

백나무로부터 분리된 quercetin-3-O- α -L-rhamnopyranoside의 알도스 환원효소 및 솔비톨 억제효과

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Inhibitory Effect of quercetin-3-O- α -L-rhamnopyranoside from *Chamaecyparis obtuse* on Aldose Reductase and Sorbitol Accumulation

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ABSTRACT : Taxifolin-3-O- β -D-xylopyranoside and quercetin-3-O- α -L-rhamnopyranoside were isolated from an EtOAc-soluble extract of the leaves of *Chamaecyparis obtuse*. Quercetin-3-O- α -L-rhamnopyranoside was found to possess a potent inhibitory activity of human recombinant aldose reductase *in vitro*, its IC₅₀ value being 11.5 μ M. Kinetic analysis showed that quercetin-3-O- α -L-rhamnopyranoside exhibited uncompetitive inhibition against DL-glyceraldehyde. Also, quercetin-3-O- α -L-rhamnopyranoside suppresses sorbitol accumulation in rat lens under high glucose conditions, demonstrating the potential to prevent sorbitol accumulation *in vivo*. These results suggest that this compound may be a promising agent in the prevention or treatment of diabetic complications.

Key Words : *Chamaecyparis obtusa*, Taxifolin-3-O- β -D-xylopyranoside, Quercetin-3-O- α -L-rhamnopyranoside, Diabetic Complications, Human Recombinant Aldose Reductase, Sorbitol Accumulation

INTRODUCTION

Aldose reductase (AR, EC 1.1.1.21) catalyzes the reduction of glucose to the corresponding sugar alcohol, sorbitol, which is subsequently metabolized to fructose by sorbitol dehydrogenase. This conversion of glucose into fructose constitutes the polyol pathway of glucose metabolism (Suzen and Buyukbingol, 2003). Under normal physiological conditions, this pathway plays a minor role in glucose metabolism for most tissues. In hyperglycemia associated with diabetes, however, cells undergoing insulin-independent glucose uptake produce significant quantities of sorbitol because sorbitol cannot penetrate cellular membranes and is metabolized by sorbitol dehydrogenase (Jedziniak *et al.*,

1981). The resulting hyperosmotic stress to cells is believed to be the primary cause for the development of diabetic complications such as retinopathy, cataracts, neuropathy, and nephropathy (Williamson *et al.*, 1992). These observations suggest that the inhibition of AR may be an innovative, potentially direct pharmacological approach toward the treatment of certain diabetic complications (Kador *et al.*, 1985; Jeon *et al.*, 2010).

Plants are a rich source of bioactive chemicals. Because many of the biocompounds are largely free from adverse effects and have excellent pharmacological actions, they could be used in the development of new classes of safer antidiabetic agents with fewer complications. In addition, some flavonoids and polyphenols and their sugar derivatives

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are reported to be effective (Park *et al.*, 2009). Therefore, ongoing research is focusing on the use of plants as potential agents or as lead compounds in commercial AR inhibitors.

Chamaecyparis obtusa Sieb. et Zucc., a genus in the family Cupressaceae, is native to northeast Asia. Previous phytochemical investigations of this plant isolated lignin, sesquiterpene derivatives, beyerane derivatives, and flavonoids from the leaves (Gao *et al.*, 2008; Krauze-Baranowska *et al.*, 2005). *C. obtusa* reportedly exhibits neurite outgrowth-promoting activity in PC12 cells, acaricidal activity against *Dermatophagides* spp., and antiviral against HSV-1 replication (Jang *et al.*, 2005; Kuo *et al.*, 2006; Kuroyanagi *et al.*, 2008). Reportedly, the essential oils have antimicrobial and hair-growth promoting effects (Lee *et al.*, 2010; Yang *et al.*, 2007). However, relatively little research has been done on the anti-diabetic effect of the compounds isolated from *C. obtusa*. In the present study, repeated chromatography of the ethyl acetate-soluble fraction of the leaves of *C. obtusa* led to the isolation of biologically active compound. All isolates obtained in the study were tested for their potential to inhibit human recombinant aldose reductase (hrAR) and sorbitol accumulation in rat lenses.

