

Properties of Dye Sensitized Solar Cells with Adding Nano Carbon Black into Blocking Layer

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ABSTRACT

Blocking layers with nano carbon blacks (NCBs) were prepared by adding 0.0 ~ 0.5 wt% NCBs to the TiO₂ blocking layer. Then, dye sensitized solar cells (DSSCs) were fabricated with a 0.45 cm² active area. TEM and micro-Raman spectroscopy were used to characterize the microstructure and phases of the NCBs, respectively. Optical microscopy and AFM were used to analyze the microstructure of the TiO₂ blocking layer with NCBs. UV-VIS-NIS spectroscopy was used to determine the band gap of the TiO₂ blocking layer with NCBs. A solar simulator and potentiostat were used to determine the photovoltaic properties and impedance of DSSCs with NCBs. The energy conversion efficiency (ECE) increased from 3.53 to 6.20 % when the NCB content increased from 0.0 to 0.3 wt%. This indicates that the effective surface area and electron mobility increased in the TiO₂ blocking layer with NCBs. However, the ECE decreased when the NCB content was increased to over 0.4 wt%. This change occurred because the effective electron transport area decreased with the addition of excessive NCBs to the TiO₂ blocking layer. The results of this study suggest that the ECE of DSSCs can be enhanced by adding the appropriate amount of NCBs to the TiO₂ blocking layer.

Key words : Nano carbon black, Blocking layer, Energy conversion efficiency, Dye sensitized solar cell

1. Introduction

Dye sensitized solar cells (DSSCs) have been studied intensively as a next-generation energy device since it was first introduced in 1991 by O'Regan and M. Grätzel.¹⁾ The new cell can provide low-cost and ease of fabrication into larger diameters, compared to the conventional silicon cell.

A DSSC typically consists of working electrode, electrolyte, and counter electrode to perform a redox reaction. Light enters the cell through a transparent electrode, and is absorbed by sensitizer generating an excited electron-hole pair. Then the electron can move within the conduction band of a neighboring TiO₂ semiconductor (SC). Current is generated when the transported electron moves to working electrode through interfaces of nano TiO₂. The generated hole in dye material, on the other hand, moves to counter electrode through the electrolyte, and is reduced by accepting an electron.²⁾

The working electrode consists of transparent conductive oxide (TCO), blocking layer (compact TiO₂), oxide SC layer (mesoporous TiO₂), and dye. To increase efficiency of DSSC, studies have been mostly involved in engineering the working electrode, especially the oxide SC layer. Kilic *et al.*³⁾ reported an increase in energy conversion efficiency (ECE)

up to 7.27% from 5.10% by adding Fe₂O₃ into TiO₂ SC, and thus decreasing the number of recombining electrons via band gap engineering.

Zhang *et al.*⁴⁾ accomplished 6.34% of ECE from 3.63% by incorporating carbon nano tube (CNT) into TiO₂ SC, and thus by increasing its specific surface area and electron mobility. Kim *et al.*⁵⁾ reduced potential barrier between TCO and TiO₂ SC by adding porous nano carbon into TiO₂ SC, and achieved 3.38% of ECE compared to 2.49% of the unloaded TiO₂ SC.

On the other hand, engineering on blocking layer is relatively less studied. However, Noh *et al.*⁶⁾ studied on the nano-diamonds (NDs) added blocking layer, and reported an increase in ECE recently. They proposed that the increased specific surface area of blocking layer led an increase in both the amount of absorbed dye on TiO₂ SC and the shunt resistance of TiO₂ blocking layer. The result suggests that incorporation of other carbon sources into blocking layer could also enhance ECE of DSSCs. Note that nano carbon black (NCB) has bonding of mainly sp²-orbital, like many other allotropes such as graphite, graphene, CNT, and ND^{7,8)}. Also, note that analysis of these allotropes in NCB can be conveniently carried out by Raman spectroscopy.⁹⁻¹¹⁾

We expected that NCB can effectively increase the specific surface area of TiO₂ blocking layer, which is normally in a rather compact form. Since this could lead to increase in surface area of the neighboring TiO₂ SC layer, we anticipated a higher dye loading on TiO₂ SC layer. We also

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expected a favorable widening of band gap for TiO_2 in blocking layer with NCB loading. When an electron moves from mesoporous TiO_2 layer to TCO, lower potential barrier by this change in band gap could reduce electron loss due to recombination. The goal of this study is to confirm these arguments and to accomplish an improved ECE with 0.0 ~ 0.5 wt% NCB loading into TiO_2 blocking layer in DSSC.

