

Inhibition on LDL-oxidation by Phenolic Compounds from the Fruit Body of *Phellinus linteus*

Ha-Na Lyu¹, Dae-Young Lee¹, Min-Kyung Lee¹, Moon-Hee Cho², Tae-Sook Jeong², In-Ho Kim³, Chang-Ho Lee³, and Nam-In Baek^{1*}

¹Graduate School of Biotechnology, Kyung Hee University, Yongin 446-701, Korea

²National Research Laboratory of Lipid Metabolism & Atherosclerosis, Korea Research Institute of Bioscience and Biotechnology, Daejeon 305-333, Korea

³Korea Food Research Institute, Sungnam 463-746, Korea

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The fruit body of *Phellinus linteus* was extracted with 80% aqueous MeOH, and the concentrated extract was successively partitioned using EtOAc, *n*-BuOH and H₂O. From the EtOAc and *n*-BuOH fractions, three phenolic compounds were isolated through repeated silica gel and ODS column chromatography. The chemical structures of these compounds were determined as 2-(3',4'-dihydroxyphenyl)-1,3-benzodioxole-5-aldehyde (1), 4-(3,4-dihydroxyphenyl)-3-buten-2-one (2), and protocatechuic acid methyl ester (3) by spectroscopic data including NMR, MS and IR. Compounds 1-3 exhibited low density lipoprotein (LDL) antioxidant activity with IC₅₀ values of 1.7, 0.7, and 2.4 μM, respectively.

Key words: 2-(3',4'-dihydroxyphenyl)-1,3-benzodioxole-5-aldehyde, 4-(3,4-dihydroxyphenyl)-3-buten-2-one, LDL-oxidation, *Phellinus linteus*, protocatechuic acid methyl ester, phenolic compound

Phellinus linteus is a fungus belonging to the family Hymenochaetaceae, which is distributed mainly in Asia, tropical America, and North Africa [Dai and Xu, 1998]. Extracts from this fungus have long been used as a traditional oriental medicine in Korea and Japan for the treatment of various diseases including arthritis of the knee, oral ulcers, gastroenteric disorders, lymphatic disease, and various cancers [Heo, 1993; Kang *et al.*, 2004]. It has been reported that *P. linteus* has anti-cancer [Choi *et al.*, 1996; Han *et al.*, 1999], anti-tumor [Ikekawa *et al.*, 1968], anti-mutagenic [Ji *et al.*, 2000], anti-angiogenic [Song *et al.*, 1995], anti-oxidant [Lee *et al.*, 2000], and immune activity [Kim *et al.*, 1996]. A number of researchers have reported various chemical constituents of *P. linteus* such as polysaccharides [Lee *et al.*, 2001], proteoglycans [Lee *et al.*, 2001], sphingolipids [Kang *et al.*, 2004] and hispidine [Park *et*

al., 2004], etc. Among these, β-glucan is especially important as it is known to have anti-tumor activity and immuno-stimulating effects [Han *et al.*, 1999; Lee *et al.*, 2000]. From our previous phytochemical research on the fruit body of this mushroom, we reported the presence of ergosterols [Lyu *et al.*, 2007a; 2007b; 2007c] and phenolic compounds [Lyu *et al.*, 2008]. Except for polysaccharides, however, there has been little research studying various pharmacological activities of lower molecular weight compounds from *P. linteus*. Furthermore, only a few reports have claimed that extracts of the fruit body of *P. linteus* act to prevent atherosclerosis or to inhibit LDL-oxidation.

It has been suggested that the oxidation of LDL cholesterol is an important step in the formation of atherosclerotic lesions [Steinberg *et al.*, 1989; Diaz *et al.*, 1997]. Evidence to support this hypothesis is based in part on observational studies that demonstrate associations between oxidized LDL cholesterol and the presence of both atherosclerotic lesions [Regnstrom *et al.*, 1992] and the progression of carotid artery atherosclerosis [Salonen *et al.*, 1992]. This paper describes three phenolic compounds isolated from *P. linteus* that inhibit LDL oxidation.

Extraction and isolation. The dried fruit body of *P. linteus* (1 kg) was extracted two times at room temperature with 80% aqueous MeOH (3 L×2). The extracts were successively partitioned with water (1 L), EtOAc (1 L×2) and *n*-BuOH (1 L×2). The EtOAc extract (10 g) was applied to a silica gel cc (6.5×17 cm) and was eluted with a gradient of CHCl₃:MeOH (15:1→

*Corresponding author

Phone: +82-31-201-2661; Fax: +82-31-201-2157

E-mail: nibaek@khu.ac.kr

Abbreviations: cc, column chromatography; CHCl₃, chloroform; EI-MS, electron ionization mass spectrometry; EtOAc, ethylacetate; IR, infra-red; LDL, low density lipoprotein; MeOH, methanol; *n*-BuOH, normal-butanol; ODS, octadecyl silica gel; PLB, *n*-butanol fraction of *P. linteus*; PLE, ethyl acetate fraction of *P. linteus*; TLC, thin layer chromatography; V_e/V_t, elution volume/total volume

