

A New Pterocarpan, (-)-Maackiain Sulfate, from the Roots of *Sophora subprostrata*

Jeong An Park^{1,2}, Hyoung Ja Kim¹, Changbae Jin¹, Kyung-Tae Lee², and Yong Sup Lee¹

¹Division of Life Sciences, Korea Institute of Science and Technology, P.O. Box 131, Cheongryang, Seoul 130-650, Korea and ²College of Pharmacy, Kyung-Hee University, Seoul 130-701, Korea

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A new pterocarpan, (-)-maackiain 3-sulfate (**1**) was isolated from the methanol extract of roots of *Sophora subprostrata* together with (-)-maackiain (**2**), trifolirhizin (**3**), lupeol (**4**), ononin (**5**), 7,4'-dihydroxyflavone (**6**), and (+)-syringaresinol (**7**). The structure of **1** was determined by analyses of 2D NMR and HRFABMS. Compounds **5-7** were isolated from this plant for the first time.

Key words: *Sophora subprostrata*, Pterocarpan, (-)-Maackiain 3-sulfate, Ononin, Dihydroxyflavone, Syringaresinol

INTRODUCTION

The roots of *Sophora subprostrata* Chun et T. Chen (Leguminosae) have been used as a Korean traditional medicine for the treatment of fever, inflammation, peptic ulcer and cancer (Sakamoto *et al.*, 1992). Many chemical components have been reported from this plant including triterpenoid saponins; soyasaponins I, II, A₃, kaikasaponin, abrisaponin, (Sakamoto *et al.*, 1992; Kitagawa *et al.*, 1988), flavonoids; sophoradachromene, sophoranochromene (Komatsu *et al.*, 1970), alkaloids; matrine, oxymatrine, and pterocarpan; trifolirhizin, maackiain (Kojima *et al.*, 1970).

Pterocarpan is a isoflavonoid found in many species of the Leguminosae. Since these compounds possess asymmetric carbons at 6a- and 11a-position, four diastereomeric structures are possible. However, the junction between C-6a and C-11a has been established to be a *cis*-fusion of the heterocyclic rings resulting in only one pair of naturally occurring enantiomers (Vanetten *et al.*, 1983). Pterocarpan is also well known for their antifungal activity as phytoalexins, and their occurrence in leguminous species is of taxonomic significance (Máximo *et al.*, 1998).

By means of chromatographic separation, we have

isolated a new pterocarpan, (-)-maackiain 3-sulfate (**1**) along with the known compounds; (-)-maackiain (**2**), trifolirhizin (**3**), lupeol (**4**), ononin (**5**), 7,4'-dihydroxyflavone (**6**), and (+)-syringaresinol (**7**). Among the isolates, compounds **5-7** were isolated from this plant for the first time. In this paper, we describe the isolation of compounds and the structure of which was determined by analyses of spectroscopic method (¹H-¹H COSY, DEPT, HMQC and HMBC).

MATERIALS AND METHODS

General experimental procedures

Optical rotations were determined on an Autopol III Automatic polarimeter (Rudolph Research Flanders, NJ). IR spectra were recorded on a Midac High Resolution FT-IR spectrometer in KBr disks. FABMS (positive ion mode) and CIMS were obtained on a JEOL JMS-700. Electrospray mass spectra were determined on an Api ES/MS (HP 59987A ES/5989A MS) mass spectrometer. NMR spectra were recorded on a Bruker 300 and 500 spectrometer. ¹H-¹H COSY, HMQC and HMBC NMR spectra were obtained with the usual pulse sequences. TLC and column chromatography were carried out on precoated silica gel F₂₅₄ plates (Merck, art. 5715), RP-18 F_{254S} plates (Merck, art. 15423), silica gel 60 (230-400 mesh, Merck), Sephadex LH-20 (Bead size 25-100 μ, Sigma) and LiChroprep RP-18 (40-63 μm, Merck).

Correspondence to: Yong Sup Lee, Ph. D., Medicinal Chemistry Research Center, Korea Institute of Science and Technology, P.O. Box 131, Cheongryang, Seoul 130-650, Korea
Tel: 82-2-958-5167, Fax: 82-2-958-5189
E-mail: yslee@kist.re.kr

Plant material

The roots of *Sophora subprostrata* were purchased from a local Korean herbal drug market in July 2000. The plant was identified by Prof. Chang Su Yuk at Kyung Hee University in Korea. Voucher specimens (812-15) have been deposited in the laboratory of Korea Institute of Science & Technology (KIST).

