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# Physicochemical Properties of Job's Tears (*Coix lachryma-jobi* L.) Starch Modified with Different Levels of Acid Hydrolysis

Hye Won Kim, Kwang Yeon Lee, In Young Bae, Soo Jin Jun, Ji Yeon Lee, and Hyeon Gyu Lee\* Department of Food Science and Nutrition, Hanyang University, Seoul 133-791, Korea

**Abstract** Physicochemical properties of native and acid-modified Job's tears (*Coix lachryma-jobi* L.) starches were investigated. Starch extracted from Job's tears was treated with 2.2 N hydrochloric acid for different length of time (3, 6, 12, and 18 hr). The hydrolysis pattern of starches with the acid proceeded rapidly up to 12 hr and then the approached constant values. The swelling power of acid-modified starches measured at all temperatures was lower than that of its native counterparts and the water solubility index increased as temperature and hydrolysis time increased. Rapid visco analyzer viscograms of acid-modified starches demonstrated a very low viscosity as compared with that of native starch. However, X-ray diffraction did not show any significant alteration in the crystallinity after acid-modification.

Keywords: Job's tears starch, acid-modified starch, physicochemical property, pasting property

#### Introduction

Carbohydrates constitute the principal component of cereals, legumes, tubers, and unripe fruits, accounting for up to 40-80% of dry matter (1). Starch is the main carbohydrate reserve in these plant sources and contributes to the appearance, structure, and quality of food products (2). However, native starches have many limitations such as low shear resistance and low cold storage stability that reduce its usefulness under processing conditions (e.g., temperature, pH, and shear) (3). Therefore, native starches have been modified by chemical, physical, and enzymatic methods to overcome the shortcomings and obtain desired attributes such as solubility, texture adhesion, dispersion, and heat tolerance (4,5). Among these methods, a chemical modification includes different chemical reaction: acid hydrolysis, oxidation, etherification, esterification, and crosslinking (6).

Acid hydrolysis changes the structure of starch granules and produces soluble materials (7). The typical procedure to manufacture acid-thinned starch is related to the treatment of treating concentrated starch slurry (36-40% solids) with mineral acid at a temperature below gelatinization temperature of the starch (40-60°C) for specific periods depending on the desired viscosity or degree of conversion (3,8). This hydrolysis process depends on the reaction condition such as reactant consistency, acid concentration, hydrolysis time, and temperature (9). Furthermore, it has advantages over other chemical methods, specially altering the physicochemical properties of starches without significantly destroying its granule shape (10).

Coix lachryma-jobi L. is known as well as Job's tears or adlay and a grain crop that mainly planted in Asia. The grain of Job's tears is pear-shaped, about 5 mm in diameter, and its protein content was similar to wheat starch. It has

long been used as a nourishing food and is used in food such as soup and beverage in Asia (11-13). Recently, it has been reported that Job's tears grains may have physiological effects such as antitumor activity (14) and anti-obesity (15). Although, starches separated from Job's tears have been studied for its physicochemical properties, there are no reports regarding the effect of the degree of acid hydrolysis on the structural and physicochemical properties of Job's tears starches.

In this study, Job's tears starches were subjected to acid hydrolysis under different conditions. Then, the physicochemical and structural properties of the resulting products were evaluated as a function of the degree of hydrolysis.

#### Materials and Methods

**Materials** Job's tears (*Coix lacryma Jobi* L.) used in this study were purchased at a grocery store and Job's tears starch was extracted according to the method of Li and Corke (11), and Ramirez (16). Flour of Job's tears was screened through a sieve of 100 mesh and was steeped in 0.45% Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub> in a water bath at 40°C for 48 hr. The material was reground and rescreened several times through a 45 mesh openings. To remove protein and impurities, the residues were extracted with 0.05 M NaCl and 20% toluene. The residue was washed twice with distilled water and once with acetone, followed by drying at 40°C for 24 hr. After dried, the residue was ground and passed through 100 mesh sieve.