2. Experimental Procedure

NCB was ultra-sonicated in ethanol for 1 hr, and dropped on a carbon-coated Cu grid and dried for microstructural analysis by TEM (H-7600, Hitach Co.) up to a magnification of 30,000. Its chemical composition was analysed by micro-Raman spectroscopy (UniRaman, UniThink Co.) using NCB dispersed in ethanol on a silicon wafer. Characteristic peaks from carbon allotropes in NCB were identified under exposure for 5 seconds in the range of $1200 \sim 2900 \text{ cm}^{-1}$ in an accumulation mode for 60 scans with noise elimination.

The blocking layer of DSSC was prepared by adding 0.0 ~ 0.5 wt% NCB (average dia. = 55 nm, DASHBLACK, OCI Co.) into TiO_2 layer. The process consists of; (a) dispersing NCB in a solution of Titanium(IV)bis(ethyl acetoacetato)-diisopropoxide and 1-Butanol, (b) ultra-sonication followed by spin-coating for 500 rpm-10 seconds and 2000 rpm-40 seconds, and (c) heat-treated at 500°C for 15 min.

We observed the blocking layers by an optical microscope (Model 815000, GIA Instruments Co.) under dark-field and overhead spot illuminations. A digital camera (Coolpix 4500, Nikon Co.) attached on eyepiece lens produced the necessary images of the blocking layers.

AFM (SPM25DRM, Park Scientific Instruments) was used to confirm the increase in specific surface area of TiO_2 layer with NCB addition (0 wt% and 0.5 wt%) after coating the layer on a glass substrate. Surface roughness was measured in terms of RMS (root mean square) by scanning over $2 \times 2 \mu\text{m}^2$ area in a non-contact mode.

Band gap changes of NCB-added (0 wt% and 0.5 wt%) blocking layer were confirmed by measuring transparency in the wavelength range of $300 \sim 800 \text{ nm}$ at medium scan speed with UV-VIS-NIR (Shimadzu Co., UV3105PC), followed by Tauc plot transformation (Tauc plot program, Shimadzu Co.).

TiO_2 SC film was doctor-bladed on the prepared blocking layer by using TiO_2 paste (20 nm, Dyesol DSL 18NR-T of 10) and heat-treated at 500°C for 30 min. Dye was adsorbed on TiO_2 SC by using 0.5 mM of cis-vis bis-ruthenium (II) bis-tetrabutylammonium (N719). The multi-layered working electrode was completed as shown bottom of Fig. 1, and it consists of a glass, FTO (fluorine-doped tin oxide), blocking layer (TiO_2 with NCB), TiO_2 semiconductor (SC), and dye (N719).

The counter electrode was prepared by RF sputter (MHS-1500, Moohan Co., 300 W, 13.56 MHz) to form a 100 nm-Pt film on a glass substrate with a target of 99.99% Pt. A flow

of 40 sccm Ar at pressure of 5 mtorr at room temperature (RT) was set for the process. The prepared working and counter electrodes were fixed at position and filled with electrolyte. Fig. 1 shows the final version of DSSC consisted of working electrode/electrolyte/counter electrode. Its active area was estimated as 0.45 cm^2 .

Impedance of DSSC was determined by solar simulator (PEC-L11, Peccell) and potentiostat (Iviumstat, Ivium) to verify interfacial resistance. The analysis carried out in the frequency range of 10 mHz ~ 1 MHz applying AC voltage and collecting the current responses. I-V (current-voltage) characteristic of DSSC was measured by the same instruments, but under a different setup; a 100 W Xenon lamp was the illumination source at 1 sun (100 mW/cm^2) condition. From the I-V curves, short-circuit current density, open circuit voltage, fill factor, and ECE were evaluated.