Extraction and isolation

The dried roots of *Sophora subprostrata* (3 kg) were cut into small pieces and extracted three times with MeOH at room temperature to afford 469 g of residue on removal of the solvent under reduced pressure. The methanol extract was suspended with water and then partitioned in turn with CH₂Cl₂, EtOAc, *n*-BuOH. The CH₂Cl₂ and EtOAc extracts were evaporated under reduced pressure to give 92.5 g and 23.6 g residues, respectively. The CH₂Cl₂ extract was divided by column chromatography on silica gel column eluting with CH₂Cl₂/EtOAc/MeOH (3:1:0.1) and CH₂Cl₂/MeOH/H₂O (7:1:0.1 → 4:1:0.1) gradient system to give six fractions (Fr.1~6). Fraction 5 (6.17 g) was divided by column chromatography on silica gel column eluting with CH₂Cl₂/EtOAc/MeOH (7:2:1→2:2:1) and CH₂Cl₂/MeOH/H₂O (7:1:0.1) to give eight fractions (Fr.5A~5H). Fraction 5F (1.1 g) was divided by column chromatography on silica gel column eluting with hexane/CH₂Cl₂/MeOH (3:3:0.7→3:3:1) to afford seven fractions (Fr.5FA~5FG). Fraction 5FF (193 mg) was purified by column chromatography on LiChroprep[®] RP-18 column eluting with MeOH/H₂O (3:7→8:2) gradient system to yield compound **1** (13.4 mg) in addition to compound **5** (4.8 mg) and compound **3** (11.9 mg). Fraction 5FE (131.9 mg) gave compound **3** (32.1 mg) by crystallization with methanol. Fraction 1 (54.3 g) was subjected to silica gel column chromatography using hexane/CH₂Cl₂/EtOAc (5:1:0.3→5:1:0.5) to give compound **4** (132.8 mg). Fraction 2 (9 g) was divided into three fractions (Fr.2A~2C) by gel filtration on Sephadex LH-20 using MeOH/CH₂Cl₂ (8:2). Fraction 2B (3.5 g) was partitioned with hexane/acetone, and this hexane layer was combined with fraction 2C (101.2 mg) and then subjected to silica gel column chromatography using hexane/EtOAc (1:2) and CH₂Cl₂/EtOAc/MeOH (3:3:0.1→3:3:1) to give seven sub-fractions (2BA~2BG). Sub-fraction 2BF (258.8 mg) was purified by LiChroprep[®] RP-18 TLC eluting with H₂O/CH₃CN (1:1) to afford compound **7** (5.9 mg).

The EtOAc extract (3.01 g) yielded compound **3** (232.2 mg) by crystallization, and then the rest of this extract was divided into two fractions (Fr.A~B) by gel filtration on Sephadex LH-20 eluting with MeOH. Fraction B (1.1 g) was subjected to silica gel column chromatography using with hexane/CH₂Cl₂/EtOAc (1:1:1→1:1:3) and CH₂Cl₂/MeOH (5:1) to afford six fractions (Fr.BA~BF). Fraction

BB (201.1 mg) was purified by preparative silica gel TLC developed with hexane/CH₂Cl₂/EtOAc (3:3:1) to afford compound **2** (11.3 mg). Fraction BD (36.7 mg) and fraction BF (210.9 mg) gave compound **6** (25.4 mg) and an additional compound **3** (71.3 mg), respectively.

(-)-Maackiain 3-sulfate (1)

m.p. 210-220°C (dec); [α]_D: -161.8° (c 0.67, MeOH); UV (MeOH): λ_{max} (log ε) 276 (sh, 1.92), 284 (2.28), 309 (4.23) nm; IR (KBr): ν_{max} 1620, 1476, 1262, 1150, 1058 cm⁻¹; ¹H and ¹³C-NMR data, see Table I; ESMS (negative-ion mode): *m/z* 363 [M-H]⁻; HRFABMS [M]⁺ *m/z* 364.0251, calcd for C₁₆H₁₂O₈S, 364.0253.

(-)-Maackiain (2)

m.p. 165°C; [α]_D: -242.8° (c 0.565, acetone); UV (MeOH): λ_{max} (log ε) 280 (2.25), 286 (2.34), 308 (2.91) nm; IR (KBr): ν_{max} 3400, 2910, 1620, 1472, 1150, 1034 cm⁻¹; CIMS: *m/z* 285 [M+H]⁺; ¹H- and ¹³C-NMR data, see Table I.

