## 단 신

# 오메가-(메틸술피닐)아세토페논류를 이용한 2-아릴벤조푸란 유도체의 합성

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## Synthesis of 2-Arylbenzofuran Derivatives Using ω-(Methylsulfinyl)acetophenones

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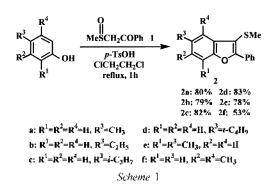
A series of benzofuran ring system bearing various substituents at the C-2 position is widely distributed in nature and has recently become of interest in biological properties. There are well known natural products having related benzofuran ring structures, particularly those isolated from *Machilus glaucescens*.<sup>1</sup> *Ophryosporus charua*.<sup>2</sup> *Ophryosporus charua*.<sup>3</sup> *Krameria ramosissima*.<sup>4</sup> and *Zanthoxylum ailanthoidol*.<sup>5</sup>

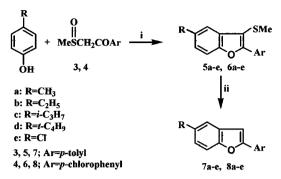
For the carbon-carbon bond formation using 1-acyl-1thiocarbocations, we found that 2-methyl- and arylbenzofurans were easily prepared by the one-pot reaction of substituted phenols with 1-acyl-1-chlorosulfides in the presence of a Lewis acid.<sup>6</sup> Also a facile one-pot procedure was developed, which offers 2-alkylbenzofurans from substituted phenols using  $\alpha$ -acylsulfoxides in the presence of *p*-toluenesulfonic acid.<sup>7</sup>

In this paper, we report a new route for synthesizing of 2-arylbenzofuran derivatives from substituted phenols with  $\alpha$ -acylsulfoxides (1.3.4) under Pummerer reaction conditions.

As preparation of the starting materials.  $\omega$ -(methylsufinyl)acetophenone (1) was obtained from the oxidation of 2-(methylthio)acetophenone with sodium metaperiodate in aqueous methanol in 80% yield. The reactions of ethyl *p*-toluate and ethyl *p*-chlorobenzoate with methylsulfinyl carbanion<sup>8</sup> afforded  $\omega$ -methylsulfinyl-*p*-methylacetophenone (**3**) and  $\omega$ -methylsulfinyl-*p*-chloroacetophenone (**4**) in 74% and 79% yields, respectively.

On the basis of our synthetic method<sup>7</sup> for 2-methylbenzofurans using  $\alpha$ -(methylsulfinyl)acetone under Pummerer reaction conditions, we first attempted the synthesis of 2-phenylbenzofurans 2 as illustrated in *Scheme* 1. Thus, treatment of equimolar amounts of substituted phenols and the sulfoxide 1 in 1,2-dichloroethane with three equivalents of anhydrous *p*-toluenesulfonic acid under reflux gave the compounds **2a-f** in moderate yields. The spectroscopic data (mp. IR, and <sup>1</sup>H NMR) were in good agreements with those reported by our pre-





Scheme 2. Reagents and conditions: (i) p-TsOH, ClCH<sub>2</sub>CH<sub>2</sub>Cl, reflux, 1 h; (ii) Raney-Ni (W-2), EtOH, 60-65 °C, 1 h.

vious work<sup>6b</sup> on the synthesis of 2-methylthio-2-phenylbenzofurans under Friedel-Crafts reaction conditions. The procedure for desulfurization of the adducts **2a-f** by heating with Rancy nickel in ethanol was reported previously.<sup>6b</sup>

Secondly, we applied the above method to syntheses of 3-methylthio-2-(p-tolyl)benzofurans (5) and 3-methylthio-2-(p-chlorophenyl)benzofurans (6), in which the sulfoxides 3 and 4 are employed as electrophiles in place of 1. The Pummerer reactions of *para*-substituted phenols with 3 and 4 were carried out as shown in *Scheme* 2.

The treatment of equimolar amounts of *para*-substituted phenols and the sulfoxide **3** in the presence of three equivalents of *p*-toluenesulfonic acid afforded the compounds **5a-e** in satisfactory yields. Also the compounds **6a-e** were obtained from the reactions of *para*-substituted phenols and the sulfoxide **4** in the presence of *p*-toluenesulfonic acid.

The adducts (5.6) given by the above Pummerer reaction can easily be desulfurized into the corresponding 2-arylbenzofurans (7.8) by heating Rancy nickel in ethanol. Thus, the adducts **5a-e** and **6a-e** were converted into 2-(*p*-tolyl)benzofurans **7a-e** and 2-(*p*-chlorophenyl)benzofurans **8a-e**, respectively, in high yields.

