

Refinement of Gd_2O_3 inclusions in the $GdBa_2Cu_3O_{7-\delta}$ films fabricated by the RCE-DR process

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(Received 17 December 2018; revised or reviewed 29 December 2018; accepted 30 December 2018)

Abstract

To improve in-field critical current densities (J_c) of $GdBa_2Cu_3O_{7-\delta}$ (GdBCO) coated conductors (CCs) fabricated by the reactive co-evaporation by deposition and reaction (RCE-DR) process, employing the nominal composition of Gd:Ba:Cu=1:1:2.5, we tried to refine the Gd_2O_3 particles trapped in the GdBCO superconducting matrix. For this purpose, we carefully selected the processing conditions on the stability phase diagram of GdBCO for this composition. By lowering the growth temperature of Gd_2O_3 in the liquid, we could refine the average particle size of Gd_2O_3 particles trapped in the GdBCO matrix and also achieve the zero-resistive transition temperatures ($T_{c,zero}$) of 92.3~94.2 K. Unfortunately, however, it was unsuccessful to achieve enhanced in-field J_c values from these samples because of an air-contamination of the amorphous precursor film before its conversion into crystalline GdBCO film, suggesting that any exposure of the amorphous precursor film to air is fatal in obtaining high performance GdBCO CCs via the RCE-DR process.

Keywords: $GdBa_2Cu_3O_{7-\delta}$, Gd_2O_3 nanoparticles, RCE-DR, critical current density, stability phase diagram

1. INTRODUCTION

The second generation high-temperature superconductors (HTS), so-called coated conductors (CCs), are commercialized as long-length $YBa_2Cu_3O_{7-\delta}$ (YBCO) and $REBa_2Cu_3O_{7-\delta}$ (REBCO, RE: rare earth elements) superconducting wires have been successfully fabricated via various processes such as metal-organic deposition (MOD) [1], pulsed laser deposition (PLD) [2], metal-organic chemical vapor deposition (MOCVD) [3], and reactive co-evaporation by deposition and reaction (RCE-DR) [4–5] processes. However, higher performance HTS CCs under high magnetic fields are still required for the electric power applications since the critical currents (I_c) of YBCO and REBCO CCs are drastically decreased in high magnetic fields [6]. Thus, many research groups have tried to improve in-field J_c values of CCs by doping [7–9], irradiation [10, 11], and defect [12].

While GdBCO CCs fabricated via the RCE-DR process exhibit the highest self-field J_c value, in-field J_c values are inferior to CCs fabricated by other processes [13] because of less effective flux pinning. According to previous reports, Gd_2O_3 particles are trapped in the GdBCO matrix [4], and act as a pinning center [14]. In this work, in order to improve the pinning properties of GdBCO CCs, we tried to refine the Gd_2O_3 particles incorporated in the GdBCO matrix by properly controlling the nucleation and growth rate of Gd_2O_3 in the liquid on the basis of the stability phase

diagram of GdBCO recently identified by our group [15]. For this study, an amorphous precursor film with the nominal composition of Gd:Ba:Cu = 1:1:2.5 was deposited by a reactive co-evaporation with an electron gun in SuNAM, and subsequent refinement of Gd_2O_3 particles trapped in the GdBCO film was performed in our laboratory.

2. EXPERIMENTAL

The amorphous precursor films were fabricated by a RCE-DR equipment of SuNAM. For the conversion from amorphous phase to superconducting phase, the conversion process was performed using a 6-zone reel-to-reel tube furnace shown in Fig. 1, in which the oxygen partial pressure of each 3 zone can be controlled differently.

Fig. 2 shows the processing routes on the stability phase diagram of GdBCO for the nominal composition of Gd:Ba:Cu = 1:1:2.5 [15]. The conversion conditions of GdBCO were fixed at 861°C in the PO_2 of 150 mTorr by

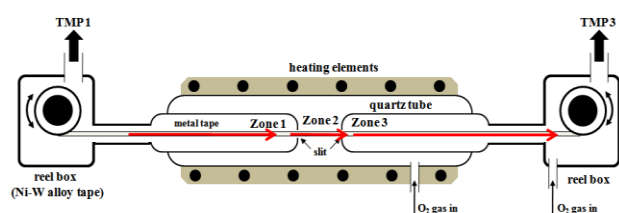


Fig. 1. The schematic of reel-to-reel tube furnace apparatus.

