

Transmission Electron Microscopy Investigation of Hot-pressed ZrB₂-SiC with B₄C Additive

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ABSTRACT

This paper reports the microstructure of hot-pressed ZrB₂-SiC ceramics with added B₄C as characterized by transmission electron microscopy. ZrB₂ has a melting point of 3245°C, a relatively low density of 6.1 g/cm³, and specific mechanical properties at an elevated temperature, making it a candidate for application to environments with ultra-high temperatures which exceed 2000°C. Due to the non-sinterability of ZrB₂-based ceramics, research on sintering aids such as B₄C or MoSi₂ has become prominent recently. From TEM investigations, an amorphous layer with contaminant oxide is observed in the vicinity of B₄C grains remaining in hot-pressed ZrB₂-SiC ceramics with B₄C as an additive. The effect of a B₄C addition on the microstructure of this system is also discussed.

Key words : Ultra-high-temperature ceramics (UHTCs), ZrB₂-SiC, B₄C addition, Densification

1. Introduction

ZrB₂-based ceramics are among the most favorable candidates for ultra-high-temperature ceramics, including thermal protection systems (TPS) for leading edge parts on hypersonic planes, re-entry space vehicles, or propulsion systems as well as conventional applications such as furnace elements and refractory crucibles.¹⁻³⁾ In order to accomplish these purposes, it is necessary to fabricate dense body parts made of ZrB₂-based ceramics. The densification of ZrB₂ powder generally requires very high temperatures due to the covalent nature of the bonds and due to the low bulk and grain boundary diffusion rates of ZrB₂.⁴⁾ Due to their non-sinterable characteristics, ZrB₂-based ceramics have been densified by a variety of methods, including hot pressing,⁵⁻⁷⁾ spark plasma sintering,⁸⁻¹⁰⁾ reactive hot pressing,^{11,12)} and pressureless sintering.¹³⁻¹⁵⁾

Furthermore, sintering additives are frequently used in the ZrB₂-based system in order to enhance the densification in conjunction with various sintering techniques.^{3,4)} For example, two of the most commonly used sintering additives are MoSi₂^{5,9)} and B₄C,^{6,13-15)} which act as a liquid former or a reducing agent, respectively, enhancing the densification characteristics during sintering. On the other hand, oxygen impurities, such as B₂O₃ and ZrO₂, have been shown to promote grain coarsening by evaporation-condensation kinetics

and to limit the final density in non-oxide ceramics.¹⁶⁾ In order to enhance the resulting density, it has been suggested that the total oxygen content must be reduced to less than 0.5 wt.% or that a strong reducing additive should be used to remove oxide contaminants. Given the current situation, B₄C is a reactive agent which enhances densification by reacting with and removing surface oxides on ZrB₂ particles. In our previous work,⁶⁾ it was reported that the hot pressing of ZrB₂-20 vol.% SiC at 1700°C for 2 h at a pressure of 30 MPa attained only a relative density of ~87%, whereas a relative density of ~95% was obtained for ZrB₂-20 vol.% SiC with an extra 5 vol.% of B₄C under identical hot pressing conditions.

Given that the mechanical properties of ZrB₂-based ceramics are closely related to the microstructure and the densification process and sintering additives used, a close examination of the microstructure is necessary to produce ZrB₂-based ceramics capable of enhanced performance levels. In this study, the microstructure of hot-pressed ZrB₂-SiC ceramics with B₄C as an additive was characterized by transmission electron microscopy. The grains and grain boundaries were analyzed using high-resolution imaging and energy dispersive X-ray spectroscopy (EDS) techniques. The effects of a B₄C addition on the densification and mechanical properties of ZrB₂-SiC are also discussed.