Of the many methods for the preparation of 2-arylbenzofuran ring, the route<sup>9</sup> through the coupling reaction of an o-halophenol with a cuprous arylacetylide have been regarded as an efficient procedure. This method requires uncommon starting materials and lengthy reaction time.

In conclusion, we developed a new one-pot method for the construction of 2-arylbenzofurans (2.5.6) using substituted phenols and  $\alpha$ -acylsulfoxides (1.3.4) in the presence of anhydrous *p*-toluenesulfonic acid. This method is generally applicable to benzofuran moiety having various aryl groups at the C-2 position.

The Pummerer reactions for utilizing  $\alpha$ -acylsulfoxides has been proved to be useful to synthesize the naturally occurring products possessing 2-arylbenzofuran skeleton.

### EXPERIMENTAL

General. All reagents and solvents were used without further purification. Melting points were determined with a Gallenkamp capillary melting point apparatus and are uncorrected. <sup>1</sup>H NMR spectra were recorded on a Hitachi R-1500 (FT, 60 MHz) spectrometer. Chemical shifts are expressed in  $\delta$  units relative to tetramethylsilane as internal standard. IR spectra were recorded by using on a JASCO FT IR-300E spectrometer. Mass spectral data were obtained on a Hewlett Packard 5970 GC MS system. Silica gel 60 (70-230 mesh, E. Merck) was used for all column chromatographic separations.

**\varpi-(Methylsufinyl)acetophenone (1).** A solution of sodium metaperiodate (5.14 g. 24 mmol) in water (30 mL) was added in small portions to a stirred solution of 2-(methylthio)acetophenone (4 g. 24 mmol) in methanol (60 mL) at 0 °C and the mixture was further stirred at room temperature for 12 h. Inorganic materials were filtered off and the filtrate was extracted with chloroform (3 30 mL). The combined organic layer was dried over MgSO<sub>1</sub>, and concentrated under reduced pressure. The residual solid was recrystallized from ethyl acetate to give 1 in 80° o yield (3.49 g). mp. 86-87 °C (lit.<sup>8</sup> 86-86.5 °C): IR (KBr) 3044, 2933, 1675 (C=O), 1577, 1422, 1299, 1193, 1030 (S=O), 978 em<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.77 (s, 3H), 4.39 (d, J=3.5Hz, 2H), 7.48-7.62 (m, 3H), 7.98 (d, J=7.6Hz, 2H).

General procedure for the synthesis of 3-methylthio-2-phenylbenzofuran (2). A solution of 1 (1.1 mmol), substituted phenol (1.1 mmol), and anhydrous *p*-toluenesulfonic acid (3.3 mmol) in 1.2-dichloroethane (15 mL) was refluxed for 1 h. Then the mixture was cooled at room temperature, washed with water to remove *p*-toluenesulfonic acid, and dried over MgSO<sub>1</sub>. The solvent was evaporated off, and the residue was purified by column chromatography (hexane/ethyl acetate=6/1) to give **2.** 2a: Yield 80°  $_{0}$ , mp 67-68 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.37 (s. 311), 2.48 (s. 311), 6.80-8.37 (m. 811), 2b: Yield 79°  $_{0}$ , mp 37-38 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.31 (t. J=7.6Hz, 3H), 2.38 (s. 311), 2.80 (q. J=7.6Hz, 2H), 7.06-8.37 (m. 811), 2c: Yield 82°  $_{0}$ , colorless liquid, <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.33 (d. J=6.5Hz, 6H), 2.38 (s. 3H), 2.84-3.31 (m. 1H), 7.10-8.36 (m. 8H), 2d: Yield 83°  $_{0}$ , colorless liquid, <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.42 (s. 9H), 2.39 (s. 3H), 7.24-8.38 (m. 8H), 2e: Yield 78°  $_{0}$ , mp 115-116 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.38 (s. 3H), 2.47 (s. 3H), 2.53 (s. 3H), 6.98-8.33 (m. 7H), 2f: Yield 53°  $_{0}$ , colorless liquid, <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.32 (s. 3H), 2.42 (s. 3H), 2.49 (s. 3H), 6.86-8.22 (m. 7H). The above spectral data for **2** are in accord with those reported.<sup>6h</sup>

