

Determination of Vitamin B₁₂ (Cyanocobalamin) in Fortified Foods by HPLC

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Abstract

This study was conducted to develop an HPLC method for determining vitamin B₁₂ in fortified foods which has typically been determined by microbiological assays according to AOAC and Korean Food Code approved methods. Vitamin B₁₂ (cyanocobalamin) was determined by reversed-phase HPLC with a triple column and UV/VIS detector (550 nm) using the column switching technique after extraction with 5 mM potassium phosphate solution by sonication without a clean-up procedure. The recovery of spiked samples and limit of detection (LOD) by HPLC were 78.6~107.5% and 2 ppb (µg/kg), respectively. The LOD of the microbiological assay (MBA) was much lower than that of HPLC. The concentrations of vitamin B₁₂ analyzed in all tested samples (n=12) confirmed compliance with declared label claims. The range of recovery ratio by the HPLC method when compared to the microbiological assay was 76.2~140.0%. There was not significant difference between the HPLC and MBA methods ($p < 0.01$) with $r=0.9791$ and linear regression $y=0.9923x-0.04$. The HPLC method for determining vitamin B₁₂ using the column-switching technique appears to be suitable for determining vitamin B₁₂ concentrations above 1 µg/100 g in fortified foods.

Key words: vitamin B₁₂, HPLC method, microbiological assay

INTRODUCTION

Vitamin B₁₂ is a water-soluble vitamin. Cobalamin is the term used to refer to compounds having vitamin B₁₂ activity, as well as to related compounds (1). Vitamin B₁₂ is present in animal products, such as meat, poultry, milk and fish, but is not present in plant products or in yeast. All of the vitamin B₁₂ in our environment originated from synthesis by bacteria, fungi and algae (1-4).

Cyanocobalamin is the permissive name for vitamin B₁₂ (4) and is the form predominantly used in vitamin preparations, supplements, medical foods and fortified foods because of its better stability compared to hydroxycobalamin (1,4).

Strict vegetarian and infants without maternal feeding should be supplied by vitamin B₁₂ through nutritional supplements or infant formulas. The Korean food code specifications for vitamin B₁₂ in infant and follow-up formulas are 0.1 µg/100 kcal, 0.15 µg/100 kcal, respectively (5).

It is necessary to monitor nutrients in fortified foods for nutrient labeling as well as process and quality control purposes. Hence simple, rapid, sensitive and repro-

ducible analytical methods are required.

Available methods for assay of vitamin B₁₂ include polarographic, spectrophotometric, various chromatographic procedures including paper, thin-layer, open column, GC and LC procedures, microbiological and radioligand binding procedures (4). Almost all available data of vitamin B₁₂ in food has been obtained by microbiological assays (4-8). However, microbiological assays require the cells to be maintained and preserved and are cumbersome and time consuming. Furthermore, other substances or contaminants may interfere growth of microorganisms, invalidating the assay.

HPLC is used for the analyses for many kinds of vitamins (8-17). But HPLC has not been successfully used for the routine analysis of vitamin B₁₂ in foods because proper protocols for sample preparation, such as the concentration of analyte and removal of interfering substances, caused by the complex sample matrices, have not been developed and confirmed.

Therefore, this study reports the development of an analytical method for the determination vitamin B₁₂ by µ-HPLC using the column-switching technique (18-23) and compares the results with a microbiological assay.

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MATERIALS AND METHODS

Reagents and materials

Vitamin B₁₂ standard was obtained from Sigma Chemical Co. (St. Louis, MO, USA). HPLC grade solvents such as acetonitrile and methanol were purchased from Merck (Frankfurter, Germany) and JT Baker (New Jersey, USA).

Vitamin B₁₂-fortified foods were purchased from the department stores in Seoul. Table 1 shows the food group, the product type and the label claim of the fortified foods.

Apparatus

An HPLC (Nanospace SI-2, Shiseido, Tokyo, Japan) equipped with an autosampler, pump, UV detector, PDA detector and valve system was used. Other equipment included a sonicator (Branson 8210, Connecticut, USA), centrifuge (CR21E, Hitacchi, Tokyo, Japan) and UV detector (Biochrom 4060, Pharmacia, Cambridge, USA).

