

First Total Synthesis of (±)-Elatadihydrochalcone and Its Derivatives

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Pyranochalcones and pyranodihydrochalcones are an abundant subclass of the flavonoids and are widely found in nature.¹ Members of the pyranochalcones and pyranodihydrochalcones have been associated with a wide variety of biological activities such as antimutagenic, antimicrobial, anti-ulcer, antitumor, anti-leishmania, and antimalarial activities. Furthermore, some plants are used as traditional medicines in China and Europe.²

Recently, naturally occurring elatadihydrochalcone bearing pyranodihydrochalcone moiety was isolated from seedpods of *Tephrosia elata*, which was distributed in East Africa (Figure 1).³ This plant is also used as traditional medicine to treat infectious diseases.⁴ The crude extract from the seedpods of this plant showed anti-plasmodia activity with IC₅₀ values of 2.8 ± 0.3 and 5.5 ± 0.3 μg/mL against chloroquine-sensitive Sierra Leone I and chloroquine-resistant Indochina I strains, respectively.³ It is the first report on the occurrence of a β-hydroxydihydrochalcone in the genus *Tephrosia* species. The potent anti-plasmodia activity has stimulated interest in the synthesis of naturally occurring elatadihydrochalcone.³

Recently, we developed a new and useful methodology for preparing a variety of benzopyrans using ethylenediamine diacetate-catalyzed reactions of resorcinols to α,β-unsaturated aldehydes.⁵ This methodology provided a rapid route for the synthesis of benzopyran derivatives with a variety of substituents on the pyranyl ring.⁵ Our efforts in developing

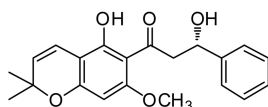
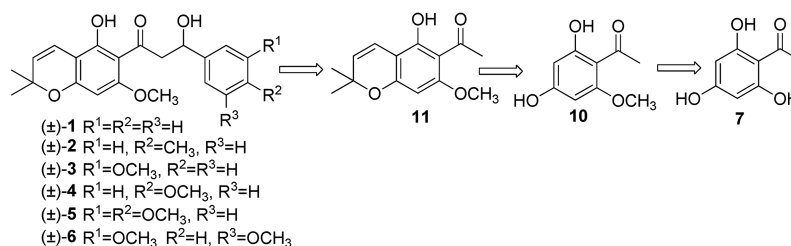


Figure 1. Naturally occurring elatadihydrochalcone (**1**) isolated from *Tephrosia elata*.



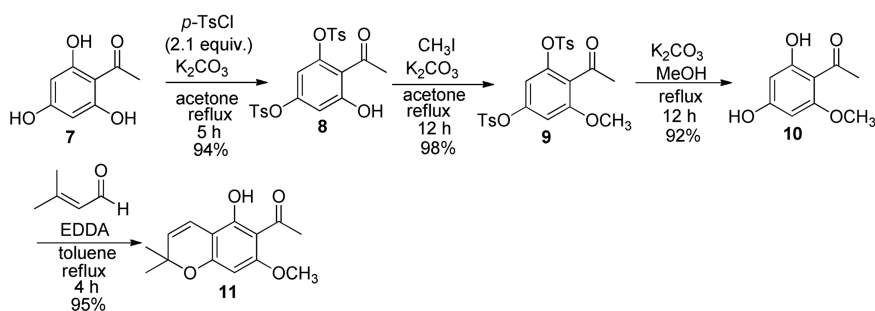
Scheme 1. Retrosynthetic analysis for a racemic elatadihydrochalcone (**1**) and its derivatives.

these methodologies led to the synthesis of biologically interesting natural products bearing pyranochalcone moiety.⁶ As an expansion and ongoing study of our previous work, we report herein the first total synthesis of elatadihydrochalcone (**1**) as a racemate, along with its unnatural derivatives.

Results and Discussion

The retrosynthetic approaches for synthesizing (±)-elatadihydrochalcone (**1**) and its unnatural derivatives (±)-**2-6** were analyzed as shown in Scheme 1. (±)-Elatadihydrochalcone (**1**) and its derivatives (±)-**2-6** could be prepared by base-catalyzed aldol reactions of **11** to the corresponding aryl aldehydes. The key intermediate benzopyran **11** could be also generated from **10** through benzopyran formation reaction. The compound **10** could be derived from commercially available 2,4,6-trihydroxyacetophenone (**7**) by selective *ortho*-methylation.