MATERIALS AND METHODS

1. Chemicals

dl-glyceraldehyde, the reduced form of nicotinamide adenine dinucleotide phosphate (NADPH), quercetin, TL-100 medium, and antibiotics were purchased from Sigma (St. Louis, MO, U.S.A.). Fetal bovine serum was from GenDEPOT (USA). hrAR was purchased from Wako Pure Chemical Industries (Osaka, Japan). Sephadex LH 20 (GE Healthcare Bio-Science AB, Sweden) was used as the column packing material.

2. Plant Materials

The leaves of *C. obtusa* were presented by Department of Forest Biomaterials Engineering, Kangwon National University. A voucher specimen (No. RIC-021) was deposited and maintained at the Herbarium of Regional Innovation Center, Hallym University, Chuncheon, Korea.

3. Extraction, fractionation, and isolation

The leaves of the dried material (2 kg) were extracted

with 70% acetone (10 L × 4) by maceration. The extracts were combined and concentrated *in vacuo* at 40 °C to give an acetone extract (322 g, 16.1%). The extract was suspended in water (2 L), and then successively extracted with *n*-hexane (2 L × 3), dichloromethane (2 L × 3), and ethylacetate (2 L × 3) to give *n*-hexane- (97.43 g), dichloromethane- (2.09 g), ethyl acetate- (66.83 g), and water-soluble extract (156.02 g), respectively. The ethyl acetate-soluble extract showed the greatest inhibitory activity on AR with an IC₅₀ value of 8.52 μg/ml. The ethyl acetate-soluble extract was chromatographed over Sephadex LH-20 column chromatography using a MeOH-H₂O (4 : 1, v/v) to afford 3 pooled fractions (E1-E3). Fraction E2 was further fractionated using Sephadex column chromatography (50% MeOH) to give compound 1 (1.12 g). Fraction E3 was subjected to Sephadex column chromatography with MeOH:H₂O (2 : 1, v/v) as the eluent to yield 3 subfractions (E3:Fr. 1 to 3). Fraction E3:Fr. 3 was chromatographed on Sephadex column with MeOH:H₂O (1 : 2, v/v) to give compound 2 (195 mg). The ¹H- and ¹³C-NMR spectra were recorded on a Bruker DPX 400 spectrometer (Karlsruhe, Germany) at 400 and 100 MHz, respectively. Chemical shifts are given in ppm using tetramethylsilane as an internal standard.

1) Taxifolin-3-O-β-D-xylopyranoside (1)

Amorphous yellow powder; m.p. 190-195 °C; $[\alpha]_D^{20} + 8.5^\circ$ (MeOH; c 0.3); IR (KBr) cm⁻¹: 3430, 2919, 1623, 1377, 1042; UV (MeOH) λ_{max} (ε): 269; Positive FAB MS: $[M+H]^+$ at *m/z* 437.

¹H-NMR (400 MHz, δ, MeOH-*d*₄): 3.06 (2H, *dd*, *J*=8.6 and 12.0 Hz, H-2" and H-5"β), 3.23 (2H, *m*, H-2" and 3"), 3.49 (1H, *m*, H-4"), 3.87 (1H, *d*, *J*=5.5 Hz, H-1"), 3.94 (1H, *dd*, *J*=4.6 and 12.0 Hz, H-5"α), 4.74 (1H, *d*, *J*=9.9 Hz, H-3), 5.19 (1H, *d*, *J*=9.9 Hz, H-2), 5.90 (1H, *d*, *J*=2.0 Hz, H-6), 5.91 (1H, *d*, *J*=2.0 Hz, H-8), 6.80 (2H, *m*, H-5' and H-6'), 6.95 (1H, *d*, *J*=2.0 Hz, H-2'). ¹³C-NMR (100 MHz, δ, MeOH-*d*₄): 65.9 (C-5"), 70.8 (C-3"), 73.4 (C-2"), 75.8 (C-3"), 77.5 (C-3), 83.6 (C-2), 96.4 (C-6), 97.4 (C-8), 102.5 (C-1"), 103.1 (C-10), 115.7 (C-2'), 116.3 (C-5'), 120.8 (C-6'), 129.0 (C-1'), 146.5 (C-3'), 147.4 (C-4'), 164.1 (C-9), 165.5 (C-5), 169.0 (C-7), 195.6 (C-4).

