

Collective effect of hydrogen in argon and Mg as ambience for the heat treatment on MgB₂

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Abstract

Magnesium diboride superconductor is still of considerable interest because of its appealing characteristics towards application mainly at around 20 K. Unlike Nb-based superconductors, MgB₂ can be operated by cryogen-free cooler which provides a cost effective alternative at low field of around 2-5 T. To explore this operating field region considerable efforts are necessary to marginally improve the superconducting properties of MgB₂. Under this situation, even the heat treatment environment during the synthesis is considered as an important factor. The addition of H₂ gas in small amount with Ar as a mixed gas during annealing has an adverse effect on the superconducting properties of MgB₂. It is although interesting to find that the presence of Mg vapor along with hydrogen during heat treatment results in the appreciable improvement in the flux pinning and the overall response of the critical current density for the ex-situ MgB₂ samples.

Keywords: Electric resistance, Magnetic fields, MgB₂ superconducting material, Critical current density

1. INTRODUCTION

Magnesium diboride superconductor has been observed to be highly sensitive to sample preparation history and composition. Since the discovery of superconductivity in this compound [1], MgB₂ has been studied by various synthesis techniques through different processes with the different levels of superconducting properties [2-6]. MgB₂, being a binary intermetallic compound, was anticipated to be an easily synthesizable compound. But the vast difference in melting point of starting materials, B and Mg make this synthesis process somewhat difficult. It showed widely different properties depending upon the synthesis process and the precursor used. A convenient example was presented in ref.[7] wherein the resistivity of various samples prepared by the different synthesis techniques is compared. It was surprising to see that the samples show 39 K of superconducting transition (T_c) even with the room temperature resistivities of little over 100 mΩ-cm [8] and also with less than 10 μΩ-cm [9]. Different level of grain connectivity among the superconducting grains depending upon the extent of impurities, secondary phases, and stoichiometry is anticipated as the reason for the discrepancies in the superconducting parameters. Various research work in this field had made it quite clear that unlike high T_c cuprate superconductors there is a strong inter-granular current flow in MgB₂ superconductor [10-12]. Keeping away the defects and grain boundary effects, the improvement of the critical current density in pure MgB₂ is highly related with the enhancement in the phase formation and the sample density [13]. Thus, the main parameter behind the discrepancies in various MgB₂

samples is the high vapor pressure of Mg which results in the instability and further depletion of Mg content. In this work, the research to improve the superconducting property of MgB₂ merely by using a combination of hydrogen and Mg annealing ambience is carried out. MgB₂ powder was used as a precursor towards nullifying the effect of preparative conditions in the synthesis of in-situ MgB₂.

2. EXPERIMENTAL

To study the collective effect of H₂ and Mg ambience on ex-situ MgB₂ samples, a commercially available magnesium diboride powder with the particle size of 100 Mesh was used. The precursor powder was hand-milled for about two hours in the agate mortar under the environment of ultra high pure Ar in glove box. This hand-milled powder was then transferred in to mould of 10 mm x 10 mm to form a pellet under the applied pressure of 10 Ton/cm². This pellet so formed was placed in Fe tube for the annealing in a tube furnace at 900 °C for 3 hours. The temperature was ramped up at a rate of 15 °C/min and different gas was flowed depending upon samples at flow rate of 2 L/min. The ambience used during annealing was Ar, Ar + 4% H₂, Ar + Mg, and Ar + 4% H₂ + Mg and the samples were accordingly named as Ar, ArH, ArMg, and ArHMg, respectively. Ar + 4% H₂ is a mixed gas of ultra high pure and used as purchased. It was also designed to maintain an equilibrium ambience of Mg around samples in a way that there is no built-up of Mg pressure in Fe tube. Hence the Fe tube was provided with a small hole at one side. This would create an equilibrium pressure with the

