

Annomocherin, Annonacin and Annomontacin: A Novel and Two Known Bioactive Mono-Tetrahydrofuran Annonaceous Acetogenins from *Annona cherimolia* Seeds

Dal Hwan Kim¹, Jong Keun Son², and Mi Hee Woo¹

¹Department of Pharmacy, College of Pharmacy, Catholic University of Daegu, Gyeongsan, 712-702, Korea, and ²Department of Pharmacy, College of Pharmacy, Yeungnam University, Gyeongsan 712-749, Korea

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A novel and two known bioactive mono-tetrahydrofuran (THF) annonaceous acetogenins, annomocherin (1), annonacin (2) and annomontacin (3), have been isolated from the fractionated ethanolic extracts of the seeds of *Annona cherimolia*, guided by the brine shrimp lethality test (BST). Their structures were elucidated on the basis of spectroscopic and chemical methods. All compounds have a relative stereochemistry of *threo/trans/threo* for the mono-THF ring with two flanking hydroxyls. Compound 1 has a double bond at C-23/24 of aliphatic chain. Compound 1 was isolated from natural sources for the first time, and was named annomocherin. Two known Compounds 2 and 3 which have never been isolated from this species before, were obtained. Compound 1 exhibited potent and selective cytotoxicities against the breast carcinoma (MCF-7) and kidney carcinoma (A-498) cell lines with 100 to 1,000 times the potency of adriamycin. In brine shrimp lethality test (BST), 1-3 exhibited cytotoxicity.

Key words: Annonaceous acetogenins, *Annona cherimolia*, Annomocherin, Annonacin, Annonacin, Cytotoxicities

INTRODUCTION

Annonaceous acetogenins are a relatively new class of natural polyketides which have promising anticancer, antiinfective, and pesticidal properties (Morre et al., 1995; Zeng et al., 1996). Annona cherimolia Mill. (Annonaceae) is the only species cultivated in Europe and is grown principally in the south of Spain, between Almuñecar and La Herradura (Granada coast) (Fies, 1959). Annona cherimolia is used in traditional medicine as an insecticide and parasiticide. In our previous studies of Annona cherimolia seeds, we isolated five new mono-THF acetogenins (annocherine, (2,4)-cis- and trans-annocherimolin (Kim et al., 2001) and five known acetogenins (cis-annonacin, (2,4)-cis- and trans-isoannonacins (Woo et al., 1999), corrosolin, and compound-2 (Kim et al., 1999)). Our continuing search

for other members of the same class in the seeds of the *Annona cherimolia* has led to the isolation of three acetogenins, one of which is a novel mono-THF acetogenin and was named annomocherin (1). The other two are the already known annonacin (2) and annomontacin (3). Their occurrence in *Annona cherimolia* is reported here for the first time.

MATERIALS AND METHODS

Instruments and reagents

Melting points were determined on a Yanaco micro melting point apparatus and were uncorrected. Optical rotations were measured on a JASCO DIP-370 digital polarimeter. IR spectra were measured on a JASCO FT/IR-300E spectrophotometer. UV spectra were obtained on a Shimadzu UV-1601PC spectrophotometer. NMR spectroscopy was taken on a Varian VXR 500S spectrophotometer in CDCl₃ using TMS as an internal standard. Mass spectra was recorded on a Quattro spectrometer. For TLC, silica gel 60 F_{254} (EM 5717) glass plates (0.25 mm) were

Correspondence to: Mi Hee Woo, Department of Pharmacy, College of Pharmacy, Catholic University of Daegu, Gyeongsan, 712-702, Korea

E-mail: woomh@cuth.cataegu.ac.kr

used and visualized by spraying with 5% phosphomolybdic acid in MeOH and subsequent heating. HPLC was performed on a Waters 600 apparatus equipped with a Waters 486 UV detector at 225 nm using the Autochrowin software system (Waters Korea, Seoul, Korea). A μ Bondapak C₁₈ columns (19 × 300 mm and 7.8 × 300 mm) were used for preparative purpose. For preparation of tetra-TMSi derivative, *N,O-bis*-(trimethylsilyl)acetamide (BSA) and pyridine in silylation grade were purchased from the Aldrich Company.