Trifolirhizin (3)

m.p. 145-148°C; UV (MeOH): λ_{max} (log ε) 284 (1.29), 308 (1.62) nm; ESMS: *m/z* 447 [M+H]⁺; ¹H-NMR (500 MHz, CD₃OD+pyridine-*d*₅): δ 7.20 (1H, d, *J*=8.4 Hz, H-1), 6.70 (1H, d, *J*=6.5 Hz, H-2), 6.63 (1H, s, H-7), 6.59 (1H, br s, H-4), 6.26 (1H, s, H-10), 5.68 (2H, d, *J*=14.9 Hz, OCH₂O), 5.26 (1H, d, *J*=7.1 Hz, H-11a), 5.01 (1H, d, *J*=7.3 Hz, H-1'), 4.03 (1H, dd, *J*=4.3, 10.5 Hz, H-6), 3.94 (1H, d, *J*=11.5 Hz, H-6'), 3.77 (1H, dd, *J*=5.1, 11.9 Hz, H-6'), 3.67 (1H, m, H-3'), 3.63 (1H, m, H-2'), 3.59 (1H, m, H-4'), 3.54 (1H, m, H-5'), 3.38 (1H, m, H-6), 3.30 (1H, m, H-6a); ¹³C-NMR (125 MHz, CD₃OD+pyridine-*d*₅): δ 160.1 (C-3), 157.7 (C-4a), 155.3 (C-10a), 149.2 (C-9), 142.8 (C-8), 133.0 (C-1), 119.3 (C-6b), 115.5 (C-11b), 111.6 (C-2), 106.0 (C-7), 105.6 (C-4), 102.3 (OCH₂O), 102.1 (C-1'), 94.2 (C-10), 79.3 (C-11a), 78.6 (C-5'), 78.2 (C-3'), 74.9 (C-2'), 71.3 (C-4'), 67.3 (C-6), 62.4 (C-6'), 41.3 (C-6a).

Lupeol (4)

m.p. 190-193°C; [α]_D: +26.4° (c 0.7, CHCl₃); GCMS: *m/z* 426 [M]⁺; ¹H-NMR (300 MHz, CDCl₃): δ 4.61 (1H, s, H-29), 4.49 (1H, s, H-29), 3.11 (1H, t, *J*=5.1 Hz, H-3), 2.30 (1H, dt, *J*=5.5, 11.0 Hz, H-19), 1.82 (1H, m, H-21); ¹³C-NMR (75 MHz, CDCl₃): δ 151.4 (C-20), 109.7 (C-29), 79.4 (C-3), 55.6 (C-5), 50.8 (C-9), 48.6 (C-18), 48.3 (C-19), 43.3 (C-17), 43.2 (C-14), 41.2 (C-8), 40.3 (C-22), 39.2 (C-4), 39.0 (C-1), 38.4 (C-13), 37.5 (C-10), 35.9 (C-16), 34.6 (C-7), 30.2 (C-21), 28.3 (C-23), 27.8 (C-2, 15), 25.5 (C-12), 21.3 (C-11), 19.7 (C-30), 18.7 (C-6), 18.4 (C-28), 16.5 (C-25), 16.3 (C-26), 15.7 (C-24), 14.9 (C-27).

Ononin (5)

m.p. 205-208°C; UV (MeOH): λ_{max} (log ε) 259 (4.80), 300

(sh, 1.62) nm; FABMS: m/z 431 [M+H]⁺; ¹H-NMR (500 MHz, CD₃OD+pyridine-*d*₅): δ 8.17 (1H, s, H-2), 8.13 (1H, d, $J=8.8$ Hz, H-5), 7.50 (2H, d, $J=8.6$ Hz, H-2', 6'), 7.24 (1H, d, $J=2.0$ Hz, H-8), 7.21 (1H, dd, $J=2.2, 8.8$ Hz, H-6), 6.99 (2H, d, $J=8.7$ Hz, H-3', 5'), 5.20 (1H, d, $J=5.8$ Hz, H-1"), 4.03 (1H, d, $J=12.0$ Hz, H-6"), 3.83 (1H, m, H-6"), 3.80 (3H, s, OMe) 3.65-3.68 (3H, m, H-2", 3", 5"), 3.56 (1H, m, H-4"); ¹³C-NMR (75 MHz, CD₃OD): δ 177.1 (C-4), 163.1 (C-7), 160.8 (C-4'), 158.7 (C-9), 154.5 (C-2), 131.2 (C-2', 6'), 128.1 (C-5), 125.5 (C-1'), 125.3 (C-3), 120.0 (C-10), 116.7 (C-6), 114.7 (C-3', 5'), 104.7 (C-8), 101.8 (C-1"), 78.6 (C-5"), 78.0 (C-3"), 74.7 (C-2"), 71.2 (C-4'), 62.4 (C-6"), 55.6 (OMe).