The total synthesis of (±)-elatadihydrochalcone (**1**) and its derivatives (±)-**2-6** was next attempted starting from 2,4,6-trihydroxyacetophenone (**7**) (Scheme 2). Conversion of **7** to **10** was first investigated through a literature survey. Previously reported *ortho*-methylation of **7** was achieved by using excess of (trimethylsilyl)diazomethane (TMSCHN₂)⁷ or dimethyl sulfate in the presence of potassium carbonate.⁸ Under these conditions, the desired product **10** was only obtained in 3-10% yield, together with the other products. As another synthetic route, Tsukayama prepared compound **10** in 3 steps starting from **7** by selective dibenylation followed by *O*-methylation and subsequent dedibenylation.⁹ We previously used this method for the synthesis of **10**.¹⁰ Although the overall yield is satisfactory (3 steps, 60%), more facile and general synthetic routes to **10** are



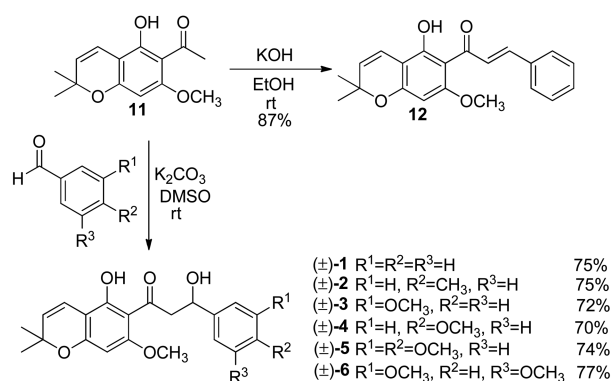
Scheme 2

needed. In order to efficiently synthesize **10** with high yields (3 steps, 85%), we modified Tsukayama's method with *p*-TsCl. As shown in Scheme 2, selective diprotection of **7** with 2.1 equivalents of *p*-TsCl in the presence of K_2CO_3 afforded **8** in 94% yield, which reacted with methyl iodide to give **9** in 98% yield. Deprotection of ditosyl groups on **9** in the presence of K_2CO_3 in refluxing methanol gave **10** in 92% yield. Treatment of **10** with 3-methyl-2-butenal in the presence of 20 mol % of ethylenediamine diacetate in refluxing toluene for 4 h provided isoevodionol (**11**) in 95% yield, which was isolated from *Evodia leptota*.¹¹

To complete the total synthesis of (±)-elatadihydrochalcone (**1**) and its derivatives (±)-**2-6**, an aldol reaction was next attempted (Scheme 3). A reaction of compound **11** with benzaldehyde using KOH in ethanol at room temperature for 24 h provided compound **12** with chalcone moiety in 87% yield. Although several successful aldol reactions of acetophenone with aryl aldehydes in the presence of strong base such as NaOH, $Ca(OH)_2$, or KOH to give β-hydroxydihydrochalcones have been reported in moderate yield,¹² but reactions of **11** with benzaldehyde in the presence of NaOH or KOH in ethanol or water to give β-hydroxydihydrochalcones were unsuccessful. In 2004, Wang reported that reactions of acetophenone with aryl aldehydes in the presence of Na_2CO_3 (0.25 equiv.) in water gave β-hydroxy ketones in very high yields.¹³ Using this reaction condition, we tried aldol reaction of **11** with benzaldehyde. Treatment of **11** with benzaldehyde in the presence of Na_2CO_3 (0.5 equiv.) in water at room temperature for 8 h gave the desired product (±)-**1** in 65% yield, together a trace of chalcone. Instead of Na_2CO_3 , treatment of **11** with benzaldehyde in the presence of K_2CO_3 (0.5 equiv.) in DMSO at room temperature for 8 h afforded (±)-**1** in increased yield (75%), together with a trace of chalcone. The spectral data of synthetic material (±)-**1** were in agreement with those of the natural product reported in the literature.³

Similarly, reaction of **11** with the corresponding aryl aldehydes in the presence of K_2CO_3 (0.5 equiv.) in DMSO at room temperature for 8–15 h provided β-hydroxydihydrochalcones (±)-**2-6** in 75, 72, 70, 74 and 77% yield, respectively.

