J. KIEEME

Vol. 27, No. 5, pp. 297-302 May 2014 DOI: http://dx.doi.org/10.4313/JKEM.2014.27.5.297 ISSN 1226-7945 (Print), 2288-3258 (Online)

Growth of ZnSnO₃ Thin Films on c-Al₂O₃ (0001) Substrate by Pulsed Laser Deposition

Trung Tran Manh¹, Jae-Ryong Lim¹, and Soon-Gil Yoon^{1,a}

¹ Department of Materials Science and Engineering, Chungnam National University, Daejeon 305-764, Korea

(Received April 10, 2014; Revised April 21, 2014; Accepted April 23, 2014)

Abstract: La_{0.5}Sr_{0.5}CoO₃ (LSCO) electrode thin films with a resistivity of $\sim 1,600~\mu\Omega$ cm were grown on c-Al₂O₃ (0001) substrate. ZnSnO₃ (ZTO) thin films with different thicknesses were directly grown on LSCO/c-Al₂O₃ (0001) substrates at a substrate temperature that ranged from 550 to 750°C using Pulsed Laser Deposition (PLD). The secondary phase Zn₂SnO₄ occurred during the growth of ZTO films and it became more significant with further increasing substrate temperature. Polarization-electric-field (P-E) hysteresis characteristics, with a remnant polarization and coercive field of 0.05 μ C/cm² and 48 kV/cm, respectively, were obtained in the ZTO film grown at 700°C in 200 mTorr.

Keywords: Piezoelectric, ZnSnO3 thin film, Ferroelectric thin films

1. INTRODUCTION

NCS (non-centrosymmetric) oxides [1] have attracted considerable attention due to their unique symmetry-dependent and spontaneous polarization properties, which are technologically significant and the basis of numerous applications ferroelectricity, piezo-electricity, nonlinear and optics [2]. ZnO is well known as environmentally friendly and piezoelectric material [3]. Besides ZnO, there are still many lead-free piezoelectric materials in the NCS group that have been discovered recently, especially in terms of perovskite structure such as: BaTiO₃ [4],(K,Na)NbO₃ $(1-x)Ba(Zr_{0.2}Ti_{0.8})O_3-x(Ba_{0.7}Ca_{0.3})TiO_3$ [8], and etc. In spite of the success in lead-based piezoelectric

method. In the present study, 500-nm thick ZTO films were deposited on $c-Al_2O_3$ (0001) substrates

Copyright ©2014 KIEEME. All rights reserved. This is an Open-Access article distributed under the terms of the Creative Commons Attribution Non-Commercial License (http://creativecommons.org/licenses/by-nc/3.0) which permits unrestricted non-commercial use, distribution, and reproduction in any medium, provided the original work is properly cited.

materials (i.e. lead zirconate titanate, $0.65PbMg_{1/3}Nb_{2/3}O_3-0.35PbTiO_3$ [9,10]) due to their high polarization and piezoelectric performance as transducers [11], a consistent effort has been taken

297

to replace Pb-based materials with some as yet undiscovered lead-free materials [12] that would be more environmentally friendly and enable new piezoelectric applications for mechanical energy

harvesting. Recently, Inaguma reported a lead-free LiNbO₃ - type ZnSnO₃ using a high-pressure

(~7 GPa) synthesized environment [2] and Jyh Ming Wu successfully fabricated ZnSnO₃ microbelts given an output voltage and current of 100 mV

Until now, there are only few reports on the

and 30 nA, respectively [13].

preparation and electrical properties of ZnSnO₃ thin films [14–16]. To the best of our knowledge, this is the first report on the growth of ZTO thin films on c-Al₂O₃ (0001) substrates using the PLD

a. Corresponding author; sgyoon@cnu.ac.kr

at various substrate temperatures by pulsed laser deposition, and the electrical properties of the heterostructure $ZTO/LSCO/c-Al_2O_3$ (0001) films grown at various temperatures are discussed.