Plant material

The seeds of Annona cherimolia were obtained in 1996 from fruits grown commercially in plantations in southern California and purchased from Hurov Botanicals and Seeds located in Chula Vista, California.

Extraction and isolation

The dried seeds of *Annona cherimolia* (8 kg) were repeatedly percolated with 95% EtOH to yield 700 g of an extract (F001), on removal of solvent. F001 was partitioned between CH_2Cl_2 - H_2O (1:1) to yield the H_2O -soluble fraction (F002, 300 g) and the CH_2Cl_2 -soluble fraction

(F003, 400 g). F003 was then partitioned between 90% aqueous MeOH-hexane (1:1) to yield a hexane-soluble fraction (F006, 150 g) and an aqueous MeOH-soluble fraction (F005, 250 g). All fractions were subjected to the brine shrimp lethality (BST), with the most active fraction being F005 (BST LC₅₀=1.13 \times 10⁻² µg/mL). 250 g of F005 was subjected to open column chromatography over Si gel (2.8 kg) eluted with a gradient of hexane/CHCl₃/MeOH. Fractions (F₁-1 to F₁-18) were collected and pooled according to their similar TLC patterns. The BST active pool, F₁-12 (130 g), was further resolved on another Si gel (1.5 kg) open column, eluted with a gradient of hexane/ CHCl₃/MeOH. Fractions were collected into 13 pools (F₂-1 to F₂-13) on the basis of similar TLC patterns. Further purification of the most bioactive BST fractions (F₂-6, BST $LC_{50}=2.00\times10^{-4}$ µg/mL) were carried out by HPLC eluted with 85% acetonitrile in water to afford compounds 1 (15 mg), 2 (34 mg) and 3 (20 mg).

Annomocherin (1)

white amorphous powder, mp 74.6-75.3°C, $\left[\alpha\right]_D^{25}$: +9.2° (c 0.02, CH₂Cl₂), UV λ_{max} nm (log ϵ): 220 (3.7), IR ν (cm⁻¹): 3467 (OH), 1734 (C=O of α , β -unsaturated γ -lactone), ¹H NMR see Table I.

Table I. ¹H and ¹³C NMR spectral data of 1, 2, and 3 (CDCl₃, δ)

No.		¹³ C NMR (75.5 MHz)				
	1	2	3	1	2	3
1	-	-	_	174.68	174.62	174.581
2	-	-	-	131.16	130.16	131.18
3a	2.40 dd (15.0, 8.2)	2.39 dd (15.0, 8.2)	2.39 dd (15.0, 8.2)	33.34	33.16	33.48
3b	2.52 dt (15.0, 1.5)	2.52 dt (15.0, 1.5)	2.53 dt (15.0, 1.5)			
4	3.84 m	3.80 m	3.80 m	69.93	69.66	69.91
5-9	1.26-1.57 m	1.25-1.59 m	1.25-1.59 m	22.69 - 37.35	22.57-37.24	22.67-37.43
10	3.59 m	3.59 m	3.58 m	<i>7</i> 1. <i>77</i>	71.45	71.86
11-14	1.26-1.57 m	1.25-1.59 m	1.25-1.59 m	22.69-37.35	22.57-37.24	22.67-37.43
15	3.43 m	3.40 m	1.25-1.59 m	74.02	73.98	22.67-37.43
16	3.80 m	3.80 m	1.25-1.59 m	82.66	82.64	22.67-37.43
17a	1.67 m	1.67 m	3.40 m	28.76	28.73	74.02
17b	1.98 m	1.98 m				
18a	1.67 m	1.67 m	3.80 m	28.79	28.73	82.62
18b	1.98 m	1.98 m				
19a	3.80 m	3.80 m	1.67 m	82.63	82.64	28.74
19b			1.98 m			
20a	3.43 m	3.40 m	1.67 m	73.58	73.98	28.74
20b			1.98 m			
21	1.48 m	1.25-1.59 m	3.80 m	33.38	22.57-37.24	82.62
22	2.20 m	1.25-1.59 m		23.33	22.57-37.24	74.02
23	5.37 m	1.25-1.59 m	3.40 m	128.95	22.57-37.24	23.14-37.85
24	5.37 m	1.25-1.59 m	1.25-1.59 m	130.85	22.57-37.24	23.14-37.85
25	2.04 m	1.25-1.59 m	1.25-1.59 m	27.25	22.57-37.24	23.14-37.85
26-31	1.26-1.57 m	1.25-1.59 m	1.25-1.59 m	22.69-37.35	22.57-37.24	23.14-37.85
32	0.88 t (7.0)	0.88 t (7.0)	1.25-1.59 m	14.12	14.02	23.14-37.85
33	7.19 q (1.5)	7.18 q (1.5)	1.25-1.59 m	151.91	151.85	23.14-37.85
34	5.06 dq (7.0,1.5)	5.06 dq (7.0, 1.5)	0.88 t (7.0)	78.03	77.94	14.10
35	1.43 d (6.5)	1.43 d (7.0)	7.18 q (1.5)	19.11	18.99	151.81
36			5.06 dq (7.0, 1.5)			77.97
3 <i>7</i>			1.43 d (7.0)			19.10

