

## Real Time Scale Measurement of Inorganic Phosphate Release by Fluorophore Labeled Phosphate Binding Protein

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Fluorescence change of coumarin labeled phosphate binding protein (PBP-MDCC) was monitored to measure the amount of released inorganic phosphate ( $P_i$ ) during nucleoside triphosphate (NTP) hydrolysis reaction. After purification of PBP-MDCC, fluorescence emission spectra showed that fluorescence responded linearly to  $P_i$  up to about 0.7 molar ratio of  $P_i$  to protein. The correlation of fluorescence signal and  $P_i$  standard was measured to obtain  $[P_i]$  - fluorescence intensity standard curve on the stopped-flow instrument. When T7 bacteriophage helicase, double-stranded DNA unwinding enzyme using dTTP hydrolysis as an energy source, reacted with dTTP, the change of fluorescence was able to be converted to the amount of released  $P_i$  by the  $P_i$  standard curve.  $P_i$  release results showed that single-stranded M13 DNA stimulated dTTP hydrolysis reaction several folds by T7 helicase. Instead of end point assay in NTP hydrolysis reaction, real time  $P_i$ -release assay by PBP-MDCC was proven to be very easy and convenient to measure released  $P_i$ .

**Key words** – phosphate binding protein, inorganic phosphate, T7 bacteriophage helicase, stopped-flow instrument

Inorganic phosphate ( $P_i$ ) is ubiquitous and very common constituent of living organisms. In fact, many enzymes use NTP as an energy source and obtain the driving force by NTP hydrolysis[6,7]. Actin, myosin, and motor proteins are good examples for the generation of forces by NTP hydrolysis[13,14]. Therefore, the ability to measure the amount of released  $P_i$  and monitor the changes of  $P_i$  during the NTP hydrolysis reaction contributes to understand an involved biological process. In fact, a variety of spectrophotometric assays have been developed to measure the released  $P_i$  quantitatively. They include the development of color after phosphomolybdate complex formation and enzymatic reaction by phosphorylation of fluorescent substrate[2,5]. Many kinetic studies of  $P_i$  release were done by these assays. In addition, another type of  $P_i$  release assay was also developed to follow rapid  $P_i$  release in real time. The assay method is based on *E. coli* phosphate binding protein (PBP), the product of the *phoS* gene, which is a member of proteins that are expressed and transported to the periplasm under the condition of  $P_i$  starvation[3]. PBP is a monomeric protein with a molecular mass of 35 kDa and the crystal structure for the  $P_i$  bound form revealed two domains with a  $P_i$ -binding cleft between

them[12]. The single cysteine mutant (A197C) form of PBP made it possible to label selectively with a fluorophore, N-[2-(1-maleimidyl)ethyl]-7-(diethylamino)coumarin-3-carboxamide (MDCC). The resultant MDCC labeled PBP (PBP-MDCC) undergoes a large conformation change on  $P_i$  binding, so fluorescence signal is very sensitive to binding and amount of released  $P_i$ . PBP-MDCC is reported as rapid ( $1.36 \times 10^8 \text{ M}^{-1}\text{s}^{-1}$ ) and tight ( $K_d$  of  $\sim 0.1 \mu\text{M}$ ) binding to  $P_i$ [3], so that it is very sensitive in the nanomolar range of  $P_i$ .

Here, PBP-MDCC was used to measure the amount of released  $P_i$  from dTTP hydrolysis by T7 bacteriophage helicase. T7 helicase is a motor protein that translocates along DNA during DNA replication and recombination and unwinds the complementary DNA strand[14]. T7 helicase assembles into hexameric ring in the presence of NTP, preferably dTTP, and the hexamer has a high affinity for single-stranded DNA that binds within the central channel of the ring. The energy source for T7 helicase is dTTP and, whose function is equivalent to that of ATP in other motor proteins. The chemical energy of dTTP hydrolysis is used to move the T7 helicase unidirectionally along the DNA. Therefore accurate measurement of dTTP hydrolysis product is the first step to investigate the dTTP hydrolysis mechanism of T7 helicase.

Using a real time fluorescence stopped-flow assay, re-

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leased  $P_i$  from dTTP hydrolysis by T7 helicase was measured in the absence and presence of DNA. The results showed that PBP-MDCC turned out to be a good probe to follow rapid  $P_i$  changes in real time experiments.