3. Results and Discussion

Figure 2 shows TEM image of NCB magnified to 30,000 times, verifying an average particle size of 55 nm and revealing an agglomerated, spherical, and amorphous form. No crystalline allotrope such as CNT or C_{60} was observed, thus we suggested the presence of graphite and/or graphene as the main phases in NCB.

Figure 3 is the result of Raman analysis for NCB which indicates the presence of pure graphite only. Although peaks at 1350 cm^{-1} and 1600 cm^{-1} are for both graphite and graphene, the presence of graphene is excluded since

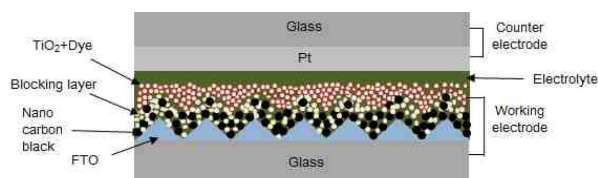


Fig. 1. Illustration of the cross sectional structure of the proposed DSSC.

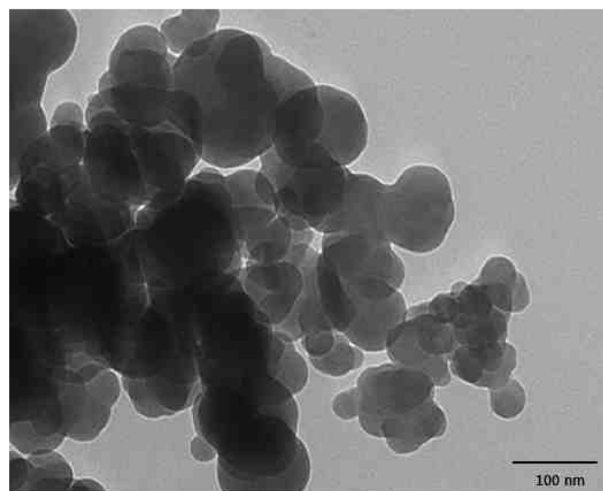


Fig. 2. TEM image of nano carbon black.

another strong peak at 2700 cm^{-1} for graphene is absent.^{7,9)}

Figure 4 shows optical images of electrodes without and with the dispersed 0.5 wt% NCB in the blocking layer. By comparing with Fig. 4(a), we confirmed that shiny contrast spots in Fig. 4(b) represent evenly-dispersed NCB. It also suggests that ultra sonicator and spin coating are appropriate processes to load NCB uniformly.

Figure 5 is AFM images of the blocking layers on a glass substrate with 0.0 wt% and 0.5 wt% NCB, which show RMS values of 2.5 nm and 6.41 nm, respectively. Thus, we verified increase in specific surface area of blocking layer by adding NCB on the blocking layer.

Figure 6 is a Nyquist diagram composed of real and imaginary terms for DSSCs with 0.0 ~ 0.5 wt% NCB under applied frequency. It is evident that the curves show three

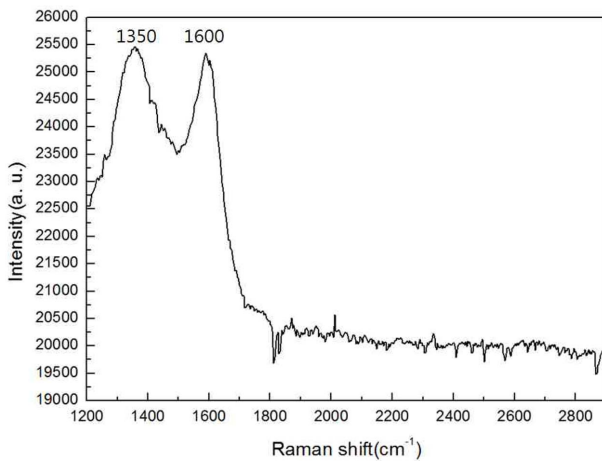


Fig. 3. Raman spectrum of nano carbon black.