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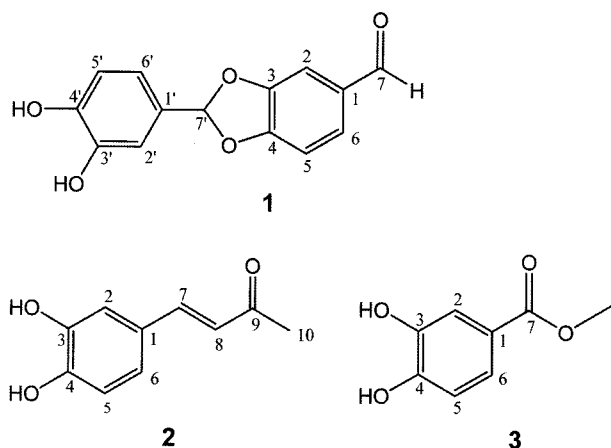


Fig. 1. Chemical structures of compounds 1~3 isolated from the fruit body of *P. linteus*.

10:1→7:1→5:1→3:1→1:1, 1 L of each) while monitoring by TLC to produce 22 fractions (PLE1 to PLE22). Fraction PLE6 (250 mg, V_e/V_t 0.18-0.20) was applied to an ODS cc (3×8 cm) and was eluted with MeOH:H₂O (1:3→1:1, 700 mL of each) to ultimately produce compound **1** [22.4 mg, V_e/V_t 0-0.03, TLC (ODS F₂₅₄) R_f 0.75, MeOH:H₂O=2:1] and compound **2** [17.5 mg, V_e/V_t 0.10-0.13, TLC (ODS F₂₅₄) R_f 0.66, MeOH:H₂O=2:1]. The concentrated *n*-BuOH fraction (PLB, 31 g) was subjected to a silica gel cc (7×23 cm) eluting with a gradient of CHCl₃:MeOH:H₂O (6:4:1→5:4:1, 4 L of each), which resulted in 19 fractions (PLB1 to PLB19). Fraction PLB2 (831 mg, V_e/V_t 0.05-0.12) was subjected to a silica gel cc (5×15 cm) and eluting with CHCl₃:MeOH (20:1→10:1, 2 L of each) to produce 10 fractions (PLB2-1 to PLB2-10). Fraction PLB2-3 (103 mg, V_e/V_t 0.25-0.40) was applied to an ODS cc (2.5×5 cm) and was eluted with MeOH:H₂O (2:3, 500 mL) yielding compound **3** [6.5 mg, V_e/V_t 0.25-0.35, TLC (ODS F₂₅₄) R_f 0.65, MeOH:H₂O=2:1].

Compound 1: Pale purple amorphous powder (MeOH); mp 131-137°C; $[\alpha]_D^{20}$ ±0° (racemic); IR (KBr, ν) 3329, 3323, 1655, 1647, 1443, 1296, 1167, 878, 754; EI-MS m/z 258 [M]⁺; ¹H-NMR (400 MHz, CD₃OD, δ) 9.67 (1H, s, H-7), 7.30 (1H, dd, $J=7.2, 2.0$, H-6), 7.29 (1H, d, $J=2.0$, H-2), 6.90 (1H, d, $J=7.2$, H-5), 6.84 (1H, d, $J=1.6$, H-2'), 6.74 (1H, d, $J=6.8$, H-5'), 6.74 (1H, d, $J=6.8, 1.6$, H-6'), 5.20 (1H, s, H-7'); ¹³C-NMR (100 MHz, CD₃OD, δ_c) 192.9 (C-7), 153.5 (C-3), 147.0 (C-4), 146.5 (C-4'), 145.9 (C-3'), 130.9 (C-1'), 130.7 (C-1), 126.3 (C-6), 119.3 (C-6'), 116.1 (C-5), 115.7 (C-5'), 115.2 (C-2), 114.7 (C-2'), 104.7 (C-7').

Compound 2: Pale purple amorphous powder (MeOH); mp 177-178°C; IR (KBr, ν) 3420, 1651, 1603, 1556, 1286, 1115; EI-MS m/z 178 [M]⁺, 163 [M-Me]⁺; ¹H-NMR (400 MHz, CD₃OD, δ) 7.51 (1H, d, $J=16.4$ Hz, H-7), 7.07 (1H, d, $J=2.4$ Hz, H-2), 6.98 (1H, dd, $J=8.4, 2.4$, H-6), 6.78 (1H, d, $J=8.4$ Hz, H-5), 6.54 (1H, d, $J=16.4$, H-8), 2.32 (3H, s, H-10); ¹³C-NMR (100 MHz,

