless); SF: Slope factor ((mg (kg d<sup>-1</sup>)<sup>-1</sup>); IUR: Inhalation unit risk factor ((μm m<sup>-3</sup>)<sup>-1</sup>); HQ: Hazard quotient (dimensionless); RfD<sub>o</sub>: Reference dose for oral exposure (mg (kg d<sup>-1</sup>)<sup>-1</sup>); RfC: Reference concentration for inhalation (mg m<sup>-3</sup>).

Previous studies reported that there is a linear relationship between the *in vitro* SBRC extracted As, Pb, and Cd concentrations and the *in vivo* relative bioavailabilities [26-28]. Similarly, the SBRC extraction method was used to investigate the Zn bioavailability of the Zn contaminated samples [29-31]. Therefore, the SBRC extracted heavy metal concentrations were used to calculate the bioavailable fractions of heavy metals in the mine waste samples (i.e., FI in Eq. (1)). Table 6 shows the SBRC extracted heavy metal concentrations and the FI values of each heavy metal of each mine waste sample.

Carcinogenic risk is a probability of an individual developing cancer from carcinogenic hazards accounted for lifetime, and the acceptable range of carcinogenic risk is between 1.0E-06 and 1.0E-04. In addition, it is widely accepted that there is non-carcinogenic risk if HI is greater than 1.0E+00. The risk calculation results are shown in Table 7. Among the four mine waste samples, only the Z mine waste and the M mine waste had risks. The Z mine waste had carcinogenic risk (i.e., CR value between 1.0E-06 and 1.0E-04) and non-carcinogenic risk (i.e., HI greater than 1.0E+00) (Table 7) for workers at mining sites. Similarly, the M mine waste had carcinogenic risk with the CR value of 8.6E-06 (Table 7). Among the heavy metals contained in the mine waste samples, As, Cd, Cr, Ni, and Pb contributed to carcinogenic risk, and As, Cd, Cr, Cu, Ni, Pb, and Zn contributed to non-carcinogenic risk. The carcinogenic risk exhibited by the Z and M mine waste samples can be attributed to As. The CR value of As in the Z mine waste was 8.8E-04, while the sum of the CR values for the other carcinogenic heavy metals (i.e., Cd, Cr, Ni, and Pb) in the Z mine waste was 6.6E-07. The high HI value of the Z mine waste can be attributed to As and Cd (Table 1). The HQ values for As and Cd were 2.8E+01 and 1.7E+00, respectively, and these were relatively higher than

