

Molecular Iodine: A Versatile Catalyst for the Synthesis of 4-Aryl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-diones in Water

Liqiang Wu,* Limin Yang, Fulin Yan, Chunguang Yang, and Lizhen Fang

School of Pharmacy, Xinxiang Medical University, Xinxiang, Henan 453003, P. R. China. *E-mail: wliq1974@sohu.com
Received November 11, 2009, Accepted February 6, 2010

Key Words: Benzo[*h*]pyrazolo[3,4-*b*]quinoline, Molecular Iodine, 3-Methyl-1-phenyl-1*H*-pyrazol-5-amine, 2-Hydroxynaphthalene-1,4-dione

Multicomponent reactions (MCRs) have attracted considerable attention since they are performed without need to isolate any intermediate during their processes; may reduce time and save both energy and raw materials.¹ They have merits over two-component reactions in several aspects including the simplicity of a one-pot procedure, possible structural variations and building up complex molecules.

Pyrazolo[3,4-*b*]quinoline derivatives are used as pharmaceutical agents,² as inhibitors of oncogenic Ras,³ and as a dopant in the multiplayer OLED fabrication.⁴ In the past several decades, three general strategies for the synthesis of pyrazolo[3,4-*b*]quinolines have been developed: 1) by the Friedlander condensation reaction of 2-aminobenzophenones and pyrazolin-5-ones.⁵ Availability of 2-aminobenzophenones limits the range of applicability of this reaction; 2) by cyclization of 4-arylidene-pyrazolin-5-ones with anilines⁶ or 5-*N*-arylpiprazoles with aromatic aldehydes.⁷ The method is complicated and has a lower yield; 3) by a three-component one-pot reaction of aromatic aldehydes, 5-amino-3-methyl-1-phenylpyrazole and dimedone under thermal⁸ or microwave condition.⁹

Organic reactions in water have become an important research area. Many reactions have been accomplished in aqueous medium.¹⁰ Water has therefore become an attractive medium for many organic reactions, not only for the advantages concerning the avoidance of expensive, catalysts and solvents, but also for some unique reactivity and selectivity. In recent years, the use of molecular iodine in organic synthesis has received considerable attention due to powerful catalytic activity for various organic transformations.¹¹ We now report a highly efficient procedure for the preparation of 4-aryl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-diones using I₂ as an efficient catalyst in water (Scheme 1).

Initial study was performed by the one-pot reaction of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (1 mmol), benzaldehyde

(1 mmol) and 2-hydroxynaphthalene-1,4-dione (1 mmol) in water in the presence of 10 mol % I₂ at reflux temperature. To our delight, we observed the formation of 4-phenyl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-dione. Complete conversion and 92% isolated yield was obtained after 5 hours.

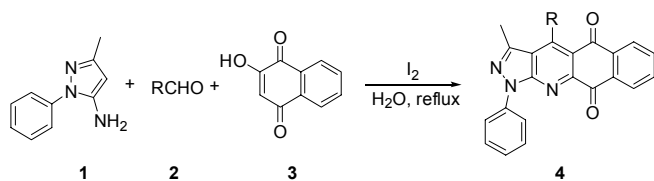
To optimize the amount of catalyst and the reaction temperature, the reaction of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (1 mmol), benzaldehyde (1 mmol) and 2-hydroxynaphthalene-1,4-dione (1 mmol) in water was selected as a model. The best result was obtained by carrying out the reaction using 10 mol % I₂ at reflux temperature.

Based on the optimized reaction conditions, a range of 4-aryl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-diones (**4**) was synthesized by the reaction of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (**1**, 1 mmol) with arylaldehydes (**2**, 1 mmol) and 2-hydroxynaphthalene-1,4-dione (**3**, 1 mmol) in H₂O. The reaction proceeded at reflux temperature within 7 h in excellent yields after the addition of the catalyst I₂ (10 mol %) (Table 2). In addition, we noticed also that when this reaction was carried out with aliphatic aldehyde such as butanal or pentanal, TLC and ¹H NMR spectra of the reaction mixture showed a combination of starting materials and numerous products, the yield of the expected product was very poor.