2) Quercetin-3-O-α-L-rhamnopyranoside (2)

Amorphous yellow powder; m.p. 271-272 °C; $[\alpha]_D^{20} -178^\circ$

(MeOH; c 0.1); IR (KBr) cm^{-1} : 3280, 1605, 1662, 1508, 1463, 1063; UV (MeOH) λ_{max} (ϵ): 264, 314, 350; Positive FAB MS: $[\text{M}+\text{H}]^+$ at m/z 449.

$^1\text{H-NMR}$ (400 MHz, δ , MeOH- d_4): 0.94 (3H, *d*, $J=6.0$, H-6"), 3.35 (1H, *m*, H-4"), 3.43 (1H, *m*, H-5"), 3.75 (1H, *dd*, $J=3.3$ and 9.3, H-3"), 4.22 (1H, *br d*, $J=3.0$, H-2"), 5.35 (1H, *d*, $J=1.0$, H-1"), 6.19 (1H, *d*, $J=1.9$, H-6), 6.36 (1H, *d*, $J=1.9$, H-8), 6.85 (1H, *d*, $J=8.2$, H-5'), 7.30 (1H, *dd*, $J=8.2$ and 1.9, H-6'), 7.34 (1H, *d*, $J=1.9$, H-2'). $^{13}\text{C-NMR}$ (100 MHz, δ , MeOH- d_4): 17.7 (C-6"), 71.9 (C-5"), 72.1 (C-3"), 72.2 (C-2"), 73.3 (C-4"), 94.8 (C-8), 99.8 (C-6), 103.7 (C-1"), 105.9 (C-10), 116.4 (C-5'), 116.5 (C-2'), 122.9 (C-6'), 123.0 (C-1'), 136.3 (C-3), 146.4 (C-3'), 149.8 (C-4'), 158.5 (C-9), 159.3 (C-2), 163.2 (C-5), 165.9 (C-7), 179.7 (C-4).

4. Assay for hrAR inhibitory activity

hrAR activities were assayed spectrophotometrically by measuring the decrease in absorption of NADPH at 340 nm over a 5 min period with DL-glyceraldehyde as a substrate. Each 1.0 μl cuvette contained equal units of enzyme, 0.10 M sodium phosphate buffer (pH 6.2), 0.3 mM NADPH, 10 mM substrate, and inhibitor. The concentration of inhibitor giving 50% inhibition of enzyme activity (IC_{50}) was calculated from the least-squares regression line of the logarithmic concentrations plotted against the residual activity (Lee *et al.*, 2008a and b).

5. Determination of inhibition-type of rhAR by active compound

Reaction mixtures consisted of 0.1 M potassium phosphate (pH 7.0), 0.16 mM NADPH and 2 mM of rhAR with varied concentrations of substrate DL-glyceraldehyde in a total volume of 200 μl . Concentrations were ranged from 0.02 to 0.2 mM for DL-glyceraldehyde. rhAR activity was assayed spectrophotometrically by measuring the decrease in absorption of NADPH at 340 nm after substrate addition using BioTek Power Wave XS spectrophotometer (Lee *et al.*, 2008a and b).