 $\omega$ -Methylsulfinyl-*p*-methylacetophenone (3).  $\wedge$  suspension of NaI1 (60% mineral oil dispersion, 2 g. 50 mmol) in DMSO (30 mL) was heated with stirring at 70-75 °C for 40 min under Ar. After cooling to room temperature. THF (15 mL) was added to the reaction mixture. Ethyl p-toluate (3.28 g, 20 mmol) was added to the mixture at 0 °C, and the stirring was continued for 90 min at the room temperature. The reaction mixture was poured into water (150 mL), acidified with aqueous HCI to a pH 3-4, and throughly extracted with chloroform (3:30 mL). The combined organic layer was washed with water (3-30 mL), dried over MgSO<sub>4</sub>, and evaporated off. The residual solid was recrystallized from ethyl acetate to give 3 in 74% vield (2.9 g), mp. 113-114 °C: IR (KBr) 2998, 2911, 2362, 2344, 1671 (C=O), 1605, 1560, 1411, 1278, 1186, 1131, 1044 (S=O), 961 em<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 2.44 (s. 3H). 2.75 (s. 3H). 4.33 (d. J=4.1Hz, 2H), 7.31 (d, J=8.2Hz, 2H), 7.89 (d, J=8.2Hz, 2H): MS m/z 196 (M<sup>+</sup>).

**ω-Methylsulfinyl-***p***-chloroacetophenone (4)**. By the same procedure as described above for the preparation of **3**, compound 4 was obtained from ethyl *p*-chlorobenzoate (3.69 g, 20 mmol), NaH (60% mineral oil dispersion, 2 g, 50 mmol), and DMSO (30 mL) in 79% yield (3.42 g). mp 126-127 °C; IR (KBr) 3033, 2988, 2921, 2377, 2344, 1666 (C=O), 1588 1422, 1288, 1033 (S=O), 767 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 2.76 (s. 3H), 4.36 (d. J=3.9Hz, 2H), 7.49 (d. J=8.2Hz, 2H), 7.94 (d. J=8.2Hz, 2H); MS m/z 216 (M<sup>+</sup>).

General procedure for the synthesis of 3-methylthio-2-(*p*-tolyl) benzofuran (5). By the same procedure as described above for the preparation of 2, compound 5 was obtained from 3 (2 mmol), substituted phenol (2 mmol), and anhydrous p-toluenesulfonic acid (6 mmol). 5a: Yield 78%, mp 97-98?; IR (KBr) 2918, 1499, 1472, 1332, 1272, 1256, 1202, 1187, 1078, 1016, 970 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 2.36 (s, 3H), 2.40 (s, 3H), 2.48 (s, 3H), 7.11-7.57 (m, 5H), 8.17 (d, J=8.2Hz, 2H); MS m z 268 (M'). 5b: Yield 83%, mp 66-67 °C: IR (KBr) 2959. 2918, 1498, 1468, 1412, 1275, 1255, 1202, 1185, 1081, 1021, 969 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ1.31 (t. J=7.6Hz, 3H), 2.37 (s. 3H), 2.41 (s. 3H), 2.79 (q. J-7.6Hz, 2H), 7.11-7.54 (m. 5H). 8.17 (d, J-8.2Hz, 2H): MS m/z 282 (M<sup>+</sup>). 5c: Yield 82%, colorless liquid, IR (neat) 2952, 2920, 1613, 1501, 1470, 1420, 1255, 1204, 1184, 1078. 966 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ1.33 (d. J-7.0Hz, 6H), 2.38 (s. 6H), 2.82-3.38 (m, 111), 7.02-7.54 (m, 5H), 8.17 (d, J=8.2Hz, 2H); MS m z 296 (M<sup>+</sup>), 5d: Yield 86%, colorless liquid. IR (neat) 2916, 2920, 2867, 1501, 1471, 1363, 1332, 1278, 1259, 1205, 1185, 1077, 1019, 969  $cm^{-1}$ ; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.42 (s, 9H), 2.37 (s, 6H), 7.11-7.69 (m. 5H), 8.17 (d. J=8.2Hz, 2H); MS m z 310 (M'). 5e: Yield 67%, mp 125-126 °C; IR (KBr) 2917, 1497, 1456, 1442, 1254, 1120, 1187, 1076, 1065, 1012, 971 cm<sup>-1</sup>; <sup>1</sup>Η NMR (CDCl<sub>3</sub>) δ2.36(s, 3H), 2.42 (s. 3H), 7.15-7.68 (m. 511), 8.17 (d, J-8.2Hz, 2H), MS m/z 288 (M<sup>+</sup>).

