

단 신

TsCl와 염기존재에서의 *N*-(2-Hydroxyethyl)-*N'*-methylthioureas의 고리화반응

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Cyclization Reaction of *N*-(2-Hydroxyethyl)-*N'*-methylthioureas in the Presence of TsCl and Base

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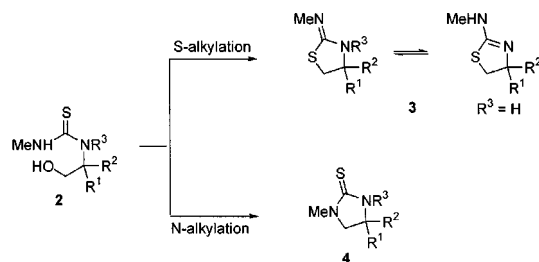
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2-Aminothiazolines have gained much interest as biologically active molecules such as potent inhibitors of human nitric oxide synthase,¹ octopaminergic-agonists,² anthelmintics,³ and anti-inflammatory agents.⁴ These compounds are usually prepared by the hydrochloric acid-catalyzed cyclization of *N*-(2-hydroxyethyl)thioureas^{2a,2b,3,5} or the cyclization of hydrogen sulfate of thioureas in aqueous basic conditions.^{2a, 6} These methods give low yields for the formation of the 2-aminothiazolines and are not applicable to acid sensitive or racemization-prone substrates due to the vigorous acidic or basic reaction conditions.

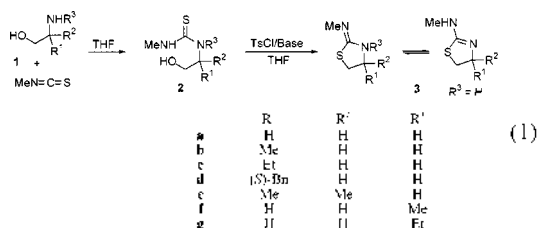
Recently, we preliminarily reported that 2-methylaminothiazolines **3** were synthesized from *N*-(2-hydroxyethyl)-*N'*-methylthioureas **2** by the intramolecular Mitsunobu reaction conditions.⁷ The Mitsunobu reaction of **2** proceeded through mild nucleophilic attack upon the oxyphosphonium intermediate either by the sulfur atom to provide 2-aminothiazoline **3** or by the nitrogen to give the 2-imidazolidinethione **4** depending on the structure of **2** (Scheme 1). With thioureas **2a-2e** prepared from *N*-unsubstituted aminoalcohols ($R^3=H$), S-cyclization to **3** was mainly observed with a trace amount of the N-cyclized products. However, the thioureas **2f** and **2g** prepared from *N*-substituted aminoalcohols ($R^3=Me$, Et) gave a mixture of 2-iminothiazolidines (S-alkylation products) and 2-imidazolidinethiones (N-alkylation prod-

ucts) in the ratio of 69:31 and 57:43, respectively. Therefore, we needed to develop a new way to 2-methylaminothiazolines to improve more selective yields of S-cyclized products in the case of **2f** and **2g**. In the course of our work in the cyclization reaction of *N*-(2-hydroxyethyl)-*N*-phenylthioureas, we found that one-pot reaction of thioureas proceeds in the presence of TsCl and NaOH to give 2-phenylaminothiazolines in good yields.⁸ These results prompted us to examine the one-pot reaction of *N*-(2-hydroxyethyl)-*N'*-methylthioureas **2** for the preparation of **3**. Thioureas **2** were readily prepared from the reaction of the corresponding 1,2-aminoalcohols with methyl isothiocyanate in tetrahydrofuran (THF) solution at room temperature in good yields, which provided exclusively the desired products under mild conditions, thus avoiding the need for O-protection. A survey of one-pot reactions by the combination of



Scheme 1.

TsCl with various basic metallic (*t*-BuOK, NaOH, and NaI) or non-metallic (Et₃N/DMAP) reagents was per-



formed to **2** in THF (Eq. 1).

One-pot reaction conditions using *t*-BuOK and TsCl were first applied to various thioureas **2**.⁹ With **2f** and **2g** prepared from *N*-substituted aminoalcohols, *N*-cyclization occurred mainly producing **4f** and **4g** in the yields of 70% and 45%, respectively while with **2a-2e** prepared from *N*-unsubstituted aminoalcohols, a small amount of 2-methylaminothiazolines **3** were produced along with unknown mixture of products. Contrary to *N*-(2-hydroxyethyl)-*N'*-phenylthioureas, the application of the reaction conditions using NaOH/TsCl also gave unacceptable results regardless of the structure of thioureas **2**. To improve the nucleophilicity of thioureas **3** the combination of more basic NaI and TsCl was explored to various thioureas **2** which resulted in unknown mixture or low selectivity and conversion yields. However, **2g** under NaI/TsCl gave only the *N*-cyclization product with a 75% conversion. The above reaction conditions in the case of **2f** and **2g** gave unsatisfactory results to prepare the 2-methylaminothiazolines, leading to *N*-cyclization to **4f** and **4g**.