Acid hydrolysis Starches were suspended in 2.2 N HCl (15.0 g starch/120 mL acid) at 35°C for various periods of time (3, 6, 12, and 18 hr). The suspension was placed in a shaking incubator at 120 rpm. After hydrolysis, starch slurries were neutralized with 2.2 N NaOH and filtered through suction (Whatman filter #2). Carbohydrates content in the filtered liquid was assayed by phenol sulfuric acid method. The residues were washed 3 times with distilled water and dried at 60°C for 24 hr. The extent of hydrolysis was determined by expressing the solubilized

\*Corresponding author: Tel: +82-2-2220-1202; Fax: +82-2-2292-1226 E-mail: hyeonlee@hanyang.ac.kr Received February 17, 2009; Revised May 11, 2009; Accepted May 27, 2009 1146 *H. W. Kim et al.* 

carbohydrates as a percentage of the initial starch (7).

Water solubility index and swelling power Water solubility index (WSI) and swelling power (SP) were determined by the method of Li and Yeh (17). A suspension of 0.1 g starch in 10 mL distilled water was heated at 55, 65, 75, 85, and 95°C in a shaking water bath for 1 hr and centrifuged  $(4,000 \times g)$  for 20 min. The supernatant was removed and the swollen granules weighed (Ws). The total solid carbohydrate (W<sub>1</sub>) of supernatant was determined by using the phenol sulfuric acid method. WSI and SP were calculated as follows;

WSI (%)=
$$(W1/0.1)\times 100$$
,  
SP (g/g)= $Ws/[0.1(100\%-WSI)]$ 

**Pasting properties** Pasting properties were determined using a rapid visco analyser (RVA, Series 4V; Newport Scientific Pty., Ltd, Warriewood, Australia). A native starch of Job's tears sample (3.0 g, d.b.) and its acid-hydrolyzed starches (10.0 g, d.b.) with 25 mL distilled water were mixed in an aluminum can with a paddle. The mixture was stirred for 1 min to facilitate dispersion before testing. The heating and cooling cycles were programmed in the following manner: The starch suspension was held at 50°C for 1 min, heated from 50 to 95°C at a rate of 12°C/min and held at 95°C for 2.5 min. Finally, it was cooled down to 50°C at a rate of 12°C/min and held at 50°C for 2 min.

X-ray diffraction measurements X-ray diffraction analysis was performed with an X-ray diffractometer (RINT 2000; Rigaku Co., Ltd., Tokyo, Japan) with target volatage-40 kV, electric current-40 mA, scanning range-4-45°, and scan speed-4°/min. Relative crystallinity of the starches was calculated using a peak fitting software (Origin-Version 7.5; Microcal Inc., Northampton, MA, USA). The starch samples were equilibrated at 25°C for 24 hr prior to the analysis.

Statistical analysis All experiments were performed in triplicate. The results were expressed as the mean±standard deviation (SD). One-way analysis of variance (ANOVA) followed by Duncan's multiple comparison test was performed. Statistical Package for the Social Science (SPSS, Version 12.0, 2004, SPSS Inc., Chicago, IL, USA) was used.

### **Results and Discussion**

**Degree of hydrolysis** The extent of hydrolysis (X) vs. time is shown in Fig. 1. The hydrolysis data plotted as  $\log_{10} [100/(100-X)]$  presented an evident 2-stage process. Acid hydrolysis of Job's tears starches proceeded rapidly up to 12 hr and then leveled off from 12 to 18 hr. Similar results were reported by Biladeris *et al.* (18) who suggested that the faster stage could be attributed to the hydrolysis of more amorphous parts of the starch granule and the later slower stage to the hydrolysis of the crystalline region. In acid modification, hydroxonium ion  $(H_3O^+)$  attacks the glycosidic oxygen atom and hydrolyzes the glucosidic linkage. Acid first acts on the surface of the starch granule before it gradually enters the inner region (3). Native

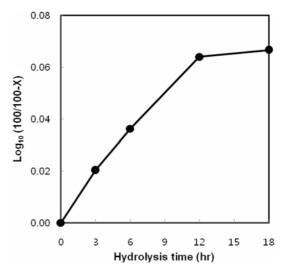


Fig. 1. Acid hydrolysis curve of Job's Tears starch which was plotted as  $log_{10}$  (100/100-X) vs time. X: The extent of hydrolysis

starches have some flaws on the surface of granules that can provide channels for the infiltration of hydrogen ions. These hydrogen ions primarily adhere to the amorphous areas and attack them (19). Wang *et al.* (20) also found that the amorphous areas of the starch granules were easier to attack with the hydrogen ions than the crystalline regions, which had a looser structure.