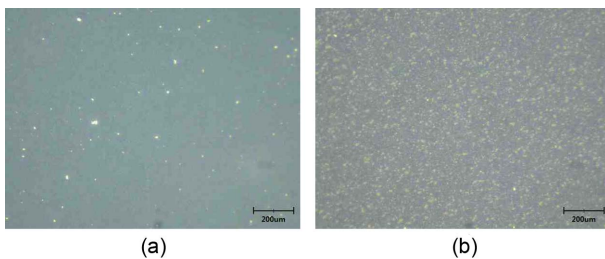


Fig. 4. Optical images of blocking layers with nano carbon black (NCB): (a) 0.0 wt% NCB, and (b) 0.5 wt% NCB.

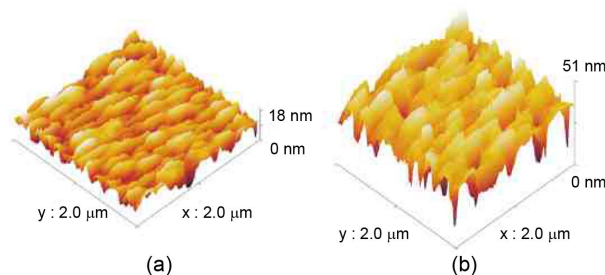


Fig. 5. AFM images of blocking layer with nano carbon black NCB: (a) 0.0 wt% NCB, and (b) 0.5 wt% NCB.

half-circle (R_1 , R_2 , R_3) like an internal resistance curve of conventional DSSC. R_1 value at $10^3 \sim 10^5$ Hz represents interfacial resistance by electron transport across TCO/ TiO_2 and electrolyte/counter electrode. The value decreased drastically from $14.6\ \Omega$ to below $2.0\ \Omega$ with NCB addition. It is attributed to increased contacts between TCO and TiO_2 layer originated from increased surface area of TiO_2 layer with NCB loading.

R_2 value at $1 \sim 10^3$ Hz represents resistance within TiO_2 SC and across TiO_2 SC/electrolyte interface. R_2 values are all about $5.5\ \Omega$, since the involved components are same. R_3 value at higher than 10^6 Hz is related to diffusing redox species within the electrolyte. R_3 values are all about $3\ \Omega$, since the electrolyte is same. Thus, we confirmed that NCB addition increased specific surface area of TiO_2 layer, and decreased interfacial resistance between TCO and TiO_2 layer.

Figure 7 is I-V data of DSSC with 0.0 ~ 0.5 wt% NCB addition. NCB addition clearly increased short-circuit cur-

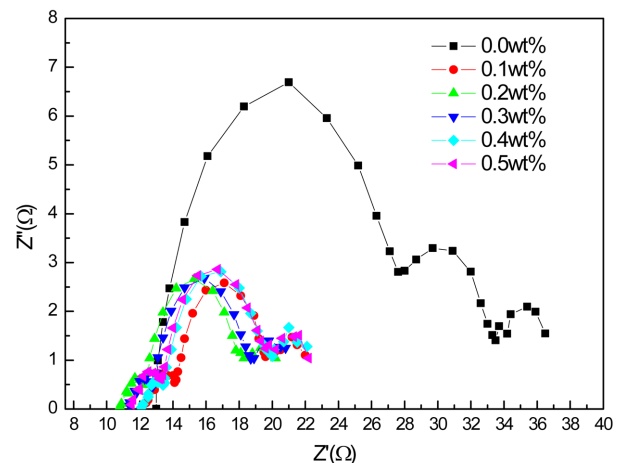


Fig. 6. Impedance of DSSCs employing blocking layer in which the amount of nano carbon blacks varies in the range of 0.0 ~ 0.5 wt%.

