# Synthesis of Novel Series of Various Substituted1-(5-(2-*p*-tolyloxyquinolin-3-yl)-2-(pyridin-4-yl)-1,3,4-oxadiazol-3(2*H*)-yl)ethanone and its Antibacterial Activity

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**ABSTRACT.** A new series of 1-(5-(2-tolyloxyquinoline-3-yl)-2-(pyridine-4-yl)-1,3,4-oxidiazol-3(2*H*)-yl)ethanones were synthesized from cyclisation of N'-((2-(*p*-tolyloxy)quinoline-3-yl)methylene) isonitonohydrazide in acetic unhydride at reflux condition for 3-4 hr. The structures of the new compounds were confirmed by elemental analyses as well as IR, <sup>1</sup>H NMR and mass spectral data. All the synthesized compounds were screened for their antibacterial activities against various bacterial strains. Several of these compounds showed potential antibacterial activity.

Key words: 1,3,4 oxadiazole, Quinoline, Isoniazide, Cyclisation, Antibacterial activity

### **INTRODUCTION**

Quinoline ring systems represent a major class of heterocycles as they occur in various natural products especially in alkaloids.<sup>1</sup> It possesses diverse biological and physiological activities such as antimalarial,<sup>2</sup> anti-inflammatory,<sup>3</sup> antitumor,<sup>4</sup> DNA binding capacity,<sup>5</sup> antibacterial properties.<sup>6</sup> Recently, quinoline has been employed in the study of bio-organic and bio-organometallic processes.<sup>7</sup> There are many quinoline-based compounds which are known to exhibit anti-TB properties. Compound TMC207<sup>8</sup> bearing a bulky biaryl side chain at position C3, is a highly potent anti-TB agent and is currently in phase II clinical trials. In particular 2-chloroquinoline-3-carbaldehyde has been used as a key intermediate for the synthesis of variety of medicinally valuable compounds.<sup>9</sup>

1,3,4-oxadiazole is widely exploited for various applications. A number of therapeutic agents such as HIV-integrase inhibitor raltegravir, a nitrofuran antibacterial furamizole, a potent PDF inhibitor BB-83698, antihypertensive agents Tiodazosin and Nesapidil are based on 1,3,4-oxadiazole moiety. The 1,3,4-oxadiazole undergoes number of reactions including elctrophillic substitution, nucleophilic substitution, thermal and photochemical. The substituted oxadiazoles are heterocyclic compounds, which serve both as biomimetic and reactive pharamacophores and many are key elements with potential biological activities<sup>10-12</sup> such as pesticidal,<sup>13</sup> antiperipheral vasomotility,<sup>14</sup> CNS stimulant, anti-inflammatory, hypertensive,<sup>15</sup> insecticidal,<sup>16</sup> bactericidal,<sup>17</sup> hypoglycemic,<sup>18-19</sup> analgesic, anticonvulsive, antiemetic, diuretic,<sup>20</sup> muscle relaxant,<sup>21-22</sup> herbicidal <sup>23-24</sup> and fungicidal activity.<sup>25-26</sup>

Isoniazide (INH) is widely applied as first-line drugs for the treatment of tuberculosis, usually in combination with other drugs. Modifying either of these molecules has been a challenge taken up by several research groups.<sup>27</sup> INH prevents a mycolic acid biosynthesis by inhibiting a 2trans-enoyl-acyl carrier protein reductase (InhA) that belongs to the FAS-II (fatty acid synthetase II) system.<sup>28</sup>

In our continuous program in the search of new potent and safe synthesis of biologically active heterocycles,<sup>29-30</sup> we have planned to synthesize some new series of various substituted 1-(5-(2-(p-tolyloxy)quinoline-3-yl)-2-(pyridine-4-yl)-1,3,4-oxadiazol-3(2H)-yl)ethanones by oxidative cyclisation of N-((2-(p-tolyloxy)quinoline-3-yl)methylene)isonitonohydrazide using by cyclisation using acetic anhydride under reflux condition and evaluating their antibacterial activity against the standard drug Streptomycin and Ampicillin.

