Effects of titanium oxide nanoparticles on *Oryzias latipes* embryos and sac-fry under different irradiation conditions

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ABSTRACT

Some phototoxicity of titanium dioxide nanoparticles (TiO₂ NPs) has been reported in recent years in studies with fish embryos or larvae. However, it is necessary to focus on the potential effects of embryonic exposure due to irreversible abnormalities and mortalities observed in sac-fry, and to expand various fish embryos to generate multiple test species. The aim of this study was to evaluate the effects of TiO₂ NPs under different irradiation conditions in exposed *Oryzias latipes* (*O. latipes*) at the embryonic and sac-fry stages. The effects of different irradiation conditions were observed using ultra-violet (UV) and visible light, and the corresponding effects were monitored by determining cumulative mortality and abnormality. *O. latipes* were exposed for 8 d to 0, 1, 5, 10, or 50 mg/L TiO₂ NPs under UV (4,818.86 mW/m² at the bottom of clear vials) or visible light, after which the embryos were transferred to NP-free embryo-rearing solution until 16 days post fertilization (dpf). Abnormalities of embryos and sac-fry increased at high TiO₂ NP concentrations under UV irradiation, compared to control samples treated with visible light or UV irradiation alone. This work provides information regarding the phototoxicity of TiO₂ NPs using *O. latipes* at the embryonic and sac-fry stages.

Keywords: Fish embryo, Oryzias latipes, Phototoxicity, TiO2 NPs, UV irradiation

1. Introduction

Titanium dioxide nanoparticles (TiO_2 NPs) have been used in beauty products, hygienic goods, and textiles [1-3]. Boxall et al. [2] presented modeling data used to estimate the TiO_2 NP concentration in water as 0.245 mg/L, based on a 100% market-penetration scenario of paints and sunscreen, particularly paints containing 5 mg/g TiO_2 NPs and sunscreen containing 50 mg/g TiO_2 NPs. TiO_2 NPs tend to become rapidly distributed throughout the environment because of their small sizes and non-biodegradability [3], and the potential exposure to bio-receptors increases as a consequence.

Several studies have been conducted in recent years to assess the toxicity of $\mathrm{TiO_2}$ NPs. Because the photoreactivity of $\mathrm{TiO_2}$ NPs can cause differing toxicities based on the bio-receptors that are activated [1, 4-5], the phototoxicities of $\mathrm{TiO_2}$ NPs have been tested in various organisms, including fish [1, 4, 6-11], waterfleas [7, 9], algae [12], and bacteria [13]. The phototoxicity of $\mathrm{TiO_2}$ NPs induced increased mortality of *Daphnia magna* [7, 9] and decreased colony formation of *Escherichia coli* and *Bacillus subtilis* [13].

Evidence suggests that TiO_2 NPs in surface water can readily absorb ultra-violet (UV) irradiation and that the absorbed energy can generate an excited state by electron-hole pair production on the surface of particles [4, 7, 8, 12].

Some authors have reported the phototoxicity of TiO2 NPs using fish embryos or larvae [1, 4, 6-7, 11]. These studies were focused on the effects of Danio rerio (D. rerio) embryos [1, 4, 11, 14] and larvae [11, 14], Oryzias latipes (O. latipes) larvae [7, 9], and Piaractus mesopotamicus juveniles [6]. Although Ma et al. [14] presented evidence to suggest that yolk-sac larvae were most sensitive to the phototoxicity of TiO2 NPs, it is still necessary to study their effects on potential embryonic exposure due to irreversible abnormalities and mortalities observed in sac-fry [15-16]. However, limited phototoxicity data exist regarding the effects of TiO2 NPs on fish embryos, and such studies have mainly focused on D. rerio. Therefore, it is necessary to expand various fish embryos for use as multiple test species in order to assess the phototoxicity of TiO2 NPs in water. O. latipes is recommended as a test fish species by Organization for Economic Co-operation and Development (OECD) No. 212 [17].