## Materials and Methods

### Reagents

Bacterial purine nucleoside phosphorylase (PNPase), 7-methylguanosine (MEG), dTTP, and the standard solution of  $P_i$  were purchased from Sigma Chemicals. N-[2-(1-maleimidyl)ethyl]-7-(diethylamino)coumarin-3-carboxamide (MDCC) was purchased from Molecular Probes.

### Purification of MDCC-labeled phosphate binding protein

The single cysteine mutant (A197C) phosphate binding protein (PBP) was expressed and purified as described[3] with the following modifications. Overnight grown cells in LB medium containing 12.5 mg/l tetracycline were placed in four liters of minimal media containing 100 mM Hepes (pH 7.5), 20 mM KCl, 15 mM  $(NH_4)_2SO_4$ , 1 mM  $MgCl_2$ , 10  $\mu M$   $FeSO_4$ , 1  $\mu g/ml$  thiamine, 0.25% (v/v) glycerol, 2 mM  $KH_2PO_4$ , and grown for two hours. 200 mM rhamnose was added to the minimal media, and cells were grown overnight at 37°C. The cells were pelleted by centrifugation and resuspended in a Tris/HCl buffer (10 mM, pH 7.6) including 30 mM NaCl. PBP was released from the cells' periplasm by osmotic shock[4], and the periplasmic extract was applied to a Q-Sepharose column (100 ml) that was equilibrated with 10 mM Tris/HCl (pH 7.6). PBP was eluted as the major protein with a linear gradient of 0 - 200 mM NaCl in the same buffer. PBP concentration was determined by absorbance measurement at 280 nm using an extinction coefficient of  $\lambda_{280nm} = 60,880 M^{-1}cm^{-1}$ .

Labeling of the A197C PBP protein with MDCC and further purification of the labeled protein (PBP-MDCC) were performed as described[3] except 200  $\mu M$  of MEG and 0.2 units/ml of PNPase were included in the labeling reaction. The labeled protein after purification showed a 280 nm/430 nm absorbance ratio of 1.6, suggesting that most of PBP was labeled with MDCC[3].

### T7 helicase protein and buffer

The T7 helicase used in this study is gp4A', which is a M64L mutant of T7 helicase-primase protein that was

over-expressed and purified as described previously[8,15]. The protein concentration was determined both by absorbance measurements at 280 nm in 8 M urea (the extinction coefficient is  $76,100 M^{-1}cm^{-1}$ ), and by the Bradford assay using bovine serum albumin as a standard. Both methods provided similar protein concentrations. Tris buffer was used throughout the experiments unless specified otherwise, which contained 50 mM Tris/Cl (pH 7.6), 40 mM NaCl, and 10% (v/v) glycerol.

### Fluorescence titrations of $P_i$ with PBP-MDCC

1 ml of PBP-MDCC solution (5  $\mu M$  PBP-MDCC, 300  $\mu M$  MEG, and 0.5 unit/ml PNPase) was titrated with standard  $P_i$  solution and fluorescence emission spectra were obtained. Fluorescence was corrected with respect to volume change. Fluorescence emission at 465 nm was plotted versus  $P_i$  concentration to create correlation curve.

### Real time measurement of released $P_i$

Fluorescence change experiments were performed using a stopped-flow instrument. Inorganic phosphate release reactions were performed in Tris buffer at 18°C. Since PBP-MDCC is sensitive to nanomolar quantities of  $P_i$  ubiquitously contaminated in all solutions, a coupled enzyme reaction ( $P_i$ -mop: 0.5 units/ml PNPase and 300  $\mu M$  MEG) was used to sequester  $P_i$  chemically. Assay conditions were adjusted to ensure that the  $P_i$ -mop did not compete with PBP-MDCC for phosphate ( $k_{cat}/K_m$  for the PNPase reaction with  $P_i$  is  $3.2 \times 10^6 M^{-1}s^{-1}$ )[1]. The fluorescence signal of PBP-MDCC was calibrated using  $P_i$  standard on the stopped-flow apparatus. The amplitude of fluorescence increase was measured by conducting a control experiment in the absence of  $P_i$  and subtracting the maximum fluorescence of the control from the one with a known concentration of  $P_i$ . The amplitude thus calculated was plotted versus  $P_i$  concentration to create the calibration curve. The calibration curve was created using the same photomultiplier tube (PMT) voltage as used in the subsequent experiment, just before a set of  $P_i$  release experiments were performed. The slope was used to convert the observed fluorescence amplitude into molar  $P_i$ .