#### **RESULT AND DISCUSSION**

#### Chemistry

The synthetic work was carried out beginning from 2chloro-3-formylquinoline 1(a-j) according to *Scheme* 1. The reaction of 2-chloro-3-formylquinoline 1(a-j) with *p*-Cresol in presence of K<sub>2</sub>CO<sub>3</sub>/DMF at 80-90 °C for 4-5 hr gave substituted 2-(*p*-tolyoxy)quinoline-3-carbaldehyde 2(a-j). Further reaction of isonicotinohydrazide 3 with substituted 2-(*p*-tolyoxy)quinoline-3-carbaldehyde 2(a-j)



Scheme 1.

gave corresponding N'-((2-(p-tolyloxy)quinoline-3-yl) methylene)isonitonohydrazide **4(a-j)** in acetic acid in ethanol at room temperature. followed by cyclisation of N-((2-(p-tolyloxy)quinoline-3-yl)methylene)isonitonohydrazide using acetic anhydride to gave our targeted product i.e. 1-(5-(2-(p-tolyloxy)quinoline-3-yl)-2-(pyridine-4yl)-1,3,4-oxadiazol-3(2H)-yl)ethanone **5(a-j**). All the synthesized compounds were characterized by their physical and spectral data (IR, <sup>1</sup>H-NMR, Mass).

### Spectral analysis

The structures of the synthesized compounds were confirmed by spectral analysis (IR, <sup>1</sup>H NMR and Mass). The IR spectrum of compound **2a** showed a peak at 1690 cm<sup>-1</sup> due to C=O stretch. In <sup>1</sup>H NMR spectrum it exhibited two singlets, one at  $\delta$  2.41 due to -CH<sub>3</sub> protons and second at  $\delta$  10.65 due to -CHO proton. Mass spectrum was consistent with assigned structure showing M+1) peak at m/z =264.1. The structure of 4a is interpreted from spectroscopic data, IR spectra of compound 4a reveals absorption band in the region 1560 cm<sup>-1</sup> corresponding to -C=N stretching at 1640 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectra of **4a** exhibits a sharp singlet at  $\delta$  2.01 for -CH<sub>3</sub> and one broad singlet at  $\delta$  10.50 of -NH. The pyridine ring protons appeared doublet at  $\delta$  8.92 and 8.75 respectively. C<sub>4</sub>- proton of quinoline nuclei appears as a singlet at  $\delta$  8.33. All other aromatic protons arise at respective position. The IR spectrum of compound **5a** showed absorption peak at 1510 cm<sup>-1</sup> due to C=N stretching vibrations. The peak at 1665 cm<sup>-1</sup> due to the N-COCH<sub>3</sub> confirms the formation of N-acyl oxadiazole ring, its <sup>1</sup>H NMR spectrum revealed a singlet at  $\delta$  4.61 due to CO-CH<sub>3</sub> also one singlet appears at 6.95 ppm due -CH of oxadiazole ring and rest of the protons appears at repective position.

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Comp. No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	Yield (%)	M.P. (°C)
5a	Н	Н	Н	51	Sticky
5b	$CH_3$	Н	Н	56	185-186
5c	Н	CH <sub>3</sub>	Н	46	Sticky
5d	Н	Н	CH <sub>3</sub>	42	161-162
5e	$OCH_3$	Н	Н	54	144-145
5f	Н	OCH <sub>3</sub>	Н	57	191-192
5g	Н	Н	$OCH_3$	54	163-164
5h	$OC_2H_5$	Н	Н	60-	157-158
5i	Н	OC <sub>2</sub> H <sub>5</sub> H	Н	61	178-179
5j	Н	Н	$OC_2H_5$	47	156-157