Table 1. Optimization of catalyst loading and reaction temperature of one-pot synthesis of 4-phenyl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-dione^a

Entry	I ₂ (mol %)	Time (h)	Temp. (°C)	Yield (%) ^b
1	0	10	reflux	26
2	5	8	80	62
3	5	8	90	67
4	5	8	reflux	72
5	10	10	r.t.	56
6	10	5	50	62
7	10	5	80	75
8	10	5	90	86
9	10	5	reflux	92
10	15	5	90	86
11	15	5	reflux	90
12	20	5	reflux	91

^aReaction conditions: 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (1 mmol); benzaldehyde (1 mmol); 2-hydroxynaphthalene-1,4-dione (1 mmol); H₂O (10 mL). ^bIsolated yield.



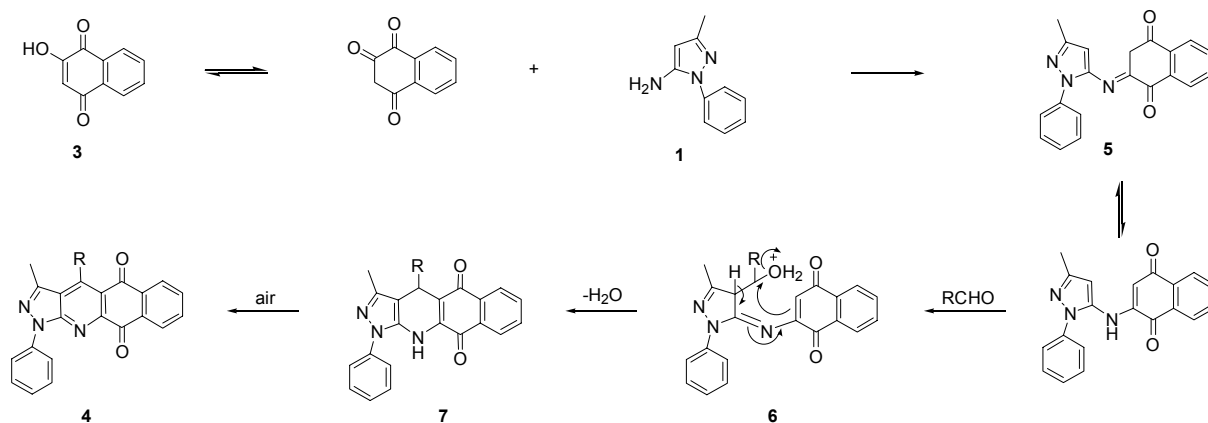
Scheme 1

In general, iodine is reaction with water to afford strong acid. Therefore the following reaction mechanism is suggested. First, the condensation of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine **1** and 2-hydroxynaphthalene-1,4-dione **3** gave the intermediate

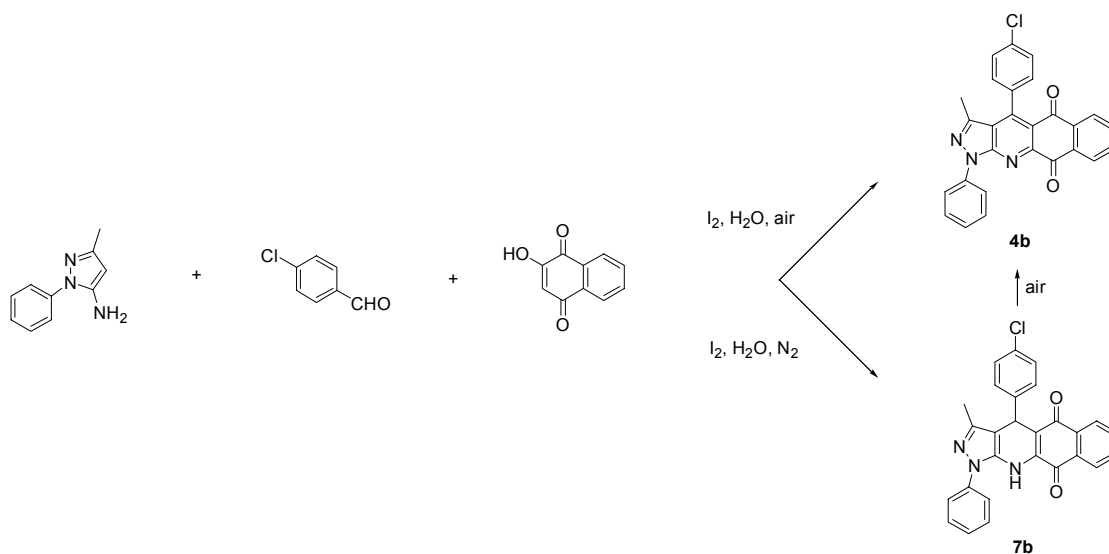
Table 2. Synthesis of 4-aryl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-diones^a