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The aim of this study was to evaluate the effects of TiO_2 NPs under different irradiation conditions in exposed O. latipes at the embryonic and sac-fry stages. The effects of different irradiation levels were observed using UV and visible light, and the corresponding effects were monitored by measuring cumulative mortality and abnormality. O. latipes was chosen as the test species due to the lack of available phototoxicity data. To the best of our knowledge, this is the first study to evaluate phototoxicity of TiO_2 NPs using O. latipes embryos.

2. Materials and Methods

2.1. Characterization and Preparation of Nanoparticles

TiO₂ NPs (particle size 21 nm, P25, white powder form) were purchased from Evonik Degussa (Germany). A field emission transmission electron microscope (FE-TEM, JEM2200FS, JEOL, Japan) was used to observe the morphology of the NPs. Specific surface area of the TiO2 NPs was analyzed by particle size analyzer (UPA-150, Microtrac, USA). An electrophoretic light scattering spectrophotometer (ELS-8000, Otuska Electronics Co., Japan) was used to measure particle size distributions of hydrodynamic diameters and zeta potential of 10 mg/L TiO2 NPs dispersed in deionized water (DW) and embryonic rearing solution (ERS). UV absorption spectra of 10 mg/L TiO2 NPs suspended in ERS was measured by a UV/Vis spectrophotometer (Libra S32 PC, Biochrom Ltd., Cambridge, UK) at 10-min intervals for 1 h, to verify aggregation of TiO₂ NPs. ERS included NaCl (1 g/L), KCl (0.03 g/L), CaCl₂·2H₂O (0.04 g/L), MgSO₄·7H₂O (0.163 g/L), and methylene blue (1.0 mg/L) to prevent fungal growth [18]. When UV spectra analysis, Methylene blue was eliminated from the ERS due to photodegradation [18]. A stock solution of 100 mg/L TiO2 NPs was prepared by dispersing the NPs in ERS with sonication for 10 min in a water bath sonicator (40 kHz frequency, Powersonic 420, Hwashin Technology, Korea). A series of exposure concentrations of TiO₂ NPs (0, 1, 5, 10, and 50 mg/L) were diluted with ERS.

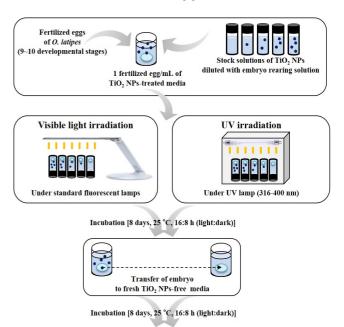
2.2. Test Species and Pre-culture Conditions

O. latipes (Japanese medaka) was received from the National Academy of Agricultural Science (NAAS, Suwon, Korea). Adult orange-red type O. latipes for broodstock were kept in dechlorinated tap water (pH 7.0 \pm 0.2; temp. 25 \pm 2°C; light:dark = 16:8 h), with a diet of commercial Tetra Min (Tetra Werke, Germany) and brine shrimp (OSI PRO80TM, USA) twice per day. On the day of the experiment, newly fertilized and healthy eggs of O. latipes were collected from the broodstock.

2.3. Phototoxicity Assay Using Fish Embryo and Sac-fry

The fish embryo and sac-fry assay was performed based on the OECD guidelines (No. 212) for chemical testing [10, 14, 15, 19]. Embryos (9-10 stages) were transferred into 2 mL clear vials with screw caps (Wheaton), one embryo per vial and one mL exposure solution per vial, and there were 10-15 replicates prepared. All experiments were repeated 1 to 4 times for treatments and 3 to 4 times for controls. The capped vials were placed upside down