2. Experimental Procedure

ZrB₂-20 vol.% SiC and ZrB₂-20 vol.% SiC with an extra 5 vol.% of B₄C were prepared for this study using raw powders of ZrB₂ (Grade F; 1.88 μm, Japan New Metals. Co. Ltd.,

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Japan), α -SiC (FCP 15C; 0.5 μm , SIKA TECH, Germany), and B₄C (Grade A; 1.5 μm , UK Abrasives, USA). The samples were sintered by hot pressing. After weighing, the powders were mixed by planetary milling with WC-Co jars and balls in isopropyl alcohol for 4 h. The slurry of the powder mixture was vaporized by a hot plate upon stirring to create cakes, and the cakes of powder mixtures were dried in an oven for 24 h. The dried powder mixtures were then sieved with a #120-mesh sieve for granulation. The prepared powder mixtures were hot-pressed in a graphite mold. The hot pressing step was carried out at 1900°C for 2 h under a pressure of 30 MPa in a flowing Ar atmosphere.

For the hot-pressed samples, the densities, microstructures, and mechanical properties were characterized. The apparent densities were measured using the Archimedes method and the relative densities were calculated by comparing the apparent densities with the theoretical densities via the rule of mixture. The specimens were also ground with a diamond wheel and then polished from 6 to 1 mm of diamond slurry for a microstructure analysis and hardness measurements. The microstructures were examined using a field-emission scanning electron microscope (FE-SEM, JSM-6701F, JEOL, Japan), and the Vickers hardness was measured by a micro-hardness tester (QM-2, Nikon, Japan).

Cross-sectional foils for the transmission electron microscopy analysis were sliced from the polished surface of the samples using a focused ion beam (FIB, Helios 600i, FEI, USA) with a Ga ion source accelerated at 30 kV. The slices were finally thinned with a fine ion beam current to less than 100 nm, which is an electron-transparent thickness in TEM. Thin foils of the samples were examined by a transmission electron microscope (TEM, Tecnai G2 F30, FEI, USA) equipped with an energy dispersive X-ray spectroscopy (EDS) analyzer.

3. Results and Discussion

In this study, the effects of an addition of B₄C on the microstructures of ZrB₂-SiC ceramics were determined using transmission electron microscopy.

Figure 1 shows the microstructure of the ZrB₂-20 vol.% SiC with or without 5 vol.% of added B₄C densified by hot pressing at 1900°C for 2 h. In the backscattered electron mode (BSE) of SEM, microstructural features with higher average atomic weights appear in brighter contrast. In Fig. 1, the gray matrix is ZrB₂, while the black dispersed particles denote the SiC. Table 1 summarizes the characteristics of the hot-pressed ZrB₂-based ceramics used in this study.

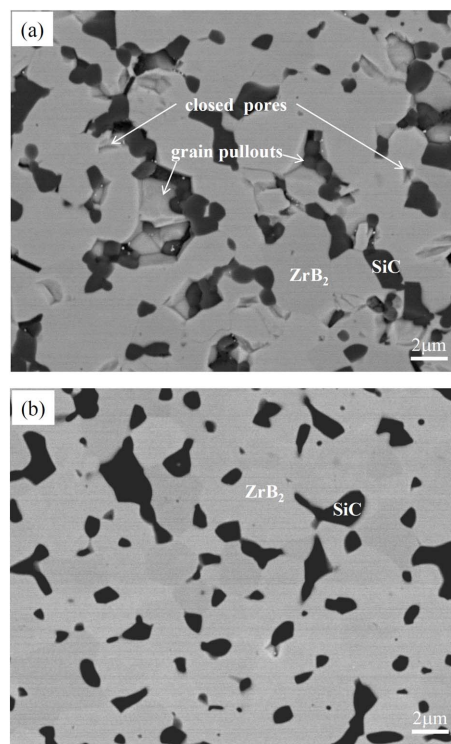


Fig. 1. SEM micrographs of ZrB₂-20 vol.% SiC with (a) no additives and (b) an extra 5 vol.% of B₄C hot-pressed at 1900°C for 2 h at 30 MPa.

