

## Note

***In vitro* Peroxynitrite Scavenging Activity of Arctic Seaweed Extracts**

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**Abstract :** In this study, antioxidant activities of acetone/dichloromethane and methanol extracts of Arctic seaweeds were investigated. The antioxidant properties of both extracts of arctic seaweed were evaluated using two different peroxynitrite tests, including scavenging power on authentic peroxynitrite and inhibitory activities on peroxynitrite generation from 3-morpholinopyridone (SIN-1) producing superoxide anion and nitric oxide simultaneously *in vitro*. At concentration of 10  $\mu\text{g/ml}$ , the acetone/dichloromethane and methanol extracts of *Odonthalia dentata* exhibited 54.6 and 64.2% inhibition against peroxynitrite generation from SIN-1 while they exhibited 24.3 and 23.0% scavenging activities on authentic peroxynitrite, respectively. On the other hand, the acetone/dichloromethane extract of *Polysiphonia stricta* exhibited 61.8% inhibition only against the induced peroxynitrite from SIN-1. Furthermore, the crude extracts of *Odonthalia dentata* and *Polysiphonia stricta* were fractionated into *n*-hexane, 85% aq. MeOH, *n*-BuOH, and H<sub>2</sub>O fractions, successively, and only 85% aq. MeOH fraction exhibited the best inhibition.

**Key words :** Arctic seaweed, peroxynitrite, *Odonthalia dentata*, *Polysiphonia stricta*

## 1. Introduction

Peroxynitrite is produced by the reaction of nitric oxide (NO) with superoxide in living organisms. Superoxide anion radical and nitric oxide belong to reactive oxygen species (ROS) and reactive nitrogen species (RNS), respectively (Patel *et al.* 1999). These activated reactive species are generated continuously via normal physiological processes, and more significantly in pathological conditions. Upon stimulation by inflammation and/or infection, macrophages and neutrophils can be activated to produce large amounts of peroxynitrite, and its overproduction exerts strong deleterious influence on homeostatic regulation of our body (Fang *et al.* 2002; Grace *et al.* 1998). Peroxynitrite is a potent oxidant (1,000 times more active than equidose hydrogen peroxide) that is able to oxidize and nitrosylate

a large variety of biomolecules. Various toxic effects of peroxynitrite have been described, such as oxidation and nitration of proteins (Virag *et al.* 2003; Reiter *et al.* 2002). Therefore, the study of peroxynitrite scavenger and its reaction mechanism might be a valuable tool for preventive therapeutic strategies.

Marine organisms are virtually untapped sources of natural products, many of them are biologically active and structurally unique. The significance of seaweed as a source for biologically active natural products is also well known (Ali *et al.* 2000). Nevertheless, secondary metabolites that scavenge peroxynitrite anion from marine algae have little been investigated and only a few compounds have been found in marine algae (Chung *et al.* 2001; Lee *et al.* 2004; Jung *et al.* 2006). In our continuous search for novel metabolites from marine resources (Lee<sup>a</sup> *et al.* 2004; Lee<sup>b</sup> *et al.* 2004; Seo *et al.* 2004; Lee and Seo 2006; Jung *et al.* 2004), we have attempted to characterize the

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antioxidative effects of arctic seaweed extracts.

## 2. Materials and Method

### Sample Collection, Extraction, and Fractionation

The seaweed samples were collected from July to August 2005 at Kongsfjorden in the Svalbard Islands, the Arctic. The taxonomic identifications of seaweed specimens were confirmed by an alga taxonomist, J. H. Kim, at the KOPRI, Korea. Shade-dried seaweed was extracted with a volumetrically equal mixture of acetone and dichloromethane for 24 hr at room temperature. This step was repeated twice. The extracted solution was then evaporated to dryness under vacuum, yielding a dark and sticky residue. The seaweed residue was then extracted once more with methanol, according to the same procedure. Each of the crude extracts was used as experimental material. The prepared samples were then stored in a refrigerator at  $-20^{\circ}\text{C}$ , for later study. Each of combined crude extracts of *Odonthalia dentata* and *Polysiphonia stricta* was suspended in  $\text{H}_2\text{O}$  and then partitioned between  $\text{CH}_2\text{Cl}_2$  and  $\text{H}_2\text{O}$ . The  $\text{CH}_2\text{Cl}_2$  fraction was further partitioned with *n*-hexane and 85% aqueous MeOH and then the  $\text{H}_2\text{O}$  fraction successively fractionated with *n*-BuOH and  $\text{H}_2\text{O}$ .