6. Lens culture and intracellular sorbitol measurement

Lenses isolated from 10-week-old male Wistar rats were cultured for 6 days in TC-199 medium that contained 15% fetal bovine serum 100 units/ml penicillin and 0.1 mg/ml streptomycin under sterile conditions in an atmosphere of 5% CO_2 and 95% air at 37°C. Samples were dissolved in

dimethylsulfoxide. The lenses were divided into 3 groups and cultured in medium containing 25.5 mM glucose and sample. Each lens was placed in a well containing 2.0 ml of medium. Sorbitol was determined by HPLC after its derivatization by reaction with benzoic acid to a fluorescent compound (Kwang-Hyok *et al.*, 2005; Lee *et al.*, 2010).

7. Benzoylation

Benzoylation of processed samples and standard sugar/sugar alcohols was performed according to the method of Shinohara with slight modification (Shinohara *et al.*, 1998). Briefly, to 70 μl of the sample, 20 μl of KH_2PO_4 (1 mol/L) solution, 3 μl benzoyl chloride and 15 μl of 8 mol/L NaOH were added in a 2.0 ml polypropylene centrifugation tube. The mixture was immediately vortex mixed for 5 min at 2500 vibrations/min in a vortex-mixer (REMI, India). Then the mixture was neutralized with 10 μl of 1.4 mol/L H_3PO_4 and after the addition of 100 μl ethyl acetate vortex mixed for an additional minute at the same speed. Finally, 75 μl of ethyl acetate phase was taken in a 1.5 ml polypropylene centrifugation tube and evaporated for 20 min in N_2 gas. The residue thus obtained was dissolved in 25 μl of acetonitrile-water (70:30, v/v) mixture. 20 μl of this sample was analyzed by HPLC.

8. Chromatographic analysis

The sample were analyzed using a Dionex HPLC System equipped with an autosampler (ASI 100), a column oven (STH 585), a UV detector (UVD 170S), quaternary pumps (P580), and Chromelone software version 4.1. The separation was achieved on a 250 4.6 mm i.d., 4 μm J'Sphoro ODS-H80 (YMC, Japan). The elution solvents were water (A) and acetonitrile (B) with the following gradient: 70% to 80% B from 0 to 5 min, 80% to 83% B from 5 to 30 min, and 83% to 100% B from 30 to 33 min, isocratic at 100% from 33 to 38 min, 100% to 70% B from 38 to 40 min, and isocratic at 70% B from 40 to 45 min to equilibrate the column for the next injection. UV chromatograms were monitored at 228 nm.

9. Statistical analysis

Values are expressed as means \pm S.D of the results of three independent experiments. Between group differences were analyzed by Student's *t*-test. The level of significance was set at less than 5%.

Table 1. Inhibitory effects of the extract and its fractions from the leaves of *C. obtusa* on hrAR.

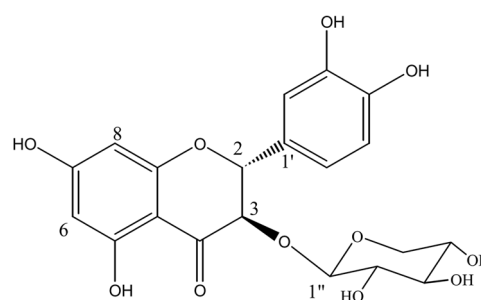
Extract and Fractions (fr.)	Concentration ($\mu\text{g}/\text{mL}$)	Inhibition (%)	IC ₅₀ ($\mu\text{g}/\text{mL}$)
Acetone extract	20	85.56 ± 0.05	11.22 ± 0.05
	10	49.22 ± 0.06	
	5	26.01 ± 0.08	
	1	11.20 ± 0.04	
<i>n</i> -Hexane fr.	10	12.85 ± 0.05	8.52 ± 0.03
Dichloromethane fr.	10	27.07 ± 0.06	
Ethyl acetate fr.	10	54.83 ± 0.07	
	5	38.53 ± 0.05	
	1	23.61 ± 0.03	
Water fr.	10	2.23 ± 0.06	