A 40  $\mu l$  solution containing T7 helicase (0.4 and 0.2  $\mu M$  hexamer in the absence and presence of single stranded DNA, respectively), EDTA (5 mM),  $P_i$ -mop (0.5 units/ml PNPase with 300  $\mu M$  MEG), and various concentrations of dTTP in the absence of DNA and presence of single-stranded (ss) M13 DNA (3 nM) was rapidly mixed with 40

μl of MgCl<sub>2</sub>, P<sub>i</sub>-mop and 10 μM PBP-MDCC in the stopped-flow instrument at 18°C. The fluorescence changes of PBP-MDCC were monitored using an excitation wavelength of 425 nm and monitoring the emission above 450 nm using a cut-off filter (Corion LL-450 F). For each experiment, at least four fluorescence traces were averaged. The fluorescence changes were converted to the concentration of released P<sub>i</sub> using the standard curve. The concentration of released P<sub>i</sub> per T7 helicase hexamer was plotted versus time of reaction. A control experiment with all the components except Mg<sup>2+</sup> was performed and subtracted. The resulting curves were fit to Eq. 1.

$$y = A[1 - \exp(-kt)] + mt \tag{Eq. 1}$$

Where *y* is observed fluorescence or molar amount of P<sub>i</sub> released at time *t*, *A* and *k* are amplitude and rate of the burst phase, respectively; and *m* is the rate of the linear steady-state phase.

## Results and Discussion

### Titration of PBP-MDCC with P<sub>i</sub>

When P<sub>i</sub> was added to PBP-MDCC and the sample was excited at 425 nm, the emission maximum shifted from 475 nm to 465 nm and the fluorescence intensity increased about 6 fold (Fig. 1A). The titration curve with P<sub>i</sub> showed that fluorescence responded linearly to P<sub>i</sub> up to about 0.7 molar ratio of P<sub>i</sub> to protein. (Fig. 1B). These measurements are in agreement with those reported previously[3]. The crystal structure of PBP in the absence and presence of P<sub>i</sub> shows that conformational change between two structures make a cleft in the absence of ligand[11,12]. PBP consists

of two domains with P<sub>i</sub> situated at the domain interface[10]. Mutagenesis of wild type PBP to A197C PBP and MDCC labeling of A197C PBP resulted in structures that were identical to wild type PBP[12,17]. Labeling of PBP with MDCC is known as not disrupting either its overall folding or P<sub>i</sub> binding site[10]. As in wild type PBP, P<sub>i</sub> is monobasic and makes 12 hydrogen bonds with PBP-MDCC[10]. An interesting feature of the PBP-MDCC structure is its similarity to the wild type PBP at the MDCC binding site, which shows no major readjustment of either main chain or side chain residues responsible for anchoring coumarin. This is consistent with the observation that the pattern of P<sub>i</sub> binding to wild type of PBP and PBP-MDCC is very similar[4]. Whereas the protein cleft is open in the absence of P<sub>i</sub>, P<sub>i</sub> binding causes cleft closure with changes in fluorescence.

Plotting fluorescence amplitude versus concentration of P<sub>i</sub> created the P<sub>i</sub> standard curve (Fig. 2). The slope was

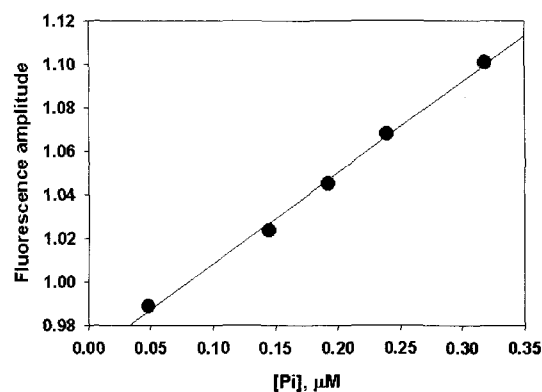


Fig. 2. P<sub>i</sub> standard curve. The dependence on [P<sub>i</sub>] of the fluorescence increase amplitude is shown. The amount of released P<sub>i</sub> can be calculated from the slope of the line. The slope is 0.42 fluorescence/[P<sub>i</sub>].