We next turned to use a non-metallic basic reagent, Et₃N/DMAP. The refluxed reaction in the presence of 5 equiv of Et₃N and 0.5 equiv of DMAP gave *S*-cyclized and *N*-cyclized mixtures in the case of **2a-2e**. With thiourea **2f** and **2g**, however, the essential 2-methylaminothiazolines were obtained in 85% and 90% yields, respectively. Thus, the use of Et₃N/DMAP in the case of **2f** and **2g** was the most effectively *S*-cyclized product with almost complete regioselectivity. Although further investigation is needed to understand these reactions, the *S*-cyclization selectivity is remarkably affected by the base employed depending on the nucleophilicity of thioureas.

Mitsunobu reaction was a condition for the regiocontrolled conversion of the only thioureas **2a-2e** derived

from *N*-unsubstituted aminoalcohols into 2-methylaminothiazolines.⁷ Most of one-pot reaction conditions of thioureas **2** using the combination of bases and TsCl produced the mixture of *S*- or *N*-cyclized products depending on the substrates and bases. However, the use of Et₃N/DMAP was the most effective condition for the regioselective conversion of the thioureas **2f** and **2g** derived from *N*-substituted aminoalcohols into the requisite *S*-cyclized products.

Experimental Section

General. ¹H NMR and ¹³C NMR spectra were recorded using 300 MHz and 75 MHz NMR spectrometer; chemical shifts are reported in ppm using CDCl₃ as solvent and TMS as an internal standard. Melting points were determined on a capillary apparatus and uncorrected. Mass spectra were recorded on a HP 5983B GC Mass spectrometer. Analytical TLC was performed on 0.25 mm precoated silica gel plates. Flash chromatography was carried out with 230-400 mesh silica gel.

General procedure for the preparation of thiourea **2.** To a stirred solution of 1,2-aminoalcohol (4.59 mmol) in THF (10 mL) under nitrogen at room temperature was added a solution of methyl isothiocyanate (0.50 mL, 4.18 mmol) in THF (5 mL) dropwise for 5 min with a syringe. The reaction mixture was stirred for 30 min and evaporated, and purified by flash column chromatography to give **2**.

***N*-(2-Hydroxyethyl)-*N'*-methylthiourea (**2a**).** Yield: 92%; mp 70-72 °C; *R*_f = 0.2-0.3 (ethyl acetate); ¹H NMR (300 MHz, CDCl₃) δ 3.85-3.82 (2H, dd, *J*=4.2, 1.2), 3.69 (2H, bs), 3.02 (3H, d, *J*=4.5).

***N*-[(2-Hydroxy-1-methylethyl)-*N'*-methylthiourea (**2b**).** Yield: 66%; *R*_f = 0.4 (ethyl acetate); ¹H NMR (300 MHz, CDCl₃) δ 3.74 (2H, dd, *J*=3.5, 11.1), 3.55 (1H, dd, *J*=6.9, 11.0), 3.01 (3H, d, *J*=4.4), 1.21 (3H, d, *J*=6.7).

***N*-[(1-Ethyl-2-hydroxy)ethyl]-*N'*-methylthiourea (**2c**).** Yield: 81%; *R*_f = 0.5 (ethyl acetate); ¹H NMR (300 MHz, CDCl₃) δ 3.78 (1H, dd, *J*=3.4, 11.1), 3.59 (2H, dd, *J*=6.8, 11.1), 3.02 (2H, d, *J*=4.5), 1.49-1.65 (2H, m), 0.98 (3H, t, *J*=7.4).

***N*-[(1*S*)-2-Hydroxy-1-phenylmethyl]ethyl]-*N'*-methylthiourea (**2d**).** Yield: 85%; *R*_f = 0.3-0.5 (ethyl acetate); ¹H NMR (300 MHz, CDCl₃) δ 7.23-7.31 (5H, m), 3.75 (1H,

dd, $J=3.6, 11.1$), 3.59 (1H, dd, $J=5.7, 11.1$), 2.82-3.01 (2H, 1H, m), 2.90 (3H, d, $J=3.3$).

***N*-[1,1-Dimethyl-2-hydroxyethyl]-*N'*-methylthiourea (2e).** Yield: 80%*v*; $R_f=0.5$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.65 (2H, s), 3.06 (3H, d, $J=4.5$), 1.32 (6H, s); ^{13}C NMR (75 MHz, CDCl_3) δ 181.4, 70.4, 57.0, 32.1, 24.5.