7,4'-Dihydroxyflavone (6)

m.p. 290°C; UV (MeOH): λ_{max} (log ε) 255 (1.26), 323 (2.41) nm; CIMS: m/z 255 [M+H]⁺; ¹H-NMR (500 MHz, pyridine-*d*₅): δ 8.45 (1H, d, $J=8.6$ Hz, H-5), 7.95 (2H, d, $J=8.4$ Hz, H-2', 6'), 7.30 (1H, d, $J=1.4$ Hz, H-8), 7.24 (3H, d, $J=8.7$ Hz, H-6, 3', 5'), 7.04 (1H, s, H-3); ¹³C-NMR (125 MHz, pyridine-*d*₅): δ 177.4 (C-4), 164.1 (C-7), 163.4 (C-2), 162.1 (C-4'), 158.6 (C-9), 128.6 (C-2', 6'), 127.4 (C-5), 123.0 (C-1'), 117.4 (C-10), 116.7 (C-3', 5'), 110.4 (C-6), 105.7 (C-3), 103.4 (C-8).

(+)-Syringaresinol (7)

[α]_D: +2.4° (c 0.245, CHCl₃); ESMS: m/z 419 [M+H]⁺; ¹H-NMR (300 MHz, CDCl₃): δ 6.58 (4H, s, H-2, 6, 2', 6'), 5.52 (2H, s, OH), 4.72 (2H, br s, H-7, 7'), 4.28 (2H, m, H-9, 9"), 3.90 (2H, m, H-9, 9'), 3.90 (12H, s, OMe), 3.09 (2H, br s, H-8, 8'); ¹³C-NMR (75 MHz, CDCl₃): δ 147.5 (C-3, 5, 3', 5'), 134.6 (C-4, 4'), 132.5 (C-1, 1'), 103.0 (C-2, 6, 2', 6'), 86.5 (C-7, 7'), 72.2 (C-9, 9'), 56.8 (OMe), 54.7 (C-8, 8').

RESULTS AND DISCUSSION

Compound **1** was obtained as an amorphous powder and its molecular formula was established as C₁₆H₁₂O₈S by HRFABMS (m/z 364.0251 [M]⁺). The IR spectrum indicated the presence of double bond (1620 cm⁻¹) and -SO₃- (1262 and 1058 cm⁻¹) groups (Xiao *et al.* 2000). The ¹H and ¹³C-NMR spectra of **1** were almost identical to those of (-)-maackiain (**2**) except for the shifts of peaks of the aromatic ring A. In the ¹H-NMR spectrum, compound **1** showed a characteristic set of peaks corresponding to a pterocarpan skeleton; the signals at δ 5.45 (d, $J=6.7$ Hz), 4.20 (dd, $J=10.2, 2.8$ Hz), 3.57 (m), and 3.50 (m) were assigned to H-11a, H-6, H-6, and H-6a, respectively (Chaudhuri *et al.*, 1995). The ¹H-NMR spectrum also displayed the doublet at δ 5.78 ($J=7.4$ Hz, -OCH₂O-) and two singlets at δ 6.30 (H-10) and 6.75 (H-7), which suggested compound **1** has a methylenedioxy moiety in the aromatic ring D. Furthermore, it showed a doublet at

Table I. ¹H- and ¹³C-NMR data of compounds **1**^a and **2**^b

Position	¹ H		¹³ C	
	1	2	1	2
1	7.35 d (8.4) ^c	7.14 d (8.4)	133.2	133.5
2	6.89 dd (8.4, 2.2)	6.37 dd (8.4, 2.4)	116.5	111.2
3			155.2	160.7
4	6.77 d (2.1)	6.19 d (2.3)	111.2	104.5
4a			157.8	158.4
6	4.20 dd (10.2, 3.8) 3.57 m	4.10 dd (10.5, 4.3) 3.44 t (10.4)	68.0	67.8
6a	3.50 m	3.34 ddd (11.0, 6.8, 4.3)	41.9	42.0
6b			120.0	120.2
7	6.75 s	6.68 s	106.5	106.4
8			143.7	143.5
9			149.8	149.8
10	6.30 s	6.26 s	94.8	94.6
10a			155.6	156.0
11a	5.45 d (6.7)	5.32 d (6.7)	80.0	80.5
11b			118.9	113.2
OCH ₂ O	5.78 d (7.4)	5.76 d (1.0), 5.74 d (1.0)	103.1	102.9

^aSpectra were measured in CD₃OD+D₂O. ^bSpectra were measured in CD₃OD. ^cValues in parenthesis are coupling constants (Hz).