#### Antibacterial activity

The antibacterial activities of the synthesized compounds were determined by the well-diffusion method.<sup>31</sup> In present work, two Gram positive bacteria, Bacillus subtilis (ATCC 6633), Staphylococcus aureus (ATCC 25923) and two Gram negative bacteria, Salmonella typhimurium (ATCC 23564), Pseudomonas aeruginosa (ATCC 27853) were used to investigate the antibacterial activities. The bacterial liquid cultures were prepared in fusion broth for their activity tests. The compounds were dissolved in DMSO at concentration of 1 mgmL<sup>-1</sup>. Antibacterial activity of DMSO against the test organisms was investigated, and was found to be nil. Molten nutrient agar (15 cm<sup>3</sup>) kept at 45 °C, was then poured into the Petri dishes and allowed to solidify. Ten millimeter diameter holes were then punched carefully using a sterile cork borer and completely filled with the test solutions. The plates were incubated for 24 hr at 37 °C. After 24 hr, the inhibition zone that appeared around the holes in each plate was measured. Antibacterial activity was determined by measuring the diameter of inhibition zone and examining the minimal inhibitory concentration (MIC). Activity of each compound was compared with streptomycin and ampicillin as standards. The observed data of antibacterial activity of compounds and the standard drugs is given in Table 2. The compounds 5e, 5f, 5i and 5j show excellent antibacterial activity against Gram positive bacterial strains. Likewise compounds 5a, 5c and 5g showed excellent activity against Gram negative bacterial strains.

#### **EXPERIMENTAL SECTION**

Melting points were recorded in open capillaries in liquid paraffin bath and are uncorrected. The progress of reaction was monitored by thin layer chromatography using silica gel (Merck). IR spectra were recorded on a SHI-

Tested Compounds	B. subtilis $ZI^{a}(MIC)^{b}$	S. aureus $ZI^{a}(MIC)^{b}$	S. typhi ZI <sup>a</sup> (MIC) <sup>b</sup>	P. aeroginosa ZI <sup>a</sup> (MIC) <sup>b</sup>			
5a	13.4(15)	13.7(15)	15.1(10)	14.8(10)			
5b	13.1(15)	13.2(15)	11.1(20)	10.4(20)			
5c	13.8(15)	13.3(15)	15.1(10)	14.9(10)			
5d	14.3(10)	14.1(10)	12.8(15)	12.7(15)			
5e	14.8(10)	14.6(10)	12.9(15)	12.8(15)			
5f	15.1(10)	14.7(10)	13.5(15)	13.4(15)			
5g	13.2(15)	13.5(15)	14.8(10)	14.5(10)			
5h	14.1(10)	14.6(10)	12.2(15)	12.6(15)			
5i	15.6(10)	15.3(10)	11.7(15)	11.2(15)			
5j	15.1(10)	15.4(10)	11.2(15)	11.3(15)			
Streptomycin	15.1(10)	14.9(10)	16.4(5)	16.1(5)			
Ampicillin	14.3(10)	14.7(10)	16.3(5)	15.9(5)			

Table 2. The Minimum Inhibitor concentrations of tested compounds 5(a-j) in µg/mL

<sup>*a*</sup>Zone of inhibition, <sup>*b*</sup>Minimum inhibitory concentration in µg/mL

MADZU-FT-IR spectrophotometer in KBr disc. <sup>1</sup>H NMR spectra were recorded on BRUKER ADVANCE-II 400 NMR spectrophotometer in DMSO-*d*<sub>6</sub> as a solvent and TMS as an internal standard. Peak values are shown in d ppm. Mass spectra were recorded on a PEP-SCIUX-APIQ pulsar (electron pre-ionization) mass spectrometer. Elemental analyses were performed on Perkin-Elmer EAL-240 elemental analyzers.

# General procedure for the synthesis of 2-(p-tolyloxy)quinoline-3-carbaldehydes 2(a-j)

To the mixture of *p*-cresol (3.7 gm, 0.034 mol) and  $K_2CO_3$  (12.8 gm, 0.093 mol) in DMF, compound **1a** (6 gm, 0.031 mol) was added and the reaction mixture was stirred at 80-90 °C for 4-5 hr. The completion of the reaction was monitored on TLC. After completion of the reaction, ice cold water (50 ml) was poured on the reaction mixture & the solid thus obtained was filtered off and washed with water, further the compound was recrystallized in ethyl acetate.