In conclusion, a concise synthetic route for biologically interesting elatadihydrochalcone (**1**) and its unnatural derivatives as racemates starting from commercially available 2,4,6-trihydroxyacetophenone in 5 steps was developed.



Scheme 3

The key strategies involved benzopyran formation by ethylenediamine diacetate-catalyzed reactions and weak base-catalyzed aldol reactions to give β-hydroxydihydrochalcones. Further synthetic approaches for naturally occurring elatadihydrochalcone and its enantiomer are currently underway.

Experimental

All the experiments were carried out in a nitrogen atmosphere. Merck precoated silica gel plates (Art. 5554) with a fluorescent indicator were used for analytical TLC. Flash column chromatography was performed using silica gel 9385 (Merck). The ¹H NMR and ¹³C NMR spectra were recorded on a Bruker Model ARX (300 and 75 MHz, respectively) spectrometer in CDCl₃ using δ 77.0 ppm as the solvent. The IR spectra were recorded on a Jasco FTIR 5300 spectrophotometer. HRMS were carried out at the Korea Basic Science Institute.

4-Acetyl-5-hydroxy-1,3-phenylene bis(4-methylbenzenesulfonate) (8). To a solution of 2,4,6-trihydroxyacetophenone (**7**) (0.840 g, 5.0 mmol) in dry acetone (30 mL) was added K_2CO_3 (3.455 g, 25.0 mmol) and *p*-toluenesulfonyl chloride (2.002 g, 10.5 mmol). The reaction mixture was refluxed under N₂ for 8 h. Evaporation of acetone, addition of water (20 mL), extraction with EtOAc (3 × 30 mL), washing with brine (20 mL), and removal of the solvent followed by flash column chromatography on silica gel using hexane/EtOAc (10:1) gave **8** (2.226 g, 94%) as a solid: mp 69–70 °C; ¹H NMR (300 MHz, CDCl₃) δ 12.73 (brs, OH), 7.70 (d, *J* = 8.1 Hz, 2H), 7.69 (d, *J* = 8.4 Hz, 2H), 7.36

(d, $J = 8.1$ Hz, 2H), 7.33 (d, $J = 8.4$ Hz, 2H), 6.42 (d, $J = 2.4$ Hz, 1H), 6.32 (d, $J = 2.4$ Hz, 1H), 2.65 (s, 3H), 2.47 (s, 3H), 2.44 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 203.2, 164.4, 153.5, 150.6, 146.6, 146.2, 131.7, 131.5, 130.3, 130.1, 128.5, 128.3, 113.8, 110.2, 107.6, 32.5, 21.8, 21.7; IR (KBr) 3070, 2934, 1628, 1611, 1377, 1253, 1188, 1114, 1045, 801, 742 cm^{-1} ; HRMS m/z [M^+] calcd for $\text{C}_{22}\text{H}_{20}\text{O}_8\text{S}_2$: 476.0600. Found: 476.0602.

4-Acetyl-5-methoxy-1,3-phenylene bis(4-methylbenzenesulfonate) (9). To a solution of **8** (2.0 g, 4.2 mmol) in dry acetone (30 mL) was added K_2CO_3 (1.452 g, 10.5 mmol) and methyl iodide (0.894 g, 6.3 mmol). The reaction mixture was refluxed under N_2 for 4 h. Evaporation of acetone, addition of water (20 mL), extraction with EtOAc (3 \times 30 mL), washing with brine (20 mL), and removal of the solvent followed by flash column chromatography on silica gel using hexane/EtOAc (8:1) gave **9** (2.015 g, 98%) as a solid: mp 85-86 $^\circ\text{C}$; ^1H NMR (300 MHz, CDCl_3) δ 7.67 (d, $J = 8.1$ Hz, 2H), 7.63 (d, $J = 8.4$ Hz, 2H), 7.34-7.29 (m, 4H), 6.51 (d, $J = 1.8$ Hz, 1H), 6.47 (d, $J = 1.8$ Hz, 1H), 3.66 (s, 3H), 2.41 (s, 6H), 2.30 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 197.9, 157.3, 150.3, 145.9, 145.8, 145.5, 131.4, 131.2, 129.8, 129.7, 128.2, 128.1, 123.8, 108.9, 104.5, 56.0, 31.5, 21.4, 21.4; IR (KBr) 3095, 3060, 2954, 1709, 1599, 1449, 1380, 1193, 1086, 994, 798, 675 cm^{-1} ; HRMS m/z [M^+] calcd for $\text{C}_{23}\text{H}_{22}\text{O}_8\text{S}_2$: 490.0756. Found: 490.0755.