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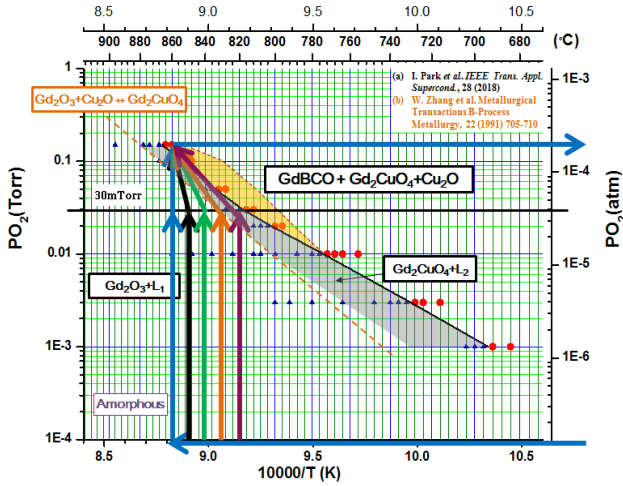


Fig. 2. The stability phase diagram of GdBCO for the nominal composition of Gd:Ba:Cu = 1:1:2.5 [15] with different fabrication process paths.

considering a undercooling temperature just below the phase boundary ($866 \pm 2^\circ\text{C}$) in the PO_2 of 150 mTorr. Also, the growth temperatures of Gd_2O_3 were varied from 821°C to 861°C and held for ~ 5 sec at the given PO_2 of 30 mTorr. The growth time of GdBCO films was ~ 1 min at 861°C in the PO_2 of 150 mTorr.

The phases of samples were analyzed by X-ray diffraction (XRD) using an X-ray diffractometer (Bruker, D8-Advance). The resistivity versus temperature (ρ - T) curves of samples were measured using the standard four point probe method. An analytical transmission electron microscope (TEM) (JEOL, JEM-2100F) was used to analyze the cross-sectional microstructures of samples. The TEM specimens were prepared by focused ion beam (FIB) (SII Nanotechnology, SMI 3050SE).

3. RESULT AND DISCUSSION

The XRD patterns of samples after oxygenation at 500°C for 1 h are shown in Fig. 3. For a comparison, the XRD pattern of the sample fabricated at the same temperature of 860°C in the PO_2 of 30 and 150 mTorr for the growth of Gd_2O_3 and GdBCO, respectively, by RCE-DR in SuNAM is also represented in Fig. 3(a). Here, it must be noted that the amorphous precursor tape prepared by reactive co-evaporation in SuNAM was exposed to air for a short term before its conversion into the superconducting phase in our lab while it was not exposed to air for the sample (a) in SuNAM. Unlike other XRD patterns including only the Gd_2O_3 phase, the XRD pattern of the sample (f) held at 821°C in 30 mTorr for the growth of Gd_2O_3 and held at 861°C in 150 mTorr for the growth of GdBCO include the Gd_2CuO_4 phase in addition to the Gd_2O_3 phase, which is marked as “5”.

The ρ - T curves of samples are shown in Fig. 4. The $T_{c, zero}$ values of samples (a) - (f) are 93.6 K, 92.3 K, 93.3 K, 94.2 K, 94.1 K, and 93.6 K, respectively. A variation in the $T_{c, zero}$ values is not so significant, but the normal-state resistivity curves show much higher than that of the