in a photoreactor (model LZC-4, Luzchem Research Inc., Ottawa, ON, Canada) under a UV lamp for UV irradiation and in the laboratory under standard fluorescent lamps for visible light irradiation, before hatching. The photoreactor has one UV lamp (spectral distribution of 316 nm to 400 nm), positioned 18 cm above the test units. The light intensity was measured by a spectroradiometer (SPR 4001, Luzchem Research Inc., Ottawa, ON, Canada) and its values were 6,198.03 mW/m2 at the top of the clear vials and 4,818.86 mW/m² at the bottom of the clear vials. At 8 days post fertilization (8 dpf, when the control embryos hatched), exposure solutions were changed with NP-free embryo-rearing solution. UV irradiated- or visible light irradiated-vials were maintained in the incubator under pre-culture conditions (without oxygen and food supply in both the control and exposure groups). Fish embryos and sac-fry were observed daily under a dissection microscope based on cumulative mortality and various abnormalities; abnormal embryos and sac-fry: Heart damage (e.g., tube heart, blood circulation disorder, pericardial necrosis, pericardial edema, and hemostasis), whole body necrosis, malformation of the pericardium and peritoneum, gallbladder edema, small eyes, tail deformity, swim bladder deformity (e.g., undeveloped, over-inflated, and deflated swim bladder), and erratic swimming activity. The dead embryos or sac-fry were immediately eliminated from the test units. The detailed experimental design of fish embryo and sac-fry assay is shown in Fig. 1. The effect of Ti ions was not measured because TiO₂ NPs are insoluble in water [3].



 ${\bf Observation} \ \ {\bf of} \ cumulative \ mortality \ and \ abnormality \ of \ embryos/sac-fry$

Fig. 1. Experimental study design. Fertilized eggs of *Oryzias latipes* were exposed to TiO₂ NPs in embryo rearing solution until 8 days post fertilization (dpf), under visible light or UV irradiation. At 8 dpf, the exposure solution was replaced with TiO2 NP-free embryo rearing solution. All tests were assessed with respect to cumulative mortality and abnormalities of embryos and sac-fry until 16 dpf (modified from Shin et al. [10, 15]).

2.4. Statistical Analysis

The results were reported as the sum of cumulative mortalities and abnormalities of embryos and sac-fry. If abnormal embryo was counted, sac-fry developed from an abnormal embryo was excluded from the calculation (even if the abnormal sac-fry manifested). The data were presented as mean \pm standard error. Means were compared using one-way analysis of variance (ANOVA) (Origin 8, OriginLab Corp.). Multiple comparisons were conducted using Fisher test and differences were considered statistically significant at p < 0.05.

3. Results and Discussion

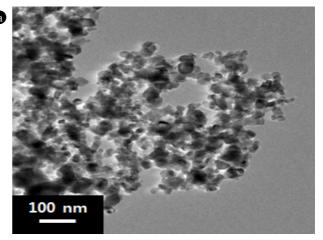
3.1. Physicochemical Characterization and Dissolution of TiO_2 NPs

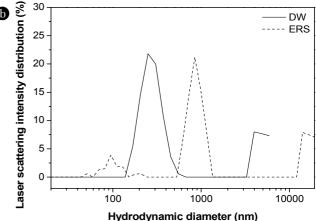
Fig. 2 shows physicochemical characterization of TiO₂ NPs. Morphological TEM image of TiO₂ NPs is shown in Fig. 2(a). Fig. 2(b) shows distributions of the hydrodynamic diameter of TiO₂ NPs suspended in DW and ERS. According to Table 1, mean hydrodynamic diameter of TiO₂ NPs dispersed in ERS was estimated to be 1,218 \pm 109 nm, which is 3.4-fold greater than the TiO₂ NPs dispersed in DW (354 \pm 15 nm). The zeta potential of TiO₂ NPs was estimated to be -7.0 \pm 2.4 mV (in ERS) and -2.0 \pm 1.4 mV (in DW). This indicates that TiO₂ NPs are rapidly coagulated or flocculated in solution. The specific surface area of TiO₂ NPs analyzed using the Brunauer-Emmett-Teller (BET) method was estimated to be 55 \pm 0.4 m²/g. The UV absorption spectra of 10 mg/L TiO₂ NPs suspended in ERS were determined at 10-min intervals for 1 h (Fig. 2(c)). TiO₂ NPs absorbed UVA light (320-400 nm) and peaked at 330-340 nm for TiO₂ NPs. The absorbance

Table 1. Glossary of Terms and Equations Used to Analyze Chlorophyll Fluorescence Transients

Tuorescence Transients	
Properties	TiO ₂ NPs
Manufacturer	Evonik Degussa
Crystal structure ^a	Anatase 72.6%, Rutile 18.4%, Amorphous 9%
Surface coating ^a	No
Mean hydrodynamic diameter $(nm)^b$	
Distilled water	354 ± 15
Embryo rearing solution	$1,218 \pm 109$
Zeta potential (mV) ^b	
Distilled water	-2.0 ± 1.4
Embryo rearing solution	-7.0 ± 2.4
Specific surface water $(m^2/g)^c$	$55~\pm~0.4$

^a supplied from the manufacturer.