### Chemicals

L-ascorbic acid, DL-penicillamine (DL-2-amino-3-mercapto-3-methyl butanoic acid) and 3-morpholino-sydnnonimine (SIN-1) were purchased from Sigma-Aldrich Chemical Company (St.Louis, MO, USA). The Dihydro-rhodamine 123 (DHR 123) and peroxy-nitrite were of the highest quality commercially available and were purchased from Molecular Probes (Eugene, Oregon, USA), and Cayman (Ann Arbor, MI, USA), respectively.

### Measurement of ONOO<sup>-</sup> Scavenging Activity

The peroxy-nitrite (ONOO<sup>-</sup>) scavenging ability was measured by monitoring the oxidation of dihydro-rhodamine 123 using a modified version of the method of Kooy *et al.* (1994). The peroxy-nitrite reacts with DHR 123, causing oxidized DHR 123 to form, and its converted chemical structure is capable of emitting fluorescence. A stock solution of DHR 123 (5 mM) purged with nitrogen was prepared in advance and stored at  $-80^{\circ}\text{C}$ . A working solution of DHR 123 (final concentration, 5 M) was diluted from the stock solution and placed on ice in the dark immediately prior to the measurement. The buffer of 90 mM sodium chloride, 50 mM sodium phosphate (pH 7.4) and 5 mM potassium chloride with 100 M (f.c.)

diethylenetriaminepentaacetic acid (DTPA) was purged with nitrogen and placed on ice before use. The ONOO<sup>-</sup> scavenging ability, based on the oxidation of DHR 123, was determined with a microplate fluorescence spectrophotometer, FL 500 (Bio-Tek instruments, USA) using the wavelengths of 485 nm and 530 nm for excitation and emission, respectively, at room temperature. The background and final fluorescent intensities were measured 5 min after treatment with or without SIN-1 (f.c. 10 M) or authentic ONOO<sup>-</sup> (f.c. 10 M) in 0.3 N sodium hydroxide. The oxidation of DHR 123 due to decomposition of the SIN-1 gradually proceeded whereas the authentic ONOO<sup>-</sup> rapidly oxidized DHR 123 with the final fluorescent intensity being stable over time. Penicillamine was used as a positive control.

## 3. Results and Discussion

The antioxidant activity of twenty seaweed extracts was determined by the scavenging effect on authentic ONOO<sup>-</sup> and inhibitory effect of ONOO<sup>-</sup> generation from SIN-1, according to the method of Kooy *et al.* (1994). Two different pathways appear to exist in the peroxy-nitrite system. One is simply scavenging authentic ONOO<sup>-</sup>. The other is to scavenge the superoxide anion ( $\cdot\text{O}_2^-$ ) and nitric oxide ( $\cdot\text{NO}$ ) generated from SIN-1 and to subsequently inhibit the formation of ONOO<sup>-</sup> from SIN-1, a generator of peroxy-nitrite through the reaction between nitric oxide and superoxide anion (Zou *et al.* 2002). As summarized in Table 1, the antioxidative capabilities of twenty arctic seaweed extracts were compared with penicillamine and L-ascorbic acid.

At 10  $\mu\text{g/ml}$  concentration of acetone/dichloromethane extract, ten seaweed extracts were comparatively effective in their inhibitory activities on SIN-1, which increased in the following order: *Chordaria flagelliformis* (34.4%), *Devaleraea ramentacea* (41.3%), and *Monostroma obscurum* (41.3%), *Chorda filum* (44.6%), *Phycodryis rubens* (45.9%), *Alaria esculenta* (46.3%), *Laminaria solidungula* (47.8%), *Spongomorpha arcta* (50.2%), *Odonthalia dentata* (54.6%), and *Polysiphonia stricta* (61.8%). The methanol extracts of the Arctic seaweeds -- except for *Odonthalia dentata* (64.2%) and *Desmarestia aculeate* (46.0%) -- did not show significant inhibitory activities in this system. On the other hand, crude extracts of seaweed showed little scavenging effect on authentic peroxy-nitrite; that of only two species, acetone/dichloromethane of *Spongomorpha arcta* (26.1%) and both extracts of *Odonthalia dentata* (24.3% and 23.0%) were found to be the best among them.