2,4-Dihydroxy-6-methoxyacetophenone (10).⁷⁻⁹ To a solution of **9** (1.80 g, 3.7 mmol) in methanol (30 mL) was added K_2CO_3 (1.523 g, 11.0 mmol). The reaction mixture was refluxed for 4 h. Evaporation of methanol, addition of water (20 mL) and 1 N HCl (10 mL), extraction with EtOAc (3 \times 30 mL), washing with brine (20 mL), and removal of the solvent followed by flash column chromatography on silica gel using hexane/EtOAc (4:1) gave **10** (0.617 g, 92%) as a solid: mp 203-204 $^\circ\text{C}$; ^1H NMR (300 MHz, acetone- d_6) δ 13.92 (s, 1H), 7.38 (brs, 1H), 6.03 (d, $J = 2.0$ Hz, 1H), 5.95 (d, $J = 2.0$ Hz, 1H), 3.90 (s, 3H), 2.55 (s, 3H); ^{13}C NMR (75 MHz, acetone- d_6) δ 203.4, 168.2, 165.7, 164.7, 105.9, 96.6, 91.7, 56.1, 33.0; IR (KBr) 3136, 1638, 1586, 1523, 1468, 1439, 1365, 1282, 1256, 1201, 1166, 1074, 959, 810 cm^{-1} .

Isoevodionol (11). To a solution of **10** (0.182 g, 1.0 mmol) and 3-methyl-2-butenal (0.168 g, 2.0 mmol) in toluene (10 mL) was added ethylenediamine diacetate (0.018 g, 0.1 mmol) at room temperature. The reaction mixture was refluxed for 10 h and then cooled to room temperature. H_2O (30 mL) was added and the mixture was extracted with EtOAc (3 \times 30 mL). Evaporation of the solvent and purification by column chromatography on silica gel hexane/EtOAc (7:1) gave **11** (0.236 g, 95%) as a solid; mp 128-129 $^\circ\text{C}$; ^1H NMR (300 MHz, CDCl_3) δ 14.28 (s, 1H), 6.62 (d, $J = 10.0$ Hz, 1H), 5.85 (s, 1H), 5.38 (d, $J = 10.0$ Hz, 1H), 3.81 (s, 3H), 2.56 (s, 3H), 1.41 (s, 6H); ^{13}C NMR (75 MHz, CDCl_3) δ 203.0, 162.8, 161.7, 160.0, 125.2, 115.9, 105.5, 102.5, 90.5, 78.0, 55.4, 32.9, 28.2; IR (KBr) 2924, 2855, 1620, 1464, 1362, 1269, 1206, 1159, 1124, 891, 831, 731 cm^{-1} ; HRMS: m/z [M^+] calcd for $\text{C}_{14}\text{H}_{16}\text{O}_4$: 248.1049; found: 248.1047.

General Procedure for the Synthesis of β -Hydroxydihydrochalcones 1-6. To a solution of **11** (100 mg, 0.40 mmol) in DMSO (5 mL) was added anhydrous K_2CO_3 (28 mg, 0.20 mmol) and aryl aldehydes (0.48 mmol). The reaction mixture was stirred at room temperature and the reaction progress was monitored by TLC for 8-15 h. After the reaction was completed, addition of water (10 mL) and 1 N HCl (2 mL), extraction with EtOAc (3 \times 20 mL), washing with brine (20 mL), and removal of the solvent followed by flash column chromatography on silica gel using hexane/EtOAc (4:1) gave the corresponding β -hydroxydihydrochalcones **1-6**.