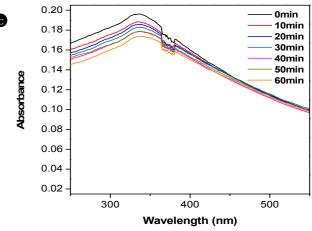


Fig. 2. Transmission electron microscope image (a), particle size distributions of hydrodynamic diameters (b), and absorption spectra (c) of TiO₂ NPs.

of TiO_2 NPs decreased as a function of increased irradiation time. This is possibly due to aggregation of NPs. When TiO_2 NPs were re-sonicated for 10 min, the absorbance of TiO_2 NPs increased (data not shown). In a previous study, Lee and An [12] reported the absorbance of TiO_2 NPs as a function of time until 1 h. They

^b measured by electrophoretic light scattering spectrophotometer (ELS).

^c measured by particle size analyzer.

reported the absorbance of ${\rm TiO_2}$ dispersed in OECD algal medium decreased as a function of increased irradiation time, similar to the results obtained in this study.

3.2. Effects of TiO₂ NPs on the Embryos and Sac-fry of O. latipes

Fig. 3-5 show embryos and sac-fry at 16 dpf of exposure to ${\rm TiO_2}$ NPs, under either visible light or UV irradiation. The abnormalities of embryos and sac-fry were increased at 10 and 50 mg/L NPs under UV irradiation, compared to the control under visible light irradiation or UV irradiation (p < 0.05) (Fig. 3). Under UV irradiation, compared to the control, at 50 mg/L ${\rm TiO_2}$ NPs, some ${\rm TiO_2}$ NP-treated embryos developed but exhibited abnormalities (p < 0.05). Tube heart, pericardial edema, and undeveloped or over-inflated swim bladder deformity were primarily induced as symptom of embryo and some exhibited hemostasis, malformation of pericardium, whole-body necrosis, pericardial edema, pericardial necrosis, and tail deformity (Fig. 4). In particular, the effect of ${\rm TiO_2}$ NPs during the embryonic stages of some fish (10% for 1 mg/L

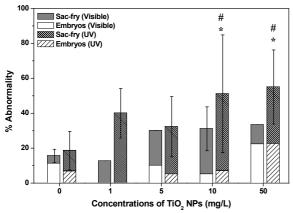


Fig. 3. Cumulative abnormality of TiO_2 NPs under visible light or UV irradiation on *Oryzias latipes*. Asterisks (*) indicate significant differences from controls under visible light irradiation (p < 0.05). Crosshatches (#) indicate significant differences from controls under UV light irradiation (p < 0.05).

and 10% for 50 mg/L) died as sac-fry, and embryonic exposure generated irreversible developmental impairment. Under visible light irradiation, some TiO₂ NP-treated embryos developed but exhibited abnormalities [e.g., deflated yolk, undeveloped or over-inflated swim bladder deformity, blood circulation disorder] at 50 mg/L TiO₂ NPs. The abnormalities shown by sac-fry included undeveloped or over-inflated swim bladder deformity and hemostasis. However, there was no significant increase of abnormality under visible light irradiation, compared to control (p < 0.05). There was no significant difference of abnormality between UV and visible light at the same exposure concentrations in the range of tested exposure concentrations (p < 0.05). No significant mortality of TiO₂ NP-treated sac-fry was observed under UV irradiation and visible light irradiation (p < 0.05) (Fig. 5).