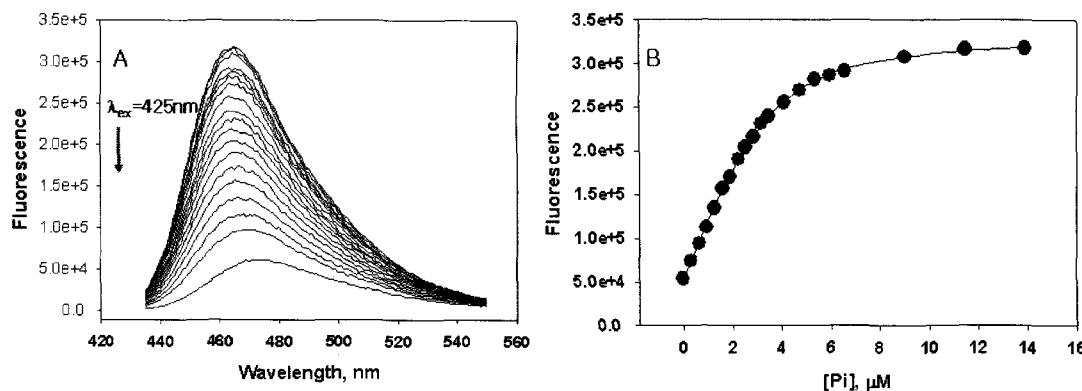


Fig. 1. Titration of PBP-MDCC with P<sub>i</sub>. (A) Fluorescence emission spectra of PBP-MDCC in the presence of P<sub>i</sub>. 5 μM PBP-MDCC in Tris buffer with 300 μM MEG and 1.0 unit/ml PNPase was titrated with standard P<sub>i</sub> solution. Samples were excited at 425 nm. (B) Correlation between [P<sub>i</sub>] and fluorescence intensity at 465 nm.

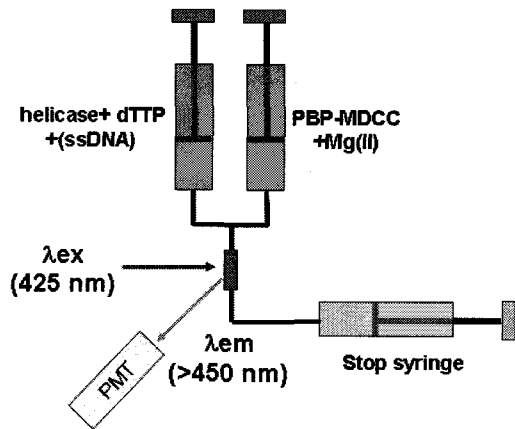


Fig. 3. Design of the stopped-flow  $P_i$ -release experiment. The two driving syringes are compressed to mix solutions, and then the mixture ages with time and the detector monitors the fluorescence change.

0.42 fluorescence/ $[P_i]$ . PMT voltage was same throughout the experiments.

#### Measurement of $P_i$ release in a stopped-flow instrument

The stopped-flow instrument rapidly mixes reactants on a millisecond time scale and optical signal is obtained. Here, the optical signal is fluorescence and the goal in a stopped-flow experiment is that fluorescence signal correlates with  $P_i$  release in the reaction. The amount of released  $P_i$  can be measured in real time in a stopped-flow instrument by following the increase in fluorescence of

MDCC[3]. In the presence of PBP-MDCC, dTTP-hydrolyzing protein, T7 helicase, was rapidly mixed in the stopped-flow instrument with  $MgCl_2$  as shown in Fig. 3. T7 helicase assembles into hexamer in the presence of dTTP and binds ssDNA[8]. However  $Mg^{2+}$  is not required for hexamer formation and dTTP hydrolysis by preformed hexameric T7 helicase is negligible in the absence of  $Mg^{2+}$  (data not shown). T7 helicase was preincubated with various concentrations of dTTP (in the absence of  $Mg^{2+}$ ) and mixed with  $MgCl_2$  and PBP-MDCC in a stopped-flow instrument at 18°C. T7 helicase hexamer·dTTP complex was preformed and dTTP hydrolysis reaction was initiated by addition of  $MgCl_2$ . The magnitude of fluorescence increase was converted to  $P_i$  concentration, which was plotted as a function of time (Fig. 4A). A burst increase of  $P_i$ -release was observed at all concentration of dTTP. This indicate that the  $P_i$ -release rate is faster than the dTTP hydrolysis turnover rate, which is governed by the dTTP release rate or dTTP rebinding. When the slopes of the linear phase were plotted as a function of  $[dTTP]$ , they increased in a hyperbolic manner (Fig. 4B). Therefore rate-limiting step shifts from dTTP binding to product release. The hyperbolic curve provided the  $V_{max}$  of  $1.82 \pm 0.06 \text{ s}^{-1}$  and  $K_m$  of  $16.5 \pm 2.5 \mu\text{M}$ .

