

Synthesis and Biological Evaluation of New Thiazolopyrimidines

Makarem Said, Khaled Abouzid¹, Ashraf Mouneer, Ali Ahmedy², and Abdel Moneim Osman³

Department of Organic Chemistry, Faculty of Pharmacy, Cairo University, ¹Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Ain-Shams University, ²Department of Microbiology, Faculty of Pharmacy, Cairo University, and ³Pharmacology Unit, NCI, Cairo University, Egypt

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In this study, a series of 4-amino-5-cyano-3-substituted-2,3-dihydrothiazol-2-thiones (1a-c), as well as their triazolo and triazinopyrimidine derivatives such as 8-substituted-3-benzyl-5-methylthiazolo[5,4-e][1,2,4] triazolo[1,5-c]pyrimidin-2-thiones (4-6, 10) and 3-benzyl-5-methyl thiazolo[5,4-e]pyrimidino[3,4-b][1,2,4]triazin-2-thiones (7a-b) were prepared as potential antimicrobial and antitumor agents. Some of the tested compounds showed promising antimicrobial activity and non of them showed any appreciable antitumor activity.

Key words: Thiazolopyrimidines, Antimicorbial activity, Antitumor activity

INTRODUCTION

Fused pyrimidines are an important class of compounds that attracted the attention of medicinal chemists as chemotherapeutic agents. The antibacterial (Bayomi et al., 1993; Holla et al., 1994) antimicrobial (El-Sherbeny et al., 1995; Hozien et al., 1997, Ghorab and El-Batal, 2002), antiviral (Kharizomenova et al., 1982; Erik De Clereq, 1986; El-Sherbeny et al., 1995) and cytotoxic properties Ogawa et al., 1986; Moharram et al., 1990; Rizk et al., 1993) are well documented. Recently, we have prepared and evaluated a new series of heterocyclic compounds containing a pyrimidine nucleus (Mouneer et al., 2002) which has been reported to possess promising antimicrobial activity. In our on going efforts for finding new compounds for the antimicrobial activity and cancer chemotherapy, a new series of 3-benzyl-5-methylthiazolo[5,4e][1,2,4]triazolo[1,5-c]pyrimidin-2-thiones and 3-benzyl-5methylthiazolo[5,4-e]pyrimidino[3,4-b][1,2,4]triazin-2-thiones have been synthesized to explore their antimicrobial and antitumor activity.

Correspondence to: Khaled A. M. Abouzid, Pharmaceutical Chemistry Department, Faculty of Pharmacy, Ain Shams University, 11566, Cairo, Egypt. Fax: 0202-5056996, 6831492

E-Mail; abouzid@yahoo.com

MATERIALS AND METHODS

Chemistry

Melting points were determined on a Griffen melting point apparatus and are uncorrected. Elemental analyses were carried out at the Microanalytical Unit of Cairo University. Infrared spectra were recorded using the Shimadzu IR 435 Spectrometer with KBr discs. Proton magnetic resonance was obtained in DMSO- d_6 using TMS as internal standard using the Jeol FX 90 Q (90 MHz) Spectrometer. Mass spectra were recorded on the HP Model-MS-5988 Mass Spectrometer. Thin layer chromatography was performed on pre-coated silica gel F-254 plates and visualized by the UV lamp.

3-Substituted-4-amino-5-Cyano-2,3-dihyrothiazol-2-thiones (1a-c)

A mixture of malononitrile (0.05 mmol), substituted isothiocyanate (0.05 mmol), finely divided sulphur (0.05 gm/ atom) in dimethylformamide (6 mL) and triethylamine (6 mL) was heated up to 50 °C for 1 h. Then the mixture was poured onto ice cold water; the solid was filtered, washed and crystallized with the proper solvent (Table I, II).

3-Benzyl-5-cyano-4-(α -ethoxyethylidene-aminothia-zolin-2-thione (2)

A mixture of compound 1a (2 mmol), triethyl orthoacetate

(1.5 mL) and acetic anhydride (6 mL) was heated under reflux for 6 h. The reaction mixture was concentrated *in vacuo* and residue obtained was triturated with petroleum ether. The separated solid was filtered and recrystallized with the appropriate solvent (Table I, II).