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environment outside the Fe tube that would prevent the formation of excess Mg pressure (2.93 Torr at 650 °C [14]) and the forceful infiltration of Mg into the samples. Amount of Mg powder introduced was equal to the weight of the pellet (which in this case ~157 mg) and was placed at a sufficient distance from the pellet. The pellet samples were characterized by X-ray diffraction (XRD) technique to study the structural variations. It was performed using Rigaku, D/Max 2200 system with Cu K α radiation. Also, Field Emission Scanning electron microscopy (FE-SEM) (using HITACHI FE-SEM S-4700) was carried out to study the effect of annealing ambience on the microstructure of samples. The superconducting transition temperatures of samples were measured through the temperature dependent resistivity using a four probe method. The zero field cooled (ZFC) and field cooled (FC) measurements was also done to confirm and compare the superconducting properties of the samples. The measurement of the magnetization as a function of magnetic field was obtained to calculate the critical current density using the modified Bean's formula [15] as given below:

$$J_c = \frac{20 \Delta M}{a \left(1 - \frac{a}{3b}\right)} \quad (1)$$

Where, J_c is critical current density in A/cm², ΔM is the difference in the magnetization in emu/cm³ for increasing and decreasing field, 'a' and 'b' is the thickness and width of the sample. The reduced flux pinning curves were plotted to study the properties of pinning mechanism of the samples.

3. RESULTS AND DISCUSSIONS

3.1. X-ray diffraction studies

Since all the samples were prepared from the MgB₂ precursor powder, they have shown XRD pattern with all the characteristic peaks corresponding to MgB₂ crystal structure as shown in Fig. 1. Although there was a presence of excess Mg during annealing, it is noteworthy to mention that the peak for unreacted Mg was not detected in samples of ArMg and ArHMg. This signifies that Mg diffusing in these samples is used up for filling the Mg deficient sites,

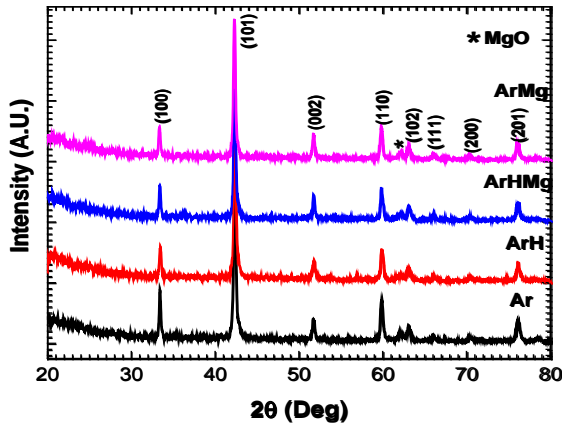


Fig. 1. X-ray diffraction patterns for samples of Ar, ArH, ArHMg, and ArMg annealed in aforementioned ambience.

rather than settling down on grain boundary or sample surface. There is a small peak for MgO, which cannot be avoided due to high affinity of oxygen towards Mg. There is neither any increase in the MgO peak in the samples annealed in Mg environment. This again shows that there might be a very scarce amount of bare Mg, if not much, settled on the surface otherwise it would have easily got oxidized in the due course giving rise to the MgO concentration.

3.2. Morphological Studies

Fig. 2(a) and (b) shows the FE-SEM images of all the samples. From the microstructure, it can be clearly understood that samples annealed in H₂ environment have a surface covered with the nm-sized tiny granules. These sites are even more prominent for the samples annealed in combination of Mg with Ar+4%H₂. From XRD, it is clear that there are no peaks observed for unreacted Mg. Also, such kind of sites is observed only in case of samples annealed in H₂ environment. Thus it can be argued that H₂ being very easily diffusible in the bulk, has some kind of reducing effect on the samples. Since it acts as a strong reducing environment, at high temperatures (900 °C), it may create some fresh boron or Mg from its oxides or even MgB₂ [14] and results in a new nucleation sites at the surfaces. Since MgO is hard to be reduced and elemental Mg has high vapor pressure, there may be Mg depletion from ArH sample. While, in the case of ArHMg sample, the presence of Mg in annealing ambience may support the increase of such kind of sites over the sample surface (As may be seen in Fig. 2(b)). This is in agreement with the

