Antioxidant and Cholinesterase Inhibitory Activities of the By-products of Three Pandalid Shrimps

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Abstract

The antioxidant and cholinesterase inhibitory activities of the acetone and dichloromethane (CH_2Cl_2) extracts of the by-products (heads, shells, and tails) of *Pandalus borealis*, *Pandalus hypsinotus*, and *Pandalopsis japonica* belonging to the family Pandalidae were investigated and their bioactivities were compared. The antioxidant and cholinesterase inhibitory activities of the organic solvent extracts of three shrimp by-products were evaluated using 2,2-diphenyl-1-picrylhydrazyl (DPPH) and 2,2'-Azino-bis[3-ethylbenzothiazoline-6-sulfonic acid] (ABTS⁺) radical scavenging activities, reducing power and xanthine oxidase (XO) inhibitory activity assays and Ellman's colorimetric method. The extracts of *P. hypsinotus* exhibited the highest antioxidant and cholinesterase inhibitory activities. The acetone extracts showed more potent activities toward antioxidant and cholinesterase inhibition compared with the CH_2Cl_2 extracts. Furthermore, the total carotenoid contents of the acetone extracts were higher than those of the CH_2Cl_2 extracts. Thus, the carotenoid contents may affect antioxidant and cholinesterase inhibition. Our results suggest that the shrimp by-products could act as a nutraceutical agent to prevent oxidative stress and Alzheimer's disease.

Key words: Shrimp by-products, Pandalus borealis, Pandalus hypsinotus, Pandalopsis japonica, Antioxidant, Cholinesterase

Introduction

Alzheimer's disease (AD), a progressive neurodegenerative disease, is the most common type of senior dementia, characterized pathologically by the presence of senile plaques, neurofibrillary tangles, and extensive neuronal loss (Giannakopoulos et al., 1996). The cause of AD has not yet been clarified, but one of the most convincing theories is the cholinergic hypothesis of deficiency in the levels of neurotransmitters including acetylcholine (ACh) and butyrylcholine (BCh) (Schneider, 2001). Furthermore, excessive free radicals produced by oxidative stress are associated with the pathological changes observed in AD (Praticò and Delanty, 2000). Accordingly, the activation of cholinergic functions by inhibition of cholinesterase which hydrolyzes the cholinergic neuromediators, and reduction of oxidative stress has been used as a key approach in the treatment of AD (Schneider, 2001). Shrimp of the family Pandalidae including *Pandalus borealis, Pandalus hypsinotus,* and *Pandalopsis japonica* are distributed widely in the deep sea at depths of 40-1,300 m in the waters around the eastern coast of Korea (Bauer, 2004). They are among the most popular shrimp species in Korea and are rich in nutrients such as protein, calcium, and vitamins. However, the inedible parts of the shrimp, the head, shell, and tail portions, account for ~50% of the catch, and the high volume of these by-products is causing environmental pollution. The shrimp by-products comprise various bioactive materials such as carotenoprotein, pigments, chitin, and chitosan (Chakrabarti, 2002; Babu et al., 2008; Younes et al., 2014), and have a potentially high economic value as a useful industrial raw material. Thus, the utilization of shrimp by-products due to their valuable constituents has attracted much interest among

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Fig. 1. Photographs of the family Pandlidae.

Pandalus hypsinotus

Pandalopsis japonica

researchers.

Several carotenoids, including β -carotene, α -carotene, lycopene, and lutein, derived from plants or animals, are effective in preventing cancer, cardiovascular disease, osteoporosis, and hypertension (Rao and Balachandran, 2003). However, there are few studies to date on the potential of shrimp by-products of the family Pandalidae and their carotenoid contents (Rao and Rao, 2007) in treating AD.

The aim of this study was to determine the effects on the antioxidant and cholinesterase (ChEs) activities of the acetone and CH₂Cl₂ extracts of the by-products of three shrimp species, and to investigate the correlation between the carotenoid contents of the extracts and their bioactivities.

