

Asymmetric Synthesis of α -Alkyl- α -phenylglycinesSeung-Han Lee,[†] Eun-Kyung Lee, and Soo-Min Jeun

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α,α -Disubstituted α -amino acids are popular replacements for naturally occurring amino acids in peptides.¹ Peptide analogs, containing these substitutions, often have useful enzyme inhibitory and other important biological properties.² In recent years, a number of methods to construct chiral α,α -disubstituted α -amino acids have been developed.³ In most approaches, the stereogenic center is established in alkylation reactions of chiral, nonracemic enolates, *e.g.*, those derived from 5,6-diphenyl,¹ 5-phenyl,⁵ and 6-phenyl-1,4-oxazin-2-ones.⁶ However, removal of the chiral auxiliaries is known to be problematic in routes where α -alkyl- α -aryl-glycines are the targets.

Recent studies in our laboratory have led to the development of a new route for asymmetric synthesis of α -alkyl- α -phenylglycines. The sequence involves sequential arylation of the Williams oxazinone **1** generating 3-phenyloxazinone **2**, alkylation to form intermediate 3-alkyl-3-phenyloxazinones **3**, and stepwise removal of *N*-BOC group and the chiral auxiliary.

To begin the sequence, (3*S*)-3-phenyloxazinone **2** is prepared by using the reported method.⁷ Treatment of **2** with NaHMDS or KHMDS at -78 °C in THF, followed by addition of the alkyl halide, stirring at room temperature, and quenching with saturated aq. NH₄Cl at -78 °C, yields the corresponding 3-alkyl-3-phenyloxazinones **3** in moderate to high yields and with high diastereomeric purities (Scheme 1, Table 1).⁹ High diastereoselectivities are observed even in processes where small alkyl halides (*e.g.*, methyl iodide) are employed. Interestingly, attempts to run the alkylation reac-

tion at -78 °C met with failure with only epimerized (3*R*)-3-phenyloxazinone **2** being recovered.^{6,8}

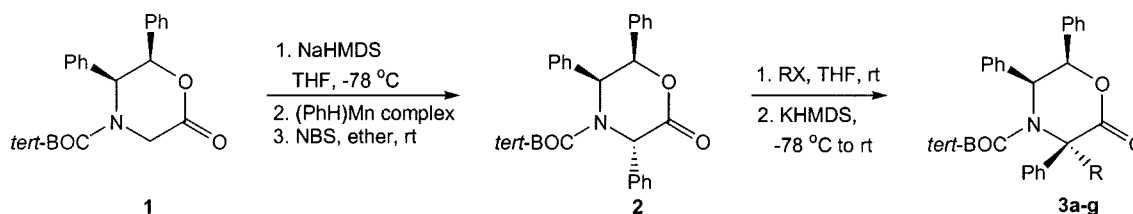
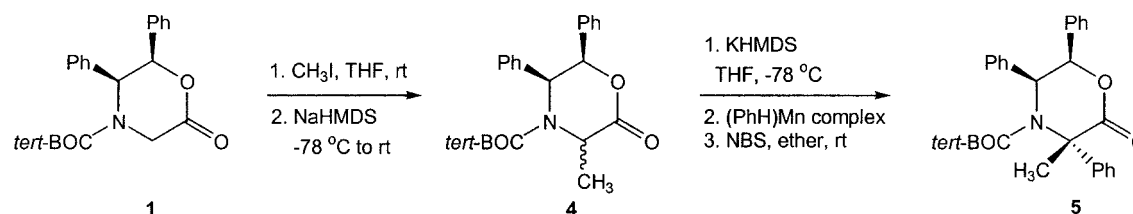
For the purpose of determining the level of diastereoselectivity associated with the alkylation reactions, (3*S*)-3-methyl-3-phenyloxazinone **5** was prepared by using the reverse sequence (*i.e.*, alkylation followed by arylation) to introduce the 3-substituent (Scheme 2). Analysis of the ¹H NMR spectra of the oxazinones **3a** and **5** showed that both methylation of **2** and phenylation of **4** yield single diastereomers (**3a** and **5**, respectively) of the 3,3-disubstituted products. However, in contrast to the high efficiency of the methylation reaction of **2**, phenylation of **4** is a low yielding (44%) process.

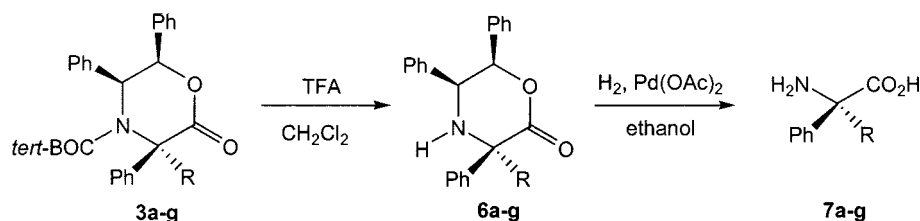
When the 5,6-diphenyloxazinone template is used for α -aryl-glycines synthesis, new methods are needed to bring