Entry	R	Time (h)	Product	Yield (%) ^b
1	C ₆ H ₅	5	4a	92
2	4-Cl-C ₆ H ₄	6	4b	93
3	4-MeO-C ₆ H ₄	5	4c	90
4	4-Me-C ₆ H ₄	5	4d	88
5	4-NO ₂ -C ₆ H ₄	6	4e	91
6	4-F-C ₆ H ₄	6	4f	94
7	3-NO ₂ -C ₆ H ₄	7	4g	87
8	2-Cl-C ₆ H ₄	5	4h	85
9	3,4-Cl ₂ -C ₆ H ₃	7	4i	89

^aReaction conditions: 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (1 mmol); arylaldehyde (1 mmol); 2-hydroxynaphthalene-1,4-dione (1 mmol); I₂ (10 mol %); H₂O (10 mL); reflux. ^bIsolated yield.



Scheme 2



Scheme 3

product **5**. The addition of **5** to aldehyde **2** then furnished the intermediate product **6**, which on intermolecular cyclization and dehydration gave rise to **7**. In the last step, the intermediate-product **7** aromatized by air-oxidation to product **4**.

In order to determine air-oxidation proceeded in the reaction, multicomponent condensation reaction of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine, 4-chlorobenzaldehyde and 2-hydroxynaphthalene-1,4-dione in the presence of molecular iodine was selected as a model. Compound **7b** was a principal product in nitrogen atmosphere. The compound **7b** was easily oxidized by air bubbling, forming the fully aromatic product **4b** (Scheme 3).

In summary, an efficient methodology for the synthesis of 4-aryl-3-methyl-1-phenyl-1*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,10-diones has been developed. To our best knowledge, this is the first report for the synthesis of these compounds by multicomponent condensation of 3-methyl-1-phenyl-1*H*-pyrazol-5-amine, aldehydes and 2-hydroxynaphthalene-1,4-dione in the presence of molecular iodine as a catalyst in water. The simple experimental procedure, utilization of an inexpensive and readily available catalyst, and excellent yields are the advantages of the present method.

Experimental Section

4-Aryl-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-diones (4a-4i). A mixture of 3-methyl-1-phenyl-1H-pyrazol-5-amine (1 mmol), aldehyde (1 mmol), 2-hydroxynaphthalene-1,4-dione (1 mmol) and I₂ (0.1 mmol) in H₂O (10 mL) was heated at reflux temperature for the appropriate time. The reaction was monitored by TLC. After completion, the mixture was treated with aqueous Na₂S₂O₃ solution, extracted with CH₂Cl₂ (2 × 10 mL). The extract was dried over sodium sulfate, filtered and solvent was evaporated in vacuo. Products **4** were purified by recrystallizing from ethanol.

4-Phenyl-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4a): Yellow crystals, mp 266 ~ 267 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.82 (d, 1H, *J* = 8.0 Hz), 8.31 (d, 2H, *J* = 7.6 Hz), 8.16 (d, 1H, *J* = 7.6 Hz), 7.88-7.84 (m, 1H), 7.64-7.52 (m, 6H), 7.43-7.39 (m, 1H), 7.32-7.30 (m, 2H), 1.95 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 180.5, 179.9, 153.7, 152.2, 150.5, 146.5, 138.6, 137.3, 136.0, 135.9, 131.7, 131.3, 129.2, 129.1, 128.4, 127.2, 127.0, 126.5, 121.3, 119.8, 117.0, 14.3; Anal. calcd for C₂₇H₁₇N₃O₂: C 78.06, H 4.12, N 10.11; found: C 78.19, H 4.06, N 10.08.