Materials and methods

Materials

P. borealis, P. hypsinotus, and P. japonica were purchased from the market of Donghae-si, Gangwondo, Korea (Fig. 1). 2,2-diphenyl-1-picrylhydrazyl (DPPH), 2,2'-azino-bis[3ethylbenzothiazoline-6-sulfonic acid] (ABTS⁺), dimethyl sulfoxide (DMSO), potassium ferricyanide (K₃Fe(CN)₆), trichloroacetic acid (TCA), acetylcholinesterase (AChE), butyrylcholinesterase (BChE), acetylthiocholine iodide (ACh), butylthiocholine iodide (BCh), eserine, L-ascorbic acid, trolox, 5,5'-dithiobis [2-nitrobenzoic acid] (DTNB), ethylenediaminetretracetic acid (EDTA), CH₂Cl₂, n-hexane, and 3,5'-di-tert-butylhydroxytoluene (BHT) were purchased from Sigma-Aldrich Chemical Co. (St. Louis, MO, USA) and iron (III) chloride (FeCl₂) was obtained from Junsei Chemical Co. (Tokyo, Japan).

Preparation of samples

The shrimp by-products were lyophilized (1 kg) and stored at -20°C until. The lyophilized samples (20 g) were extracted three times with 50 mL acetone or CH₂Cl₂, which is used in the analysis of various lipophilic materials including carotenoids (Sachindra et al., 2005; Rayees and Manish 2014). The yields of the shrimp by-product extracts were: P. hypsinotus acetone extract (PHA, 2.72 g), P. hypsinotus CH₂Cl₂ extract (PHC, 2.49 g), P. borealis acetone extract (PBA, 2.97 g), P. borealis CH₂Cl₂ extract (PBC, 2.52 g), P. japonica acetone extract (PJA, 2.62 g), and P. japonica CH₂Cl₂ extract (PJC, 2.18 g).

DPPH radical scavenging activity

The DPPH radical-scavenging activity was measured using a modification of the method of Blois (1958). DPPH solution (6.5 mM, 160 µL) was added to 40 µL sample. The samples were incubated in the dark for 10 min at room temperature. The absorbance was measured by spectrophotometer (BIO-TEK US/MQX 200, USA) at 540 nm. L-ascorbic acid was used as a positive control and the DPPH radical scavenging activity of each sample was expressed as the 50% inhibition value (IC_{50}).

ABTS⁺ radical scavenging activity

The ABTS+ radical scavenging activity was determined using the method of Re et al. (1999). ABTS+ solution was diluted with water to an absorbance of 0.75 ± 0.03 at 734 nm. Then, 200 µL of ABTS⁺ solution was added to 10 µL samples of different concentrations. The samples were incubated in the dark for 10 min and measured the absorbance at 734 nm. Trolox was used as a positive control and the ABTS+ radical scavenging activity of each sample was expressed as an IC₅₀ value.

Reducing power

The reducing power of the shrimp-shell extracts was measured using a modification of the method of Jayaprakasha et al. (2001). Various concentrations of samples in 10% DMSO (50 μ L) were mixed with 50 μ L 0.2 M sodium phosphate buffer (pH 6.6) and 50 µL potassium ferricyanide (10 mg/mL). The mixtures were incubated at 50°C for 20 min. Then, 50 µL of TCA (100 mg/mL) were added and the mixture was centrifuged at 2,000 \times g for 10 min. After centrifugation, 100 µL supernatant was mixed with 20 µL iron (III) chloride (1 mg/ mL). The sample concentration resulting in 0.5 of absorbance (EC_{50}) was calculated from the graph of absorbance at 700 nm, and L-ascorbic acid was used as a positive control.

Xanthine oxidase (XO) inhibitory activity

XO inhibition assays were performed using a modification of the method of Nongonierma and FitzGerald (2012). 50 μ L of sample were treated in a 96-well plate containing EDTA (12.5 μ M), hydroxylamine phosphate (25 μ M), and xanthine (0.125 mM). In addition, 50 μ L XO (0.1 U/mL) was treated in a 96-well plate and incubated at 37°C for 30 min. The absorbance of the uric acid formed was measured using a spectrophotometer (BIO-TEK US/MQX 200, USA) at 290 nm. BHA was used as a positive control and the XO inhibitory activity of each sample was expressed as an IC_{s0} value.

ChEs inhibitory activity

ChEs inhibition was measured using the spectrophotometric method of Ellman et al. (1961). The reaction mixture contained 140 μ L 100 mM sodium phosphate buffer (pH 8.0), 20 μ L sample, and 20 μ L of either AChE (0.36 U/mL) or BChE (0.36 U/mL). The solution was placed in a 96-well plate and mixed. After incubation at room temperature for 15 min, 10 μ L DTNB solution and 10 μ L ACh or BCh were added. The absorbance of all reactions was measured using a spectrophotometer (BIO-TEK US/MQX 200, USA) at 412 nm and eserine was used as a positive control. The ChEs inhibitory activity of each sample was expressed as an IC₅₀ value.