6-Amino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetrahydrothiazolo[4,5-d] pyrimidin-2-thione (3)

Hydrazine hydrate (1 mL) was added to a solution of 3-benzyl-5-cyano-4-(α -ethoxyethylideneaminothiazolin-2-thione (2) (10 mmol) in ethanol (25 mL). The reaction mixture was stirred for 4 h below 30 °C. The separated solid was filtered, washed with water, dried and recrystallized from the appropriate solvent (Table I, II).

Table I. Physical, analytical and IR spectral data of the synthesized compounds

Comp. No.	m.p. (°C) Solvent of crystallization	Yield %	Mol. Form. (Mol. Wt.)	Elememtal analysis % calcd./found			IR VPa (see 1)					
				С	Н	N	- KBr (cm ⁻¹)					
1a	210 (ethanol)	56	$C_{11}H_9N_3S_2$ (247)	53.44 53.40	3.64 3.60		3200-3300 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 2205 (C=N), 1590-1450 (C=C str. of aromatic ring)					
1b	183 (ethanol)	58	$C_{10}H_6BrN_3S_2$ (312)	38.46 38.40	1.92 1.90		3200-3300 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 2205 (C=N), 1590-1450 (C=C str. of aromatic ring)					
1c	285 (ethanol)	50	$C_{10}H_{13}N_3S_2$ (239)	50.20 50.10	5.43 5.42	17.57 17.50	3200-3300 (NH), 3050 (CH aromatic), 2950-2700 (CH aliph.), 2205 (C=N)					
2	178(pet. ether/ methylene chloride)	56	C ₁₅ H ₁₅ N ₃ OS ₂ (317)	56.78 56.76	4.73 4.70		3050 (CH aromatic), 2950-2850 (CH aliph.), 2200 (C=N), 1590-1450 (C=C str. of aromatic ring)					
3	220 (ethanol/methylene chloride)	61	C ₁₃ H ₁₃ N ₅ S ₂ (303)	51.48 51.30	4.29 4.10		3300-3100 (NH, NH ₂), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
4	143 (ethanol)	57	C ₁₅ H ₁₃ N ₅ S ₂ (327)	55.04 54.90	3.97 3.80		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
5	180 (chloroform)	64	C ₁₄ H ₁₁ N ₅ S ₂ (313)	53.67 53.50	3.51 3.40		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
6	238 (ethanol/DMF	64	C ₂₀ H ₁₆ N ₆ S ₂ (404)	59.40 59.30	3.96 3.80		3280 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
7a	138 (ethanol)	61	C ₂₁ H ₁₇ N ₅ S ₂ (403)	62.53 62.40	4.21 4.20		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
7b	145 (ethanol)	60	C ₂₁ H ₁₆ BrN ₅ S ₂ (482)	52.28 52.20	3.31 3.30		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
8	232 (pet. Ether)	76	C ₁₆ H ₁₇ N ₅ S ₂ (343)	55.97 55.70	4.95 7.80		3250 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1635 (C=N), 1590-1450 (C=C str. of aromatic ring)					
9a	236 (ethanol)	80	C ₂₀ H ₁₆ BrN ₅ S ₂ (470)	51.06 51.00	3.40 3.30		3250 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1630 (C=N), 1590-1450 (C=C str. of aromatic ring)					
9b	261 (ethanol)	85	C ₂₀ H ₁₆ CIN ₅ S ₂ (425.5)	56.40 56.30	3.76 3.70	16.45 16.30	3250 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1630 (C=N), 1590-1450 (C=C str. of aromatic ring)					
9c	>300 (ethanol)	79	C ₂₀ H ₁₆ N ₆ O ₂ S ₂ (436)	55.04 55.00	3.66 3.50		3250 (NH), 3050 (CH aromatic), 2950-2850 (CH aliph.), 1630 (C=N), 1590-1450 (C=C str. of aromatic ring)					
10a	265 (ethanol)	60	C ₂₀ H ₁₄ CIN ₅ S ₂ (423.5)	56.67 56.50	3.30 3.20		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
10b	288 (ethanol)	64	C ₂₀ H ₁₄ FN ₅ S ₂ (407)	58.96 58.80	3.43 3.30		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
10c	270 (ethanol)	62	C ₂₀ H ₁₄ N ₆ O ₂ S ₂ (434)	55.29 55.10	3.22 3.10		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					
10d	279 (ethanol)	60	C ₁₈ H ₁₃ N ₅ S ₃ (395)	54.68 54.50	3.29 3.20		3050 (CH aromatic), 2950-2850 (CH aliph.), 1590-1450 (C=C str. of aromatic ring)					