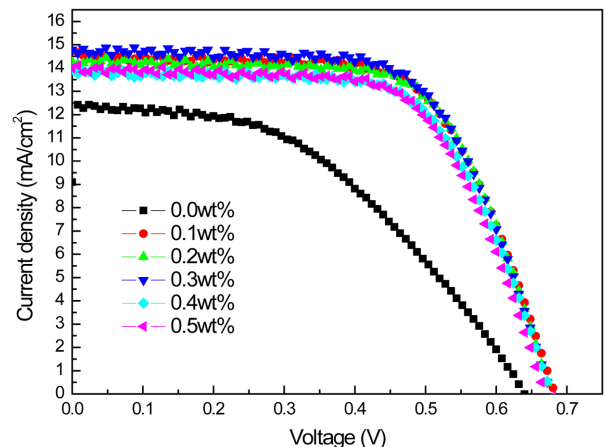


Fig. 7. Current-voltage (I-V) characteristic of DSSCs employing blocking layer in which the amount of nano carbon blacks varies in the range of 0.0 ~ 0.5 wt%.

rent density, open circuit voltage, and fill factor, and thus increased ECE.

Table 1 is a summary of results for I-V curves of Fig. 7. NCB addition of 0.3 wt%, especially, increased ECE up to 6.20%, which corresponds to enhancement of 1.75 times. We believe that increase in surface area and change within the band gap by NCB addition are responsible for the result.

Firstly, this argument supported by the surface roughness measurement as shown in Fig. 5 indicates that increase in surface area of blocking layer can induce the same effect on the mesoporous TiO_2 above, and eventually can give a higher dye absorption. Also, impedance results as shown in Fig. 6 and the decrease in value of R_1 , which is interfacial resistance between TCO and TiO_2 , indicate increase in specific surface area. Thus, the result showed reduction in the value of R_1 as both specific surface area of mesoporous TiO_2 , which was formed on the blocking layer and adsorption capacity increased at the same time according to increased specific surface area of the blocking layer itself by adding NCB on the blocking layer. Meanwhile, decreasing ECE with excessive NCB (> 0.4 wt%) can be understood by the same argument; with over-loaded NCB, the effective path for electron transport was rather decreased due to the relatively-reduced portion of TiO_2 in the blocking layer that is responsible for the electron move.

Secondly, the possibility of increase in ECE by band gap change according to addition of NCB can be verified from the values of short-circuit current density and open circuit voltage as shown in Table 1. Short-circuit current density increased certainly by addition of NCB. This means the possibility for formation of new band which reduces recombination during electron transport across existing TiO_2 SC layer to blocking layer. Also, the possibility can be verified by increase in open circuit voltage which is related to the Fermi level of TiO_2 electrode. Minor increase in band gaps is verified according to NCB addition as we obtained more direct evidence from UV-VIS-NIR spectroscopy that shows band gaps of 3.60 and 3.61 eV with 0.0 wt% and 0.3 wt% NCB additions, respectively. We decided that the increase in band gaps reduces electron loss when produced electrons from the dye move to the blocking layer from mesoporous TiO_2 which is oxide SC layer.

In summary, appropriate amount of NCB addition resulted in an increase in specific surface area both on

blocking layer and on TiO_2 SC layer. We also created a slightly wider band gap which provides a reduced electron loss during electron transport, and successfully enhanced ECE of DSSC.

4. Conclusions

We employed 0.0 ~ 0.5 wt% of NCB into the blocking layers of working electrode of DSSC, and evaluated their physical and opto-electrical properties. NCB increased the specific surface area of blocking layer and decreased the interfacial resistance between TCO and TiO_2 . We also verified that the ECE increases up to 0.3 wt% addition of NCB. This attributes to the increase in specific surface area of blocking layer and to the newly-formed band of which can decrease interfacial resistance. However, when more than 0.4 wt% of NCB was added, the efficiency decreased due to the decrease in short-circuit current density coming from the diminished TiO_2 portion which is responsible for electron transport. By employing the appropriate amount of NCB, we demonstrated an improvement on ECE of DSSC device.

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Table 1. Photovoltaic Performance and Energy Conversion Efficiency of DSSC

Sample	Voc (V)	FF	Jsc (mA/cm ²)	η (%)
0.0 wt%	0.632	0.437	12.81	3.53
0.1 wt%	0.689	0.616	14.73	6.10
0.2 wt%	0.682	0.632	14.24	6.11
0.3 wt%	0.682	0.621	14.78	6.20
0.4 wt%	0.684	0.620	13.92	5.80
0.5 wt%	0.675	0.616	13.98	5.74

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