4-(*p*-Chloro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4b): Yellow crystals, mp 243 ~ 245 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.80 (d, 1H, *J* = 8.0 Hz), 8.30 (d, 2H, *J* = 8.0 Hz), 8.15 (d, 1H, *J* = 7.6 Hz), 7.87-7.83 (m, 1H), 7.64-7.51 (m, 5H), 7.43-7.40 (m, 1H), 7.28-7.24 (m, 2H), 2.00 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 180.3, 179.9, 153.8, 150.8, 150.5, 146.2, 138.5, 137.1, 136.0, 134.6, 134.3, 131.6, 131.4, 129.3, 129.2, 129.1, 128.7, 128.5, 127.2, 126.7, 121.4, 121.0, 119.7, 116.8, 14.5; Anal. calcd for C₂₇H₁₆ClN₃O₂: C 72.08, H 3.58, N 9.34; found: C 72.35, H 3.39, N 9.50.

4-(*p*-Methoxy-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4c): Yellow crystals, mp 274 ~ 275 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.82 (d, 1H, *J* = 8.0 Hz), 8.32 (d, 2H, *J* = 8.0 Hz), 8.16 (d, 1H, *J* = 8 Hz), 7.88-7.84 (m, 1H), 7.64-7.60 (m, 3H), 7.43-7.39 (m, 1H), 7.23 (d, 2H, *J* = 8.4 Hz), 7.06 (d, 2H, *J* = 8.4 Hz), 3.93 (s, 3H), 2.03 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 180.8, 180.3, 159.8, 153.8, 152.4, 150.5, 146.6, 138.6, 137.4, 136.0, 131.6, 131.3, 129.2, 129.0, 128.6, 127.7, 127.2, 126.5, 121.4, 120.2, 117.3, 113.9, 55.3, 14.6; Anal. calcd for C₂₈H₁₉N₃O₃: C 75.49, H 4.30, N 9.43; found: C 75.41, H 4.33, N 9.51.

4-(*p*-Methyl-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4d): Yellow crystals, mp 276 ~ 277 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.82 (d, 1H, *J* = 8.0 Hz), 8.31 (d, 2H, *J* = 8.0 Hz), 8.15 (d, 1H, *J* = 7.6 Hz), 7.85 (t, 1H, *J* = 7.8 Hz), 7.63-7.60 (m, 3H), 7.41 (t, 1H, *J* = 7.2 Hz), 7.33 (d, 2H, *J* = 7.6 Hz), 7.18 (d, 2H, *J* = 7.6 Hz), 2.49 (s, 3H), 1.99 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 180.7, 180.1, 153.7, 152.6, 150.5, 146.7, 138.6, 138.2, 137.4, 135.9, 132.8, 131.6, 131.3, 129.2, 129.1, 129.0, 127.2, 126.9, 126.5, 121.4, 120.0, 117.1, 21.5, 14.4; Anal. calcd for C₂₈H₁₉N₃O₂: C 78.31, H 4.46, N 9.78; found: C 78.50, H 4.42, N 9.82.

4-(*p*-Nitro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4e): Yellow crystals, mp 326 ~ 328 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.82 (d, 1H, *J* = 8.0 Hz), 8.42 (d, 2H, *J* = 8.4 Hz), 8.30 (d, 2H, *J* = 8.0 Hz), 8.18

(d, 1H, *J* = 7.6 Hz), 7.88 (t, 1H, *J* = 7.8 Hz), 7.67-7.62 (m, 3H), 7.51 (d, 2H, *J* = 8.8 Hz), 7.44 (t, 1H, *J* = 7.4 Hz), 1.97 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 179.8, 179.6, 153.9, 150.5, 149.2, 147.9, 145.7, 143.1, 138.4, 136.9, 136.2, 131.7, 131.6, 129.4, 129.3, 128.2, 127.3, 126.9, 123.8, 121.5, 119.2, 116.1, 14.5; Anal. calcd for C₂₇H₁₆N₄O₄: C 70.43, H 3.50, N 12.17; found: C 70.48, H 3.38, N 12.25.