(\pm)-Elatadihydrochalcone (1): Yield 75%, a solid, mp 118-120 $^\circ\text{C}$; ^1H NMR (300 MHz, CDCl_3) δ 13.93 (brs, OH), 7.36-7.17 (m, 5H), 6.58 (d, $J = 9.9$ Hz, 1H), 5.79 (s, 1H), 5.38 (d, $J = 9.9$ Hz, 1H), 5.19 (dd, $J = 9.0, 3.0$ Hz, 1H), 3.69 (s, 3H), 3.42 (dd, $J = 18.3, 3.0$ Hz, 1H), 3.27 (dd, $J = 18.3, 9.0$ Hz, 1H), 1.37 (s, 6H); ^{13}C NMR (75 MHz, CDCl_3) δ 204.1, 162.9, 161.8, 160.6, 143.3, 128.4, 127.3, 125.8, 125.4, 115.8, 105.5, 102.7, 91.3, 78.3, 70.1, 55.6, 52.7, 28.3; IR (KBr) 3566, 2973, 2935, 1608, 1383, 1281, 1198, 1136, 873, 699 cm^{-1} ; HRMS m/z [M^+] calcd for $\text{C}_{21}\text{H}_{22}\text{O}_5$: 354.1467. Found: 354.1469.

(\pm)-3-Hydroxy-1-(5-hydroxy-7-methoxy-2,2-dimethyl-2H-chromen-6-yl)-3-*p*-tolylpropan-1-one (2): Yield 75%, an oil; ^1H NMR (300 MHz, CDCl_3) δ 13.43 (brs, OH), 7.33 (d, $J = 8.1$ Hz, 2H), 7.21 (d, $J = 7.8$ Hz, 2H), 6.56 (d, $J = 10.2$ Hz, 1H), 6.03 (s, 1H), 5.43 (d, $J = 10.2$ Hz, 1H), 5.36 (dd, $J = 12.9, 2.7$ Hz, 1H), 3.87 (s, 3H), 2.97 (dd, $J = 16.5, 12.9$ Hz, 1H), 2.77 (dd, $J = 16.5, 2.7$ Hz, 1H), 2.36 (s, 3H), 1.44 (s, 3H), 1.42 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 189.5, 162.0, 159.9, 158.8, 138.3, 135.9, 129.3, 126.2, 125.9, 115.9, 105.6, 102.8, 93.6, 78.8, 78.0, 56.1, 45.5, 28.4, 28.1, 21.2; IR (neat) 3456, 3444, 3053, 2973, 2931, 1674, 1634, 1604, 1571, 1466, 1348, 1202, 1150, 1119, 911, 817, 734 cm^{-1} ; HRMS m/z [$\text{M}-\text{H}_2\text{O}$] $^+$ calcd for $\text{C}_{22}\text{H}_{22}\text{O}_4$: 350.1518. Found: 350.1515.

(\pm)-3-Hydroxy-1-(5-hydroxy-7-methoxy-2,2-dimethyl-2H-chromen-6-yl)-3-(3-methoxyphenyl)propan-1-one (3): Yield 72%, an oil; ^1H NMR (300 MHz, CDCl_3) δ 13.76 (brs, OH), 7.31 (d, $J = 7.8$ Hz, 1H), 7.01-6.98 (m, 2H), 6.89-6.86 (m, 1H), 6.58 (d, $J = 9.9$ Hz, 1H), 6.03 (s, 1H), 5.44 (d, $J = 9.9$ Hz, 1H), 5.36 (dd, $J = 12.9, 3.0$ Hz, 1H), 3.87 (s, 3H), 3.81 (s, 3H), 2.95 (dd, $J = 16.5, 12.9$ Hz, 1H), 2.78 (dd, $J = 16.5, 3.0$ Hz, 1H), 1.44 (s, 3H), 1.42 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 189.3, 162.0, 159.9, 159.7, 158.7, 140.5, 129.8, 126.3, 118.1, 115.9, 113.6, 111.6, 105.5, 102.9, 93.7, 78.7, 78.0, 56.2, 55.3, 45.6, 28.4, 28.1; IR (Neat) 3442, 3054, 2970, 2936, 1673, 1635, 1604, 1463, 1348, 1266, 1201, 1151, 1118, 1047, 884, 779, 735 cm^{-1} ; HRMS m/z [$\text{M}-\text{H}_2\text{O}$] $^+$ calcd for $\text{C}_{22}\text{H}_{22}\text{O}_5$: 366.1467. Found: 366.1465.