#### 2-(p-tolyloxy)quinoline-3-carbaldehyde (2a)

IR-(KBr, cm<sup>-1</sup>): 1690(C=O), 1610, 1525(C=C); <sup>1</sup>H NMR-(400 Hz, DMSO- $d_6$ ):  $\delta$  10.65(s, 1H, -CHO), 8.72(s, 1H, Ar-H), 7.89(d, J = 8 Hz, 1H, Ar-H), 7.73(d, J = 8 Hz, 1H, Ar-H), 7.70(t, 1H, Ar-H), 7.45(t, 1H, Ar-H), 7.27(d, J = 8Hz, 2H, Ar-H), 7.18(d, J = 8 Hz, 2H, Ar-H), 2.41(s, 3H, -CH<sub>3</sub>); ES-MS (m/z): 264.4 (M+1).

# General procedure for the synthesis of (E)-N'-((2-tolyloxy) quinoline-3-yl) methylene) isonicotinohydrazide 4(a-j)

An equimolar mixture of 2-chloro-3-formyl quinoline

**1a** (0.004 mol) and isonicotinohydrazide **3** (0.004 mol) in 10 mL ethanol containing few drops of glacial acetic acid was strired at room temperature. After completion of reaction (checked by TLC), the excess of solvent was removed on rotary evaporator to yield solid which was washed with petroleum ether followed by crystallization from ethanol.

### (E)-N'-((2-tolyloxy) quinoline-3-yl) methylene)isonicotinohydrazide (4a)

IR (KBr, cm<sup>-1</sup>): 2980 (-NH), 1560 (C=N), 1513 (C=C); <sup>1</sup>H NMR (400 Hz, DMSO- $d_6$ ):  $\delta$  10.72 (s, 1H, -NH), 8.90 (s, 1H), 8.77 (d, J= 7.92 Hz, 3H, Ar-H), 7.75-7.80 (m, 4H, Ar-H), 7.26 (t, J= 10.94 Hz, 2H, Ar-H), 7.19 (dd, J= 8.21 & 4.24 Hz, 4H, Ar-H) 2.30 (s, 3H, Ar-CH<sub>3</sub>), ES-MS (m/z): 383(M+1); Anal. Calcd. C<sub>23</sub>H<sub>18</sub>N<sub>4</sub>O<sub>2</sub> C, 72.24; H, 4.74; N, 14.65, found C,72.22; H, 4.80; N, 14.56%.

# (E)-N'-((2-tolyloxy)-6-ethoxyquinoline-3-yl)methylene)isonicotinohydrazide (4h)

IR (KBr, cm<sup>-1</sup>) : 3009 (-NH), 1577 (C=N), 1510 (C=C); <sup>1</sup>H NMR (400 Hz, DMSO- $d_6$ ):  $\delta$  10.79 (s, 1H, -NH), 8.92 (s, 1H), 8.73 (d, J= 8.21 Hz, 2H, Ar-H) 7.84-7.89 (dd, 4H, Ar-H), 7.21 (t, J= 11.30 Hz, 2H, Ar-H), 7.16 (dd, J= 8.44 & 4.66Hz, 4H, Ar-H), 2.67 (s, 3H, Ar-CH<sub>3</sub>), 1.41 (s, 3H, -CH<sub>3</sub>); ES-MS (m/z): 427 (M+1); Anal. Calcd. C<sub>25</sub>H<sub>22</sub>N<sub>4</sub>O<sub>3</sub> C, 70.41; H, 5.20; N, 13.14, found C, 70.43; H, 5.30; N, 13.21%.

# General Procedure for the Synthesis of Substituted 1-(5-(2-(p-tolyloxy)quinoline-3-yl)-2-(pyridine-4-yl)-1,3,4oxadiazol-3(2H)-yl)ethanones 5(a-j)

N'-((2-(p-tolyloxy)quinoline-3-yl)methylene)isonicitinohydrazide (0.1 gm, 0.00028 mol) **4a** was dissolved in 10 ml of acetic anhydride. The reaction mixture was heated in an oil bath for 3-4 h at 120 °C and left for overnight. 10 ml cold water was slowly added to the flask and the separated product was filtered and washed with water, dried under vacuum and recrystallized from ethanol.