Table II. 1H-NMR and EIMS spectral data of the synthesized novel compounds

Compound No.	Spectral Data
1a	1 H-NMR (DMSO-d ₆): δ2.8 (s, 2H, CH ₂ Ar), 3.6 (br, 2H, NH ₂ , D ₂ O, exchangeable), 7.3-7.6 (m, 5H, Ar-H). EIMS m/Z: 247M ⁺ , (14.8%), 91 (100%).
1c	1 H-NMR (DMSO-d ₆): δ2.1-2.3 (m, 6H, 3CH ₂) 2.4-2.5 (m, 4H, 2CH ₂), 2.7 (m, 1H, CH), 2.8 (s, 2H, CH ₂ Ar), 3.5 (br, 2H, NH ₂ , D ₂ O, exchangeable).
2	$^{1}\text{HNMR (DMSO-d}_{6}\text{): }\delta\ 0.6-0.9\ (t,\ 3H,\ CH_{2}CH_{3}\text{), }2.00\ (s,\ 3H,\ CH_{3}\text{), }2.8\ (s,\ 2H,\ CH_{2}Ar\text{), }3.1-3.2\ (q,\ 2H,\ CH_{2}CH_{3}\text{), }7.2-7.6\ (m,\ 5H,\ Ar-H).$
3	1 HNMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar) 3.3 (s, 2H, NH ₂) D ₂ O, exchangeable), 6.1 (br, 1H, NH, D ₂ O exchangeable, 7.0-7.5 (m, 5H, Ar-H). EIMS m/Z: 303 M ⁺ , (6.12%), 91.05 (100%).
4	¹ H-NMR (DMSO-d ₆): δ? 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar), 3.1 (s, 3H, C ₈ -CH ₃), 6.9-7.5 (m, 5H, Ar-H).
5	1 H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar) 6.8 (s, 1H, CH) 7.1-7.5 (m, 5H, Ar-H). EIMS m/Z: 313 M ⁺ , (23.7%), 91 (100%).
6	¹ H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar), 6.5 (s, 1H, NH, D ₂ O, exchangeable), 7.0-7.5 (m, 5H, Ar-H).
7a	¹ H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar) 3.3 (s, 2H, CH ₂) 6.9-7.5 (m, 10H, Ar-H).
8	$^{1}\text{H-NMR (DMSO-d}_{6}\text{): }\delta~1.3~(s,~6\text{H},~2\text{CH}_{3}\text{), }2.3~(s,~3\text{H},~\text{CH}_{3}\text{) }2.8~(s,~2\text{H},~\text{CH}_{2}\text{Ar}\text{), }6.3~(s,~1\text{H},~\text{NH},~\text{D}_{2}\text{O},~\text{exchangeable})~7.1-7.5~(m,~5\text{H},~\text{Ar-H}).$
9a	$^{1}\text{H-NMR (DMSO-d}_{6}\text{): }\delta\ 2.3\ (\text{s, 3H, CH}_{3}\text{), }2.8\ (\text{s, 2H, CH}_{2}\text{Ar}\text{), }6.3\ (\text{s, 1H, NH, D}_{2}\text{O, exchangeable})\ 7.0-7.5\ (\text{m, 10H; 9H, Ar-H; 1H, N=CH}).$
9b	1 H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar), 6.4 (s, 1H, NH, D ₂ O, exchangeable) 7.0-7.5 (m, 10H; 9H, Ar-H; 1H, N=CH). EIMS m/Z: 425 M ⁺ , 17.0%, 427 (8.3%), 91 (100%).
10a	¹ H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar), 6.9-7.5 (m, 9H, Ar-H).
10b	1 H-NMR (DMSO-d ₆): δ 2.3 (s, 3H, CH ₃), 2.8 (s, 2H, CH ₂ Ar), 6.9-7.4 (m, 9H, Ar-H). EIMS m/Z: 409 M $^{+2}$, (25.8%), 91 (100%).