(\pm)-3-Hydroxy-1-(5-hydroxy-7-methoxy-2,2-dimethyl-2H-chromen-6-yl)-3-(4-methoxyphenyl)propan-1-one (4): Yield 70%, an oil; ^1H NMR (300 MHz, CDCl_3) δ 13.98 (brs, OH), 7.36 (d, $J = 8.7$ Hz, 2H), 6.92 (d, $J = 8.7$ Hz, 2H), 6.56 (d, $J = 10.2$ Hz, 1H), 6.03 (s, 1H), 5.44 (d, $J = 10.2$ Hz, 1H), 5.34 (dd, $J = 12.9, 3.0$ Hz, 1H), 3.87 (s, 3H), 3.82 (s, 3H),

2.98 (dd, $J = 16.5, 12.9$ Hz, 1H), 2.76 (dd, $J = 16.5, 3.0$ Hz, 1H), 1.44 (s, 3H), 1.42 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 189.6, 162.1, 160.0, 159.7, 158.9, 130.9, 127.5, 126.2, 116.0, 114.0, 105.6, 102.8, 93.7, 78.7, 78.0, 56.2, 55.3, 45.4, 28.5, 28.1; IR (neat) 3583, 3467, 3055, 2970, 2933, 1674, 1604, 1570, 1513, 1464, 1347, 1202, 1150, 1118, 1034, 884, 830, 735 cm^{-1} ; HRMS m/z $[\text{M}-\text{H}_2\text{O}]^+$ calcd for $\text{C}_{22}\text{H}_{22}\text{O}_5$: 366.1467. Found: 366.1465.

(±)-3-(3,4-Dimethoxyphenyl)-3-hydroxy-1-(5-hydroxy-7-methoxy-2,2-dimethyl-2H-chromen-6-yl)propan-1-one (5): Yield 74%, an oil; ^1H NMR (300 MHz, CDCl_3) δ 13.96 (brs, OH), 6.97 (s, 1H), 6.90 (d, $J = 8.7$ Hz, 1H), 6.82 (d, $J = 8.7$ Hz, 1H), 6.62 (d, $J = 9.9$ Hz, 1H), 5.84 (s, 1H), 5.43 (d, $J = 9.9$ Hz, 1H), 5.19 (dd, $J = 8.7, 3.0$ Hz, 1H), 3.88 (s, 3H), 3.85 (s, 3H), 3.76 (s, 3H), 3.40 (dd, $J = 18.3, 3.0$ Hz, 1H), 3.30 (dd, $J = 18.3, 8.7$ Hz, 1H), 1.41 (s, 6H); ^{13}C NMR (75 MHz, CDCl_3) δ 204.2, 162.9, 161.8, 160.6, 148.9, 148.2, 136.0, 125.5, 117.9, 115.7, 110.9, 109.0, 105.5, 102.7, 91.3, 78.3, 69.9, 55.8, 55.7, 55.6, 52.8, 28.3; IR (neat) 3513, 2959, 1612, 1512, 1443, 1422, 1261, 1205, 1140, 1033, 876, 817, 737 cm^{-1} ; HRMS m/z $[\text{M}^+]$ calcd for $\text{C}_{23}\text{H}_{26}\text{O}_7$: 414.1679. Found: 414.1681.