2. EXPERIMENTS

The ZnSnO₃/LSCO/c-Al₂O₃ thin films were grown at a substrate temperature within the range of 550 to 750°C on c-Al₂O₃ (0001) substrates by pulsed laser deposition using a pulsed KrF excimer laser (248nm, Lamda Physik COMPex-Pro 201). A ZnSnO₃ ceramic target with a diameter of 2.54 cm was used for deposition of the ZTO thin films. The ZnSnO₃ target was fabricated by processing stoichiometric mixture of ZnO (Aldrich 99.999%) and SnO₂ (Aldrich 99.995%) powder at the desired 1/1 ratio. Powder was ball-milled in ethanol for 4 h to insure homogeneous distribution. Pellets with a diameter of 1 inch were unaxially pressed without binder. The samples were calcined at 850°C and sintered at 1,300°C in ambient air for 12 h. After the base pressure reached 5×10^{-6} Torr, the substrate temperature was set to various temperatures with an oxygen partial pressure of 200 mTorr during the deposition of a ZnSnO3 thin film. The c- Al_2O_3 (0001) substrates were attached to a heated plate with silver paste in order to maintain a uniform temperature. The details of the conditions for the deposition of the ZTO/LSCO/ c-Al₂O₃ substrates are summarized in Table 1. After deposition, high-purity oxvgen introduced into the growth chamber up to 5 Torr and then the films were cooled naturally to room temperature.

Film thicknesses were measured using cross-sectional scanning electron microscopy (SEM) images. The phase of the heterostructure was investigated by $\theta{\sim}2$ θ , using a high-resolution X-ray diffraction (HRXRD, Rigaku RINT2000). The resistivity of the LSCO thin films was measured by an electrometer (CMT-SR 1000) using a four-point probe.

Table 1. Deposition conditions of ZTO, LSCO films grown on c-Al₂O₃ (0001) substrates.

Deposition parameters	LSCO	ZTO
Target	La _{0.5} Sr _{0.5} CoO ₃	ZnSnO ₃
Deposition temperature ($^{\circ}$ C)	450~650	$550 \sim 750$
Film thickness (nm)	100, 200	500
Deposition pressure (Torr)	0.3	0.2
Energy density (J/cm ²)	1.5	1.5

Conventional photolithography was used to pattern the Pt - squares top electrodes of 100 μ m in length and sputter-deposition implemented in a commercial dc magnetron sputtering system. The polarization vs. electric field (P–E) curve of the films was measured using an RT66A ferroelectric tester (Radiant Technology) operated in the virtual ground mode.

3. RESULTS AND DISCUSSION

The LSCO films were prepared on $c\text{-}Al_2O_3$ (0001) substrates to determine the influences of substrate temperature. The studied ranges were 450 \sim 650°C under 300 mTorr oxygen pressure.

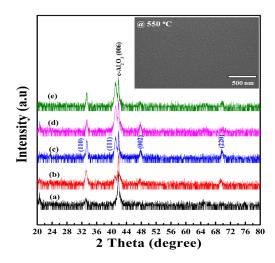


Fig. 1. X-ray diffraction patterns of LSCO/c-Al₂O₃.

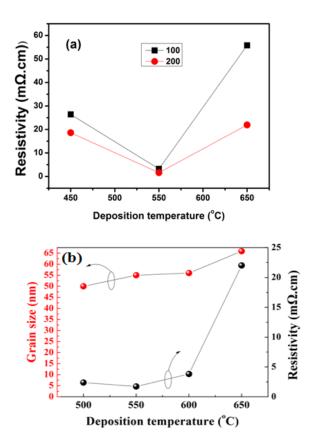


Fig. 2. (a) Surface resistivity of the LSCO thin films deposited in 300 mTorr oxygen at different substrate temperatures with various thicknesses and (b) grain size and surface resistivities of the LSCO thin films deposited in 300 mTorr oxygen at different substrate temperatures, measured at room temperature.

(0001) deposited at oxygen pressures of 300 mTorr and different temperatures (a) 450° C, (b) 500° C, (c) 550° C, (d) 600° C, and (e) 650° C, the inset shows the morphology of LSCO film grown at 550° C.

Fig. 1 shows $\theta \sim 2~\theta$ XRD patterns for a series of LSCO thin films deposited on c-Al₂O₃ (0001) substrates at different temperatures. The film grown at 500°C exhibits a small peak (111) from LSCO, and become higher with further increasing temperatures. The films present different room temperature electrical resistivity values depending on the processing conditions, and a low resistivity value is important for the use of LSCO as an electrode.