6.77 ($J=2.1$ Hz, H-4), a doublet of doublets at δ 6.89 (1H, $J = 8.4, 2.2$ Hz, H-2) and a doublet at δ 7.35 (1H, $J=8.4$ Hz, H-1) assignable to the ABX system for the ring A. In their ¹³C-NMR spectra, most of the signals of **1** and **2** (Table I) had very similar chemical shifts each other except for the signals at C-2, 3, 4 and -11b. Significant differences between the spectra of **1** and **2** were the large downfield shift (4-7 ppm) of the C-2, -4 and -11b signal and upfield shifts for C-3 (5 ppm) in **1** owing to an electron withdrawing group caused by sulfation of the 3-hydroxy group. *O*-Sulfate leads to a decreased electron density of *ortho* and *para* carbons and an increased electron density of the carbon carrying the sulfate group (Barron *et al.* 1988). The ¹³C-NMR chemical shift of **1** was assigned based on further analyses of its HMQC and HMBC spectra. The cross-peaks between H-1/C-3, H-1/C-11a, H-2/C-3, H-2/C-11b, H-4/C-3, H-4/C-11b in the HMBC spectrum confirmed the planar structure of **1**. Moreover, the signal at δ 5.78 (OCH₂O) showed a long-range correlation with C-8 and -9 at δ 143.7 and 149.8, respectively. On the basis of the foregoing observations, compound **1** was elucidated to be a (-)-maackiain 3-sulfate.

Six known compounds, (-)-maackiain (**2**) (Chaudhuri *et al.*, 1995; Máximo *et al.*, 1998; Mizuno *et al.*, 1990), trifolirhizin (**3**) (Yamamoto *et al.*, 1991), lupeol (**4**) (Sholichin *et al.*, 1980), ononin (**5**) (Barrero *et al.*, 1989), 7,4'-dihydroxyflavone (**6**) (Agrawal, 1989), and (+)-syringaresinol

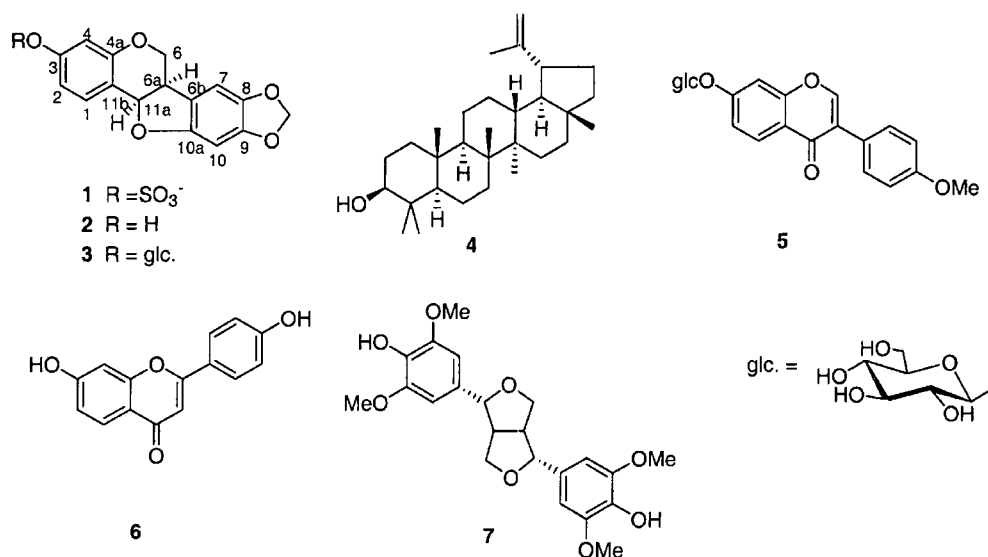


Fig. 1. Structure of compounds 1-7 isolated from *S. subprostrata*

(7) (Badawi et al., 1983; Kwon et al., 1999) were identified by comparing the physical and spectroscopic data with the literature values. Among the isolated compounds, 5-7 were isolated from this plant for the first time.

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