Swelling power and water solubility index The SP and WSI of native and acid-modified starches were measured at 10°C intervals from 55 and 95°C (Fig. 2). The swelling power of acid-modified starch decreased with an increase in hydrolysis time at all temperature tested in this study. It decreased significantly between 12 and 18 hr. These results were in good agreement with those of Wang and Wang (21) and Sánchez-Rivera et al. (5). It is well known that swelling power of starches have been influenced by various factors such as amylopectin molecular structure and amylose content (22,23). Morrison et al. (24) suggested that swelling factor is primarily a property of amylopectin and amylose acts as an inhibitor of swelling, especially in the presence of lipids. Zhou et al. (25) suggested that lower swelling factor of wrinkled pea starches could be attributed to its lower amylopectin content and/or strong interactions between amylose chains. Therefore, the modification of Job's tears starches with acid led to the depolymerization of amylopectin, appearing to make the starch lose the ability to hold absorbed water (21). The WSI increased as both temperature and hydrolysis time increased. A similar pattern was reported by Wang et al. (26) for corn starches modified with acid at different chlorine concentrations. This result indicated that the prolonged acid hydrolysis allowed the starch to be decomposed greatly, resulting in increased solubility (2). John et al. (27) also suggested that acid-modified starch granules could be comprised of a high proportion of soluble dextrins of both small and medium chain lengths, which increased solubility.

**Pasting properties** The pasting properties of native and acid-modified starches are shown in Table 1. Acid-

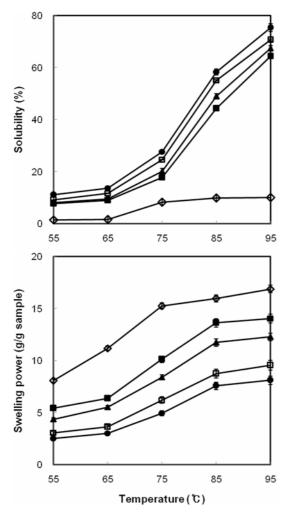


Fig. 2. Solubility and swelling power of native and acid-modified Job's tears starches. ♦, 0 hr; ■, 3 hr; ▲, 6 hr; □, 12 hr; ●, 18 hr.

modified starches were very thin, thus its RVA viscogram could be observed only at high concentration (10 g starch/25 mL distilled water) while the viscosity of native starch was too high to be determined by the RVA at same concentration. This result was similar to that reported by Atichokudomchai *et al.* (28) in which higher acid hydrolysis time provided tapioca starches with a very low viscosity. Pasting temperatures of native starches were reduced by the acid modification. However, the pasting temperature after 6 hr acid hydrolysis could not be

detected. The peak and final viscosity of acid-modified starches decreased with increasing hydrolysis time. This observation was in agreement with the finding of Chung et al. (29) who mentioned that acid hydrolysis decreased paste viscosity, possibly due to the shorter chains produced. The decrease in viscosities was caused by partial cleavage glucosidic linkage form extensive hydrolysis, resulting in a decrease in molecular weight of starch molecules (4). Kuakpetoon and Wang (30) also proposed that the partially degraded network was not resistant to shear and thereby produced a lower viscosity. The breakdown viscosity (BV) decreased with an increase in hydrolysis time. The BV of samples treated for 18 hr was 1.3 RVU compared to 139.3 RVU for native starch. A higher BV presents granule disruption or the tendency of starch to succumb to shear force during heating (31). As hydrolysis time increased, the peak time (time required to peak viscosity) and setback decreased. All RVA viscograms of acid modified starches decreased with an increase in the hydrolysis time. This observation was in agreement with the finding of Koksel et al. (32) who reported that all the viscosity values of acidmodified corn starches are lower than native corn starch, and those of the hydrolysates progressively decreased to a significant extent with an increase in the hydrolysis time.