General procedure for the synthesis of 3-methylthio-2-(p-chlorophenyl)benzofuran (6). By the same procedure as described above for the preparation of 2, compound 6 was obtained from 4 (2 mmol), substituted phenol (2 mmol), and anhydrous p-toluenesulfonic acid (6 mmol), 6a; Yield 87%, mp 107-108 °C; IR (KBr) 2920. 1486, 1470, 1401, 1254, 1202, 1090, 1074, 1010, 971 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ2.37 (s. 3H), 2.49 (s. 3H), 7.07-7.52 (m. 5H), 8.26 (d. J=8.7Hz, 2H); MS m/z 288 (M1). 6b: Yield 90%, mp 64-65 °C; IR (KBr) 2955, 2925. 2865, 1544, 1485, 1467, 1400, 1337, 1254, 1202, 1092. 1076, 1009, 966 cm<sup>-1</sup>: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.31 (t, J-7.6Hz, 3H), 2.37 (s. 3H), 2.79 (q. J-7.6Hz, 2H), 7.11-7.52 (m, 5H), 8.25 (d, J-8.7Hz, 2H): MS m z 302 (M<sup>-</sup>). 6c: Yield 84%, mp 81-82 C; IR (KBr) 2952, 2919. 2862, 1545, 1486, 1463, 1422, 1402, 1383, 1254, 1204, 1177, 1094, 1077, 1012, 969 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>4</sub>) δ 1.33 (d. J=7.0Hz, 6H), 2.38 (s. 3H), 2.71-3.19 (m. 1H), 7.10-5.53 (m. 5H). 8.25 (d. J=8.7Hz, 2H): MS m z 316 (M'). 6d: Yield 90%, mp 83-84 °C: IR (KBr) 2959,

1472, 1400, 1360, 1273, 1257, 1206, 1090, 1076, 1028, 1011 cm<sup>-1</sup>: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.42 (s. 9H), 2.38 (s. 311), 7.16-7.71 (m. 511), 8.25 (d. J=8.211z, 2H); MS m z 330 (M<sup>+</sup>). **6e**: Yield 75%, mp 136-137 °C: IR (KBr) 2942, 2929, 1483, 1456, 1442, 1400, 1253, 1198, 1092, 1065, 1009, 971 cm<sup>-1</sup>: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.37 (s. 3H), 7.18-7.69 (m. 5H), 8.25 (d. J=8.2Hz, 2H); MS m z 308 (M<sup>-</sup>).

General procedure for the synthesis of 2-(p-tolyl)benzofuran (7). Compound 5 (1.2 mmol) was heated at 60-65 °C in ethanol (30 mL) containing Raney nickel (W-2, 1.8 g) for 1h. The Raney nickel was removed by filtration and the solvent was evaporated off. The residual solid was recrystallized from ethanol to give 7, 7a: Yield 95%, mp 154-155 °C: IR (KBr) 2962, 1587, 1505, 1467, 1290, 1266, 1207, 1121, 1036, 1015, 912 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 2.39 (s, 311), 2.43 (s, 314), 6.89 (s, 114), 6.9 7-7.47 (m. 5H), 7.75 (d. J=8.2Hz, 2H); MS m/z 222 (M<sup>-</sup>), 7b: Yield 95%, mp 108-109 °C; IR (KBr) 2911, 1505, 1463, 1289, 1265, 1209, 1195, 1111, 1037, 1015, 932 em<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.28 (t, J=7.6Hz, 3H), 2.38 (s, 3H), 2.74 (q. J=7.6Hz, 2H), 6.89 (s. 1H), 7.01-7.51 (m. 5H), 7.74 (d, J=8.2Hz, 2H); MS m/z 236 (M<sup>1</sup>), 7c: Yield 95% o, mp 97-98 °C; IR (KBr) 2956, 2866, 1506, 1471, 1381, 1354, 1288, 1264, 1208, 1160, 1112, 1034, 1015, 916 cm<sup>-1</sup>; <sup>1</sup>H NMR (CIXCl<sub>3</sub>)  $\delta$  1.18 (d, J=7.0Hz, 6H), 2.38 (s, 311), 2.68-3.23 (m, 111), 6.90 (s, 114), 7.03-7.50 (m, 511), 7.74 (d, J-8.2Hz, 2H); MS m/z 250 (M<sup>-</sup>), 7d: Yield 95%, mp 124-125 °C; IR (KBr) 2956, 1505, 1473, 1458, 1364, 1328, 1277, 1208, 1166, 1127, 1036, 1016, 913 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>2</sub>) 1.39 (s. 9H), 2.38 (s. 3H), 6.92 (s. 1H), 7.06-7.66 (m. 5H), 7.75 (d. J=8, 2Hz, 2H); MS m z 264 (M<sup>+</sup>), 7e: Yield 95%, mp 177-178 °C: IR (KBr) 2955, 1585, 1504, 1444, 1262, 1206, 1163, 1114, 1060, 1035, 1015, 927 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) 82.39 (s. 911), 6.94 (s. 1H), 7.01-7.58 (m. 5H), 7.74 (d. J-8.2Hz, 211); MS m z 242 (M<sup>-</sup>).