***N*-(2-Hydroxyethyl)-*N*-methyl-*N'*-methylthiourea (2f).** Yield: 75%*v*; $R_f=0.3$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.88 (4H, s), 3.23 (3H, s), 3.12 (3H, d, $J=4.5$).

***N*-Ethyl-*N'*-(2-hydroxyethyl)-*N'*-methylthiourea (2g).** Yield: 95%*v*; $R_f=0.4$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.85-3.88 (2H, m), 3.80-3.71 (4H, m), 3.09 (3H, d, $J=4.5$), 1.24 (3H, t, $J=7.2$).

General procedure for the preparation of 2-methylaminothiazolines 3

TsCl/Metallic Base Conditions: To a stirred solution thiourea **2** (0.88 mmol) and base (2.2 mmol) in THF (10 mL) under nitrogen at room temperature was added a solution of TsCl (0.18 g, 0.97 mmol) in THF (5 mL) dropwise for 5 min with a syringe. The reaction mixture was stirred for 30 min, added with water (30 mL), and extracted with ether (50 mL \times 3). The organic layer was dried, filtered, evaporated, and purified by flash column chromatography to give **3** or **4**.

TsCl/Et₃N/DMAP Conditions: To a stirred solution thiourea **2** (0.88 mmol) and triethylamine (0.61 mL, 4.4 mmol) and 4-(dimethylamino)pyridine (49 mg, 0.44 mmol) in THF (10 mL) under nitrogen at room temperature was added a solution of TsCl (0.18 g, 0.97 mmol) in THF (5 mL) dropwise for 5 min with a syringe. The reaction mixture was refluxed over night, added with water (30 mL), and extracted with ether (50 mL \times 3). The organic layer was dried, filtered, evaporated, and purified by flash column chromatography to give **3** or **4**.

4,5-Dihydro-*N*-methyl-2-thiazolamine (3a). mp 90 °C; $R_f=0.1-0.3$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 4.00(2H, t, $J=7.4$), 3.34(2H, t, $J=7.4$), 2.93(3H, s); ^{13}C NMR (75 MHz, CDCl_3) δ 162.9, 59.8, 35.3, 31.3; HRMS (EI) calcd for $\text{C}_4\text{H}_8\text{N}_2\text{S}$ 116.0408 found 116.0428.

4,5-Dihydro-4-methyl-*N*-methyl-2-thiazolamine (3b). mp 72-75 °C; $R_f=0.1-0.2$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 4.37-4.44 (1H, m), 3.56 (1H, dd, $J=3.6, 10.8$), 3.10 (1H, dd, $J=3.9, 10.8$), 3.02 (3H, s), 1.45 (3H, d, $J=5.1$); ^{13}C NMR (75 MHz, CDCl_3) δ 161.3, 67.3, 41.2,

31.3, 21.3; HRMS (EI) calcd for $\text{C}_5\text{H}_{10}\text{N}_2\text{S}$ 130.0564, found 130.0545.

4,5-Dihydro-4-ethyl-*N*-methyl-2-thiazolamine (3c). mp 61 °C; $R_f=0.1-0.2$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 4.09-4.22 (1H, m), 3.40 (1H, dd, $J=7.2, 10.5$), 3.00 (1H, dd, $J=7.3, 10.5$), 2.93 (3H, s), 1.71-1.85 (1H, m), 1.49-1.64 (1H, m), 0.99 (3H, t, $J=7.4$); ^{13}C NMR (75 MHz, CDCl_3) δ 161.0, 73.9, 39.1, 31.5, 28.7, 10.9; HRMS (EI) calcd for $\text{C}_6\text{H}_{12}\text{N}_2\text{S}$ 144.0721, found 140.0709.

(4S)-4,5-Dihydro-*N*-methyl-4-phenylmethyl-2-thiazolamine (3d). mp 105 °C; $R_f=0.1-0.2$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 7.19-7.33 (5H, m), 4.42-4.51 (1H, m), 3.23 (1H, dd, $J=7.2, 10.8$), 3.15 (1H, dd, $J=4.8, 13.5$), 3.06 (1H, dd, $J=5.7, 10.8$), 2.17 (1H, dd, $J=9.3, 13.5$), 2.95 (3H, s); ^{13}C NMR (75 MHz, CDCl_3) δ 167.7, 138.9, 129.1, 128.3, 126.1, 73.3, 41.3, 38.4, 31.4; HRMS (EI) calcd for $\text{C}_{11}\text{H}_{11}\text{N}_2\text{S}$ 206.0877, found 206.0838.