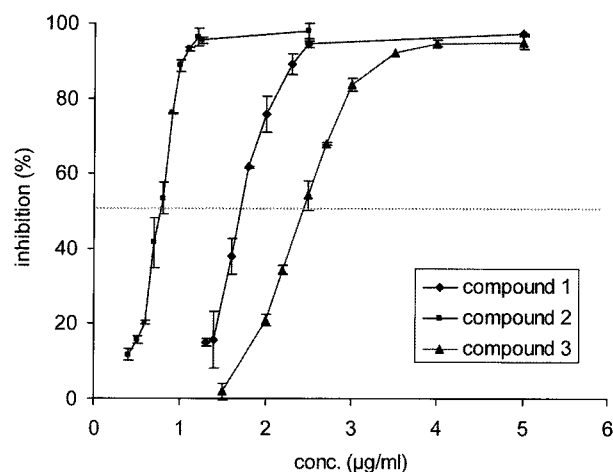


Fig. 2. Inhibition on LDL-oxidation by compounds 1~3 isolated from the fruit body of *P. linteus*. *The data are presented as mean±standard deviation of three replications. **Positive control of LDL-oxidation (BHT): 85.9% (±0.5) inhibition at 4 μ M

CD₃OD, δ_c) 201.39 (C-9), 149.83 (C-4), 146.74 (C-3), 146.74 (C-7), 127.56 (C-1), 124.55 (C-8), 123.44 (C-6), 116.42 (C-5), 115.14 (C-2), 27.02 (C-10).

Compound 3: Pale purple amorphous powder (MeOH); mp 136°C; IR (KBr, ν) 3338, 1685, 1604, 1521, 1438, 1301; EI-MS m/z 168 [M]⁺, 137, 109, 81, 63, 53; ¹H-NMR (400 MHz, CD₃OD, δ) 7.38 (1H, d, $J=8.4, 2.0$, H-6), 7.39 (1H, d, $J=2.0$, H-2), 6.77 (1H, d, $J=8.4$, H-5), 3.80 (3H, s, H-7-OMe); ¹³C-NMR (100 MHz, CD₃OD, δ_c) 168.9 (C-7), 151.7 (C-4), 146.2 (C-3), 123.6 (C-6), 122.5 (C-1), 117.4 (C-2), 115.8 (C-5), 52.7 (C-7-OMe).

Low-density lipoprotein isolation and oxidation assay. Plasma was obtained from fasted, healthy normolipidemic volunteers. The LDL was isolated by a standard procedure with a slight modification [Havel *et al.*, 1955]. The TBARS assay of Buege and Aust [Buege and Aust, 1978] was used with a slight modification. Thus, an LDL solution (250 μ L, 50-100 μ g of protein) in 10 mM PBS (pH 7.4) was supplemented with 10 μ M CuSO₄. The oxidation was performed in a screw-capped 5 mL glass vial at 37°C in a shaking water bath. After 4 h incubation, the reaction was terminated by the addition of 1 mL of 20% trichloroacetic acid. Following precipitation, 1 mL of 0.67% thiobarbituric acid (TBA) in 0.05 N NaOH was added and vortexed, and the final mixture was heated for 5 min at 95 μ , cooled on ice, and then centrifuged for 2 min at 1,000×g. The optical density of the produced malondialdehyde (MDA) was measured at 532 nm. Calibration was completed with an MDA standard prepared from tetramethoxypropane.

When the MeOH extract of the fruit body of *P. linteus* was developed on a silica gel TLC, visualization by spraying with a 10% H₂SO₄ solution and then heating the TLC plate gave purple spots, indicating the presence of phenolic compounds in the

extract. The MeOH extracts were partitioned into EtOAc, *n*-BuOH, and H₂O layers through solvent fractionation. Three phenolic compounds (compounds 1~3) were isolated through repeated silica gel and ODS cc of the EtOAc and *n*-BuOH fractions, and their structures were determined to be 2-(3',4'-dihydroxyphenyl)-1,3-benzodioxole-5-aldehyde (1), 4-(3,4-dihydroxyphenyl)-3-buten-2-one (2), and protocatechuic acid methyl ester (3) by interpretation of extensive spectroscopic data and comparisons with literature data [Tagashira *et al.*, 1998; Lyu *et al.*, 2008]. Especially, the stereostructure of the double bond in compound 2 was determined as *trans*-configuration from the coupling constant ($J=16.4$ Hz) observed between H-7 and H-8 in the ¹H-NMR spectrum.

In order to determine whether these compounds might be effective in the development of hypercholesterolemic or antiatherogenic agents, their potential for inhibiting LDL oxidation was evaluated. Compounds 1~3 demonstrated LDL antioxidant activity with IC₅₀ values of 1.7, 0.7, and 2.4 μM, respectively, and were more potent than the positive control, BHT, which had an IC₅₀ value of 3.0±0.4 mM [Hung *et al.*, 2006]. In this study, we have demonstrated that compounds 1, 2, and 3 isolated from the fruit body of *P. linteus* inhibited LDL oxidation activity.

LDL oxidation inhibitors from natural sources have been rarely reported. Because compounds 1-3 not only exhibit activity similar to the positive control used in this study but are derived from natural sources which have been used as an oriental medicine in Korea, they may prove useful for treating hypercholesterolemia and atherosclerosis.

Acknowledgments

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