(±)-3-(3,5-Dimethoxyphenyl)-3-hydroxy-1-(5-hydroxy-7-methoxy-2,2-dimethyl-2H-chromen-6-yl)propan-1-one (6): Yield 77%, a solid, mp 111-112 °C; ^1H NMR (300 MHz, CDCl_3) δ 13.96 (brs, OH), 6.62 (d, $J = 9.9$ Hz, 1H), 6.56 (d, $J = 2.1$ Hz, 2H), 6.35 (d, $J = 2.1$ Hz, 1H), 5.84 (s, 1H), 5.43 (d, $J = 9.9$ Hz, 1H), 5.18 (dd, $J = 9.0, 3.0$ Hz, 1H), 3.77 (s, 6H), 3.76 (s, 3H), 3.40 (dd, $J = 18.0, 3.0$ Hz, 1H), 3.28 (dd, $J = 18.0, 9.0$ Hz, 1H), 1.42 (s, 6H); ^{13}C NMR (75 MHz, CDCl_3) δ 204.0, 162.9, 161.8, 160.7, 160.5, 145.9, 125.4, 115.7, 105.4, 103.7, 102.7, 99.2, 91.2, 78.2, 70.1, 55.6, 55.2, 52.7, 28.3; IR (neat) 3522, 2970, 2939, 2839, 1610, 1461, 1424, 1285, 1150, 1060, 884, 841, 735 cm^{-1} ; HRMS m/z $[\text{M}^+]$ calcd for $\text{C}_{23}\text{H}_{26}\text{O}_7$: 414.1679. Found: 414.1679.