3-Benzyl-5,8-dimethyl-2,3-dihydrothiazolo[5,4-e] [1,2,4]triazolo[1,5-c] pyrimidin-2-thione (4)

A mixture of 6-amino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetra hydrothiazolo[4,5-d]pyrimidin-2-thione (3) (2.5 mmol), glacial acetic (4 mL) and acetic anhydride (4 mL) was heated under reflux for 4 h. The separated solid was collected and recrystallized from the appropriate solvent (Table I, II).

3-Benzyl-5-methyl-2,3-dihydrothiazolo[5,4-e] [1,2,4] triazolo[1,5-c]pyrimidin-2-thione (5)

A solution of compound **3** (2.5 mmol) in formic acid (5 mL) was heated under reflux for 10 h. The reaction mixture was concentrated under reduced pressure, water (20 mL) was added and the solution was neutralized with sodium carbonate. The obtained solid was filtered, washed with water and recrystallized with the proper solvent (Table I, II).

3-Benzyl-5-methyl-8-phenylamino-2,3-dihydrothiazolo[5,4-e][1,2,4]triazolo [1,5-c]pyrimidin-2-thione (6)

A mixture of compound **3** (1 mmol) and phenyl isothiocyanate (1 mmol) in absolute ethanol (20 mL) was heated under reflux for 12 h. The product thus formed was collected by filtration and recrystallized from the appropriate solvent

(Table I, II).

8-Aryl-3-benzyl-5-methyl-2,3,9,10-tetrahydrothiazolo [5,4-e]pyrimidino[3,4-b] [1,2,4] triazin-2-thiones (7a & b)

A mixture of compound 3 (5 mmol), the appropriate α -bromoacetophenone (5 mmol) and sodium bicarbonate (5 mmol) in ethanol (10 mL) was heated under reflux for 8 h. After cooling, the separated solid was filtered, washed with water, dried and recrystallized from the appropriate solvent (Table I, II).

6-(Acetone-2-ylidenamino-3-benzyl-7 imino-5-methyl) -2,3,6,7-tetrahydrothiazolo [4, 5-d]pyrimidino-2-thione (8)

A mixture of compound **3** (1 mmol) and acetone (20 mL) was heated under reflux for 4 h. After cooling, the reaction mixture was concentrated to one third of the volume. The product obtained was filtered and recrystallized from the appropriate solvent (Table I, II).

6-Arylidineamino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetrahydrothiazolo[4,5-d]pyrimidin-2-thiones (9a-c)

A mixture of compound 3 (1 mmol), selected aromatic aldehydes (1 mmol) was refluxed in absolute ethanol (20

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mL) for 5 h. After cooling, the product was filtered and recrystallized from the proper solvent (Table I, II).

8-Aryl-3-benzyl-5-methyl-2,3-dihydrothiazolo[5,4-e] [1,2,4]triazolo[1,5-c] pyrimidin-2-thiones (10a-d) Method A:

A mixture of **3** (5 mmol) and the appropriate aldehydes (5 mmol) in acetic acid (15 mL) was heated under reflux for 12 h. The separated solid, upon cooling, was filtered, dried and recrystallized from the proper solvent (Table I, II).

Method B: (for compounds 10b and 10c)

A mixture of **9b** or **9c** (5 mmol) in acetic acid (10 mL) was heated under reflux for 7 h. After cooling, the separated solid was filtered, dried and recrystallized from the appropriate solvent to afford **10b** and **10c** respectively. (Table I, II).

Antimicrobial activity Microorganisms

In vitro antimicrobial activity of the tested compounds was determined by the serial agar micro dilution method (Ahmady et al., 1985). Microorganisms used in the antimicrobial assay were; Mycobacterium phlei, Bacillus subtilis, Staphylococcus aureus, Sarcina lutea, Escherichia coli, proteus vulgaris, Pseudomonas aeruginosa, Candida albicans, Candida tropicals and Tourulopsis glabrata.

Culture media

The Mueller-Hinton agar was used for diluting the bacteria suspension and for two-fold agar dilution of the tested compounds and the Sabouraud liquid medium was used for the yeast-like fungi.

