



Technical Note

Experimental investigation of effective atomic numbers for some binary alloys



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ABSTRACT

In the present work, the gamma ray backscattering technique was used to determine the effective atomic numbers for certain binary alloys. With the help of a muffle furnace, the binary alloys were synthesized using the melt quenching technique with different compositions of ^{82}Pb , ^{50}Sn , and ^{30}Zn . The intensity distribution of backscattered photons from radioactive isotope ^{22}Na (511 keV) was recorded with the help of GAMMARAD5 [76 mm \times 76 mm NaI(Tl) scintillator detector] and analyzed as a function of both atomic number and thickness of the target material. The effective atomic numbers for the same binary alloys were also computed theoretically using the atomic to electronic cross-section method with the help of the mass attenuation coefficient database of WinXCom (2001). Good agreement was observed between theoretical and experimental results for the effective atomic numbers of all the selected alloys.

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1. Introduction

The backscattering of gamma ray photons plays an important role in radiation physics. Information regarding the thickness and density of a material can be obtained by the gamma ray backscattering technique. Gamma ray backscattering results correspond to the incident gamma photons, which are scattered at an angle of 180° with respect to the incident direction. The gamma backscattering technique is a nondestructive method of testing a material, in which there is no direct contact with the material under study or with the detector assembly [1]. When the ultrasonic method fails to work, the gamma backscattering method is very useful for estimating the thickness of hot objects and unclean and corroded surfaces [2]. The gamma ray backscattering method can be successfully used to measure corrosion in insulated pipe [3]. The importance of backscattered photons and multiple scattered photons from different types of materials were explored by different researchers [4–10].

The parameter “effective atomic number” for compounds/mixtures is analogous to the “atomic number” of elements. It signifies that, at a given energy, photons will interact with a compound/

mixture in a similar way as a single element of atomic number equivalent to that compound/mixture. This number provides the characteristics of the material, knowledge of which is very helpful in numerous applications such as technological, nuclear industry, space research program, and engineering applications. This number plays an important role in the field of nondestructive testing of samples of medical (radiology, nuclear medicines, and radiotherapy), agricultural (characterization of various types of woods and inhomogeneity in different parts of woods), and industrial (to check for void and insertion in the inner part of the material) interest. Hence, its measurement and computation with high accuracy are required.

An alloy is a mixture of two or more elements of different characteristic properties, created to enhance properties such as tensile strength, hardness, resistance to corrosion, etc. An alloy can be a combination of metals with metals or metals with nonmetals to achieve the desired characteristics. Since an alloy can be synthesized from combinations of various known elements in different compositions, these can be classified in numerous ways (on the basis of number of elements, on the basis of elements of major contribution, etc.). Here, binary alloys refer to alloys that are made up of two elements. In the present work, effective atomic numbers for different compositions of Pb–Sn, Pb–Zn, and Zn–Sn alloys were measured by the gamma ray backscattering technique; values were also computed using the electronic to atomic cross-section ratio method.

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Table 1
Chemical composition of prepared alloy samples.

Sample no.	Selected alloys	Chemical composition (fractional weight)
1.	Pb–Sn	$^{80}\text{Pb}_{20}\text{Sn}$ Pb = 0.80, Sn = 0.20
		$^{60}\text{Pb}_{40}\text{Sn}$ Pb = 0.60, Sn = 0.40
		$^{40}\text{Pb}_{60}\text{Sn}$ Pb = 0.40, Sn = 0.60
		$^{20}\text{Pb}_{80}\text{Sn}$ Pb = 0.20, Sn = 0.80
2.	Pb–Zn	$^{80}\text{Pb}_{20}\text{Zn}$ Pb = 0.80, Zn = 0.20
		$^{50}\text{Pb}_{50}\text{Zn}$ Pb = 0.50, Zn = 0.50
		$^{40}\text{Pb}_{60}\text{Zn}$ Pb = 0.40, Zn = 0.60
		$^{80}\text{Zn}_{20}\text{Sn}$ Zn = 0.80, Sn = 0.20
3.	Zn–Sn	$^{60}\text{Zn}_{40}\text{Sn}$ Zn = 0.60, Sn = 0.40
		$^{70}\text{Zn}_{30}\text{Sn}$ Zn = 0.70, Sn = 0.30

Table 2
Properties of the prepared alloy samples.