## 1-(5-(2-(p-tolyloxy)quinolin-3-yl)-2-(pyridin-4-yl)-1,3,4oxadiazol-3(2H)-yl)ethanone (5a)

IR KBr: 1665 (N-CO-CH<sub>3</sub>); 1610 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 2.332 (s, 3H, Ar-CH<sub>3</sub>); 4.621 (s, 3H, N-CO-CH<sub>3</sub>); 6.956 (s; 1H, -CH); 7.115 (d, *J*=8.0 Hz, 2H, Ar-H); 7.244 (d, *J*=8.0 Hz, 2H, Ar-H); 7.623-7.851 (m, 6H, Ar-H); 8.151 (s, 1H, quinoline C<sub>4</sub>-H); 9.022 (s, 2H, pyridine); Mass(m/z): 425 (M+1).

## 1-(5-(2-(p-tolyloxy)-6-methylquinolin-3-yl)-2-(pyridin-4-yl)-1,3,4-oxadiazol-3(2H)-yl) ethanone (5b)

IR KBr: 1688 (N-CO-CH<sub>3</sub>); 1593 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  2.344 (s, 3H, Ar-CH<sub>3</sub>); 2.275 (s, 3H, Ar-CH<sub>3</sub>); 3.829 (s, 3H, NCOCH<sub>3</sub>); 6.945 (s, 1H, -CH); 7.107-7.166 (m, 3H, Ar-H); 7.275 (d, *J*=8.4 Hz, 2H, Ar-H); 7.861 (dd, *J*=4.6 & 6.0 Hz, 3H, Ar-H); 8.026 (d, *J*=8.8 Hz, 1H, Ar-H); 8.772-8.821(d, 2H, pyridine); 8.962(s, 1H, quinoline C<sub>4</sub>-H); Mass(m/z): 439.0(M+1).

## 1-(5-(2-(p-tolyloxy)-7-methylquinolin-3-yl)-2-(pyridin-4-yl)-1,3,4-oxadiazol-3(2H)-yl)ethanone (5c)

IR KBr: 1649 (N-CO-CH<sub>3</sub>); 1556 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  2.154 (s, 3H, Ar-CH<sub>3</sub>); 2.165 (s, 3H, Ar-CH<sub>3</sub>); 3.984 (s, 3H, NCOCH<sub>3</sub>); 6.546 (s, 1H, -CH); 7.166-7.245 (m, 4H, Ar-H); 7.315 (d, *J*=8.4 Hz, 1H, Ar-H); 7.748 (dd, *J*=3.4 & 7.2 Hz, 3H, Ar-H); 8.126 (d, *J*=7.8 Hz, 1H, Ar-H); 8.845-8.875 (d, 2H, pyridine); 8.987 (s, 1H, quinoline C<sub>4</sub>-H); Mass(m/z): 439.3(M+1).

## 1-(5-(2-(p-tolyloxy)-7-methoxyquinolin-3-yl)-2-(pyridin-4-yl)-1,3,4-oxadiazol-3(2H)-yl)ethanone (5g)

IR KBr: 1687 (N-CO-CH<sub>3</sub>); 1510 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  2.54 (s, 3H, Ar-CH<sub>3</sub>); 3.651 (s, 3H, Ar-OCH<sub>3</sub>); 3.914 (s, 3H, NCOCH<sub>3</sub>); 6.646 (s, 1H, -CH); 6.924-7.145 (m, 3H, Ar-H); 7.245 (d, *J*=8.4 Hz, 2H, Ar-H); 7.658 (dd, *J*=2.6 & 6.8 Hz, 3H, Ar-H); 8.646 (d, *J*=8.8 Hz, 1H, Ar-H); 8.845-8.895 (d, 2H, pyridine); 8.912 (s, 1H, quinoline C<sub>4</sub>-H); Mass(m/z): 455.1(M+1).