3.3. Comparison of TiO_2 NPs on the Embryos and Sac-fry of *O. latipes* under UV and Visible Light

We studied the phototoxicity of TiO₂ NPs under UV and visible light using O. latipes embryos and sac-fry, and found that there are no significant differences at low exposure concentrations (0-50 mg/L). Likewise, Bar-Ilan et al. [4] studied the phototoxicity of anatase/rutile TiO₂ NPs using zebrafish embryos. They reported that there are no significant differences in the survival of embryos under UV or visible light at TiO2 NP concentrations of 0, 1, and 10 mg/L. However, their significant differences of survival are observed at 100, 500, and 1,000 mg/L. Clemente et al. [1] studied the phototoxicity of anatase and anatase/rutile TiO2 NPs using zebrafish eggs and larvae. They reported that there are no significant differences in the survival of eggs and larvae under UV or visible light at TiO2 NP concentrations of 0, 1, and 10 mg/L. However, their significant differences of survival are observed at 100 mg/L. Meanwhile, Faria et al. [5] studied the phototoxicity of rutile and anatase/rutile TiO2 NPs using zebrafish embryos. They reported that rutile TiO2 NPs did not cause significant mortality at a concentration of 1 mg/mL under solar radiation. However, anatase/rutile TiO₂ NPs caused significant mortality at the same concentration and radiation. Felix et al. [11] studied the phototoxicity of anatase/rutile and anatase poly (acrylic acid)-coated TiO2 NPs using

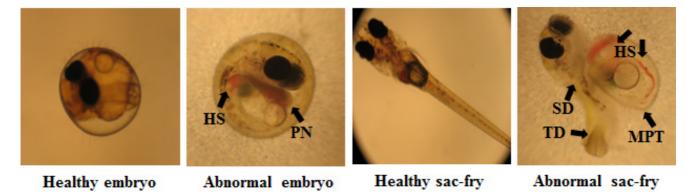


Fig. 4. Morphological appearance of *Oryzias latipes* exposed to TiO₂ NPs. HS: hemostasis, PN: pericardial necrosis, SD: swim bladder deformity, TD: tail deformity, and MPT: malformation of the pericardium and peritoneum.

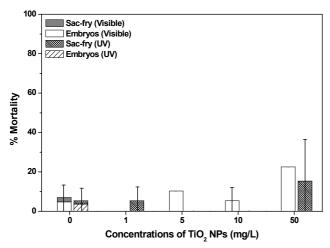


Fig. 5. Cumulative mortality of TiO₂ NPs under visible light or UV irradiation on *Oryzias latipes*.

zebrafish embryos and larvae. They reported that 10 mg/L uncoated ${\rm TiO_2}$ NPs induced hydroxyl radicals, production of lipid peroxidation, and increase of total glutathione level under UV illumination, unlike poly (acrylic acid)-coated ${\rm TiO_2}$ NPs, due to surface coating. It has been suggested that phototoxicity is due to the production of reaction oxygen species (ROS) that are generated by the exposure of ${\rm TiO_2}$ NPs to UV light and damage the receptors [4, 7-8, 12]. However, the results showed that lethal phototoxicity occurs only at relatively high exposure concentrations (100-1,000 mg/L).

4. Conclusions

This study showed that ${\rm TiO_2}$ NPs under UV or visible light cause abnormalities to O. latipes embryos and sac-fry at tested concentrations (10 and 50 mg/L), while the mortality of ${\rm TiO_2}$ NP-treated sac-fry was not significant under UV or visible light. Additionally, no significant differences were identified under UV or visible light at the same low exposure concentrations (0-50 mg/L) of bare ${\rm TiO_2}$ NPs. Therefore, further research is needed to assess the effects of higher exposure concentrations of ${\rm TiO_2}$ NPs than those tested in this study and the effects of different surface coating of nanoparticles. Further research is also needed to evaluate the toxicity of NPs in natural surface waters for investigating their potential photo-activation and risk involved in biotic partitioning of NPs.