T7 helicase with dTTP in the presence of single-stranded M13 DNA was mixed with  $MgCl_2$  and PBP-MDCC in a stopped-flow instrument at 18°C, and the  $P_i$ -release was measured as a function of time. Fig. 5A shows time tra-

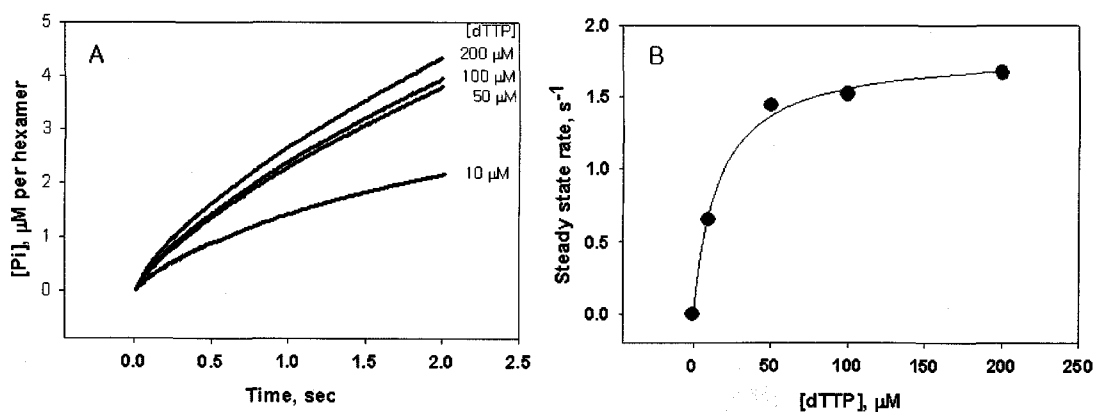


Fig. 4.  $P_i$ -release from dTTP hydrolysis by T7 helicase in the absence of DNA. (A) T7 helicase (0.4  $\mu\text{M}$  hexamer), EDTA (5.0 mM),  $P_i$ -mop, and dTTP (5~400  $\mu\text{M}$ ) was mixed with  $MgCl_2$  (9.1 - 9.4 mM) and PBP-MDCC (10  $\mu\text{M}$ ) in a stopped-flow instrument at 18°C. The final concentrations of helicase and free  $MgCl_2$  were 0.2  $\mu\text{M}$  hexamer and 2 mM, respectively. The fluorescence ( $\text{em} > 450 \text{ nm}$ ) upon excitation at 425 nm was measured as a function of time. Molar amounts of  $P_i$  released per mole of hexameric helicase were shown. The  $P_i$  release curve fit best to the burst equation (Eq. 1). (B) The  $[dTTP]$  dependence of the steady state rate is shown. It shows a hyperbolic trend toward saturation with increasing  $[dTTP]$ .  $V_{max}$  and  $K_m$  are  $1.82 \pm 0.06 \text{ s}^{-1}$  and  $16.5 \pm 2.5 \mu\text{M}$ , respectively.