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References

- (a) Wagner, H.; Farkas, L. In *The Flavonoids*; Harborne, J. B., Mabry, T. J., Mabry, H., Eds.; Academic Press: New York, 1975; p 127. (b) Gripenberg, J. In *The Chemistry of Flavonoid Compounds*; Geissman, T. A., Ed.; MacMillan: New York, 1962; p 409. (c) Wollenweber, E. In *The Flavonoids*; Harborne, J. B., Ed.; Chapman & Hall: London, 1994; p 259. (d) In *The Handbook of Natural Products*; Harborne, J. B., Baxter, H., Eds.; John Wiley & Sons Ltd.: London, 1999, Vol 2, p 1. (e) Fang, N.; Casida, J. E. *J. Nat. Prod.* **1999**, *62*, 205. (f) Saini, T. R.; Pathak, V. P.; Khanna, R. N. *J. Nat. Prod.* **1983**, *46*, 936. (g) Pathak, V. P.; Saini, T. R.; Khanna, R. N. *Phytochemistry* **1983**, *22*, 1303. (h) Subrahmanyam, K.; Madhusudhana Rao, V.; Jagannadha Rao, K. V. *Indian J. Chem.* **1997**, *15B*, 12. (i) Rao, M. S.; Rao, P. S.; Tóth, G.; Balázs, B.; Duddeck, H. *J. Nat. Prod.* **1998**, *61*, 1148. (j) Krohn, K.; Steingröver, K.; Rao, M. S. *Phytochemistry* **2002**, *61*, 931. (k) Narender, T.; Tanvir, S. K.; Rao, M. S.; Srivastava, K.; Puri, S. K. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 2453.
- (a) Welton, A. F.; Tobias, L. D.; Fiedler-Nagy, C.; Anderson, W.; Hope, W.; Meyers, K.; Coffey, J. W. In *Plant Flavonoids in Biology and Medicine*; Cody, V., Middleton, Jr., E., Harborne, J. B., Eds.; Alan R. Liss: New York, 1986; p 231. (b) Miranda, C. L.; Stevens, J. F.; Helmrich, A.; Henderson, M. C.; Rodriguez, R. J.; Yang, Y.-H.; Deinzer, M. L.; Barnes, D. W.; Buhler, D. R. *Food Chem. Toxicol.* **1999**, *37*, 271. (c) Han, A.-R.; Kang, Y.-J.; Windono, T.; Lee, S. K.; Seo, E.-K. *J. Nat. Chem.* **2006**, *69*, 719. (d) Demizu, S.; Kajiyama, K.; Hiraga, Y.; Kinoshida, T.; Koyama, K.; Takahashi, K.; Tamura, Y.; Okada, K.; Kinoshida, T. *Chem. Pharm. Bull.* **1992**, *40*, 392. (e) Narender, T.; Gupta, S.; Gupta, S. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3913.
- Muiva, L. M.; Yenesew, A.; Dereese, S.; Heydenreich, M.; Peter, M. G.; Akala, H. M.; Eyase, F.; Waters, N. C.; Mutai, C.; Keriko, J. M.; Walsh, D. *Phytochem. Lett.* **2009**, *2*, 99.
- Kokwaro, J. O. In *Medicinal Plants of East Africa*; East Africa Literature Bureau, 1993, p 255.
- (a) Xia, L.; Lee, Y. R. *Synlett* **2008**, 1643. (b) Lee, Y. R.; Hung, T. V. *Tetrahedron* **2008**, *64*, 7338. (c) Lee, Y. R.; Xia, L. *Tetrahedron Lett.* **2008**, *49*, 3283. (d) Lee, Y. R.; Kim, Y. M.; Kim, S. H. *Tetrahedron* **2009**, *65*, 101. (e) Jung, D. H.; Lee, Y. R.; Kim, S. H. *Helv. Chim. Acta* **2010**, *93*, 635. (f) Lee, H. J.; Kim, S. H.; Lee, Y. R.; Wang, X.; Lyoo, W. S. *Bull. Korean Chem. Soc.* **2010**, *31*, 3027. (g) Park, B. H.; Lee, Y. R. *Bull. Korean Chem. Soc.* **2010**, *31*, 2712. (h) Hari, G. H.; Lee, Y. R.; Wang, X.; Lyoo, W. S.; Kim, S. H. *Korean Chem. Soc.* **2010**, *31*, 2406. (i) Xia, L.; Narasimhulu, M.; Li, X.; Shim, J.-J.; Lee, Y. R. *Korean Chem. Soc.* **2010**, *31*, 664. (j) Jung, E. J.; Park, B. H.; Lee, Y. R. *Green Chem.* **2010**, *12*, 2003. (k) Park, B. H.; Lee, H. J.; Lee, Y. R. *J. Nat. Prod.* **2011**, *74*, 644.
- (a) Lee, Y. R.; Kim, D. H. *Synthesis* **2006**, 603. (b) Lee, Y. R.; Wang, X.; Kim, B. S. *Synth. Commun.* **2006**, *36*, 2017. (c) Lee, Y. R.; Li, X.; Kim, J. H.; *J. Org. Chem.* **2008**, *73*, 4313. (d) Lee, Y. R.; Wang, X. *Tetrahedron* **2009**, *65*, 10125. (e) Xia, L.; Lee, Y. R. *Bull. Korean Chem. Soc.* **2011**, *32*, 2921. (f) Wang, X.; Lee, Y. R. *Tetrahedron* **2011**, *67*, 9179.
- Aponte, J. C.; Castillo, D.; Estevez, Y.; Gonzalez, G.; Arevalo, J.; Hammond, G. B.; Sauvain, M. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 100.
- (a) Sultana, R.; Hossain, R.; Miah, A. J.; Islam, A. *Orient. J. Chem.* **2006**, *22*, 77. (b) Alam, S.; Miah, M. A. J.; Islam, A. *ACGC Chem. Res. Commun.* **2005**, *18*, 1. (c) Srivastava, S. D.; Srivastava, S. K. *J. Indian Chem. Soc.* **1987**, *64*, 253.
- Tsukayama, M.; Kawamura, Y.; Tamaki, H.; Kubo, T.; Horie, T. *Bull. Chem. Soc. Jpn.* **1989**, *62*, 826.
- Lee, Y. R.; Xia, L. *Synthesis* **2007**, 3240.
- Li, G.-L.; Zeng, J.-F.; Song, C.-Q.; Zhu, D.-Y. *Phytochemistry* **1997**, *44*, 1175.
- (a) Singh, P.; Bhardwaj, A. *J. Med. Chem.* **2010**, *53*, 3707. (b) Gravotto, G.; Demetri, A.; Nano, G. M.; Palmisano, G.; Penoni, A.; Tagliapietra, S. *Eur. J. Org. Chem.* **2003**, 4438. (c) Karmee, S. K.; Hanefeld, U. *ChemSusChem* **2011**, *4*, 1118. (d) Mitchell, P. W. D. US Patent 4874900; AN 1989:574454, 1989. (f) Weissmehl, K.; Arpe, H.-J. *Industrial Organic Chemistry*, 4th ed.; Wiley-VCH: Weinheim, 2003.
- Wang, G.; Zhang, Z.; Dong, Y. *Org. Process Res. Dev.* **2004**, *8*, 18.