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References

- 1. Clemente Z, Castro VLSS, Moura MAM, Jonsson CM, Fraceto LF. Toxicity assessment of TiO_2 nanoparticles in zebrafish embryos under different exposure conditions. *Aquat. Toxicol.* 2014;147:129-139.
- Boxall A, Chaudhry Q, Sinclair C, et al. Current and future predicted environmental exposure to engineered nanoparticles. Central Science Laboratory; 2007.
- Wang H, Wick RL, Xing B. Toxicity of nanoparticulate and bulk ZnO, Al₂O₃ and TiO₂ to the nematode Caenorhabditis elegans. Environ. Pollut. 2009;157:1171-1177.
- Bar-Ilan O, Louis KM, Yang SP, et al. Titanium dioxide nanoparticles produce phototoxicity in the developing zebrafish. Nanotoxicology 2012;6:670-679.
- Faria M, Navas JM, Soares AMVM, Barata C. Oxidative stress effects of titanium dioxide nanoparticle aggregates in zebrafish embryos. Sci. Total Environ. 2014;470-471:379-389.
- Clemente Z, Castro VL, Feitosa LO, et al. Fish exposure to nano-TiO₂ under different experimental conditions methodological aspects for nanoecotoxicology investigations. Sci. Total Environ. 2013;463-464:647-656.
- 7. Li S, Pan X, Wallis LK, Fan Z, Chen Z, Diamond SA. Comparison of TiO₂ nanoparticle and graphene-TiO₂ nanoparticle composite phototoxicity to *Daphnia magna* and *Oryzias latipes*. *Chemosphere* 2014;112:62-69.
- Ma H, Brennan A, Diamond SA. Photocatalytic reactive oxygen species production and phototoxicity of titanium dioxide nanoparticles are dependent on the solar ultraviolet radiation spectrum. *Environ. Toxicol. Chem.* 2012a;31:2099-2107.
- 9. Ma H, Brennan A, Diamond SA. Phototoxicity of ${\rm TiO_2}$ nanoparticles under solar radiation to two aquatic species: *Daphnia magna* and Japanese medaka. *Environ. Toxicol. Chem.* 2012b;31:1621-1629.
- Shin YJ, Nam SH, An YJ. Continuous UV irradiation increases the adverse effects of photoreactive nanoparticles on the early development of *Oryzias latipes*. Environ. Toxicol. Chem. 2015;35:1195-1200.
- Lindsey CF, Erik JF, Yuhe H, Greg GG. Poly(acrylic acid)-coated titanium dioxide nanoparticle and ultraviolet light co-exposure has minimal effect on developing zebrafish (*Danio rerio*). *Environ. Sci. Nano.* 2017;4:658-669.
- 12. Lee WM, An YJ. Effects of zinc oxide and titanium dioxide nanoparticles on green algae under visible, UVA, and UVB irradiations: No evidence of enhanced algal toxicity under UV pre-irradiation. Chemosphere 2013;91:536-544.
- Kim SW, An YJ. Effect of ZnO and TiO₂ nanoparticles preilluminated with UVA and UVB light on Escherichia coli and Bacillus subtilis. Appl. Microbiol. Biotechnol. 2012;95:243-253.
- Ma H, Diamond SA, Phototoxicity of TiO₂ nanoparticles to zebrafish (*Danio rerio*) is dependent on life stage. *Environ. Toxicol. Chem.* 2013;32:2139-2143.

- 15. Shin YJ, Nam SH, An YJ. Japanese medaka exposed to gold nanoparticles: Only embryonic exposure generates irreversible hatching failure, developmental failure, and mortality of sac-fry. Comp. Biochem. Phys. C. 2014;161:26-32.
- 16. Truong L, Saili KS, Miller JM, Hutchison JE, Tanguay RL. Persistent adult zebrafish behavioral deficits results from acute embryonic exposure to gold nanoparticles. *Comp. Biochem. Phys. C.* 2012;155:269-274.
- 17. Organization for Economic Co-operation and Development
- (OECD). Guideline for testing of chemicals no. 212. Fish, short-term toxicity test on embryo and sac-fry stages. 1998.
- 18. Kirchen RV, West WR. The Japanese medaka: Its care and development [Ichthyology; Fishes]. Burlington, North Carolina: Carolina Biological Supply Co.; 1976.
- 19. Shen W, Li Z, Wang H, Liu Y, Guo Q, Zhang Y. Photocatalytic degradation for methylene blue using zinc oxide prepared by codeposition and sol–gel methods. *J. Hazard. Mater.* 2008;152:172-175.