Total carotenoid contents

A spectrophotometric method was used to evaluate the total carotenoid contents following a modification of the method of Tolasa et al. (2005). Astaxanthin standard (3.0 mg) and BHT (100 mg) were dissolved in 10 mL CH_2Cl_2 . Subsequently, 1 mL of this stock solution was diluted to 10 mL with *n*-hexane, and the absorbance was measured in a UV-visible spectrophotometer (BioMate 5) at a wavelength between 350 and 600 nm. The maximum absorbance was observed at 472 nm and the concentration of astaxanthin in the solution was measured and corrected according to the following formula:

$$C_{astaxanthin} (\mu g/mL) = A \times 10,000/E,$$

where $C_{astaxanthin}$ is the total carotenoid contents, A is the absorbance at 472 nm, E is the extinction coefficient, and 10,000 is the scale factor.

To provide data to plot the standard curve, 0.1, 0.25, 0.50, 0.75, 1.0, 1.25, and 1.5 mL of stock solution were placed in separate 10 mL flasks and made up to the appropriate volume with *n*-hexane. The absorbance was measured at 472 nm.

Statistical analysis

The data were analyzed using analysis of variance following the general linear model procedure (SAS Institute, Cary, NC, USA). Duncan's multiple-range test was applied to determine the significance of differences between means (P < 0.05).

Results and Discussion

Antioxidant activity

Reactive oxygen species (ROS) including superoxide (O^2), hydrogen peroxide (H_2O_2), and hydroxyl radical (HO) are generated by respiration and normal metabolism (Yen and Chen, 1995). Their production in the body causes various diseases such as neurodegenerative diseases (e.g., AD), cancer, rheumatoid arthritis, and atherosclerosis through cell or DNA damage (Cerutti, 1985; Squadrito and Pryor, 1998; Cooke et al., 2003).

We investigated the antioxidant and ChEs inhibitory activities of the acetone and CH₂Cl₂ extracts of the by-products of three shrimp species based on the DPPH and ABTS⁺ radical scavenging, xanthine oxidase inhibitory activities, and reducing power assays (Tables 1 and 2). Of these, PHA showed the most potent DPPH and ABTS+ radical scavenging activities with IC₅₀ values of 1.43 ± 0.02 and 1.44 ± 0.02 mg/mL, respectively. However, all extracts exhibited lower DPPH and ABTS⁺ radical scavenging activities than those of the positive controls, L-ascorbic acid and trolox. The xanthine oxidase inhibitory activities of the acetone and CH2Cl2 extracts of three shrimp by-products were (Table 2): PHA (IC₅₀ = 0.03 ± 0.00 mg/mL) > PHC (IC₅₀ = 0.11 \pm 0.00 mg/mL) > PBA (IC₅₀ = $0.12 \pm 0.01 \text{ mg/mL}) > \text{PBC} (\text{IC}_{50} = 0.57 \pm 0.01 \text{ mg/mL}) >$ PJA (IC₅₀ = 1.03 \pm 0.03 mg/mL) > PJC (IC₅₀ = 1.30 \pm 0.03 mg/mL). In particular, PHA was found to be 47-fold stronger than the positive control, BHA (IC₅₀ = 0.14 ± 0.00 mg/mL). Moreover, the inhibition of PBA toward the xanthine oxidase

 Table 1. DPPH and ABTS* radical scavenging activities of the acetone and dichloromethane extracts of three shrimp by-products

Samples	DPPH (IC ₅₀ , mg/mL)	ABTS ⁺ (IC ₅₀ , mg/mL)
PHA	$1.43 \pm 0.02^{1,\mathrm{f}}$	$1.44\pm0.02^{\rm f}$
PHC	3.22 ± 0.03^{b}	$3.34\pm0.00^{\text{e}}$
PBA	$2.49\pm0.01^{\text{d}}$	$6.76\pm0.02^{\circ}$
PBC	$3.45\pm0.01^{\rm a}$	$6.96\pm0.03^{\text{b}}$
PJA	2.12 ± 0.01^{e}	$6.46\pm0.03^{\text{d}}$
PJC	$2.98 \pm 0.01^{\circ}$	$7.94\pm0.01^{\text{a}}$
L-ascorbic acid	$0.02\pm0.00^{\text{g}}$	
Trolox		$0.11\pm0.00^{\rm g}$

PHA, P. hypsinotus acetone extract; PHC, P. hypsinotus CH₂Cl₂ extract; PBA, P. borealis acetone extract; PBC, P. borealis CH₂Cl₂ extract; PJA, P. japonica acetone extract; PJC, P. japonica CH₂Cl₂ extract.