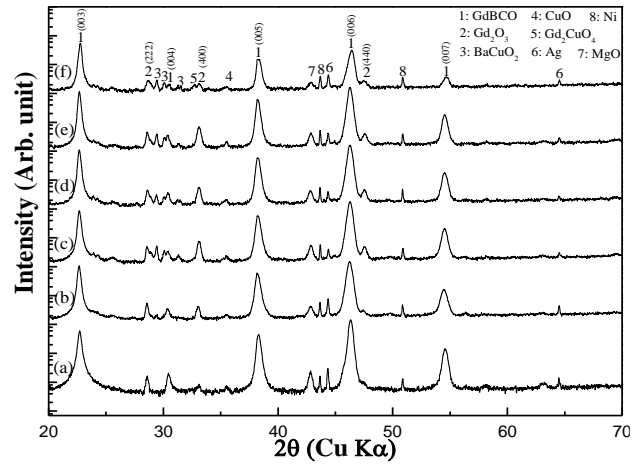


Fig. 3. The XRD patterns of the sample fabricated by RCE-DR in SuNAM (a), and the samples after growth at (b) 861°C , (c) 851°C , (d) 841°C , (e) 831°C , and (f) 821°C in 30 mTorr O_2 for the growth of Gd_2O_3 and at 861°C in the PO_2 of 150 mTorr for the conversion into GdBCO.

RCE-DR sample from SuNAM as shown in the inset of Fig. 4. The reason for this difference is most probably attributed to an air-exposure of the amorphous precursor films before those were set in the tube furnace for the conversion process in our lab. Further, the magnetic J_c - B curves (though not represent here), evaluated by measuring the magnetization hysteresis curves, showed almost one order of magnitude lower than those of the RCE-DR sample (a) from SuNAM. An air-contamination might degrade the subgrain boundaries of as-grown GdBCO films into weak links resulting in higher normal-state resistivity values and also degraded in-field J_c values.

However, regardless of the above disappointing result, the effect of the growth conditions of Gd_2O_3 on its average particle sizes could be characterized by analyzing the microstructures taken from scanning transmission electron

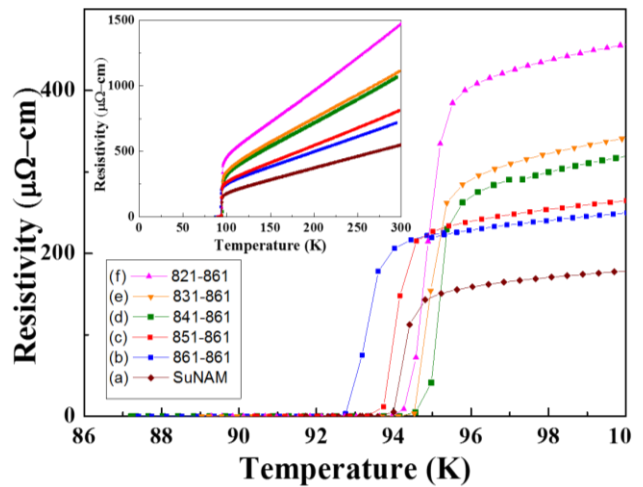


Fig. 4. The ρ - T curves for the GdBCO CC fabricated by RCE-DR in SuNAM (a), and also for GdBCO films grown at 861°C in 150 mTorr after the heat-treatment at (b) 861°C , (c) 851°C , (d) 841°C , (e) 831°C , and (f) 821°C in 30 mTorr for the size control of Gd_2O_3 particles.

microscopy (STEM)-energy dispersive X-ray spectroscopy (EDS) elemental mapping for 861-861, 841-861, and 821-861 samples, where the sample IDs denote the growth temperature of Gd₂O₃ in 30 mTorr - that of GdBCO in 150 mTorr. The cross-sectional TEM micrographs for these samples are represented in Fig. 5.

According to Figs. 5 (a)-(1), (2), and (3), bright round particles are Gd₂O₃ since Ba and Cu are free except Gd, and dark round particles are CuO or BaCuO₂ since Gd is free. Also, the matrix is GdBCO since all three elements of Gd, Ba, and Cu are detectable. These analysis results are exactly same with our previous report [4]. For the sample 861-861, the average particle size is 245.24 ± 50.5 nm. In the case of the sample 841-861, it can be seen that the Gd₂O₃ particles are relatively smaller than those in (a), and its average particle size is 128.09 ± 36.6 nm. The TEM micrograph of sample 821-861 in Fig. 5 (c) shows that the Gd₂O₃ particles seem to be rather bigger even though the growth temperature is lowered. However, the EDS mapping data in Figs. 5 (c)-(1), (2), and (3) show that there exist not only Gd but also Ba and Cu elements together,

indicating that the bright region is not the Gd₂O₃ phase but the GdBCO phase.