4-(*p*-Fluoro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4f): Yellow crystals, mp 282 ~ 283 °C; ¹H NMR (DMSO-*d*₆, 400 MHz) δ 8.77 (d, 1H, *J* = 8.0 Hz), 8.30 (d, 2H, *J* = 7.6 Hz), 8.05 (d, 1H, *J* = 7.2 Hz), 7.98-7.94 (m, 1H), 7.74-7.66 (m, 3H), 7.44-7.36 (m, 5H), 1.90 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 180.5, 180.0, 164.1, 161.6, 153.8, 151.1, 150.2, 146.3, 138.6, 137.2, 136.0, 131.7, 131.6, 131.4, 129.3, 129.2, 129.1, 129.0, 128.9, 127.2, 126.7, 121.4, 119.9, 117.0, 115.7, 115.5, 14.4; Anal. calcd for C₂₇H₁₆N₃O₂: C 74.82, H 3.72, N 9.69; found: C 75.02, H 3.68, N 9.76.

4-(*m*-Nitro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4g): Yellow crystals, mp 288 ~ 289 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.83 (d, 1H, *J* = 8.0 Hz), 8.41 (d, 1H, *J* = 8.4 Hz), 8.30 (d, 2H, *J* = 8.0 Hz), 8.24 (s, 1H), 8.18 (d, 1H, *J* = 7.6 Hz), 7.89 (t, 1H, *J* = 7.2 Hz), 7.75 (t, 1H, *J* = 7.8 Hz), 7.69-7.61 (m, 4H), 7.44 (t, 1H, *J* = 14.8 Hz), 1.97 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 179.9, 179.7, 153.9, 150.6, 148.7, 148.2, 145.6, 138.4, 137.7, 136.9, 136.2, 133.3, 131.7, 131.6, 129.6, 129.3, 129.2, 127.3, 126.9, 123.4, 122.4, 121.5, 119.4, 116.5, 14.6; Anal. calcd for C₂₇H₁₆N₄O₄: C 70.43, H 3.50, N 12.17; found: C 70.54, H 3.32, N 12.21.

4-(*o*-Chloro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4h): Yellow crystals, mp 229 ~ 230 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.84 (d, 1H, *J* = 8.0 Hz), 8.33 (d, 2H, *J* = 8.0 Hz), 8.19 (d, 1H, *J* = 7.6 Hz), 7.89-7.85 (m, 1H), 7.65-7.40 (m, 7H), 7.25-7.23 (m, 1H), 2.00 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 179.9, 179.3, 153.8, 150.8, 148.5, 146.2, 138.6, 137.2, 136.0, 135.1, 131.7, 131.6, 131.4, 129.9, 129.5, 129.3, 129.2, 128.3, 127.2, 127.0, 126.6, 121.4, 119.6, 116.5, 13.6; Anal. calcd for C₂₇H₁₆ClN₃O₂: C 72.08, H 3.58, N 9.34; found: C 72.19, H 3.70, N 9.28.

4-(3,4-Dichloro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (4i): Yellow crystals, mp 269 ~ 270 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.78 (d, 1H, *J* = 8.0 Hz), 8.28 (d, 2H, *J* = 8.0 Hz), 8.15 (d, 1H, *J* = 7.6 Hz), 7.87-7.83 (m, 1H), 7.63-7.59 (m, 4H), 7.43-7.40 (m, 2H), 7.16 (d, 1H, *J* = 8.0 Hz), 2.03 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 179.9, 179.5, 153.7, 150.4, 148.9, 145.9, 138.4, 136.9, 136.1, 135.7, 132.8, 131.5, 130.5, 129.2, 129.0, 127.2, 126.8, 126.6, 121.3, 119.4, 116.5, 14.7; Anal. calcd for C₂₇H₁₅Cl₂N₃O₂: C 66.96, H 3.12, N 8.68; found: C 67.03, H 3.20, N 8.58.