Sample preparation

Each sample (equivalent to 200 ng of vitamin B₁₂) was suspended in 5 mM potassium phosphate solution. The mixture was extracted by sonication for 10 min and then made to volume (50 mL) with phosphate solution and centrifuged at 23,000×g for 10 min. The middle

layer was collected and again centrifuged at 23,000×g for 10 min. The middle layer was transferred to a test tube, chloroform (3 mL) was added to remove lipids and centrifuged at 1,000×g for 10 min. The aqueous layer was then collected and centrifuged at 23,000×g for 10 min and the clear layer was passed through a 0.45 μm membrane filter and analyzed by HPLC.

HPLC analysis

The μ-HPLC system used in this study is illustrated in Fig. 1 and the operating conditions are shown in Table 2. A triple column was used for pretreatment, concentration and separation. Pumps 1 and 2 were used to deliver eluent A at a flow rate of 120 μL/min and eluent B at a flow rate of 500 μL/min, respectively. In the initial step, a sample solution was introduced to the pretreatment column via the autosampler using eluent B. By switching the valve as shown in Fig. 1 (B), vitamin B₁₂ is eluted from the pretreatment column and introduced to the concentration column. Finally vitamin B₁₂ adsorbed in the concentration column is introduced to the separation column by switching the valve as shown in Fig. 1 (A) using eluent A. Total analytical time was 40 minutes per a sample.

Microbiological assay

Vitamin B₁₂ was also analyzed by the microbiological as

Table 1. Selected samples for analysis

Sample	Food group	Product type	Label claims (μg/100 g)
1	Infant formula	Powder	2.0
2	Infant formula	Powder	2.0
3	Follow up formula	Powder	2.0
4	Follow up formula	Powder	3.7
5	Cereal based infant formula	Powder	2.0
6	Cereal based infant formula	Powder	1.1
7	Cereal based infant formula	Powder	2.0
8	Cereal based infant formula	Powder	2.0
9	Medical food	Powder	2.0
10	Nutritional supplement product	Tablet	50 (0.3 μg/1 tablet)
11	Nutritional supplement product	Tablet	1689 (25 μg/1 tablet)
12	Snack	Flake	1.5

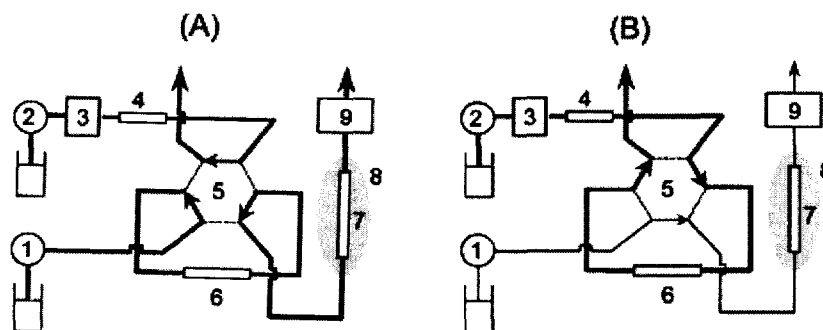


Fig. 1. Schematic diagram of the μ-HPLC system (23). (A) Separation position, (B) Concentration position. 1, pump (for separation); 2, pump (for pretreatment); 3, autosampler; 4, pretreatment column; 5, switching valve; 6, concentration column; 7, analytical column; 8, column oven; 9, UV detector.

Table 2. HPLC operating conditions for vitamin B₁₂ (cyanocobalamin) determination

Detector	UV (550 nm)
Mobile phase	A: 5 mM KH ₂ PO ₄ /MeOH = 80/20 B: 5 mM KH ₂ PO ₄
Flow rate	120 µL/min for pump 1 500 µL/min for pump 2
Temp.	40°C
Column	Pre-separation column: Capcellpak MF C ₈ (4.6 mm × 150 mm, 5 µm) Focusing column: Capcellpak MG C ₁₈ (2.0 mm × 35mm, 5 µm) Analytical column: Capcellpak UG C ₁₈ (1.5 mm × 250 mm, 5 µm)
Inj. vol.	400 µL

say as described in the procedures of Korean Food Code (6).

RESULTS AND DISCUSSION

The extraction and analytical conditions

Because vitamin B₁₂ (cyanocobalamin) is water soluble and occurs in a free form, potassium phosphate solution at concentration 5 mM was sufficient to extract analyte. A 30 minute extraction time was sufficient to obtain maximum recovery.