# 1-(5-(2-(p-tolyloxy)-7-ethoxyquinolin-3-yl)-2-(pyridin-4-yl)-1,3,4-oxadiazol-3(2H)-yl)ethanone (5h)

IR KBr: 1667 (N-CO-CH<sub>3</sub>); 1556 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, DMSO): δ 2.464 (s, 3H, Ar-CH<sub>3</sub>); 2.563 (t,

3H,CH<sub>2</sub><u>CH<sub>3</sub></u>); 3.456 (q, 2H, OCH<sub>2</sub>); 3.814 (s, 3H, NCOCH<sub>3</sub>); 6.512 (s, 1H, -CH); 6.958-7.165 (m, 3H, Ar-H); 7.145 (d, *J*=7.2 Hz, 2H, Ar-H); 7.456 (dd, *J*=4.2 & 7.6 Hz, 3H, Ar-H); 8.245 (d, *J*=8.2 Hz, 1H, Ar-H); 8.312 (s, 1H, quinoline C<sub>4</sub>-H); 8.614-8.814 (d, 2H, pyridine); Mass(m/z): 469.1 (M+1).

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### REFERENCES

- (a) MethCohn, O.; Narine, B.; Tarnowski, B.; Hayes, R.; Keyzad, A.; Rhouti, S.; Robinson, A. *J. Chem. Soc. Perkin Trans.* **1981**, *1*, 2509 (b) Bhaduri, A. P. *Synlett.* **1990**, 557.
- (a) Mccormick, J. L.; Mckee, T. C.; Cardellina, J. H.; Boyd, M. R. *J. Nat. Prod.* **1996**, *59*, 469 (b) Chen, I. S.; Chen, H. F.; Cheng, M. J.; Chang, Y. L.; Teng, C. M.; Tsutomu, I.; Chen, J. J.; Tsai, I. L. *J. Nat. Prod.* **2001**, *64*, 1143 (c) Nadaraj, V.; Selvi, S. T.; Sasi, R. *Arkivoc.* **2006**, *x*, 82.
- 3. Craig, J. C.; Person, P. E. J. Med. Chem. 1971, 14, 1221.
- Dillard, R. D.; Pavey, D. E.; Benslay, D. N. J. Med Chem. 1973, 16, 251.
- Sukhova, N. M.; Lidak, M.; Zidermane, A.; Pelevina, I. S.; Voronia, S. S. *Khim. Farm. Zh.* **1989**, *23*, 1226.
- Atwell, G. J.; Bangaley, B. C.; Denny, W. A. J. Med. Chem. 1989, 32, 396.
- Patel, H. V.; Vyas, K. V.; Fernandes, P. S. Ind. J. Chem. 1990, 29(B), 836.
- Saito, I.; Sando, S.; Nakatani, K. *Bioorg. Med. Chem.* 2001, 9, 2381.
- Andries, K.; Verhasselt, P.; Guillemont, J.; Goehlmann, H. W. H.; Neefs, J. M.; Winkler, H.; Van Gestel, J.; Timmerman, P.; Zhu, M.; Lee, E.; Williams, P.; De Chaffoy, D.; Huitric, E.; Hoffner, S.; Cambau, E.; Truffot-Pernot, C.; Lounis, N.; Jarlier, V. A. *Science*, **2005**, *307*, 223.
- Joseph, J. P.; Harry, L. Y. U.S.Patent 3,141,022, 1961; *Chem. Abstr.* 1964, *61*, 8317b.
- 11. Hokfelt, B.; Jonsson, A. J. Med. Chem. 1962, 5, 247.
- Ansel, P. S. U.S.Patent 2,883,391, 1959. Chem. Abstr. 1959, 53, 16157g.
- 13. Hiroshi, K.; Isaq, H.; Shigeki, O.; Zassokenkyn 1969, 8, 46; Chem. Abstr. 1970, 73, 108544b.
- 14. Derappe, C.; Rips, R.; Albert, O. Aurousseau, M. Chim. Ther. 1968, 3, 181; Chem. Abstr. 1968, 69, 106626y.
- 15. Deshmukh, A. A.; Sattur, P. B.; Sheth, U. K. Indian. J. Exp. Biol. 1976, 4, 166.