In vitro antimicrobial assay

A series of two-fold agar dilution for each compound was made by dispersing the stock solutions in the remaining wells. The inoculum was prepared by diluting the overnight culture of each microorganism to 10⁶ CFU/mL, and 10 μL of each culture media were transferred on the surface of the died plates (10⁴ CFU/inoculum). The bacterial plates were incubated for 24 h at 37 °C and yeast plates for 48 h at 25 °C. Ofloxacine (OFX) for bacteria and Amphotericin B (Amp. B) for yeast, were tested under the same conditions and were used as positive controls. The minimal inhibitory concentration (MIC) was determined for the selected compounds (1a-c, 3, 7a, 8, 9a, 10b) and compared to that obtained using the positive control (Table III).

Antitumor activity Materials and methods

Animals: Weighing 18-22 g, female Swiss albino mice

Table III. Antimicrobial activity of some synthesized compounds

pd. Mi	nimal	Inhib	itory o	conce	ntrati	on (M	IC, μ	g/LI)	
s 1a	1b	1c	3	7a	8	9a	10b	OFX	AMP B
-	-	128	-	-	64	-	-	2	
-	-	64	128	128	32	128	128	2	
128	64	64	64	64	64	64	64	16	
128	32	32	64	-	32	-	-	8	
64	32	32	64	-	32	-	128	4	
32	32	32	64	64	32	-	128	2	
-	64	64	-	-	64	-	-	2	
128	64	16	128	-	64	-	-		8
64	64	32	64	-	64	-	-		4
128	64	32	64	-	64	-	-		4
	128 128 64 32 - 128 64	128 64 128 32 64 32 32 32 64 128 64 128 64 64 64	1a 1b 1c - - 128 - - 64 128 64 64 128 32 32 64 32 32 32 32 32 - 64 64 128 64 16 64 64 32	1a 1b 1c 3 - - - - 64 128 128 64 64 64 128 64 64 64 64 32 32 64 32 32 32 64 - 64 64 - 128 64 16 128 64 64 32 64	1a 1b 1c 3 7a - - - - - - - 64 128 128 128 64 64 64 64 128 32 32 64 - 64 32 32 64 64 - 64 64 - - 128 64 16 128 - 64 64 32 64 - 64 64 32 64 -	1a 1b 1c 3 7a 8 - - 128 - - 64 - - 64 128 128 32 128 64 64 64 64 64 128 32 32 64 - 32 32 32 32 64 64 32 32 32 64 64 32 - 64 64 128 64 64 64 32 64 64 64 64 32 64 64 64 64 32 64 64	1a 1b 1c 3 7a 8 9a - <td>1a 1b 1c 3 7a 8 9a 10b -<</td> <td>128 64 2 128 64 128 128 64 128 128 128 128 128 64 128 128 128 32 64 64 64 64 16 128 32 64 32 32 128 4 32 32 64 64 32 128 4 32 32 64 64 32 128 2 64 64 64 64 64 2 2 128 64 128 64 64 2 2 64 64 32 64 64 2 2</td>	1a 1b 1c 3 7a 8 9a 10b -<	128 64 2 128 64 128 128 64 128 128 128 128 128 64 128 128 128 32 64 64 64 64 16 128 32 64 32 32 128 4 32 32 64 64 32 128 4 32 32 64 64 32 128 2 64 64 64 64 64 2 2 128 64 128 64 64 2 2 64 64 32 64 64 2 2

Microorganisms selected: A (Escherichia coli) ATCC 10536, B (Proteus vulgaris) NCTC 4175, C (Pseudomonas aurginosa) CNCM A21, D (Staphylococcus aureus) ATCC 4175, E (Sarcina lutea)*, F (Bacillus subtilis) NCTC 6633, G (Mycobacterium pheli*), H (Candida albicans ATCC 60193), I (Candida tropicalis*), J (Torulopsis glabrata*) (-) no activity at MIC>128 µg/mL

OFX (Ofloxacin)

AMP B (Amphotericin B)

from the animal house of Cairo Cancer Institute, were used. Animals were sustained on standard pellet diet and water ad-lib.

Tumor: Ehrlich ascites carcinoma (EAC)

Experiment

A set of sterile test tubes was used, 2.5×10^5 tumor cells per mL (0.1 mL) were suspended in the phosphate buffer saline (0.8 mL). Each tested compound (10 mg) was dissolved in DMSO (0.5 mL) and then diluted with H₂O (9.5 mL). From this solution, 0.1 mL was taken and diluted with H₂O (0.9 mL) to give the required concentration used in the study. The test tubes were incubated for 2 h at 37 °C. The trypan blue exclusion test (Mclimans *et al.*, 1957) was then carried out to calculate the percentage of non viable cells and the activity was compared with Doxorubicin (40 μ g/mL).