Although the PDA spectrum for vitamin B₁₂ had a maximum peak at 361 nm, peak separation was very difficult because of matrix interferences. Not only vitamin B₁₂ but also other vitamins, amino acids and organic acids absorb in the UV region (9). It was reported that vitamin B₁₂ shows UV absorbance at 260 and 360 nm and visible absorbance at 550 nm. However, thiamine, riboflavin, ascorbic acid, nicotinamide, pyridoxine, folic acid, tyrosine, phenylalanine and tryptophan have

absorbances at 260 nm and several compounds, such as folic acid and riboflavin, have absorbance at 360 nm (4, 8,9). Hence, monitoring at 550 nm was suitable for the determination of vitamin B₁₂.

The determination of vitamin B₁₂ was performed by the column-switching technique using a triple column system as described in earlier reports (18,19), and appears to be a sensitive and effective technique.

Recovery, detection limits, and comparison with a microbiological method

The calibration curve from the vitamin B₁₂ standard showed good linearity ($r=0.9999$) in the range of 1.0~200 µg/kg. The results in Table 3 demonstrate that the recovery of vitamin B₁₂ was 78.6~107.5% by standard addition method. The spiking concentrations were varied with the amount of vitamin B₁₂ in the samples; the amounts (20~1000 ng) are shown in Table 3.

The limit of detection (LOD, S/N=3) was 2 ppb (µg/kg) in all tested samples. The HPLC chromatograms of the vitamin B₁₂ standard and samples are shown in Fig. 2; the vitamin B₁₂ was obtained at 19 min. The limit of quantification, five times of LOD, was 1 µg/100 g, therefore this HPLC method could be applicable for assays of foods fortified with vitamin B₁₂ above this level.

The HPLC method was found to have a higher LOD than the microbiological assay and showed large differences among samples in high concentrations of analyte. However, the concentration range with linearity was narrow and errors caused by dilution effects may be problematic in the microbiological assay.

The concentrations of vitamin B₁₂ analyzed by both methods in all tested samples confirmed compliance with declared label claims.

The accuracy of HPLC analysis was assessed by com-

Table 3. Recoveries and detection limits for vitamin B₁₂ (cyanocobalamin)

Sample	Microbiological assay			HPLC method		
	Added amount (ng)	Recovery (%)	LOD (µg/kg) ¹⁾	Added Amount (ng)	Recovery (%)	LOD (µg/kg)
1	30	82.2 ± 6.2 ²⁾	0.0012	200	89.1 ± 6.2	2.0
2	30	106.1 ± 6.3	0.0012	200	84.6 ± 8.5	2.0
3	30	81.9 ± 4.3	0.0012	200	90.1 ± 5.1	2.0
4	30	101.7 ± 9.9	0.0012	400	107.5 ± 6.1	2.0
5	30	87.0 ± 9.7	0.0012	200	78.6 ± 6.5	2.0
6	30	108.5 ± 4.8	0.0012	200	85.0 ± 4.3	2.0
7	30	104.0 ± 1.4	0.0012	200	93.9 ± 4.1	2.0
8	30	106.4 ± 7.6	0.0012	200	86.6 ± 2.6	2.0
9	30	82.8 ± 4.6	0.0012	200	80.0 ± 2.2	2.0
10	30	102.2 ± 7.2	0.0012	1000	104.1 ± 8.5	2.0
11	30	82.3 ± 6.8	0.0012	2000	87.7 ± 7.1	2.0
12	30	105.6 ± 4.9	0.0012	200	88.9 ± 6.5	2.0

¹⁾LOD: The limit of detection for S/N=3.

²⁾Mean ± SD, results of three replicates.