Sample no.	Selected alloys	Thickness (cm)	Volume (cm ³)	Mass (g)	Density (g/cm ³)
1.	Pb–Sn	$^{80}\text{Pb}_{20}\text{Sn}$ 0.75	2.94	27.19	9.25
		$^{60}\text{Pb}_{40}\text{Sn}$ 0.82	3.46	27.87	8.07
		$^{40}\text{Pb}_{60}\text{Sn}$ 0.877	3.22	27.07	8.41
		$^{20}\text{Pb}_{80}\text{Sn}$ 0.87	3.50	24.76	7.08
2.	Pb–Zn	$^{80}\text{Pb}_{20}\text{Zn}$ 0.54	2.66	19.92	9.62
		$^{50}\text{Pb}_{50}\text{Zn}$ 0.40	1.55	11.44	7.36
		$^{40}\text{Pb}_{60}\text{Zn}$ 0.60	2.24	12.0	5.33
		$^{80}\text{Zn}_{20}\text{Sn}$ 0.63	2.07	16.8	6.67
3.	Zn–Sn	$^{60}\text{Zn}_{40}\text{Sn}$ 0.53	1.55	14.33	6.57
		$^{70}\text{Zn}_{30}\text{Sn}$ 0.60	2.25	12.41	6.85

2. Materials and methods

The melt-quench technique has been employed to synthesize some binary alloys of ^{82}Pb , ^{50}Sn , and ^{30}Zn in different compositions, as shown in Table 1, using a muffle furnace. High purity (> 99.5%) metallic granules of Zn (melting point: 419°C), Sn (melting point: 231°C), and Pb (melting point: 327°C) were procured from Nice Chemicals (P) Ltd, Cochin, Kerala, India. These metallic granules were weighed in required quantities using an electronic digital balance (least count: 1 mg and maximum capacity: 500 g); then, the mixture was heated in an alumina crucible at 450°C for 10 min in a muffle furnace. The melt was then quickly poured into a cast iron mold of inner dimensions $2 \times 2 \times 2 \text{ cm}^3$ at room temperature. The physical properties of the prepared alloy samples are listed in Table 2.

Next, to measure the effective atomic numbers of the synthesized binary alloys in terms of back scattered photons, an experimental setup was designed and is shown in Fig. 1. It consists of radioactive isotopes ^{22}Na (photon energy = 511 keV, half-life = 2.6 years, strength = 1.2 μCi ; procured from BRIT, Mumbai, India), GAMMARAD5 [NaI(Tl) scintillator detector of dimensions 76 mm \times 76 mm and energy resolution is 7% at 662 keV, which is coupled with a multichannel analyzer; procured from Amptek Inc., Bedford, MA 01730, USA], and lead housing/collimators (which are used to minimize the noise or unwanted signal). The scintillator detector was placed in front of the ^{22}Na gamma ray source at a distance of 9.5 cm.

Before initializing the investigation, the detector was calibrated for energy scale. The pulse height spectra were recorded for different calibrating sources of ^{57}Co (122 keV), ^{133}Ba (81 keV, 302 keV, and 356 keV), ^{22}Na (511 keV), ^{137}Cs (662 keV), and ^{60}Co (1,173 keV and 1,332 keV). After calibration of the detector for energy scale (converting channel number into energy value along the x-axis), an attempt was made to calibrate the geometry to determine the thickness and atomic number of the interacting material. For this reason, the back scattered photon spectra of ^{22}Na (511 keV) from different thicknesses of certain metals (^{28}Ni , ^{47}Ag , ^{50}Sn , and ^{82}Pb) behind the source were recorded. The thickness of ^{28}Ni was increased by placing the Ni metal sheets one by one behind the previously placed Ni metallic sheet, without disturbing the experimental geometry. In this way, at a fixed energy (511 keV) and fixed atomic number ($Z = 28$), the dependence of the thickness on the backscattered count was analyzed. This procedure was repeated with samples of other metals (Ag, Sn, and Pb). All the spectra were recorded for a time period of 600 s, so as to yield a sufficient number of counts under the backscattered peak; hence, the associated statistical error is within the limit of $\pm 5\%$. The contributions of backscattered photons from particular samples were obtained by subtracting the backscattered photon spectrum obtained without the sample from the same spectrum obtained with the sample. Finally, the backscattered photon spectra for the synthesized alloys were recorded for the same time duration.

3. Results

After analyzing the recorded spectra, it was observed that the backscattered peak and the photo peak appear at 170 keV and 511 keV, respectively. Spectra recorded without and with a silver sheet (thickness 9.5 mm) at a scattering angle of 180° are shown in Fig. 2.