**Table 5.** Input Parameters and Exposure Factors Used in This Study for Human Health Risk Assessment

	Units	Value	Reference
<b>Input parameters</b>			
C <sub>S</sub> (Total content of metal)	mg kg <sup>-1</sup>	-	Table 1
IR (Mine waste ingestion rate)	mg d <sup>-1</sup>	50	[1]
FI (Bioavailable fraction)	%	-	Table 6
EF (Exposure frequency)	day y <sup>-1</sup>	100	[2]
ED (Exposure duration)	y	25	[1]
AF (Mine waste-to-skin adherence factor)	mg cm <sup>-2</sup>	0.07	[1]
SA (Skin surface area)	cm <sup>2</sup> d <sup>-1</sup>	4,271	[1]
TSP (Total suspended particle)	mg m <sup>-3</sup>	0.07	[1]
frs (Fraction of mine waste in air)	-	0.5	[1]
FR (Retention factor of mine tailings in lungs)	-	0.75	[1]
CF (Conversion factor)	kg mg <sup>-1</sup>	10 <sup>-6</sup>	-
BW (Average body weight)	kg	62.8	[1]
AT <sub>c</sub> (Averaging time for carcinogens)	d	28,470	[2]
AT <sub>n</sub> (Averaging time for non-carcinogens)	d	9,125	[2]
CR <sub>w</sub> (Daily intake of groundwater)	L d <sup>-1</sup>	Adult: 2 Children: 1	[1]
BCF (Bio-concentration factor)	mg kg-crop <sup>-1</sup> (mg kg-soil <sup>-1</sup> ) <sup>-1</sup>	As: 0.03 Cr: 0.02 Pb: 0.03	[1]
CR <sub>p</sub> (Daily intake of crop (leaf vegetable))	kg d <sup>-1</sup>	Adult: 0.027 Child: 0.008	[1]
<b>Exposure factors</b>			
SF (Slope factor)	(mg (kg d) <sup>-1</sup> ) <sup>-1</sup>	As: 1.50E+00 Pb: 8.50E-03	[2]
ABS <sub>D</sub> (Dermal absorption factor)	-	As: 3.0E-02 Cd: 1.0E-03	[2]
IUR (Inhalation unit risk factor)	(µm m <sup>-3</sup> ) <sup>-1</sup>	As: 4.3E-03 Cd: 1.8E-03 Cr: 8.4E-02 Ni: 2.6E-04 Pb: 1.2E-05	[2]
RfD <sub>O</sub> (Reference dose for oral exposure)	mg (kg d) <sup>-1</sup>	As: 3.0E-04 Cd: 1.0E-03 Cu: 4.0E-02 Pb: 3.6E-03 Zn: 3.0E-01	[2]
RfC (Reference concentration for inhalation)	mg m <sup>-3</sup>	As: 1.5E-05 Cd: 1.0E-05 Cr: 1.0E-04 Ni: 9.0E-05	[2]

1) (KMOE 2015)

2) (USEPA 2011)

the sum of the HQ values for the other non-carcinogenic heavy metals (i.e., Cr, Cu, Ni, Pb, and Zn), which was 6.9E-02. The high CR value of the M mine waste sample can also be explained by the relatively higher CR value for As, which was 8.4E-06.

Also, among the three exposure routes considered in the risk calculation, the route considering the dermal contact with mine wastes was the major route for the carcinogenic risk for both the Z and M mine wastes. On the other hand, the human health

**Table 6.** Bioavailable Metal Concentrations and Bioavailable Fractions (FI) Determined by Using the Solubility/Bioavailability Research Consortium (SBRC) Method

			As	Cd	Cu	Cr	Pb	Zn
Z	SBRC	(mg kg <sup>-1</sup> )	257.9 ± 44.9	14.5 ± 2.8	22.8 ± 6.3	3.4 ± 0.1	1,056.7 ± 119.0	422.4 ± 56.0
	FI <sup>1)</sup>	(%)	0.568	0.632	3.50	2.86	39.6	2.84
M	SBRC	(mg kg <sup>-1</sup> )	8.6 ± 4.7	ND	3.2 ± 0.5	11.9 ± 0.3	3.5 ± 0.3	26.8 ± 2.0
	FI <sup>1)</sup>	(%)	2.14	-	2.00	1.87	0.298	1.15
Y	SBRC	(mg kg <sup>-1</sup> )	ND	ND	7.2 ± 0.4	2.5 ± 0.5	9.1 ± 2.8	27.4 ± 0.1
	FI <sup>1)</sup>	(%)	-	-	27.0	8.08	10.5	7.03
H	SBRC	(mg kg <sup>-1</sup> )	ND	1.0 ± 0.5	ND	3.8 ± 0.1	7.7 ± 1.2	67.1 ± 5.5
	FI <sup>1)</sup>	(%)	-	7.93	-	2.69	0.651	3.75

1) FI: (concentration of metal in SBRC leachate)/(total content of the metal)

**Table 7.** Human Health Risk on Workers in Industrial Areas Where Mine Wastes Are Reused

Mine waste sample	Z	M	Y	H
Carcinogenic risk (CR)	8.8E-04	8.6E-06	1.2E-08	3.3E-08
Non-carcinogenic risk (HI)	3.0E+01	3.2E-01	3.0E-03	3.5E-02

risks calculated for workers at the Y and H mine sites showed that there is neither carcinogenic risk nor non-carcinogenic risk at the Y and H mine sites (Table 7). The CR values were lower than 1.0E-6 and the HI values were lower than 1.0E+00.