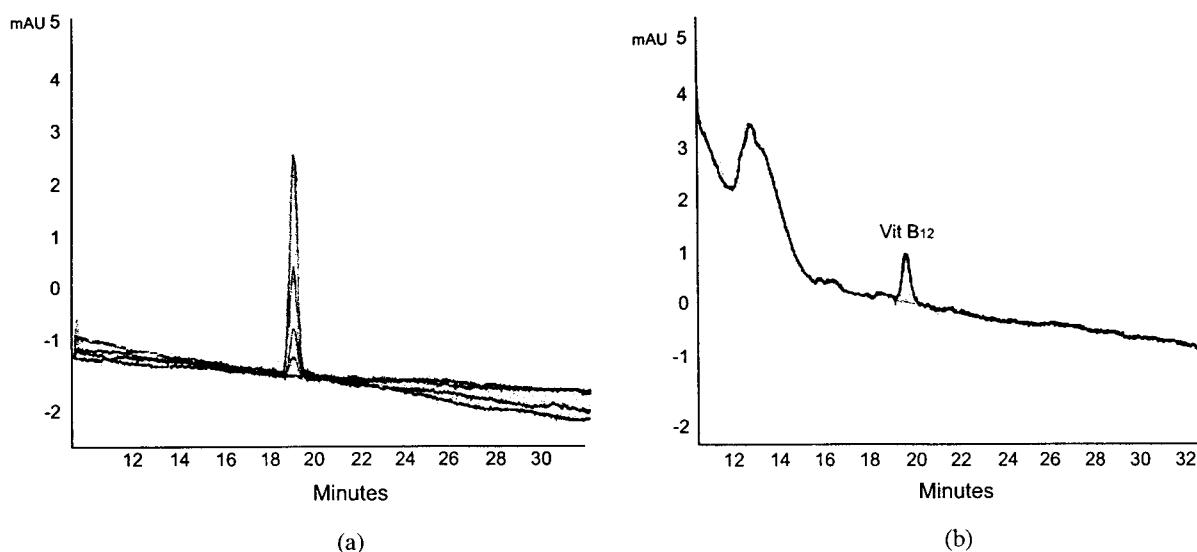


Fig. 2. HPLC chromatograms of vitamin B₁₂ using a UV/VIS detector at 550 nm: (a) standard (2~20 ppb) chromatogram, (b) sample No. 6 chromatogram obtained with triple column system.

parison of vitamin B₁₂ levels against the microbiological assay for a range of fortified foods and the data are shown in Table 4. The recovery by the HPLC method compared to microbiological method was 76.2~140.0% and there was no significant difference ($p < 0.01$), with $r = 0.9791$ and linear regression $y = 0.9923x - 0.04$, between the two methods (Fig. 3).

In conclusion, HPLC using the column-switching technique was a useful tool for the separation and determination of vitamin B₁₂ in foods because it was simple and rapid with a high degree of recovery. Moreover, this method was shown to have a good correlation with microbiological assays which has been accepted as an official method.

As a result, this method could be used in monitoring and quality control of infant formulas, follow-up for-

Table 4. Comparisons for vitamin B₁₂ between microbiological assay (MBA) and HPLC method

SA	Label claim ($\mu\text{g}/100\text{ g}$)	MBA ¹⁾ ($\mu\text{g}/100\text{ g}$)	HPLC ($\mu\text{g}/100\text{ g}$)	HPLC/MBA (%)
1	2.0	$2.6 \pm 0.3^{2)}$	2.8 ± 1.0	107.7
2	2.0	3.4 ± 0.2	2.9 ± 0.5	85.3
3	2.0	2.4 ± 0.1	2.7 ± 0.1	112.5
4	3.7	5.2 ± 0.4	5.4 ± 1.0	103.8
5	2.0	2.1 ± 0.3	2.1 ± 0.2	100.0
6	1.1	2.1 ± 0.3	1.6 ± 0.7	76.2
7	2.0	2.1 ± 0.1	2.0 ± 0.7	95.2
8	2.0	4.1 ± 0.2	3.8 ± 0.2	92.7
9	2.0	2.1 ± 0.1	2.2 ± 0.7	104.8
10	50.0	98.0 ± 7.7	101.7 ± 56.0	103.8
11	1689	1803 ± 46	2089 ± 330	115.9
12	1.5	4.0 ± 0.2	5.6 ± 0.2	140.0

¹⁾Microbiological assay.

²⁾Mean \pm SD, results of three replicates.

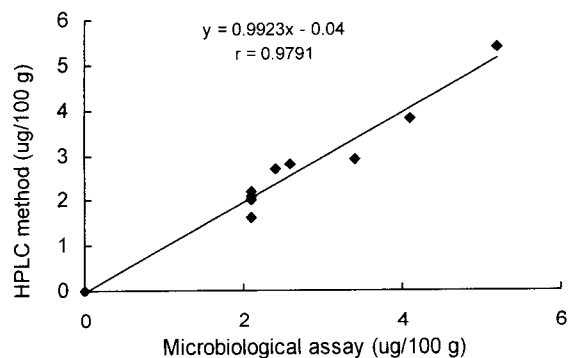


Fig. 3. The correlation between microbiological assay (MBA) and HPLC method. The values of sample No. 10 & 11 were excluded in this regression curve because of their high concentrations.

mulas, and nutritional supplement products because of its efficiency.

Further studies on concentration and clean-up techniques would be required so that this HPLC method could be used to analyze vitamin B₁₂ in all forms of foods.