The values of DPPH and ABTS $^+$ radical scavenging activities were expressed as the means \pm SD. L-ascorbic acid and trolox were used as positive controls.

¹Means within the same row with different superscripts are significantly different by Duncan's multiple range test (P < 0.05).

activity was similar to that of BHA when used as a positive control. However, PJA exerted no notable activity on xanthine oxidase inhibition.

The electron donation capacity of the acetone and CH_2Cl_2 extracts are shown in Table 2. PHA ($EC_{50} = 0.57 \pm 0.01$ mg/mL) exhibited the greatest reducing power and the other extracts showed moderate electron donation capacity. However, all extracts showed lower reducing power than that of L-ascorbic acid, used as a positive control.

These results indicate that the acetone extracts of the shrimp by-products possessed potent scavenging activity toward DPPH and ABTS⁺ radicals, as well as xanthine oxidase inhibitory activity and reducing power, and their activities were higher than those of the CH_2Cl_2 extracts.

ChEs inhibitory activity

The neurotransmitters including ACh and BCh were decomposed by ChEs (Schneider, 2001). Loss of neurotransmitters at the synaptic site of the AD patient's brain is associated with memory and cognitive deficits (Johnson et al., 2000). Therefore, prevention of neurotransmitter loss through inhibition of ChEs activation has been accepted as the most effective treatment in the pathology of AD (Giacobini, 2004).

The ChEs inhibitory activity of the acetone and CH₂Cl₂ extracts of the shrimp by-products was measured using AChE and BChE inhibitory activity assays (Table 3). Among the extracts, PHA showed strong inhibitory activity against both AChE and BChE, with IC₅₀ values of 0.20 ± 0.01 and 0.67 ± 0.03 mg/mL, respectively. PJA (IC₅₀ = 0.30 ± 0.03 , 1.16 ± 0.02 mg/mL) and PBA (IC₅₀ = 0.36 ± 0.01 , 1.18 ± 0.02 mg/mL) also exhibited ChEs inhibitory activity. The CH₂Cl₂ extracts, PHC (0.56 ± 0.02 and 0.76 ± 0.03 mg/mL), PBC (0.64 ± 0.02 and 2.05 ± 0.02 mg/mL), and PJC (0.94 ± 0.03 and 1.76 ± 0.04 mg/mL), showed moderate ChEs inhibitory activity. However, all extracts showed lower ChEs inhibitory activities than that of Eserine, which was used as a positive control (IC₅₀ = 0.00011 ± 0.00000 and 0.00013 ± 0.00000 mg/mL).

Total carotenoid contents

Carotenoids exist as natural pigments in terrestrial plants, seaweeds, microorganisms, and crustaceans, and exert various biological effects, such as an inhibitory effect against oxidative damage, and prevention of cardiovascular disease and cancer (Davies, 1985; Kohlmeier and Hastings, 1995; Stahl et al., 1998; Fraser and Bramley, 2004). Therefore, we evaluated the total carotenoid contents of the acetone and CH_2Cl_2 extracts and investigated the correlation between the carotenoid contents and their biological activities.

The total carotenoid contents of the acetone and CH_2Cl_2 extracts (Table 4): PHA (11.12 ± 0.11 mg/g) > PBA (11.06 ± 0.00 mg/g) > PBC (9.51 ± 0.39 mg/g) > PJA (8.93 ± 0.00 mg/g) > PHC (7.76 ± 0.19 mg/g) > PJC (6.79 ± 0.19 mg/g).