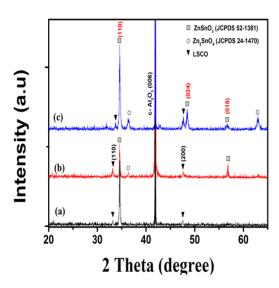
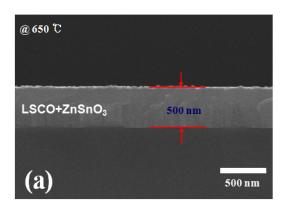


Fig. 3. X-ray diffraction patterns of ZTO/LSCO/c-Al $_2$ O $_3$ (0001) heterostructure deposited at oxygen pressures of 200 mTorr and different temperatures (a) 600°C, (b) 700°C, and (c) 750°C.

Fig. 2(a) shows the electrical property corresponding to the series of temperatures and thickness of film measured by the classical four-probe method at room temperature. As can be seen in this figure, the thicker (200 nm) films show lower resistivity than the thinner (100 nm) films with different deposition temperatures. A plot of the resistivity and grain size as a function of growth temperature of LSCO films as shown in Fig. 2(b). The resistivity value for LSCO films reach the lowest point at 550°C. This result coincides with the preparation of LSCO bottom electrode in Juan Jiang et. al. [9]. The grain size of particles is calculated from Fig. 1 by using Debye-Scherrer's equation. The results are in range, from 50~70 nm and grain size becomes bigger with further increasing of temperature. If grain size increases, the electron scattering at grain boundary seems to decreasing which will result in an increased resistivity, following by relationship between grain size and resistivity in Fig. 2(b) within the range of 550 to 650°C. Thus, high temperature is more important to obtain good conductivity.



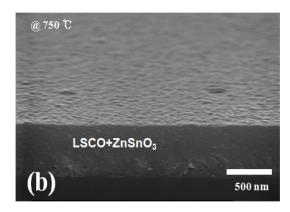


Fig. 4. Cross - sectional images at (a) $650\,^{\circ}$ C and (b) $750\,^{\circ}$ C as of the heterostructure ZTO/LSCO/c-Al₂O₃ deposited in 200 mTorr.

Fig. 3 shows the $\theta \sim 2$ θ XRD patterns of a ZTO/LSCO/c-Al₂O₃ heterostructure at substrate temperatures. As can be seen from this figure, there were only ZnSnO₃ peak observed at the lower temperature (600°C) . The second phase Zn₂SnO₄ were founded in the ZTO films grown at the higher temperature (700°C), and become stronger with further increasing substrate temperature (750°C). The co-existence of the secondary phase Zn₂SnO₄ were also observed at high temperatures in some reports before [2,13]. As can be seen in Fig. 4(a), the thickness of both LSCO and ZTO film was about 500 nm grown at 650°C in 200 mTorr and was thicker when grown at 750°C in same pressure and deposition time.

Fig. 5 shows the surface resistivity of the LSCO bottom electrodes after growing ZTO layers (A part

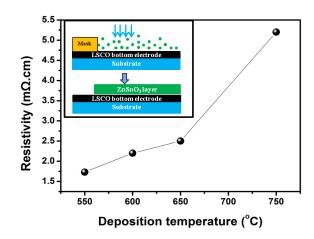
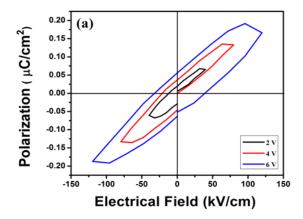


Fig. 5. Surface resistivities of the LSCO electrode after growing ZTO layer in 200 mTorr oxygen at different substrate temperatures, measured at room temperature.



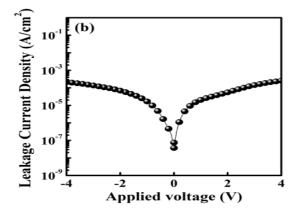


Fig. 6. (a) Saturated P-E hysteresis loop of the 500 nm thick ZTO grown at 700°C and (b) leakage current density versus electrical field.

of LSCO electrode films was covered by a mask). From this figure, it can be seen that, the resistivity is unchanged at 550°C in comparison with its value in Fig. 1(b) and increased with higher substrate temperature and reach the highest value at 750°C, it means the conductivity of LSCO bottom electrode become worse with increasing deposition temperature. This may be due to the appearance of the secondary Zn₂SnO₄ peaks in Fig. 3. However, the resistivity change is not significantly high and they still show low values, conductivities of LSCO mean good electrodes.