X-ray diffraction measurements The X-ray diffraction patterns of native and acid-modified starches are shown in Fig. 3. There was no pronounced difference between the native and modified starches. That is, the native and modified starches showed diffraction patterns with peaks at 15, 17, 18, and 23° (2θ), which are typical diffraction lines for A-type crystallites (33). This result was in accordance with that of other acid-hydrolyzed starches which exhibited the same crystalline type comparison to the unmodified starches (34). However, as the starches were subjected to a longer period of acid hydrolysis (i.e., 18 hr), the crystallinity of acid-modified starches increased slightly with sharper peaks at  $2\theta = 17^{\circ}$ . These results are consistent with reports by Wang et al. (20), Chung et al. (29), and Wang and Wang (34) who found that the crystallinity of starches with acid hydrolysis increased slightly with increasing extent of acid modification. It has been suggested that cleavage of starch chains in the amorphous regions allows extensive recording of the chain segments to give a more crystalline structure with a sharper X-ray pattern (35). Therefore, it may be that starch chains modified with more hydrolysis time showed an increased intensity of peak because of more extensive hydrolysis and subsequently more recording (27).

In this study, native Job's tears starches were modified

Table 1. Pasting properties of native and acid-modified Job's tears starches

Hydrolysis time (hr)	Pasting temp. (°C)	Peak viscosity (RVU)	Peak time (min)	Final viscosity (RVU)	Breakdown (RVU)	Setback (RVU)
0	74.68±0.33	323.96±19.57 <sup>a1)</sup>	4.07±0.25a	211.28±0.39 <sup>a</sup>	139.31±1.83 <sup>a</sup>	$26.56\pm0.20^{a}$
3	$70.45 \pm 0.05$	$71.09\pm3.36^{b}$	$3.50\pm0.15^{b}$	$68.02\!\pm\!1.78^{b}$	64.50±2.21 b	$13.58\pm2.55^{b}$
6	ND	$8.33 \pm 1.47^{c}$	$3.44{\pm}0.08^{bc}$	$4.26\pm0.11^{c}$	$6.52\pm0.18^{c}$	$2.50\pm1.57^{c}$
12	ND	$4.22 \pm 1.01^{c}$	$3.23{\pm}0.03^{cd}$	$1.17\pm0.02^d$	$3.03\pm0.10^{d}$	$1.10\pm0.60^{c}$
18	ND	$1.78 \pm 0.55^{c}$	$3.13{\pm}0.03^d$	$0.58{\pm}0.04^{\rm d}$	$1.32 \pm 0.03^d$	$0.72\pm0.23^{c}$

<sup>&</sup>lt;sup>1)</sup>Different subscripts within the same column are significantly different among samples  $\alpha$ =0.05 level by Duncan's multiple range test; ND, not detected.

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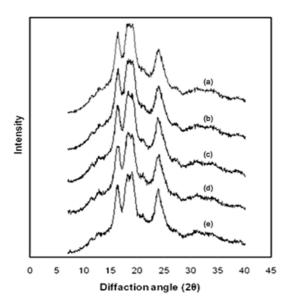


Fig. 3. X-ray diffraction pattern of Job's tears starches after acid modification. (a) native, (b) 3 hr, (c) 6 hr, (d) 12 hr, and (e) 18 hr

with hydrochloric acid under various hydrolysis times. Then, the physicochemical and pasting properties of the modified starches were investigated as a function of the degree of hydrolysis. The acid modification significantly altered the physicochemical properties of Job's tears such as water solubility and swelling power which depended on the extent of acid hydrolysis. Also, significant reduction in pasting profiles of Job's tears starches could be readily observed, depending on the degree of acid hydrolysis. The X-ray pattern of the granule, however, remained unchanged as a result of acid modification. Thus, the physicochemical properties of the modified Job's tears starches measured in this study can be a clue for its practical use in food products. Further research would be necessary to evaluate its rheological and textural performances in a food system when incorporated into food formulations.

## Acknowledgment

This work was supported by Seoul Research & Business Development (Seoul R&BD) Program (Project No. 10625), Korea.

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