General procedure for the synthesis of 2-(*p*-chlorophenyl) benzofuran (8). By the same procedure as described above for the preparation of 7, compound 8 was obtained from 6 (1 mmol), Raney nickel (1.5 g), and ethanol (25 mL). The residual solid was recrystallized from ethanol to give 8. 8a: Yield 88% of mp 186-187 °C;

IR (KBr) 2913, 1579, 1488, 1462, 1404, 1261, 1210, 1195, 1092, 1034, 1009, 914 cm<sup>-1</sup>, <sup>1</sup>Π NMR (CDCI<sub>0</sub>) δ 2.43 (s. 311), 6.91 (s. 111), 6.99-7.55 (m. 5H), 7.77 (d. J-8.8Hz, 2H): MS m/z 242 (M<sup>+</sup>). 8b: Yield 84%, mp 149-150 °C; IR (KBr) 2970, 1487, 1466, 1403, 1264. 1193, 1089, 1033, 1009, 911 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.29 (t. J=7.0Hz, 3H), 2.75 (q. J=7.6Hz, 2H), 6.94 (s. 1H), 7.01-7.50 (m, 5H), 7.78 (d, J=8.2Hz, 2H); MS m/ z 256 (M'). 8c: Yield 89%, mp 152-153 °C; IR (KBr) 2958, 1580, 1488, 1469, 1403, 1355, 1267, 1160, 1117, 1103, 1091, 1031, 1009, 915 cm<sup>-1</sup>, <sup>1</sup>Π NMR (CDCl<sub>i</sub>) δ 1.30 (d. J-7.0Hz, 6H), 2.80-3.23 (m. 111), 6.94 (s, 111), 7.09-7.56 (m. 5H), 7.78 (d, J-8.8Hz, 2H); MS m/z 270 (M\*). 8d: Yield 90%, mp 154-155 °C; IR (KBr) 2956, 1579, 1471, 1402, 1366, 1328, 1275, 1166, 1127, 1088. 1032, 1009, 913 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>4</sub>) δ 1.39 (s. 9H), 6.96 (s. 1H), 7.10-7.57 (m. 5H), 7.78 (d. J=8.8Hz, 2H); MS m/z 284 (M<sup>+</sup>), 8e: Yield 76%, mp 151-152 °C; IR (KBr) 2957, 1599, 1581, 1486, 1443, 1324, 1273, 1260, 1163, 1091, 1061, 1033, 1011, 926 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 6.93 (s. 1H), 7.15-7.86 (m, 7H); MS m/z 262 (M<sup>+</sup>).

#### REFERENCES

- Talapatra, B.; Ray, T.; Talapatra, S. K. J. Indian Chem. Soc. 1978, 55, 1204.
- Levan, N.; Pham, T. V. Phytochemistry 1981, 20, 485.
- Bohlmann, F.; Ahmed, M.; Robinson, H.; King, R. M. Phytochemistry 1981, 20, 1493.
- Achenbach, H.; Gross, J.; Dominguez, X. A.; Star, J. V.; Salgado, F. *Phytochemistry* 1987, 26, 2041.
- Sheen, W. S.; Tsai, I. L.; Teng, C. M.; Chen, I. S. Phytochemistry 1994, 36, 213.
- (a) Choi, H. D.; Seo, P. J.; Son, B. W. J. Korean Chem. Soc. 1999, 43, 237. (b) Choi, H. D.; Seo, P. J.; Son, B. W. J. Korean Chem. Soc. 1999, 43, 606. (c) Seo, P. J.; Ha, M. C.; Choi, H. D.; Son, B. W. J. Korean Chem. Soc. 2000, 44, 391.
- Choi, H. D.; Seo, P. J. J. Korean Chem. Soc. 2001, 45, 274.
- Corey, E. J.; Chaykovsky, M. J. Am. Chem. Soc. 1965, 87, 1345.
- Schneiders, G. E.; Stevenson, R. J. Org. Chem. 1979, 44, 4710.