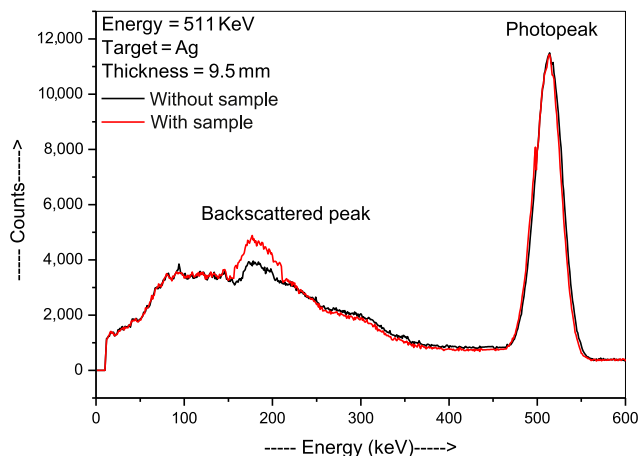


Fig. 2. Recorded spectra for ^{22}Na radioactive isotope with and without sample.

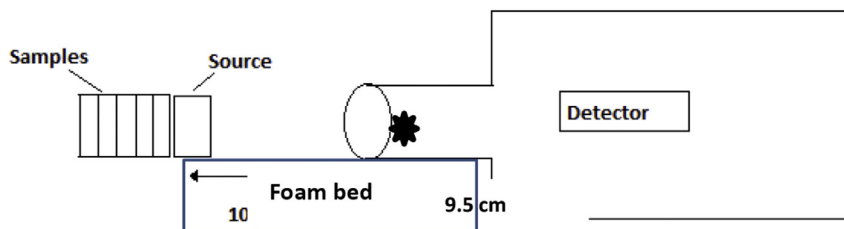


Fig. 1. Experimental setup (not to scale).

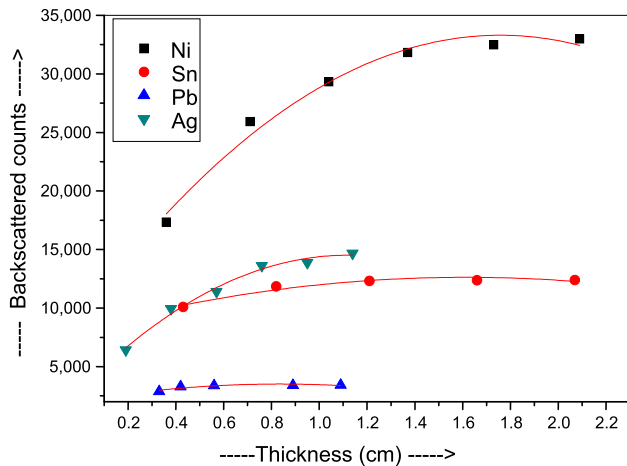


Fig. 3. Variation of backscattered counts with target thickness for some metals (Ni, Ag, Sn, and Pb).

Table 3

Effective atomic numbers for selected alloys at 511 keV gamma photons.

Sample no.	Selected alloys	No. of backscattered counts	Z_{eff} (theoretical)	Z_{eff} (experimental)
1.	Pb–Sn 80Pb ₂₀ Sn	3681	75.6	76.03 ± 0.38
	60Pb ₄₀ Sn	4469	69.2	70.40 ± 0.35
	40Pb ₆₀ Sn	6044	62.7	63.70 ± 0.32
	20Pb ₈₀ Sn	8265	56.4	57.40 ± 0.28
2.	Pb–Zn 80Pb ₂₀ Zn	3974	71.9	73.79 ± 0.36
	50Pb ₅₀ Zn	7820	56.7	57.40 ± 0.29
	40Pb ₆₀ Zn	9856	48.9	51.60 ± 0.25
	Zn–Sn 80Zn ₂₀ Sn	17146	34.2	35.4 ± 0.17
3.	60Zn ₄₀ Sn	15398	38.3	39.5 ± 0.19
	70Zn ₃₀ Sn	15730	36.9	38.6 ± 0.19

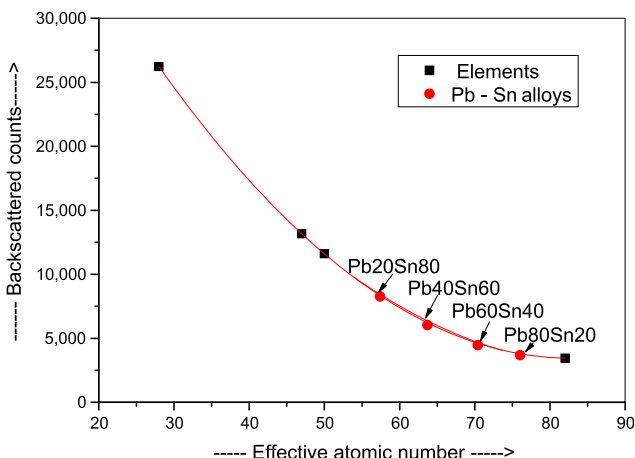


Fig. 4. Variation of backscattered counts with effective atomic number for Pb–Sn alloys of different compositions.