- SenGupta, A. K.; Garg, M.; Chandra, U. J. Indian Chem. Soc. 1979, 56, 1230.
- Chiyomaru, I.; Takita, K.; Ito, H.; Kumiai Chem. Ind. Co. Ltd. Jap. Pat. 1972, 72 07, 549; Chem. Abstr. 1972, 77, 549.
- O Neal, J. B.; Rosen, H.; Russel, P. B.; Adams, A. C.; Blumenthal, A. J. Med. Pharm. Chem. 1962, 5, 617; Chem. Abstr. 1962, 57, 9168c.
- 19. Kurzer, F. Org. Compd. Sulphur, Selenium, Tellurium, 1974, 4, 417.
- 20. Thomas, J. Ger. Pat. 2,403, 357/1974, Chem. Abstr. 1974, 81, 136153g.
- 21. Yale, H. L.; Losee, K. J. Med. Chem. 1966, 9, 478.
- 22. Turner, S. R. Colman Products Ltd. Ger. Pat. 1978, 2, 727, 146; Chem. Abstr. 1978, 88, 105357s.
- 23. Hodogaya, Chemical Co. Ltd., Jap. Pat. 1980, 80, 27024; *Chem. Abstr.* **1980**, *93*, 232719q.
- 24. Hakko Chem. Ind. Co. Ltd., Brit. Pat. 1,266,542/1972; *Chem. Abstr.* 1972, 77, 5474g.
- 25. Singh, H.; Yadav, L. D. S. Agric. Biol. Chem. 1976, 40, 759.
- Misato, T.; Ko, K.; Honma, Y.; Konno, K.; Taniyama, E. Inst. Phys. Chem. Res. Jap. Pat. 1977, 772508; *Chem. Abstr.* 1977, 87, 147054.

- Imramovsky; S. Polanc; J. Vinsova; M. Kocevar; J. Jampýlek;
  Z. Reckova; J. Kaustova; *Bioorg. Med. Chem.* 2007, 15, 2551.
- H. Marrakchi; G. Laneelle; A. Quemard, *Microbiology* 2000, 146, 296.
- 29. (a) Joshi, R. S.; Mandhane, P. G.; Diwakar, S. D.; Gill, C. H., Ultrason. Sonochem. 2010, 17, 298 (b) Joshi, R. S., Mandhane, P. G., Badadhe, P. V., Gill, C. H.; Ultrason. Sonochem. 2011, 18, 735 (c) Mandhane, P. G.; Joshi, R. S.; Nagargoje, D. R.; Gill, C. H.; Tett. Letts. 2010, 51, 1490 (d) Joshi, R. S.; Mandhane, P. G.; Dabhade S. K.; Gill, C. H. J. Chin. Chem. Soc. 2010, 57, 1227 (e) Joshi, R. S.; Mandhane, P. G.; Shaikh, M. U.; Kale, R. P.; Gill, C. H. Chin. Chem. Lett. 2010, 21, 432.
- (a) Joshi, R. S.; Mandhane, P. G.; Diwakar, S. D.; Dabhade, S. K. Gill, C. H., *Bioorg. Med. Chem. Letts.* 2010, 20, 3721 (b) Joshi, R. S., Mandhane, P. G., Khan, W., Gill, C. H. *Bull. Korean Chem. Soc.* 2010, 31, 2341. (c) Joshi, R. S., Mandhane, P. G., Dabhade, S. K., Gill, C. H.; *Green Chem. Lett. Rev.* 2010, 3(3), 191 (d) Joshi, R. S.; Mandhane, P. G.; Khan, W.; Gill, C. H. *J. Het. Chem.* DOI 10.1002/jhet.653.
- Christine, H. F.; Michael, H. C. Antimicrob. Agents Chemother. 1986, 29, 386.