The ferroelectric nature of the ZTO thin films was verified by the polarization-electrical field (P-E) hysteresis behavior. Fig. 6(a) shows the P-E temperature for the loop taken at room heterostructure films. The film exhibits well saturated hysteresis with remnant and saturation polarization of 0.05 μ C/cm² and 0.15 μ C/cm², respectively, and a coercive field of 48 kV/cm. The little asymmetric shape of the P-E hysteresis loop and the leakage current characteristics in Fig. 6(b) is attributed to the existence of an internal biasfield which has been built up in the ZTO film, due to an asymmetric space-charge distribution among the top and the bottom electrodes.

4. CONCLUSIONS

In summary, the 500-nm thick ZTO films were directly grown a on LSCO/c-Al₂O₃ (0001) substrates at various ranges of temperature under 200 mTorr by PLD technique. The LSCO thin film grown at 550°C in 300 mTorr showed the highest conductivity value and its resistivity increase with higher substrate temperature when depositing ZTO films. There were not significant changes in the resistivity of the LSCO bottom electrode during ZTO deposition in the range of 550~750°C. The hysteresis loop for the ZTO film grown at 700°C demonstrated the ferroelectric behavior of the film with a remnant polarization of 0.05 μ C/cm² and a

coercive field of 48 kV/cm. The present study exhibited growth of ZTO films on an LSCO/ c-Al₂O₃ substrate with a ferroelectric behavior.

ACKNOWLEDGMENT

This work was supported by the National Research Foundation of Korea (NRF) grant funded by the Korea government (MEST) (No. NRF-2013R1A4A1069528).

REFERENCES

- [1] P. S. Halasyamani and K. R. Poeppelmeier, *Chem. Mater.*, **10**, 2753 (1998).
- [2] Y. Inaguma, M. Yoshida, and T. Katsumata, J. Am. Chem. Soc., 130, 6704 (2008).
- [3] Z. L. Wang and J. H. Song, Science, 32, 242 (2006).
- [4] K. J. Choi, M. Biegalski, Y. L. Li, A. Sharan, J. Schubert, R. Uecker, P. Reiche, Y. B. Chen, X. Q. Pan, V. Gopalan, L. Q. Chen, D. G. Schlom, and C. B. Eom, *Science*, 306, 1005 (2004).
- [5] Y. Wakasaa, I. Kannoa, R. Yokokawaa, H. Koteraa, K. Shibatab, and T. Mishima, Sensor. Actuat., Phys. A, 171, 223 (2011).
- [6] I. Kanno, T. Mino, S. Kuwajima, T. Suzuki, H. Kotera, and K. Wasa, IEEE Trans. Ultrason. Ferroelectr. Freq. Control, 54, 2562 (2007).
- [7] J. Y. Son, G. Lee, M. H. Jo, H. Kim, H. M. Jang, and Y. H. Shin, *J. Am. Chem. Soc.*, **131**, 8386 (2009).
- [8] G. Kang, K. Yao, and J. Wang, Journal of the American Ceramic Society, 95, 986 (2012).
- [9] J. Jiang, H. J. Jung, and S. G. Yoon, J. Alloy. Compd., 509, 6924 (2011).
- [10] J. Jiang and S. G. Yoon, J. Alloy. Compd., 509, 3065 (2011).
- [11] A. I. Kingon and S. Srinivasan, *Nat. Mater.*, **4**, 233 (2005).
- [12] R. J. Zeches, M. D. Rossell, J. X. Zhang, A. J. Hatt, Q. He, C. H. Yang, A. Kumar, C. H. Wang, A. Melville, C. Adamo, G. Sheng, Y. H. Chu, J. F. Ihlefeld, R. Erni, C. Ederer, V. Gopalan, L. Q. Chen, D. G. Schlom, N. A. Spaldin, L. W. Martin, and R. Ramesh, Science, 326, 977 (2009).
- [13] J. M. Wu, C. Xu, Y. Zhang, and Z. L. Wang, ACS

Nano, 6, 4335 (2012).

- [14] J. H. Jung, M. Lee, J. I. Hong, Y. Ding, C. Y. Chen, L. J. Chou, and Z. L. Wang, ACS Nano, 5, 10041 (2011).
- [15] H. Q. Chiang, J. F. Wager, R. L. Hoffman, J. Jeong,
- and D. A. Keszler, *Appl. Phys. Lett.*, **86**, 013503 (2004).
- [16] I. G. Pathan, D. N. Suryawanshi, A. R. Bari, D. S. Rane, and L. A. Patil, Advanced Nanomaterials and Nanotechnology, 143, 143 (2013).