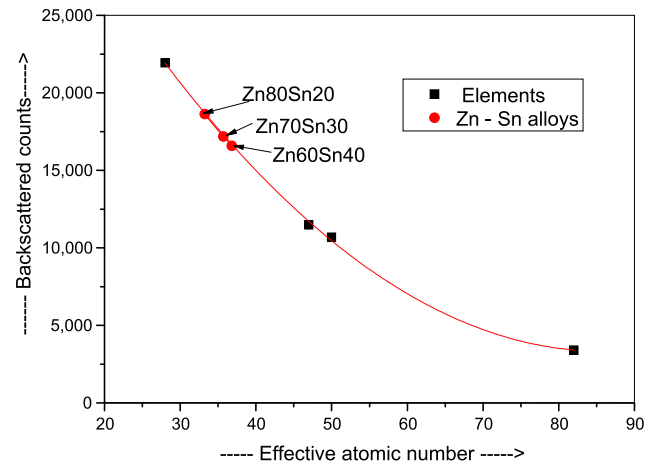


Fig. 5. Variation of backscattered counts with effective atomic number for Zn–Sn alloys of different compositions.

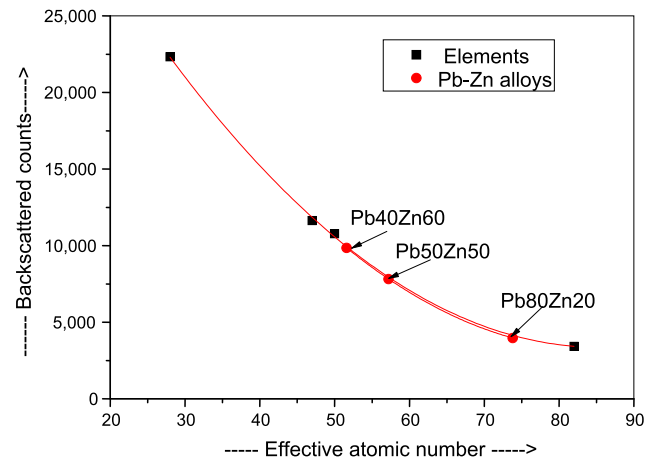


Fig. 6. Variation of backscattered counts with effective atomic number for Pb–Zn alloys of different compositions.

4. Discussion

Since the distance between the source and the detector remains the same, the number of counts under the photo peak also remains the same and, due to the increase/presence of material behind the source, the number of the backscattered count increases.

It has been observed that the number of backscattered photons depends not only on the atomic number of the interacting material, but also on the thickness of the interacting material, as shown in Fig. 3. The number of backscattered counts increases with an increase in the thickness of the interacting material and saturates afterwards at different thicknesses. The number of the backscattered count increases because an increase in the sample thickness results in a higher number of scattering centers for the interaction of primary gamma rays with the sample material. However, it can be clearly seen that for a fixed incident photon energy and almost similar thickness range, the variation in the number of backscattered photons with the atomic number of the interacting material behaves differently.

Using curve fitting techniques, the data points shown in Fig. 3 were further used to obtain polynomial equations for the

backscattered counts and thickness, so that for a fixed energy and atomic number, the number of backscattered counts can be obtained for any thickness (within the range of the experiment). In this way, the geometry was calibrated on energy as well as atomic number scale (for fixed energy and atomic number of the interacting material, a one-to-one relation was obtained between the thickness and the backscattered counts).

Then, the geometry was used to assign effective atomic numbers to the unknown samples (within a range of atomic number of 28–82 and at photon energy of 511 keV). The metallic sheets were replaced by the synthesized binary alloys and, again, the backscattered photon spectra were recorded for the same time duration. The contribution of the backscattered counts observed for the selected alloys (also mentioned in Table 3) are marked on the calibration line in Figs. 4–6. The corresponding effective number values are interpolated along the X-axis. These values are the effective atomic numbers of the alloys under study.

The theoretical values obtained using the atomic to electronic cross-section method, with the help of the mass attenuation coefficient database of XCom [11] and updated version WinXCom [12]; shows good agreement with the experimental measured values.

Conflicts of interest

All authors have no conflicts of interest to declare.

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