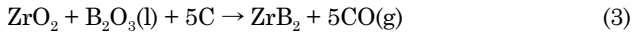
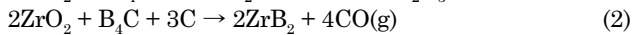
Both samples exhibited almost full densities. While a considerable number of pores can be observed in Fig. 1(a), much of the apparent porosity is thought to be due to grain pullout during polishing, which resulted not from densification behavior but from the machining process.¹⁴⁾ Furthermore, the remaining B₄C is not discernible in this SEM micrograph.

Figure 2 shows typical bright-field TEM images of ZrB₂-20 vol.% SiC with an extra 5 vol.% of B₄C densified by hot pressing at 1900°C for 2 h. The microstructure was composed of three constituent phases, which were equiaxed ZrB₂ or SiC and elongated B₄C. Pores were rarely observed, except for those in the vicinity of B₄C particles. In the bright-field TEM images, diffraction contrast should be observed for the crystalline phases in the microstructure due to the periodicity of the atomic structure. In contrast, amorphous phases show lower contrast in bright-field images due to the absence of crystallite. In Fig. 2(b), an amorphous layer can be observed, especially near the B₄C particles.

Table 1. Composition, Sintering Condition, Density, and Vickers Hardness of Hot-pressed ZrB₂-based Ceramics Used in This Study

Compositions	Hot-pressing condition	Apparent density (g/cm ³)	Theoretical density (g/cm ³)	Relative density (% T.D.)	Vickers hardness (GPa)	
					R.T.	900°C
ZrB ₂ -20 vol.% SiC	1900°C/2 h/30 MPa /flowing Ar	5.47	5.51	99.3	19.7	8.5
ZrB ₂ -20 vol.% SiC with extra 5 vol.% B ₄ C		5.37	5.37	100	21.3	9.0

B_4C is one of the most frequently used sintering additives for the densification of ZrB_2 -based ceramics.¹³⁻¹⁵ The full densification of ZrB_2 -based ceramics is known to be scarcely achievable due to the covalent nature of the bonding or the low diffusion rates.^{3,4} Oxide layers on the surfaces of the starting powder have been shown to inhibit densification and to promote grain growth in boride ceramics systems.¹⁶ The addition of B_4C to a ZrB_2 -based system is known to enhance the densification by removing oxides during the heat treatment. These reactions proceed as follows¹⁴:



In our previous work,⁶ an addition of B_4C to a ZrB_2 -SiC system enhanced the densification, especially at a relatively low temperature of 1700°C. On the other hand, the differences in the relative densities of ZrB_2 -SiC ceramics hot-pressed at 1900°C for 2 h were not obvious, as shown in

