# The Effect of the Sintering Additives on the Fabrication and Thermal Conductivity of Porous Sintered RBSN

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#### ABSTRACT

The nitriding and post-sintering behavior of silicon powder compact containing sintering additives of 2.3 wt% and 7 wt% were investigated in this study. Regardless of the liquid phase content, elongated large grains of a typical morphology evolved in the post-sintered specimens. Phase analysis revealed a complete phase transformation into  $\beta$ -Si<sub>3</sub>N<sub>4</sub> in both porous systems. Oxynitride second phases (mellilite) precipitated in the latter, while those were free in the former containing less amount of liquid phase. The post-sintering condition that yielded a favorable microstructure for a filter application was achieved when the specimens were soaked at 1800°C for 2 h. It was found that the thermal conductivity of porous Si<sub>3</sub>N<sub>4</sub> ceramics is dominated by the porosity more than this factor is influenced by the addition of Al<sub>2</sub>O<sub>3</sub>.

Key words: RBSN, SRBSN, DPF, Porous, Thermal conductivity

#### 1. Introduction

In addition to providing machinability, porous ceramic materials have many industrial applications as filtering media including high-temperature gas filters, separation membranes, and catalyst supports. Recently in Europe, the market share of diesel vehicles among newly built passenger cars exceeded 50 percents and this rate continues to increase. A diesel particulate filter (DPF) is a component that traps and eliminates particulate matter (PM) contained in the exhaust gas of diesel-powered vehicles. In advanced countries, the attachment of DPFs onto diesel vehicles is required by law. Nevertheless, only limited efforts have been made to replace the currently commercialized cordierite and/or SiC filter in DPFs with Si<sub>3</sub>N<sub>4</sub> ceramics. (2)

The addition of sintering aids is indispensable for the densification of  $\mathrm{Si}_3\mathrm{N}_4$  ceramics fabricated by both gas pressure sintering and pressureless sintering. The composition and quantity of sintering aids decisively controls the extent of grain growth, grain morphology and the precipitation of second phases. The authors, however, could not identify previous reports dealing with the aforementioned phenomena investigated in porous  $\mathrm{Si}_3\mathrm{N}_4$  ceramics.

In this study, metal silicon powder, which is much cheaper than Si<sub>3</sub>N<sub>4</sub> powder, was selected as a starting material, and a reaction-bonding process<sup>4,5)</sup> followed by post-

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sintering<sup>6,7)</sup> was employed to produce a porous  $Si_3N_4$  filter substrate. Based on the typical intermingled columnar microstructure of  $Si_3N_4$  ceramics, the goal of this research is to develop a filter substrate exerting the maximum permeability and soot trap performance simultaneously. Emphasis was placed on the relationship between the additive content and the microstructure, as well as on the effect of the additive composition on the thermal conductivity of porous reaction-bonded silicon nitride (RBSN) and sintered-RBSN (SRBSN).

# 2. Experimental Procedure

The raw Si materials used in this study were commercially available. Commonly used sintering additives for the sintering of  $\mathrm{Si_3N_4}$  ceramics,  $\mathrm{Y_2O_3}(\mathrm{H.C.}$  Starck, Grade C) and  $\mathrm{Al_2O_3}$  (Sumitomo, AKP-30), were selected. Powder mixtures were made either by a single addition of rare earth oxide (designated as Y) or a concurrent addition of  $\mathrm{Y_2O_3}$  and  $\mathrm{Al_2O_3}$  (designated as YA). The total amount of sintering additives was fixed at 2.3 wt% and 7 wt%, respectively, for the two aforementioned cases. A spherical-shaped pore former (PMMA, Aldrich) was blended in order to control the porosity level of the sintered body. Various combinations of the sintering additive and pore former are listed in Table 1 along with their nomenclature.

Approximately 50 g of powder mixture consisting of Si, sintering additives and pore former were thoroughly milled in a nylon jar with  $\mathrm{Si_3N_4}$  balls and ethanol as milling media for 7 h. After the powder mixtures were dried using a rotating evaporator, they were sieved to a particle size < 150  $\mu$ m.