RESULTS AND DISCUSSION

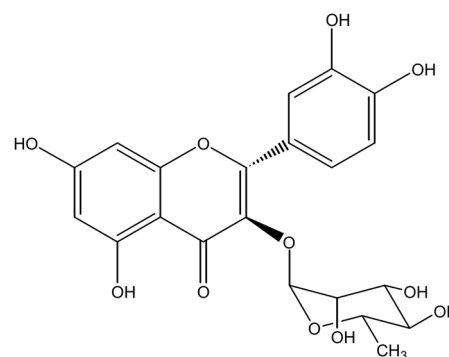
In our attempt to identify powerful, nontoxic and natural AR inhibitors (ARIs), the leaves of *C. obtusa* were evaluated for potential inhibitory effects on rhAR. In this study, the acetone extract of *C. obtusa* leaves was found to exhibit AR inhibitory activity with an IC₅₀ value of 10.25 $\mu\text{g}/\text{mL}$. The acetone extract was further partitioned successively with methylene chloride, ethylacetate, and water. As shown in Table 1, the ethylacetate fraction exhibited strong rhAR inhibitory activity, with an IC₅₀ value of 8.52 $\mu\text{g}/\text{mL}$. This finding suggested that this fraction was likely to contain many ARIs.

Therefore, this study focused specifically on the isolation of ARI compounds within this fraction. The ethyl acetate-soluble portion was subjected to repeated Sephadex LH-20 chromatography and resulted in the isolation of two compounds, quercetin-3-*O*- α -L-rhamnopyranoside and taxifolin-3-*O*- β -D-xylopyranoside. These were identified on the basis of the 1D (¹H- and ¹³C-NMR) and 2D NMR (HMOC and HMBC) spectral data and on comparison with published spectral data and/or by direct comparison with actual samples (Ishiguro *et al.*, 1991a and b; Baderschneider and Winterhalter, 2001). Their chemical structures are shown in Fig. 1.

Quercetin-3-*O*- α -L-rhamnopyranoside exhibited strong hrAR inhibitory activity with an IC₅₀ value of 11.5 μM (Table 2). However, taxifolin-3-*O*- β -D-xylopyranoside showed no inhibitory effect. Previous study demonstrated the possible relationships of structure to the inhibitory activities of flavonoids. It was reported that the glycoside of a flavonol has a higher or lower inhibitory activity than its parent aglycone, depending upon the nature of the glycosylation sugar at 3-



Taxifolin-3-*O*- β -D-xylopyranoside (1)



Quercetin-3-*O*- α -L-rhamnopyranoside (2)

Fig. 1. Chemical structures of compounds isolated from the leaves of *C. obtusa*.

Table 2. Inhibitory effects of compounds isolated from *C. obtusa* on hrAR.

Compounds	Inhibition (IC ₅₀ , μmol)
Taxifolin-3- <i>O</i> - β -D-xylopyranoside (1)	187.7 ± 0.08
Quercetin-3- <i>O</i> - α -L-rhamnopyranoside (2)	11.5 ± 0.05
Quercetin	15.9 ± 0.06

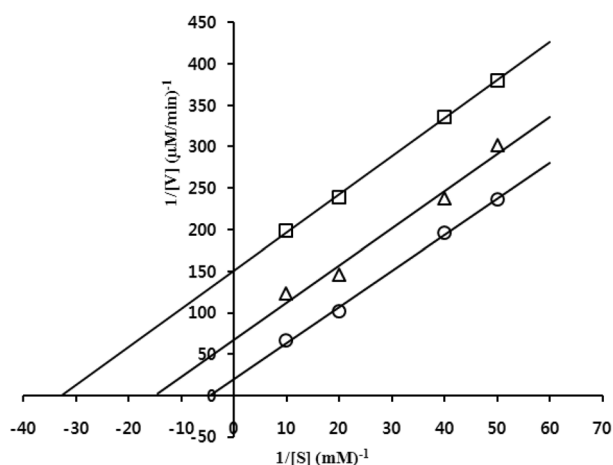


Fig. 2. Inhibitory effect of quercetin-3-O-α-L-rhamnopyranoside on hrAR. Lineweaver-Burk plots in the absence (○) and in the presence (△, 11 μM; □, 22 μM) of quercetin-3-O-α-L-rhamnopyranoside.