The results indicate that when the Z and M mine waste samples are reused in industrial areas, the workers at the mine sites who deal with the mine wastes or in the industrial areas where the mine wastes are reused are likely to have carcinogenic and non-carcinogenic risks largely due to As in the mine waste samples via dermal contact. If the Z and M mine waste samples are to be reused in industrial areas, strategies that can prevent the dermal contact of the mine waste samples such as the use of protective clothing can be adopted for workers. Also, the bioavailable fraction of As and Cd in the Z mine waste sample and As in the M mine waste sample can be reduced by applying techniques such as immobilization. On the other hand, the Y and H mine wastes might be reused without imposing human health risks on the workers at mine sites or in the industrial areas where the mine wastes are reused.

### 3.4. Human Health Risk Assessment for Residents in Residential Areas Where the Mine Wastes Are Reused

The human health risks on residents in residential areas where the Y or H mine wastes are used as construction materials were calculated since the Y and H mine wastes did not impose carcinogenic or non-carcinogenic risks. For human health risk at residential areas, exposure through ingestion of crops grown on the mine wastes (i.e., worst case scenario) and exposure through ingestion of groundwater were considered additionally. Carcinogenic risk and non-carcinogenic risk of each metal in the mine waste samples can be calculated using Eq. (1)-(12) considering the ADD via oral ingestion, dermal contact, ingestion of crop, and ingestion of groundwater and the ADE via inhalation of fugitive dust [24, 25]. The input parameters and exposure

factors used in risk calculation are shown in Table 5.

$$\text{Ingestion of crop grown on mine wastes: ADD} \quad (8)$$

$$= (Cs \times AF \times BCF \times CR_p \times SA \times EF \times ED) / (BW \times AT)$$

$$\text{Ingestion of groundwater: ADD} \quad (9)$$

$$= (C_w \times CR_w \times EF \times ED) / (BW \times AT)$$

$$\text{CR for ingestion of crop grown on mine wastes: CR} \quad (10)$$

$$= \text{ADD} \times \text{SF}$$

$$\text{CR for ingestion of groundwater: CR} \quad (11)$$

$$= \text{ADD} \times \text{IUR}$$

$$\text{Non-CR for ingestion of crop grown on mine} \quad (12)$$

$$\text{wastes and of groundwater: HQ} = \text{ADD} / \text{RfD}_o$$

where ADD: average daily dose (mg kg<sup>-1</sup> d<sup>-1</sup>); Cs: total content of metal (mg kg<sup>-1</sup>); AF: Mine waste-to-skin adherence factor (mg cm<sup>-2</sup>); BCF: Bio-concentration factor (mg kg-crop<sup>-1</sup> (mg kg-soil<sup>-1</sup>)<sup>-1</sup>); CR<sub>p</sub>: Daily intake of crop (kg d<sup>-1</sup>); SA: Skin surface area (cm<sup>2</sup> d<sup>-1</sup>); EF: exposure frequency (day y<sup>-1</sup>); ED: exposure duration (y); BW: body weight (kg); AT: averaging time (d); C<sub>w</sub>: metal concentration in groundwater (mg L<sup>-1</sup>); CR<sub>w</sub>: Daily intake of groundwater (L d<sup>-1</sup>); CR: Carcinogenic risk (dimensionless); SF: Slope factor ((mg (kg d<sup>-1</sup>)<sup>-1</sup>); IUR: Inhalation unit risk factor ((μm m<sup>-3</sup>)<sup>-1</sup>); HQ: Hazard quotient (dimensionless); RfD<sub>o</sub>: Reference dose for oral exposure (mg (kg d<sup>-1</sup>)<sup>-1</sup>).