Table 1. Alkylation Reactions of 3,5,6-Triphenyloxazinone (**2**)

Entry	RX	Product	Yield (%) ^a	%de
1	CH ₃ I	3a	99	>95
2	CH ₃ CH ₂ CH ₂ I	3b	82	>95
3	H ₂ C=CHCH ₂ Br	3c	95	>95
4	HC≡CCH ₂ Br	3d	98	>95
5 ^b	CH ₃ OCH ₂ Cl	3e	75	>95
6	BrCH ₂ CO ₂ CH ₂ CH ₃	3f	88	>95
7 ^c	C ₆ H ₅ I	3g	96	>95

^aIsolated yields. ^bSome starting material is recovered. ^cSince *n*-butyl bromide does not react, the corresponding iodide, prepared by treatment of the bromide with NaI in acetone, was used instead.

**Scheme 1****Scheme 2**



Scheme 3

about selective cleave of the benzylic C-O and C-N bonds in order to liberate the amino acid targets. Williams has developed both a dissolving metal reduction and a catalytic hydrogenolysis method for this purpose.^{4d} Also, Remuzon showed that the chiral auxiliary in 3-alkyl-3,6-diphenyl-1,4-oxazin-2-ones is removed selectively by catalytic hydrogenolysis.⁶ Finally, Hegedus reported that *syn*-3,5,6-triphenyl-oxazinone isomers are selectively cleaved to form the amino acid in high yield under mild reductive conditions (1 atm of H₂, PdCl₂).¹⁰

However, our attempts to use the hydrogenolysis process for removal of the chiral auxiliary present in *tert*-BOC protected 3-alkyl-3,5,6-triphenyloxazinones **3** gave none or only low yields (*ca.* 10%) of the desired products. We believed that this process might be more successful if it were applied to *N*-deprotected oxazinones. We first tried to remove the *tert*-BOC group by using TMSI in CH₂Cl₂,^{4d} but low yields (40% and 32%) were encountered for the allyl and propargyl-oxazinones. **3c** and **3d**. Moreover, deprotection of oxazinones **3e** and **3f** failed completely. In contrast, removal of *tert*-BOC group with TFA in dichloromethane^{4e} in all cases cleanly furnishes the deprotected 3-alkyl-3-phenyl-1,4-oxazin-2-ones **6** (Scheme 3 and Table 2). Importantly, hydrogenolysis of the deprotected oxazinones **6** under mild conditions (1 atm H₂, 0.5 equiv Pd(OAc)₂, 25 °C, 4h) affords the desired α -alkyl- α -phenylglycines **7** in moderate yields (Scheme 3 and Table 2).¹¹ The enantiomeric purity of (*R*)- α -methyl- α -phenylglycine **7a** ($[\alpha]_D^{25}$ -94.5), generated by use of this route, was determined to be *ca.* 100% by comparison of its specific rotation to the reported value (lit.^{6,12} $[\alpha]_D^{25}$).

In conclusion, we have demonstrated that 3-alkyl-3-phenyl-oxazinones can be prepared in high yields and with high diastereomeric purities by alkylation reactions of chiral 3-phenyloxazinone. Also, removal of *tert*-BOC group follow-

ed by selective hydrogenolysis is an effective procedure to efficiently transform 3-alkyl-3,5,6-triphenyloxazinones to the corresponding α -alkyl- α -phenylglycines without accompanying racemization.

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- Spectroscopic data for (3*R*,5*S*,6*R*)-4-(*tert*-butyloxycarbonyl)-2,3,5,6-tetrahydro-3-methyl-3,5,6-triphenyl-1,4-oxazin-2-one (**3a**): mp 72-74 °C; $[\alpha]_D^{25}$ -18.8 (c 0.8, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 7.50-7.10 (m, 15H), 6.23 (d, *J* = 2.8 Hz, 1H), 5.28 (s, 1H), 2.39 (s, 3H), 1.09 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 171.72, 153.89, 142.35, 135.16, 134.80, 129.82, 128.32, 128.17, 128.13, 128.09, 127.74, 127.00, 126.81, 125.93, 81.64, 80.39, 64.73, 59.56, 27.83, 26.42; IR (KBr) 3076, 2975, 1751, 1688, 1454, 1347, 1164, 1082, 880, 698 cm⁻¹; Anal. Calcd for C₂₈H₂₉NO₄: C, 75.82; H, 6.59; N, 3.16. Found: C, 75.81; H, 6.61; N, 3.21.
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Table 2. *N*-BOC Removal and Hydrogenolysis of **3**

Entry	R	6 (%) ^a	7 (%) ^a
1	CH ₃	94	75
2	CH ₂ CH ₂ CH ₃	90	82
3	H ₂ C=CHCH ₂	90	82 ^b
4	HC≡CCH ₃	93	75 ^b
5	CH ₂ OCH ₃	87	83
6	CH ₂ CO ₂ CH ₂ CH ₃	90	75
7	C ₆ H ₅	92	71

^aIsolated yields. ^bAllyl and propargyl groups are reduced to propyl group under this reaction conditions.