Table 1. Composition of the Si Powder Mixture

(Unit : g) Si**PMMA**  $Y_9O_3$ Al<sub>2</sub>O<sub>3</sub> 2.3YAP0 3.12 100 0.88 2.3YP0 100 2.3YAP20 100 3.12 0.88 20 2.3YP20 100 20 4 **7YAP20** 100 9.66 2.73 20

The resulting powder mixtures were uniaxially pressed into rectangular pellets measuring 10 mm × 10 mm × 1.3 mm and were subsequently cold isostatically pressed at 200 MPa. After shaping, the samples were heated in a 95% N<sub>2</sub>+5% H<sub>2</sub> flowing atmosphere at temperatures ranging from 1250°C to 1350°C using a horizontal tube furnace in which reaction-bonding occurred. A maximum of eight specimens were placed upon a suspended BN-coated graphite plate. The specimens were erected upright on their edges and made to fit into a small hole on a graphite plate so that both the chance for contamination and the temperature gradient in the specimen were minimized with minimum contact between the specimens and the graphite plate. The percent of nitridation was calculated from the weight change before and after the nitriding process. In preparation for the post-sintering of the nitrided specimens, samples taken from the nitriding furnace were placed upon a BN-coated graphite crucible with Si<sub>2</sub>N<sub>4</sub> powder filling the volume of the crucible. Post-sintering was carried out in a graphite resistance furnace both at 1700°C and 1800°C under a static nitrogen gas pressure of 0.9 MPa. SEM (Jeol, JSM-6700F, Japan) was used to observe the fracture surfaces, and X-ray diffraction (Rigaku, D/Max 2200, Japan) was performed for the phase identification. The thermal conductivity (K) at room temperature was calculated from the measured thermal diffusivity (α) using a micro-flash machine (MicroFlash, LFA437, Germany) together with the measured density (ρ) and known heat capacity  $(c_n)$ ,  $K(W/mK) = \rho \cdot c_n \cdot \alpha$ .

## 3. Results and Discussion

A volume increase of 21.4% and weight gain of 66.5% occurred by the nitriding reaction of  $3\mathrm{Si}(s)+2\mathrm{N}_2(g)~\mathrm{Si}_s\mathrm{N}_4(s)$ . Assuming perfect nitridation of raw Si powder and zero loss during the entire process, the theoretical porosity of RBSN and SRBSN can be calculated.<sup>8)</sup> In Fig. 1, the measured porosity values (scattered dots) are plotted together with the calculated values (solid lines) for 20 parts of pore former content with a fixed sintering additive of 2.3 wt%. More than 45% porosity was readily achieved by post-sintering up to 1700°C, at which temperature shrinkage was restricted within 1%~2%. A constant tendency was found in which the measured values are somewhat larger than the calculated values. This discrepancy can be attributed to the

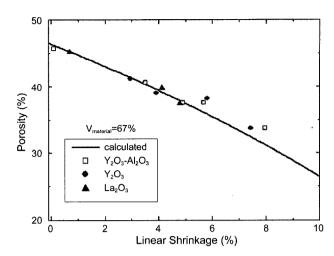


Fig. 1. Measurement (scattered dots) and calculation (solid line) of the porosity for specimens blended with 20 parts of pore former: The amount of sintering additive was fixed at 2.3 wt%.

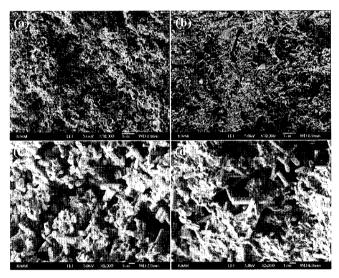


Fig. 2. Fracture surface of RBSN (a,b) and SRBSN (c,d): Insets (a,c) and (b,d) are for 2.3YAP20 and 7YAP20, respectively. Reaction bonding of Si compact was conducted at 1350°C and subsequent post-sintering was conducted at 1800°C for 2 h in a 0.5 MPa nitrogen atmosphere.

weight loss caused by evaporation during the process.