Since the sample 821-861 in Fig. 5(c) shows a peculiar microstructure different from those of the samples 861-861 and 841-861 in Fig. 5 (a) and (b), and further a round particle (red box) in the bright region shown in Fig. 5 (c) is also observed, we performed a high resolution TEM analysis on this particle as shown in Fig. 6. In Figs. 6 (c)-(e), the bright region is GdBCO, and dark round particles must be the Gd₂CuO₄ phase particle since Gd and Cu are detectable but Ba is free. It is in good agreement with the XRD pattern in Fig. 3 (f) and the stability phase diagram of GdBCO in Fig. 2. Since the processing routes passed the stable region of "Gd₂CuO₄ + L" (purple line in Fig. 2), the GdBCO phase might be formed from not only from Gd₂O₃ + L but also from Gd₂CuO₄ plus liquid, and thus Gd₂CuO₄ particle might be trapped in the GdBCO matrix phase. We speculate that Gd₂CuO₄ might react with the liquid (Ba-Cu-Oxide) phase by the peritectic recombination to form the GdBCO phase with relatively slower conversion rate compared with the peritectic recombination of Gd₂O₃

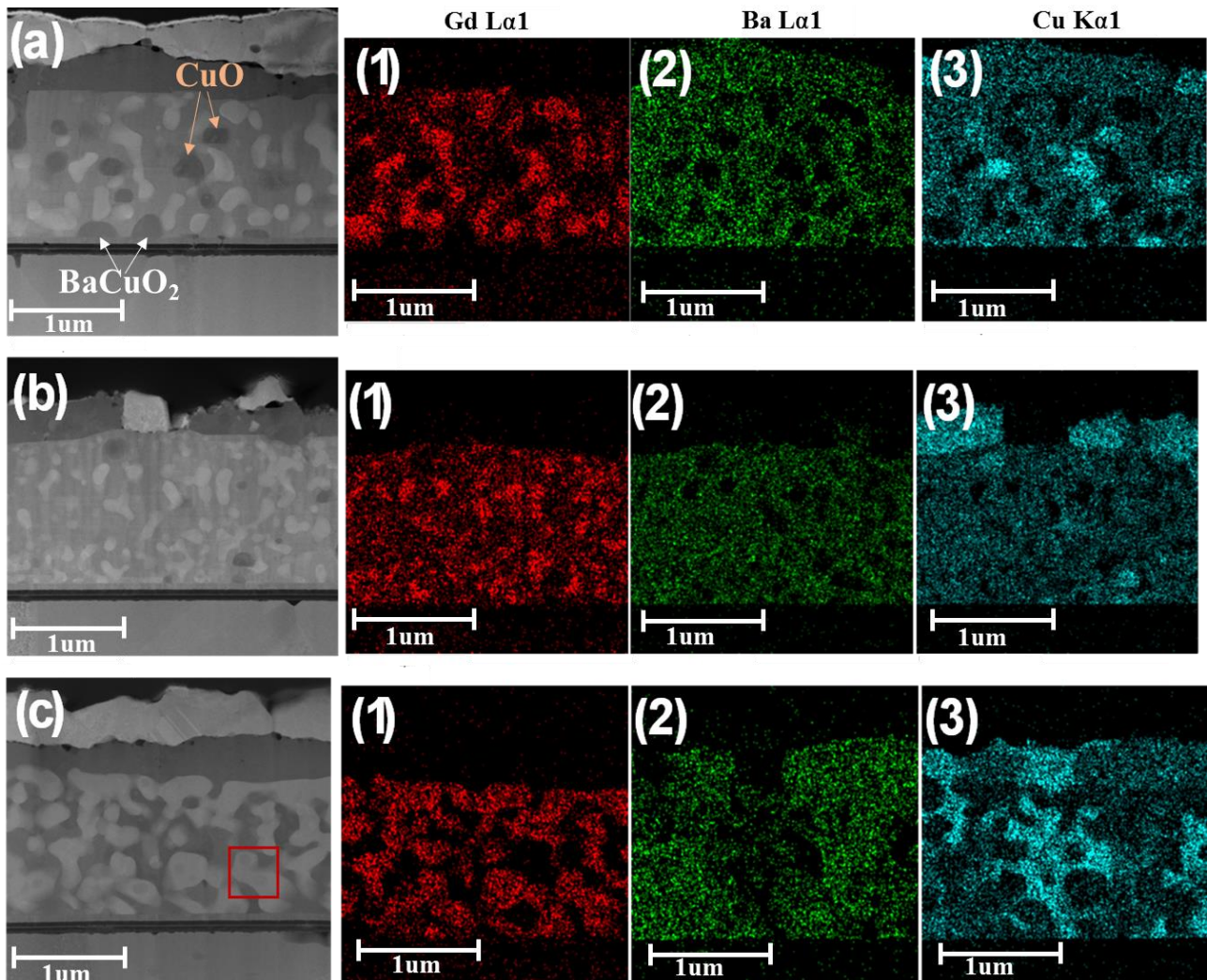


Fig. 5. Z-contrast STEM micrographs of the sample (a) 861-861, (b) 841-861, and (c) 821-861, (1) Gd, (2) Ba, and (3) Cu elements spectral images analysed for the its samples, respectively.

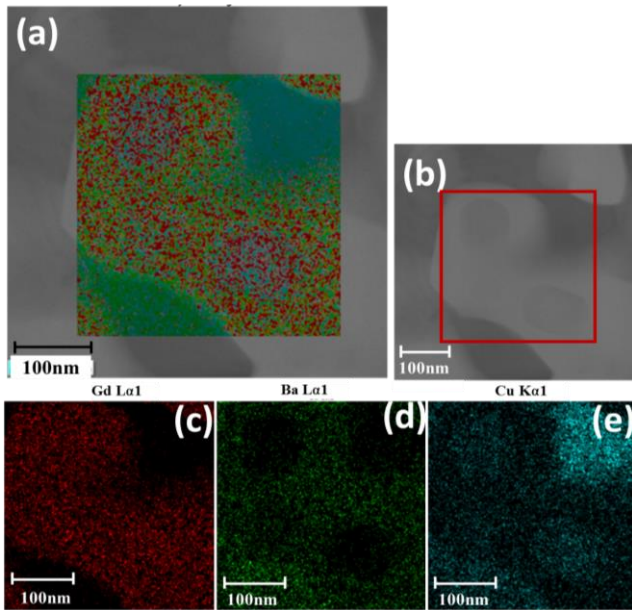