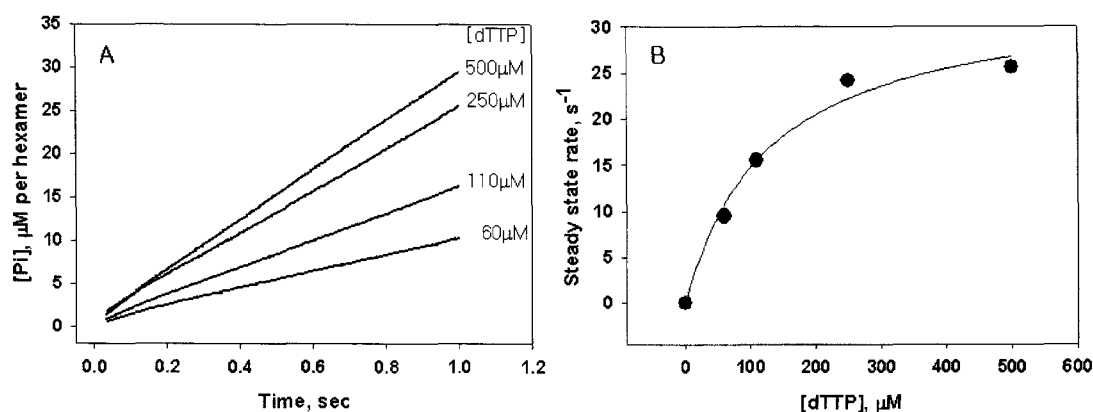


Fig. 5.  $P_i$ -release from dTTP hydrolysis by T7 helicase in the presence of single stranded M13 DNA. (A) T7 helicase (0.2  $\mu\text{M}$  hexamer), EDTA (5.0 mM),  $P_i$ -mop, M13 ssDNA (3 nM), and dTTP (120~1000  $\mu\text{M}$ ) was mixed with  $\text{MgCl}_2$  (45.2~48.2 mM) and PBP-MDCC (10  $\mu\text{M}$ ) in a stopped-flow instrument at 18°C. The final concentrations of helicase and free  $\text{MgCl}_2$  were 0.1  $\mu\text{M}$  hexamer and 20 mM, respectively. The fluorescence ( $\text{em}>450$  nm) upon excitation at 425 nm was measured as a function of time. Molar amounts of  $P_i$  released per mole of hexameric helicase were shown. (B) The steady state rates from slope of linear phase in (A). It shows a hyperbolic trend toward saturation with increasing [dTTP].  $V_{\text{max}}$  and  $K_m$  are  $33.7 \pm 3.1 \text{ s}^{-1}$  and  $129.5 \pm 31.9 \mu\text{M}$ , respectively.

jectories of  $P_i$  release at different [dTTP]. When the slopes of the linear phase were plotted as a function of [dTTP], they increased in a hyperbolic manner (Fig. 5B). The hyperbolic curve provided the  $V_{\text{max}}$  of  $33.7 \pm 3.1 \text{ s}^{-1}$  and  $K_m$  of  $129.5 \pm 31.9 \mu\text{M}$ .

Stopped-flow experiments showed that  $P_i$  release can be measured even millisecond scale of dTTP hydrolysis reaction. As shown in Fig. 4A and Fig. 5A, released  $P_i$  was about 2.3  $\mu\text{M}$  at 1.0 s of 100  $\mu\text{M}$  dTTP hydrolysis reaction in the absence of DNA and 16.3  $\mu\text{M}$  at 1.0 s of 110  $\mu\text{M}$  dTTP in the presence of M13 ssDNA, respectively. This indicate that ssDNA stimulate dTTP hydrolysis about 7-fold at 1.0 s reaction. Stimulation by ssDNA is clearly shown in Fig. 5B. Steady state rate reached  $33 \text{ s}^{-1}$  when [dTTP] increased up to 500  $\mu\text{M}$  in the presence of M13 ssDNA. However steady state rate reached only  $1.7 \text{ s}^{-1}$  in the absence of DNA. The dTTP hydrolysis activity of T7 helicase is greatly stimulated by the presence of ssDNA, which is consistent with previous studies[15,16].

Here, the procedures and applications of PBP-MDCC as a  $P_i$  probe were described by dTTP hydrolysis reaction of T7 helicase. Instead of using radioactive material, fluorescence change made it possible to monitor real-time  $P_i$  release reaction by PBP-MDCC. Therefore it is expected to study the mechanism of nucleotide hydrolyzing enzymes in detail on a rapid time scale by real time measurement of  $P_i$  release.

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#### 초록 : 형광단이 붙어 있는 인산결합 단백질에 의한 인산 배출의 실시간 측정

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Coumarine이 부착된 인산결합 단백질 (PBP-MDCC)의 형광변화가 뉴클레오타이드 삼인산 가수분해과정에서 배출된 무기 인산의 양을 측정하기 위해 관찰되었다. PBP-MDCC 정제후, 형광 방출 스펙트럼은 형광세기 PBP-MDCC의 몰비율로 약 70%까지 직선형태로 증가하는 것을 보였다. 형광 신호와 인산 기준물질과의 상호관계 측정이 인산 농도-형광세기 표준곡선을 구하기 위하여 stopped-flow 기구에서 행하여졌다. dTTP 가수분해로부터 나오는 에너지를 이용하여 이중나선 DNA를 풀어주는 단백질인 T7 박테리오파지 나선효소를 dTTP와 반응시켰을때, 형광변화를 배출된 인산의 양으로 전환할수 있었다. 인산 배출 결과는 단일가닥 M13 DNA가 T7 나선효소에 의한 dTTP 가수분해반응을 여러배 증가시키는 것을 보인다. 뉴클레오타이드 삼인산 가수분해 반응에 있어서 종말점 분석 대신에, PBP-MDCC에 의한 연속적인 인산 배출 분석이 배출된 인산을 측정하는데 있어서 쉽고 편리한 방법임을 보였다.