^{*}Strains of laboratory collection.

RESULTS AND DISCUSSION

Chemistry

A series of 4-amino-5-cyano-3-substituted 2,3-dihydrothiazol-2-thiones (**1a-c**) was prepared according to the reported method (Gewald, 1966). 4-Amino-3-benzyl-5-cyano-2,3-dihydrothiazol-2-thione (**1a**) was reacted with triethyl orthoacetate in acetic anhydride to yield 3-benzyl-5-cyano-4-(α -ethoxyethylideneamino)thiazolin-2-thione (**2**). Subsequently, the latter was cyclocondensed with hydrazine hydrate to afford 6-amino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetrahydrothiazolo[4,5- α]pyrimidin-2-thione (**3**) (Scheme 1).

Treatment of the latter with a mixture of acetic anhy-

Scheme 1. Synthesis of compounds 3a-c

Scheme 2. Sythesis of new thiazolopyrimidines from 3

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dride and glacial acetic acid under reflux yielded 3-benzyl-5,8-dimethyl 2,3-dihydrothiazolo[5,4-e][1,2,4]triazolo[1,5-c] pyrimidin-2-thione (**4**). On the other hand, when compound **3** was allowed to react with formic acid, it produced 3-benzyl-5-methyl-2,3-dihydrothiazolo[5,4-e][1,2,4]triazolo [1,5-c] pyrimidin-2-thione (**5**). Treatment of 3 with phenyl isothiocyanate, afforded 3-benzyl-5-methyl-8-phenylamino-2,3-dihydrothiazolo[5,4-e][1,2,4]triazolo[1,5-c]pyrimidin-2-thione (**6**).

8-Aryl-3-benzyl-5-methyl-2,3,9,10-tetrahydrothiazolo [5,4-e]pyrimidino[3,4-b][1,2,4]triazin-2-thiones (**7a** and **7b**), Scheme II, were obtained by reacting compound **3** with the appropriately 4-substituted α-bromoacetophenones in the presence of sodium bicarbonate. The latter reaction proceeds in basic medium, *via N*-phenacylimino intermediates which finally cyclized to the corresponding triazine derivatives **7a** and **7b**. Condensation of compound **3** with acetone in ethanol brought about 6-acetone-2-ylidenamino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetrahydrothiazolo[4,5-*d*]pyrimidin-2-thione (**8**) (Scheme II). On the other hand, reacting compound **3** with certain aldehydes in ethanol gave 6-arylidineamino-3-benzyl-7-imino-5-methyl-2,3,6,7-tetrahydrothiazolo[4,5-*d*]pyrimidin-2-thiones (**9a-c**).

8-Aryl-3-benzyl-5-methyl-2,3-dihydrothiazolo[5,4-e][1,2,4] triazolo[1,5-c]pyrimidin-2-thiones (10a-d) were obtained by two pathways. The first pathway involved reflux of 3 with the appropriate aldehyde in acetic acid for 12 h. The second method involved heating compound 9a or 9c under reflux in the same solvent for 7 h. Compounds 10a and 10c obtained by the two pathways were identical and verified by spectral data. Compounds 10b and 10d were prepared as depicted in Scheme II.

Antimicrobial activity

The results of antimicrobial testing revealed that compounds (1c, 3, and 8) were the most active and also showed a broad spectrum antimicrobial activity against both Gram negative, Gram positive as well as fungal species. Both compounds 1c and 8 showed moderate activity against mycobacterial species. On the other hand, compounds 1a and 1b exhibit also moderate antimicrobial activity, of which was more active on Gram positive than on Gram-negative bacteria. On the other hand, products 7a, 9a, and 10b showed little or no activity against the Gram-positive bacteria and moderate effect against the Gram negative ones. It was interesting to find out that all the tested compounds possessed anti-pseudomonal activity.

Antitumor activity

Some representative compounds (1a-c, 3, 6, 7a, 8, 9a, 9c, and 10a-c) have been tested for their antitumor activity and they showed no actions against the growth of Ehrlich ascites carcinoma (EAC) cells even at high

compound concentration levels.

Antitumor activity of the tested drugsusing (EAC)

None of the tested compounds showed any appreciable activity on EAC.

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