Table 2. Xanthine oxidase inhibitory activity and reducing power of the acetone and dichloromethane extracts of three shrimp by-products

Samples	Xanthine oxidase (IC50, mg/mL)	Reducing power (EC50, mg/mL)
РНА	$0.03\pm 0.00^{\rm 1,f}$	$0.57\pm0.01^{\text{e}}$
РНС	$0.11 \pm 0.00^{\rm e}$	$1.08\pm0.27^{\text{d}}$
PBA	$0.12\pm0.01^{\text{de}}$	$1.56\pm0.01^{\circ}$
PBC	$0.57\pm0.01^{\circ}$	$6.02\pm0.00^{\text{ a}}$
PJA	$1.03\pm0.03^{\text{b}}$	$1.60 \pm 0.00^{\circ}$
PJC	$1.30\pm0.03^{\rm a}$	$3.35 \pm 0.00^{\mathrm{b}}$
BHA	$0.14\pm0.00^{\text{d}}$	
L-ascorbic acid		$0.03\pm0.00^{\rm f}$

PHA, P. hypsinotus acetone extract; PHC, P. hypsinotus CH₂Cl₂ extract; PBA, P. borealis acetone extract; PBC, P. borealis CH₂Cl₂ extract; PJA, P. japonica acetone extract; PJC, P. japonica CH₂Cl₂ extract.

The values of xanthine oxidase inhibitory activity and reducing power were expressed as the means \pm SD. BHA and L-ascorbic acid were used as positive controls.

¹Means within the same row with different superscripts are significantly different by Duncan's multiple range test (P < 0.05).

 Table 3. Cholinesterase inhibitory activity of the acetone and dichloromethane extracts of three shrimp by-products

Samples	AChE (IC ₅₀ , mg/mL)	BChE (IC ₅₀ , mg/mL)
PHA	$0.20\pm 0.01^{\rm 1,f}$	0.67 ± 0.03^{e}
РНС	$0.56\pm0.02^{\rm c}$	$0.76\pm0.03^{\rm d}$
PBA	$0.36\pm0.01^{\text{d}}$	$1.18\pm0.02^{\rm c}$
PBC	$0.64\pm0.02^{\rm b}$	$2.05\pm0.02^{\text{a}}$
PJA	$0.30\pm0.03^{\text{e}}$	$1.16 \pm 0.02^{\circ}$
PJC	$0.94\pm0.03^{\text{a}}$	$1.76\pm0.04^{\text{b}}$
Eserine	0.00011 ± 0.00000^{g}	$0.00013 \pm 0.00000^{\rm f}$

PHA, P. hypsinotus acetone extract; PHC, P. hypsinotus CH₂Cl₂ extract; PBA, P. borealis acetone extract; PBC, P. borealis CH₂Cl₂ extract; PJA, P. japonica acetone extract; PJC, P. japonica CH₂Cl₂ extract.

The values of cholinesterase inhibitory activity were expressed as the means \pm SD. Eserine was used as a positive control.

¹Means within the same row with different superscripts are significantly different by Duncan's multiple range test (P < 0.05).

 Table 4. Total carotenoid contents of the acetone and dichloromethane

 extracts of three shrimp by-products

Samples	Total carotenoid contents (mg/g)
PHA	$11.12 \pm 0.11^{1,a}$
PHC	7.76 ± 0.19^{d}
PBA	11.06 ± 0.00^{a}
PBC	9.51 ± 0.39^{b}
PJA	$8.93 \pm 0.00^{\circ}$
PJC	$6.79 \pm 0.19^{\rm e}$

PHA, P. hypsinotus acetone extract; PHC, P. hypsinotus CH₂Cl₂ extract; PBA, P. borealis acetone extract; PBC, P. borealis CH₂Cl₂ extract; PJA, P. japonica acetone extract; PJC, P. japonica CH₂Cl₂ extract.

¹Means within the same row with different superscripts are significantly different by Duncan's multiple range test (P < 0.05).

The acetone extracts including PHA, PBA, and PJA exhibited higher total carotenoid contents compared with the CH₂Cl₂ extracts. In particular, PHA exhibited the highest total carotenoid contents.

In the present study, total carotenoid contents was suggested to be related to antioxidant and inhibitory activities. Recent researches have suggested several possible roles of carotenoids in neurodegenerative diseases, including AD and oxidative stress (Rao and Balachandran, 2003). High levels of reactive oxygen species and low antioxidant capacity cause damage to the human brain, which is vulnerable to oxidative damage, and trigger development of neurodegenerative diseases. Moreover, significant reductions in the levels of carotenoids were reported in Alzheimer's and parkinson's diseases and vascular dementia patients (Foy et al., 1999).

Thus, the antioxidant and ChE inhibitory activities of shrimp by-product extracts are possibly attributable to carotenoids. Further study should focus on isolating the active components of shrimp by-products and investigating the underlying mechanisms.

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