**Table 1.** Peroxynitrite inhibitory/scavenging activities of seaweed extracts (10 µg/ml)

Species (10 µg/ml)	SIN-1		Authentic peroxynitrite	
	Acetone/CH <sub>2</sub> Cl <sub>2</sub> ext.	MeOH ext.	Acetone/CH <sub>2</sub> Cl <sub>2</sub> ext.	MeOH ext.
Green algae				
<i>Spongomorpha arcta</i>	50.2	28.0	26.0	2.7
<i>Monostroma obscurum</i>	41.3	3.0	5.5	-
Brown algae				
<i>Alaria esculenta</i>	46.3	34.0	12.1	10.6
<i>Laminaria digitata</i>	32.8	29.5	-	-
<i>Laminaria saccharina</i>	25.6	16.3	-	-
<i>Desmarestia aculeata</i>	33.4	46.0	-	-
<i>Laminaria solidungula</i>	47.8	29.1	8.7	-
<i>Chorda filum</i>	44.6	24.8	-	5.0
<i>Fucus distichus ssp edentatus</i>	32.4	19.4	7.4	-
<i>Chordaria flagelliformis</i>	43.4	15.5	13.0	-
<i>Laminaria sp. 1</i>	24.3	5.8	6.1	5.8
<i>Laminaria sp. 2</i>	25.4	12.3	3.3	0.6
<i>Laminaria hyperborea</i>	26.5	8.4	-	-
Red algae				
<i>Ptilota seriata</i>	38.8	6.4	4.5	7.4
<i>Polysiphonia stricta</i>	61.8	32.4	3.5	17.2
<i>Phycodrys rubens</i>	45.9	20.4	-	6.5
<i>Devaleraea ramentacea</i>	41.3	27.4	-	1.7
<i>Odonthalia dentata</i>	54.6	64.2	24.3	23.0
<i>Callophlilis cristata</i>	28.5	19.6	15.2	-
<i>Palmaria palmata</i>	21.2	-	3.0	-
Penicillamine	88.2		90.4	
L-ascorbic acid	93.5		98.1	

**Table 2.** Effects of solvent fractions from polysiphonia stricta and Odonthalia dentata on peroxynitrite from decomposition of SIN-1 and authentic peroxynitrite (10 µg/ml)

Seaweeds	ONOO <sup>-</sup> from decomposition of SIN-1				Authentic peroxynitrite			
	<i>n</i> -hexane	85% aq. MeOH	<i>n</i> -BuOH	H <sub>2</sub> O	<i>n</i> -hexane	85% aq. MeOH	<i>n</i> -BuOH	H <sub>2</sub> O
<i>Polysiphonia stricta</i>	38.8	82.1	60.2	32.0	14.7	52.9	60.7	20.2
<i>Odonthalia dentata</i>	31.3	82.9	57.7	25.6	17.4	37.8	27.0	-
Penicillamine		88.2				90.4		
L-ascorbic acid		93.5				98.1		

On the basis of the above results, these two seaweed extracts which are the most active were fractionated with *n*-hexane, 85% aq. MeOH, *n*-BuOH and H<sub>2</sub>O and then all fractions were tested on peroxynitrite *in vitro* (Table 2). The 85% aq. MeOH fractions of both *Odonthalia dentata* and *Polysiphonia stricta* exhibited the best activities for the induced peroxynitrite from SIN-1 and authentic peroxynitrite, thus suggesting these fractions may contain

antioxidative components with moderately polar chemical features.

In <sup>1</sup>H NMR spectrum of each of their 85% aq. MeOH fractions, several signals in the region of δ 3.2-4.5 as well as one strong signal at ~δ 1.3 showed that compounds which have sugar moieties and long aliphatic chains exist as major components. In addition, a few peaks in downfield region of δ 6.4-7.4 indicated a possibility that

aromatic compounds, probably phenol derivatives, may exist.

Although secondary metabolites from the red alga *Polysiphonia stricta* have never been reported, brominated phenols from other species of the genus of *Polysiphonia* were isolated as secondary metabolites (Pedersen 1978; Akinin et al. 1992; Flodin and Whitfield 2000). In case of the red alga *Odonthalia dentata*, brominated phenols including lanosol were also isolated and biological activities of the lanosol were investigated (Craigie and Gruenig 1967; Kurata and Taniguchii 1997; DeBusk et al. 2000). Phenol derivatives are well-known as free radical scavengers which play an important role in preventing oxidative damage (Yokozawa et al. 1998). The presence of a hydroxy group in the phenol ring plays an important role in the antioxidant activity in the ONOO systems (Rice-Evans et al. 1996; Choi et al. 2002).

The results of this study indicate that *Odonthalia dentata* and *Polysiphonia stricta* may be useful as source for the treatment of oxidative damage by reactive nitrogen species, particularly in pathological conditions. To the best of our knowledge, this is the first report on the antioxidative activity of seaweeds collected from the Arctic. However, the components responsible for the antioxidant activity of both seaweed species are currently unclear at this stage. Therefore, it is suggested that further work should be performed on the isolation and identification of the antioxidant components in their active extracts.

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