Table 8 shows the human health risks on residents in residential areas where the Y and H mine wastes are reused. The CR values for the Y mine waste reused area (i.e., 4.7E-06) and the H mine waste reused area (i.e., 6.8E-06) both exceeded 1.0E-6 indicating that there is a carcinogenic risk (Table 8). The major exposure route of heavy metals in the Y and the H mine wastes was the ingestion of crops grown on the mine wastes. The major

**Table 8.** Human Health Risk on Residents in Residential Areas Where Mine Wastes Are Reused

			Ingestion of mine waste	Ingestion of crop grown on mine waste	Dermal contact with mine waste	Inhalation of fugitive dust	Ingestion of groundwater	Total
Y	Carcinogenic risk (CR)	Adult	1.9E-08	3.6E-06	0.0E+00	2.1E-08	0.0E+00	3.7E-06
		Children	4.0E-08	9.7E-07	0.0E+00	5.1E-09	0.0E+00	1.0E-06
		Total	5.9E-08	4.6E-06	0.0E+00	2.6E-08	0.0E+00	4.7E-06
	Non-carcinogenic risk (HI)	Adult	2.8E-03	5.3E-02	0.0E+00	7.9E-03	0.0E+00	6.4E-02
		Children	2.4E-02	6.7E-02	0.0E+00	7.9E-03	0.0E+00	9.9E-02
		Total	2.7E-02	1.2E-01	0.0E+00	1.6E-02	0.0E+00	1.6E-01
H	Carcinogenic risk (CR)	Adult	1.6E-08	5.2E-06	0.0E+00	9.6E-08	0.0E+00	5.3E-06
		Children	3.4E-08	1.4E-06	0.0E+00	2.3E-08	0.0E+00	1.5E-06
		Total	5.0E-08	6.6E-06	0.0E+00	1.2E-07	0.0E+00	6.8E-06
	Non-carcinogenic risk (HI)	Adult	3.5E-03	3.5E-01	5.7E-05	1.2E-01	0.0E+00	4.7E-01
		Children	3.1E-02	4.0E-01	2.6E-04	1.2E-01	0.0E+00	5.5E-01
		Total	3.5E-02	7.4E-01	3.2E-04	2.4E-01	0.0E+00	1.0E+00

contributor to the CR of the Y and H mine waste samples was Cr - the CR value for Cr in the Y mine waste sample was 4.2E-06, while it was 5.1E-06 in the H mine waste sample. However, this route assumed that the crops are grown on soil consisting of only the mine waste samples to simulate the worst case scenario, and this will not happen in reality. Also, the exposure via ingestion of mine wastes or inhalation of fugitive dust is unlikely if the mine wastes are reused as construction materials that do not have direct contact with residents. The Y and H mine waste samples did not impose non-carcinogenic risks on residents in residential areas (Table 8). The results suggest that the Y and H mine waste samples may impose risks if reused in residential areas. If the Y and H mine wastes are to be reused in residential areas, the reuse of mine wastes as filling materials of agricultural areas should be avoided or Cr treatment may be necessary.

#### 4. Conclusions

The metal mine waste samples used in this study contained several heavy metals such as As, Cd, and Cr that can impose human health risks. However, the heavy metal concentrations in the leachates generated via the STM and SPLP showed that the heavy metal concentrations in the leachates did not depend on the total heavy metal contents of the mine wastes. This can largely be attributed to the crystal structures of the mine wastes. The leachates studied did not have toxic effects on *D. magna*. The leaching test-based assessment can be used to screen out the potentially reusable mine wastes from those that are not reusable. This can be followed by the risk assessment to assess the risk associated with reusing mine wastes in industrial or residential areas. If the potentially reusable mine wastes do not impose any human health risks, then they can be reused. If the potentially reusable mine wastes may impose risks, then they can be discarded or they can be reused after preventing

the major risk route and/or treating the heavy metal of concern. With this two-stage assessment method for mine wastes, the reusability of mine wastes can be determined, and this will contribute towards protecting environment and resources.

#### Acknowledgments

This research was supported by Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Science, ICT & Future Planning (NRF-2014R1A1A1003762 and NRF-2016R1D1A1A02936998). This work was also supported by Hankuk University of Foreign Studies Research Fund (2018).

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