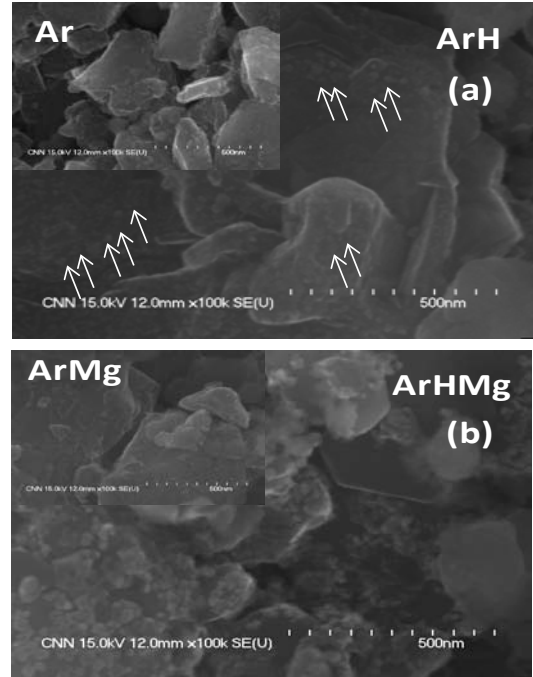


Fig. 2. FE-SEM image of samples (a) annealed in Ar (inset) and Ar+4%H₂ (b) annealed in presence of Mg along with Ar (inset) and Ar+4%H₂. Arrow indicates the presence of nucleation sites observed in the samples annealed in presence of H₂ ambience. (Scale bar is for 500 nm).

results observed for malic acid doped MgB₂ wherein the weight fraction of MgB₂ was observed to decrease with increase of H₂ content in argon [16].

3.3. Resistivity behavior

Fig. 3 shows the reduced as well as the original (in inset) resistivity plots for all the samples. From inset it can be easily understood that the overall resistivity decreases from Ar to ArH sample, while it is highly reduced for the samples annealed with Mg ambience. The values for effective current carrying area, A_f calculated using Rowell analysis [17-18], onset ($T_{c, on}$), offset ($T_{c, off}$) and width (ΔT) of the superconducting transition is given in table 1. The decrease in the resistivity of ArH samples supports our argument of H₂ with some reducing effect, which more or less cleans the grain as well as the grain boundary from the oxides rendering the improved overall connectivity and hence the reduced resistivity. This explains the exact scaling of resistivity plots for ArH samples on Ar in the reduced resistivity plot in Fig. 3.

Meanwhile, in case of samples with Mg ambience, the resistivity shows much improved power law dependence in temperature with a reduced transition width of 1 K. This can indicate the increased dependence on electron-phonon scattering due to the decrease in defect and improved intra-grain as well as inter-grain connectivity [18]. Further, it can be observed that the actual superconducting current carrying fraction is much higher for ArMg sample as compared to ArHMg sample. This difference is not clearly understood now but can be attributed to the slightly increased grain boundary density of ArHMg due to the formation of new grains at the nucleation sites created by H₂ gas as seen in SEM image (Fig. 2(b)).

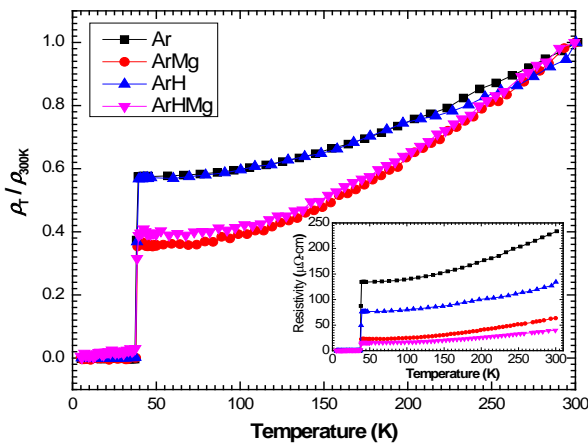


Fig. 3. Reduced and original (inset) temperature dependent resistivity behavior for all the MgB₂ samples.

TABLE I

MEASURED VALUES FOR SUPERCONDUCTING AREA FRACTION AND SUPERCONDUCTING TRANSITION PARAMETERS.