4,5-Dihydro-4,4-dimethyl-*N*-methyl-2-thiazolamine (3e). mp 110 °C; $R_f=0.1-0.2$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.26 (2H, s), 2.94 (3H, s), 1.42 (6H, s); ^{13}C NMR (75 MHz, CDCl_3) δ 159.4, 73.1, 46.2, 31.4, 28.3; HRMS (EI) calcd for $\text{C}_6\text{H}_{12}\text{N}_2\text{S}$ 144.0721, found 144.0737.

3-Methyl-2-methyliminothiazolidine (3f). ^1H NMR (300 MHz, CDCl_3) δ 3.42 (2H, t, $J=6.6$), 3.12 (2H, t, $J=6.6$), 3.04 (3H, s), 2.85 (3H, s); ^{13}C NMR (75 MHz, CDCl_3) δ 160.8, 53.2, 41.5, 33.8, 26.8.

1,3-Dimethyl-2-imidazolidinethione (4f). $R_f=0.7$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.54 (4H, s), 3.13 (6H, s); ^{13}C NMR (75 MHz, CDCl_3) 183.4, 48.2, 35.0.

3-Ethyl-2-methyliminothiazolidine (3g). ^1H NMR (300 MHz, CDCl_3) δ 3.46 (2H, t, $J=6.6$), 3.37 (2H, q, $J=7.2$), 3.13 (2H, t, $J=6.6$), 3.04 (3H, s), 1.14 (3H, t, $J=7.2$); ^{13}C NMR (75 MHz, CDCl_3) δ 156.7, 50.2, 41.4, 41.0, 26.7, 12.0.

1-Ethyl-3-methyl-2-imidazolidinethione (4g). $R_f=0.7$ (ethyl acetate); ^1H NMR (300 MHz, CDCl_3) δ 3.67 (2H, q, $J=7.2$), 3.54 (4H, s), 3.13 (3H, s), 1.17 (3H, t, $J=7.2$); ^{13}C NMR (75 MHz, CDCl_3) δ 182.6, 48.3, 45.3, 42.4, 34.8, 12.0.

REFERENCES

- (a) Southan, G. J.; Zingarelli, B.; O'Connor, M.; Salz-

- man, A. L.; Szabo, C. *J. Pharmacol.* **1996**, *117*, 619. (b) Moore, W. M.; Webber, R. K.; Fok, K. F.; Jerome, G. M.; Connor, J. R.; Manning, P. T.; Wyatt, P. S.; Misko, T. P.; Tjoeng, F. S.; Currie, M. G. *J. Med. Chem.* **1996**, *39*, 669.
2. (a) Hirashima, A.; Yoshii, Y.; Eto, M. *Agric. Biol. Chem.* **1991**, *55*, 2537. (b) Hirashima, A.; Yoshii, Y.; Eto, M. *Biosci. Biotech. Biochem.* **1992**, *56*, 1062. (c) Hirashima, A.; Tomita, J.; Pan, C.; Taniguchi, E.; Eto, M. *Bioorg. & Med. Chem.* **1997**, *5*, 2121.
3. Caujolle, R.; Amarouch, H.; Payard, M.; Loiseau, P. R.; Borics, C.; Loiseau, P. M.; Garyral, P. *Eur. J. Med. Chem.* **1989**, *24*, 287.
4. Bender, P. E.; Hill, D. T.; Offen, P. H.; Razgaitis, K.; Lavanchy, P.; Stringer, O. D.; Sutton, B. M.; Griswold, D. E.; DiMartino, M.; Walz, D. T.; Lantos, I.; Ladd, C. B. *J. Med. Chem.* **1985**, *28*, 1169.
5. (a) Cherbuliez, E.; Baehler, B.; Espejo, O.; Jindra, H.; Willahn, B.; Rabinovitz, J. *Heb. Chim. Acta* **1967**, *50*, 331. (b) Cherbuliez, E.; Baehler, B.; Jaccard, S.; Jindra, H.; Weber, G.; Wyss, G.; Rabinovitz, J. *Helv. Chim. Acta* **1966**, *49*, 807.
6. Dewey, C. S.; Bafford, R. A. *J. Org. Chem.* **1965**, *30*, 491.
7. Kim, T. H.; Cha, M.-H. *Tetrahedron Lett.* **1999**, *40*, 3125.
8. (a) Kim, T. H.; Min, J. K.; Lee, G.-J. *Tetrahedron Lett.* **1999**, *40*, 8201. (b) Kim, T. H.; Min, J. K.; Lee, G.-J. *Bull. Korean Chem. Soc.* **2000**, *21*, 919.
9. Kim, T. H.; Lee, G.-J. *J. Org. Chem.* **1999**, *64*, 2941.
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