Fig. 6. Z-contrast STEM micrograph of the sample (a) and (b) 821-861, (c) Gd, (d) Ba, and (e) Cu elements spectral images analysed for the sample 821-861.

plus liquid, and hence the GdBCO matrix phase are not fully connected each other.

4. CONCLUSION

For the enhancement of the pinning properties, we tried to refine the Gd_2O_3 inclusions in the GdBCO films by controlling conversion processing conditions of the amorphous precursor film into the GdBCO film. In conclusion, although it was unsuccessful to confirm the enhancement of pinning properties due to the air-contamination by the refinement of Gd_2O_3 particles trapped in the GdBCO superconducting matrix, we could control their average particle size since, at relatively lower growth temperature of Gd_2O_3 in the liquid, more refined Gd_2O_3 particles were observable in the GdBCO matrix by TEM analysis. Therefore, we believe that higher in-field J_c values will be achievable from the samples with more refined Gd_2O_3 particles if we avoid the air contamination problem of the amorphous precursor film like the normal process of RCE-DR in SuNAM since the pinning properties are expected to become more effective.

ACKNOWLEDGMENT

This work was supported by the Korea Institute of Energy Technology Evaluation and Planning(KETEP) and the Ministry of Trade, Industry & Energy(MOTIE) of the Republic of Korea (No. 20131010501800).

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