Sample	A_f	$T_{c, on}$	$T_{c, off}$	ΔT
Ar	0.07	39.2	37.2	1.98
ArH	0.13	39.5	37.5	2
ArHMg	0.18	38.8	37.8	1
ArMg	0.30	38.5	37.5	1

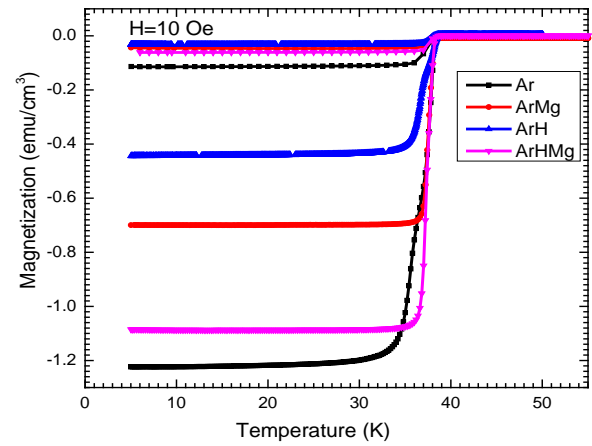


Fig. 4. FC-ZFC plots for all the MgB₂ samples.

3.4. Zero Field Cooled (ZFC) and Field Cooled (FC) study

The FC-ZFC plots for all the samples are in a good agreement with the resistivity plots in terms of the superconducting transition temperature. However it is interesting to note that the sample ArH, which shows a lower resistivity values relative to sample Ar, shows very low extent of diamagnetism in Fig. 4. This may indicate that the sample has the lowest density amongst all the samples. This is in support with our argument that H₂ has some kind of reducing effect and deplete the level of Mg, which results in the decreased density of the samples. The increase in the extent of diamagnetism in sample ArHMg further consolidates this fact.

3.5. Magnetization and Critical Current Density

The magnetization with respect to field (M-H) measurements at 5 K is shown in Fig. 5. The plots can be conveniently divided in samples prepared with and without Mg ambience. It can be seen that the samples annealed in Mg ambience shows wider magnetization loop than those annealed without Mg. Although both the samples show similar M-H response, sample ArHMg seems to be slightly superior than ArMg. Thus even though sample ArH shows very poor performance, the combined effect of Mg and H₂ highly improve the superconducting properties. This effect is reflected straight away in the critical current density curves calculated using equation 1 as shown in Fig. 6.

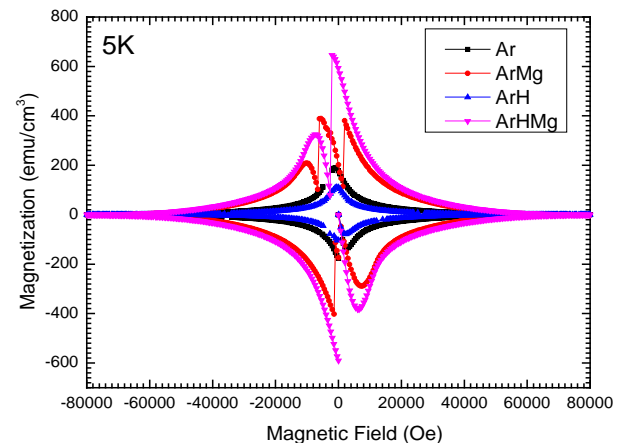


Fig. 5. Magnetization Vs field behavior for all the samples at 5K.

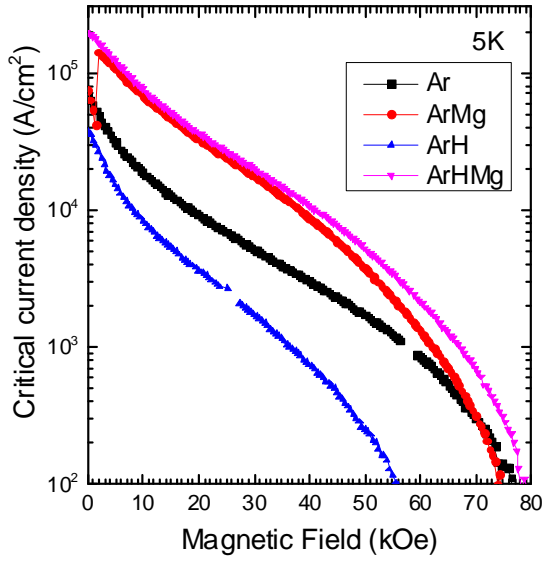


Fig. 6. Critical current density plots at 5K.