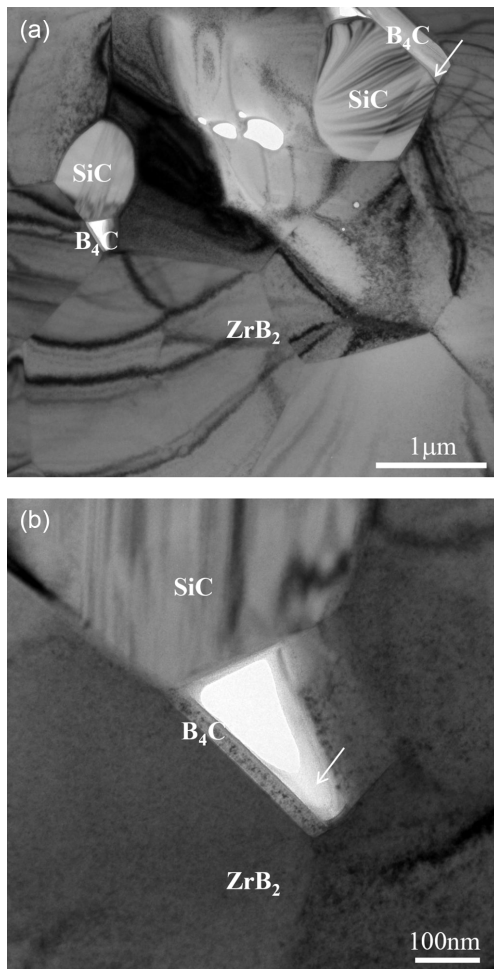


Fig. 2. (a) Bright-field transmission electron microscopy image and (b) a higher-magnification image of hot-pressed ZrB_2 -20 vol.% SiC with an extra 5 vol.% of B_4C . (The arrow indicates the amorphous region near the B_4C grain.)

Table 1. In addition, ZrB_2 -SiC ceramics with a B_4C addition hot-pressed at 1900°C for 2 h exhibited moderately higher density and hardness values.

In Fig. 3, the high-resolution TEM microstructures are shown to be present near the grain boundaries in hot-pressed ZrB_2 -SiC with an addition of B_4C . In the HR images, crystalline phases, which are well aligned with the electron beam, show the lattice fringes. Each constituent phase, in this case ZrB_2 , SiC, or B_4C , exhibited lattices fringes, of which the width was in agreement with the d-spacing of the relevant lattice plane, as denoted in Fig. 3. In particular, lattice fringes from SiC represented the ordered sequence of 6H-SiC, which corresponds to α -SiC. In addition, amorphous layers were observed especially near the B_4C grains in both images.

In order to identify the chemical composition of the amorphous layer, an energy dispersive X-ray spectroscopy analysis was carried out. Fig. 4 shows an STEM (scanning transmission electron microscopy) image and the EDS spectra obtained at the grain boundary as well as the constitu-

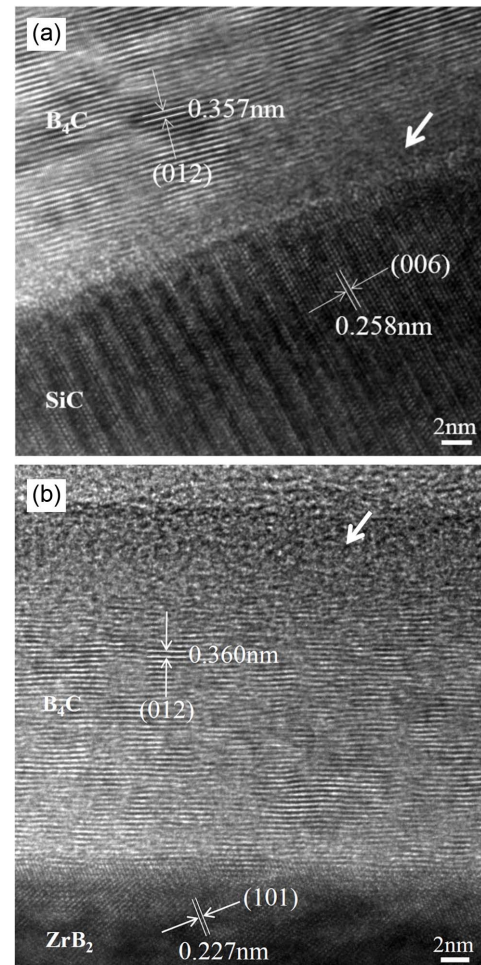


Fig. 3. High-resolution transmission electron microscopy images of grain boundaries in hot-pressed ZrB_2 -20 vol.% SiC with an extra 5 vol.% of B_4C : (a) B_4C -SiC interface and (b) B_4C - ZrB_2 interfaces. (Arrows indicate the amorphous region near the B_4C grains.)