position (Matsuda *et al.*, 2002). For example, quercetin-3-O-α-L-rhamnopyranoside has a higher activity than its aglycone quercetin. On the other hand, the glucose and galactose of quercetin, hyperin and isoquercitrin are less active than the aglycone. Also, taxifolin is a flavanone with the same hydroxylation pattern of quercetin but its C ring lacks the 2,3-double bond. Toxicofolin, the aglycone of astilbin, has been reported to be an inhibitor of AR, which showed less activity than its rhamnoside (Haraguchi *et al.*, 1997). From this finding, the substitution of a 3-xylosyl moiety to toxicofolin was found to decrease inhibitory activity against AR. Therefore, inhibitory effect of *C. obtusa* on AR may be due to quercetin-3-O-α-L-rhamnopyranoside, one of major constituents of this plant.

To determine the inhibition type for quercetin-3-O-α-L-rhamnopyranoside, we conducted a kinetic study using DL-glyceraldehyde as the substrate at a concentration of 0.2-2 mM. A kinetic analysis of AR inhibition by quercetin-3-O-α-L-rhamnopyranoside using Lineweaver-Burk plots of 1/velocity vs 1/concentration of the substrate is shown in Fig. 2.

When the concentration of the substrate dl-glyceraldehyde was changed, the slopes obtained for the uninhibited enzyme and two different concentrations of quercetin-3-O-α-L-rhamnopyranoside were parallel. These results show that rhAR inhibition by quercetin-3-O-α-L-rhamnopyranoside was uncompetitive, meaning that the inhibitor was unable to bind to either the substrate binding region or the NADPH binding

Table 3. Inhibitory effect of quercetin-3-O-α-L-rhamnopyranoside on sorbitol accumulation in rat lens.

Compounds	Sorbitol content† (n/ml)
Blank (Glucose free)	10.1±0.03
Control	84.1±0.05
Quercetin-3-O-α-L-rhamnopyranoside (2)	35.2±0.06*
Quercetin	49.2±0.04*

†Sorbitol was determined by HPLC after its derivatization by reaction with benzoic acid to a fluorescent compound.

*Significantly different from control group ($p < 0.01$)

region of rhAR.

We also investigated the inhibitory activities of quercetin-3-O-α-L-rhamnopyranoside on sorbitol accumulation in rat lenses. Lenses of experimentally diabetic animals contain excessive amounts of sorbitol and fructose. Sorbitol is synthesized by the action of AR on glucose and the reducing equivalents are derived from NADPH. Fructose is derived by the action of sorbitol dehydrogenase and NAD is the electron acceptor. Excessive accumulation of sorbitol and fructose in the lens fibers and epithelium is considered osmotically traumatic and leads to tissue swelling and opacity (Sakushima *et al.*, 2002; Srivastava *et al.*, 2005). Because AR catalyzes the first reaction in the polyol pathway, it is considered the key initiator of cataractogenesis in diabetes and galactosemia. Thus, we investigated the effect of quercetin-3-O-α-L-rhamnopyranoside on sorbitol in rat lenses. The effects of quercetin-3-O-α-L-rhamnopyranoside and a positive control, quercetin, against AR on sorbitol accumulation in rat lenses are shown in Table 3.

Sorbitol accumulation was 8-fold greater when cells were incubated in a high-glucose medium than in a glucose-free medium. Quercetin-3-O-α-L-rhamnopyranoside inhibited sorbitol accumulation by 58.3% at 50 μM. These results suggested that quercetin-3-O-α-L-rhamnopyranoside could prevent the accumulation of sorbitol in rat lenses. On the basis of its inhibition of AR, we conclude that quercetin-3-O-α-L-rhamnopyranoside has therapeutic potential for preventing and treating diabetic complications, although further clinical research is needed.

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