1,4-Dihydrogen-4-(*p*-chloro-phenyl)-3-methyl-1-phenyl-1H-benzo[h]pyrazolo[3,4-b]quinoline-5,10-dione (7b). In nitrogen atmosphere, a mixture of 3-methyl-1-phenyl-1H-pyrazol-5-amine (1 mmol), 4-chlorobenzaldehyde (140 mg, 1 mmol), 2-hydroxynaphthalene-1,4-dione (1 mmol) and I₂ (0.1 mmol) in H₂O (10 mL) was stirred at room temperature for 5 h. The reaction was monitored by TLC. After completion, the mixture was treated with aqueous Na₂S₂O₃ solution, extracted with CH₂Cl₂ (2 × 10 mL). The extract was dried over sodium sulfate, filtered and solvent was evaporated in vacuo. The residue was

purified by silica gel column chromatography using CHCl_3 as eluent to give **7b** as a brownish red solid (360 mg, 80%); mp $252 \sim 253^\circ\text{C}$; $^1\text{H NMR}$ (CDCl_3 , 400 MHz) δ 8.12 (t, 2H, $J = 9.4$ Hz), 7.80-7.32 (m, 10H), 7.06 (t, 2H, $J = 8.4$ Hz), 5.45 (s, 1H), 2.03 (s, 3H); Anal. calcd for $\text{C}_{27}\text{H}_{18}\text{ClN}_3\text{O}_2$: C 71.76, H 4.01, N 9.30; found: C 71.84, H 3.96, N 9.35.

Acknowledgments. We are pleased to acknowledge the financial support from Xinxiang Medical University.

References

1. Devi, I.; Bhuyan, P. J. *Tetrahedron Lett.* **2004**, *45*, 8625.
2. El-Sayed, O. A.; Aboul-Enein, H. Y. *Arch. Pharm.* **2001**, *334*, 117.
3. Wolin, R.; Wang, D.; Kelly, J.; Afonso, A.; James, L.; Kirschmeier, P.; Mcphail, A. T. *Bioorg. Med. Chem. Lett.* **1996**, *6*, 195.
4. He, Z.; Milburn, G. H. W.; Baldwin, K. J.; Smith, D. A.; Danel, A.; Tomasik, P. *J. Lumin.* **2000**, *86*, 1.
5. Tomasik, P.; Tomasik, D.; Abramovitch, R. A. *J. Heterocycl. Chem.* **1983**, *20*, 1539.
6. Chaczatryan, K.; Chaczatryan, G.; Danel, A.; Tomasik, P. *ARKIVOC* **2001**, (vi), 63.
7. Danel, A.; Chaczatryan, K.; Tomasik, P. *ARKIVOC* **2000**, (i), 51.
8. Quiroga, J.; Insuasty, B.; Saitz, C.; Jullian, C. *J. Heterocycl. Chem.* **1998**, *35*, 575.
9. Hua, G. P.; Xu, J. N.; Tu, S. J.; Wang, Q.; Zharlg, J. P.; Zhu, X. T.; Li, T. J.; Zhu, S. L.; Zhang, X. *J. Chin. Org. Chem.* **2005**, *25*, 1610.
10. Lindstrom, U. M. *Chem. Rev.* **2002**, *102*, 2751.
11. (a) Das, B.; Ravikanth, B.; Ramu, R.; Laxminarayana, K.; Rao, B. V. *J. Mol. Catal. A: Chem.* **2006**, *255*, 74. (b) Kidwai, M.; Bansal, V.; Mothsra, P.; Saxena, S.; Somvanshi, R. K.; Dey, S.; Singh, T. *J. Mol. Catal. A: Chem.* **2007**, *268*, 76. (c) Bhosale, R. S.; Magar, C. V.; Solanke, K. S.; Mane, S. B.; Choudhary, S. S.; Pawar, R. P. *Synth. Commun.* **2007**, *37*, 4353.