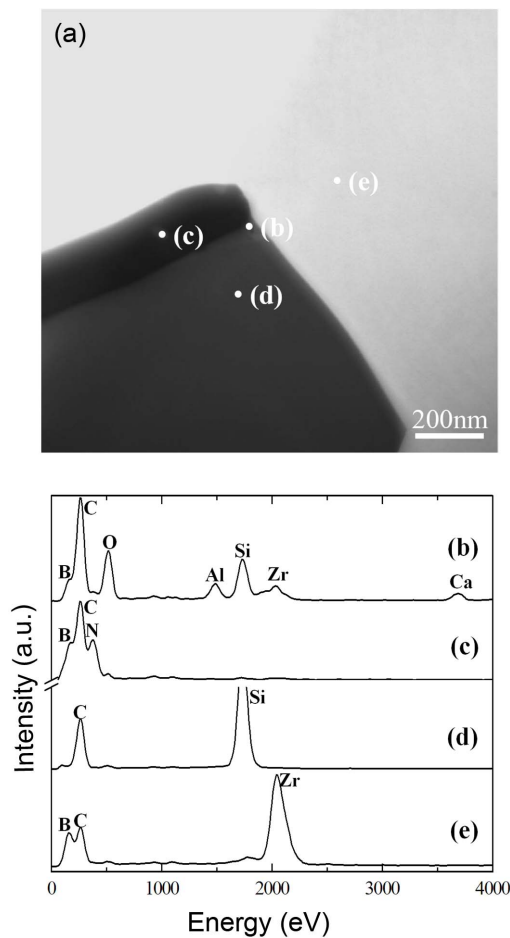


Fig. 4. (a) Scanning transmission electron microscopy image of hot-pressed ZrB₂-20 vol.% SiC with an extra 5 vol.% of B₄C and energy dispersive X-ray spectroscopy spectra from (b) the interface, (c) B₄C, (d) SiC, and (e) ZrB₂ as marked in (a).

ent phases. In the STEM images from a HAADF (high-angle annular dark field) detector, the elemental contrast could be seen similarly to the BSE mode in SEM.¹⁷ In the microstructure, the elongated black particle corresponds to the B₄C. In the EDS spectra, it was noted that all spectra had carbon peaks due to carbon depositions during the STEM analysis. Fig. 4(b) presents the EDS spectra from the amorphous layer of a triple junction between ZrB₂, SiC, and B₄C. These spectra contained peaks from constituent elements such as Zr, Si, B, and C and from contaminants such as Ca, Al and O. It has been reported that oxide impurities are observable at the grain boundary and inside the grains of hot-pressed ZrB₂, while oxygen cannot be detected either inside the ZrB₂ grains or at the grain boundaries of the specimen densified by spark plasma sintering.¹⁰ In this study, intergranular amorphous layers with oxide contaminants were observed exclusively in the vicinity of the B₄C grains. From these observations, B₄C was thought to absorb and react with oxide impurities as in the above-mentioned reactions of eq. (1) ~ (3) to enhance the densification of ZrB₂-

based ceramics. On the other hand, oxide contaminants in ZrB₂-based ceramics can degrade the mechanical properties at an elevated temperature. A second-phase addition reduces the densification temperature and the grain size,³⁾ but the resulting ceramics may undergo degradation in their mechanical properties at an elevated temperature.¹⁾ For example,¹⁾ substantial differences were observed in the flexural behavior at temperatures which exceeded 1200°C among the materials with or without additives such as carbides, nitrides, and/or oxides. This type of strength degradation is claimed to be due to the softening of the glassy intergranular phases, which contribute to subcritical crack growth. Typically, oxide impurities form a grain boundary phase or become localized at triple junctions, which reduces the strength.¹⁸⁾

4. Summary

The microstructure and the grain boundary phase of ZrB₂-SiC ceramics with added B₄C were examined by means of TEM. For hot-pressed ZrB₂-SiC ceramics with B₄C, the microstructure consisted of equiaxed ZrB₂ or SiC grains and a few elongated B₄C grains. Intergranular amorphous layers were observed especially near the B₄C grains. These amorphous layers exhibited EDS elemental peaks from constituent elements such as Zr, Si, B, and C and from contaminants such as Ca, Al and O.

Acknowledgments

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