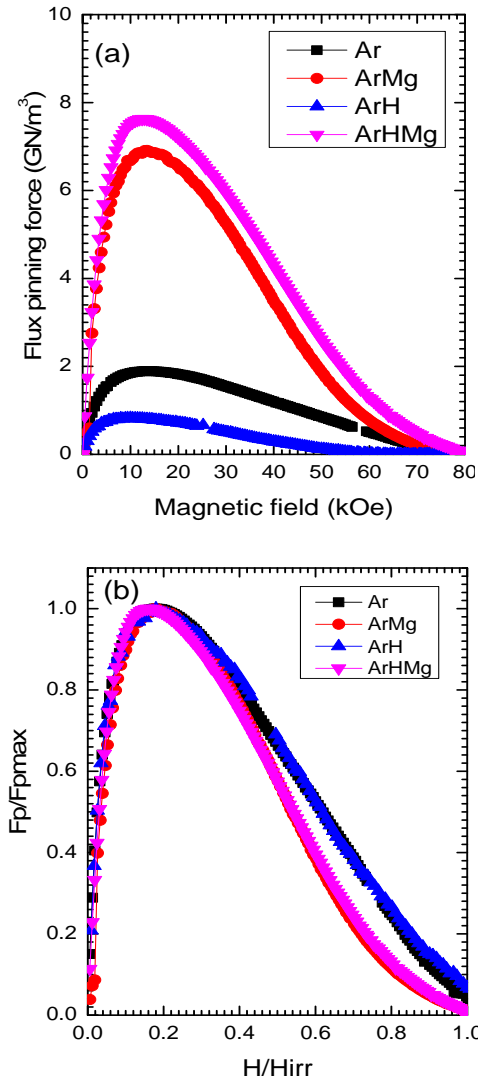


Fig. 7. (a) Flux pinning force plots and (b) Reduced flux pinning force plots for all the samples.

These plots clearly show that the samples treated in Mg ambiance show the higher performance in overall critical current density values. Here, it is noticeable that the low-field values for ArMg and ArHMg has improved and approach up to 2×10^5 A/cm² as compared to Ar and ArH which is less than 10^5 A/cm². The irreversibility field value with the 100 A/cm² criteria comes out to be about 80 kOe for all the samples except for ArH which is around 55 kOe. Thus it can be understood that although H₂ environment alone create a deteriorating effect, but when clubbed with the presence of Mg the superconducting property of the sample can be enhanced significantly. To understand the pinning mechanism in all these samples, the flux pinning curves were plotted and are as shown in Fig. 7(a) and (b). Fig. 7(a) shows the flux pinning force plots against applied magnetic field. It depicts a clear increase in the flux pinning force in ArMg and ArHMg as compared to Ar and ArH samples. The reduced flux pinning plot for all the samples is shown in Fig. 7(b). It shows maxima around 0.2 for all the samples which implies a prominent grain boundary pinning mechanism which is common for the pure MgB₂ bulk samples [19]. The sample treated in the presence of Mg environment shows an improved flux pinning force. This may be on account of the increased grain boundary density, possibly due to the new grain arrangements caused by nucleation and growth observed especially in combination of H₂ and Mg in annealing ambiance.

4. CONCLUSION

It can be concluded that the Ar+4%H₂ gas alone act as a reducing environment, which diffuse in the bulk and etch it out. This kind of effect can cause the cleaning of the insulating oxide phases in the sample, most probably at the grain boundary, but at the same time deplete the Mg from the bulk. So, it happens to decrease the overall resistivity of the sample and adversely affect the critical current density property of MgB₂ samples. However, the inclusion of Mg in this reducing environment may replenish the Mg deficiencies in the reduced samples and hence improve the overall superconducting properties of MgB₂.

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