The microstructures of the RBSN specimens (Figs. 2(a) and (b)) consist of equiaxed matrix grains and whisker-like grains with diameters of a few hundred nanometers. The starting raw Si powder ( $d_{50}=7~\mu m$ ) were shattered into smaller Si $_3N_4$  particles by the nitriding reaction. It is interesting to note the very similar same grain morphology and pore structure of SRBSN between the 2.3 wt% system and 7 wt% system. The fracture surfaces of both specimens post-sintered at 1800°C show a well-developed three-dimensional array of elongated grains. In addition, the coalescence of pores proceeded during the post-sintering stage together with the grain growth. Much of the commercial products of dense Si $_3N_4$  ceramics fall in the composition range similar to

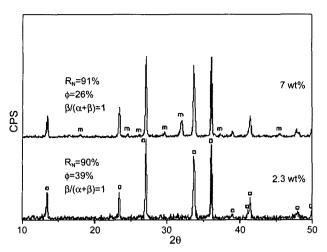


Fig. 3. Phase analysis of a post-sintered specimen: Complete phase transformation into β-Si<sub>3</sub>N<sub>4</sub> was confirmed for both 2.3YAP20 and 7YAP20. The lower additive system (2.3 wt%) was verified to be second-phase-free.

 $5~\rm wt\% Y_2O_3\text{-}2~\rm wt\% Al_2O_3$ , i.e., a 7 wt% additive system. Obtaining a similar microstructure with less sintering additive, 2.3 wt% in this research, bears some engineering importance considering that this is associated with less shrinkage due to the lower liquid phase content. A suitable porosity and pore channel structure for a filter application is apt to be acquired in the system in which shrinkage is kept to a minimum level.

A phase analysis conducted for the SRBSN specimens is plotted in Fig. 3, in which  $R_N$ ,  $\phi$  and  $\beta/(\alpha+\beta)$  are the nitridation rate, porosity and  $\beta$ -Si<sub>3</sub>N<sub>4</sub> phase fraction, respectively. For the RBSN specimens, all of the parameters mentioned above are similar each other independent of the amount of sintering additives. In contrast, a higher porosity was measured in the 2.3 wt% system compared to the 7 wt% system for the post-sintered specimens. It should be noted that the 2.3 wt% specimen was free from oxynitride second phases

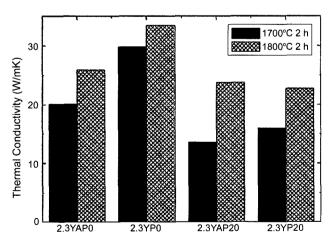


Fig. 4. Thermal conductivity of porous  $\mathrm{Si_3N_4}$  ceramics with 2.3 wt% additive: The addition of  $\mathrm{Al_2O_3}$  decreases the thermal conductivity in specimens free from pore former, while a negligible effect was revealed for the porous specimens blended with pore former.

(designated as m), which are deteriorating to the material properties. By reducing the sintering additive content from 7 wt% to 2.3 wt%, a second-phase-free and higher porosity filter substrate was obtained successfully.

The incorporation of aluminum and oxygen in the Si<sub>2</sub>N<sub>4</sub> structure decreases the thermal conductivity because substituted ions and vacancies for charge neutrality act as a phonon scattering site.<sup>9)</sup> The measurement of thermal conductivity was carried out for the 2.3 wt%-additive specimens, which are better suited for a filter application. For specimens free from pore former, the thermal conductivity of the 2.3YP0 specimen is higher than that of 2.3YAP0 specimen, which was blended with Al<sub>2</sub>O<sub>3</sub> as a sintering additive. Only a fractional difference existed in the measured thermal conductivity between  $Al_2O_3$ -added (2.3YAP20) and Al<sub>2</sub>O<sub>3</sub>-free (2.3YP20) samples when pore former was mixed to increase the porosity. It is believed that the larger porosity with the addition of pore former dominated the decrease in the thermal conductivity. The addition of Al<sub>2</sub>O<sub>3</sub> has a negligible influence on the thermal conductivity of highly porous Si<sub>3</sub>N<sub>4</sub> ceramics.

#### 4. Conclusions

In order to develop microstructure controlling technology of porous RBSN and SRBSN for a DPF application, the effect of additive contents and composition was investigated. An elongated grain microstructure was obtained in as low as a 2.3 wt% additive system. Furthermore, a second-phase-free sound microstructure consisting of only a  $\beta\text{-Si}_3N_4$  phase was fabricated by reducing the additive content. An addition of  $Al_2O_3$  for liquid phase sintering is possible and does not sacrifice the thermal conductivity for in the investigated porous body.

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