REFERENCES

1. Tom B. 1994. *Nutritional Biochemistry*. Academic Press Inc, San Diego. p 376-382.
2. The Korean Nutrition Society. 1999. *RDA for Koreans*. Seoul. p 611-625, 632-646.
3. Choi HM. 1998. *21st Century Nutrition*. Kyomoonsa, Seoul. p 226-240.
4. Eitenmiller RR, Landen WO Jr. 1999. *Vitamin Analysis for the Health and Food Sciences*. CRC Press, Boca Raton. p 467-478.
5. Korea food & drug administration. 2002. *Korean food code*. p 277-281.
6. Korea food & drug administration. 2002. *Korean food*

- code (attend.). p 344-347.
7. AOAC. 1995. *Official Methods Analysis*. 16th ed. Association of official analytical chemists, Arlington, VA. Vol 2, Chap 45, p 44.
 8. Iwase H, Ichiro O. 1997. Determination of cyanocobalamin in foods by high-performance liquid chromatography with visible detection after solid-phase extraction and membrane filtration for the precolumn separation of lipophilic species. *J Chromatogr* 771: 127-134.
 9. Iwase H. 1992. Ultramicrodetermination of cyanocobalamin in elemental diet by solid-phase extraction and high-performance liquid chromatography with visible detection. *J Chromatogr* 590: 359-363.
 10. Frankel EP, Kitchens RL, Prough R. 1979. High performance liquid chromatographic separation of cobalamins. *J Chromatogr* 174: 393-401.
 11. Amin M, Reusch J. 1987. Simultaneous determination of vitamin B₁, B₂, B₆ and B₁₂ and C, nicotinamide and folic acid in capsule preparations by ion-pair reversed-phase high-performance liquid chromatography. *Analyst* 112: 989-994.
 12. Amin M, Reusch J. 1987. High performance liquid chromatography of water-soluble vitamins. II. Simultaneous determination of vitamins B₁, B₂, B₆, and B₁₂ in pharmaceutical preparations. *J Chromatogr* 390: 448-454.
 13. Gonzalez L, Yuln G, Volonte MG. 1999. Determination of cyanocobalmin, betamethason, and diclofenac sodium in pharmaceutical formulation, by high performance liquid chromatography. *J Pharm Biomed Anal* 20: 487-492.
 14. Hubert C, Ryszard L. 1998. Determination of cobalamins and cobinamides by microbore reversed-phase HPLC with spectrophotometric, ion-spray ionization MS and inductively coupled plasma MS detection. *Analysis Chimica Acta* 395: 227-235.
 15. Thomas SH, Shyamala S, Rebecca A. 1984. Determination of pantothenic acid, biotin, and vitamin B₁₂ in nutritional products, instrumental method for the analysis of vitamins. *J AOAC International* 67: 994-998.
 16. Yumie M, Masayoshi Y, Kazuhiro O, Shiro S, Toshio M, Hiroyuki N. 1989. Simultaneous liquid chromatographic determination water-soluble vitamins, caffeine, and preservative in oral liquid tonics. *J AOAC International* 72: 244-247.
 17. Woollard DC. 1984. New ion-pair reagent for the high-performance liquid chromatographic separation of B-group vitamins in pharmaceuticals. *J Chromatogr* 301: 470-475.
 18. Park SJ, Kim HK, Hahm TS, Kim B. 2001. Determination of Vitamin B₁₂ in foods using column-switching technique in μ -HPLC. *J Korean Soc Food Sci Nutr* 28: 1208-1211.
 19. Lee YJ, Lee HS. 1990. Simultaneous determination of cefoxitin, cefuroxime, cephalixin and cepaoridirine in plasma using HPLC and column-switching technique. *Chromatographia* 30: 80-84.
 20. Wyss R, Buchell F. 1988. Quantitative analysis of retinoids in biological fluids by high performance liquid chromatography using column switching. I. Determination of isotretinoin and tretinoin and their 4-oxo metabolites in plasma. *J Chromatogr* 424: 303-314.
 21. Lee HS, Kim K, Kim JH, Do DS, Lee SK. 1997. Simultaneous determination of parathion and metabolites in serum by HPLC column switching. *Chromatographia* 44: 473-476.
 22. Nielsen SE, Dtrasted LO. 1998. Column-switching high performance liquid chromatographic assay for the determination of quercetin in human urine with ultraviolet absorbance detection. *J Chromatogr* 707: 81-89.
 23. Yasueda S, Kimura M, Ohtori A, Kakehi K. 2003. Analysis of an anti-inflammatory steroidal drug, difluprednate in aqueous humor by combination of semi-micro HPLC and column switching method. *J Pharm Biomed Anal* 30: 1735-1742.

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