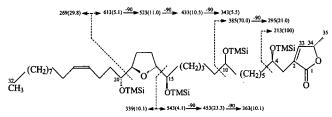


Fig. 1. Diagnostic EIMS fragmentations of tetra-TMSi derivative of **1** (intensities are indicated in parentheses).

Tetra-TMSi derivative (1a) of annomocherin Mass (EI) *m/z*: Fig. 1.

Annonacin (2)

white amorphous powder, mp 59.3-61.7°C, $[\alpha]_D^{22}$: +8.6° (c 0.07, CH₂Cl₂), UV λ_{max} nm (log ε): 205 (4.0), IR ν (cm⁻¹): 3348 (OH), 1751 (C=O of α , β -unsaturated γ -lactone), ¹H NMR see Table I.

Tetra-TMSi derivative (2a) of annonacin

Mass (EI) m/z: Fig. 2.

Annomontacin (3)

white amorphous powder, mp 65.2-67.4 °C, $\left[\alpha\right]_D^{25}$: -7.5° (c 0.06, CH₂Cl₂), UV λ_{max} nm (log ϵ): 205 (3.8), IR ν (cm⁻¹): 3567 (OH), 1736 (C=O of α , β -unsaturated γ -lactone), ¹H NMR see Table I.

Tetra-TMSi derivative (3a) of annomontacin Mass (EI) m/z: Fig. 3.

$$181(9.3) \xrightarrow{-90} 271(100) \xrightarrow{-90} 613(12.0) \xrightarrow{-90} 523(10.0) \xrightarrow{-90} 433(12.0)$$

$$OTMSi$$

Fig. 2. Diagnostic EIMS fragmentations of tetra-TMSi derivative of 2 (intensities are indicated in parentheses).

$$(CH_2)_{11} \xrightarrow{90} 551(45.3) \xrightarrow{90} 461(15.2) \xrightarrow{90} 213(71.2)$$

$$OTMSi \xrightarrow{17} (CH_2)_5 \xrightarrow{10} (CH_2)_5 \xrightarrow{10} (CH_2)_5 \xrightarrow{385(80.1)} 341(70.1) \xrightarrow{90} 481(55.9) \xrightarrow{90} 391(13.8)$$

Fig. 3. Diagnostic EIMS fragmentations of tetra-TMSi derivative of **3** (intensities are indicated in parentheses).

Table II. Characteristic ¹H NMR data of Mosher esters of **1s** and **1r** for determinations of stereochemistries

Position	1s δs	1r δ r	Δδ s-r
5	1.59	1.58	+ 0.01
4	5.32	5.36	R
3	2.58	2.65	- 0.07
33	6.73	6.97	- 0.24
34	4.85	4.92	- 0.07
35	1.29	1.32	- 0.03
14	1.62	1.56	+ 0.06
15	4.98	5.01	R
16	3.93	4.05	- 0.12
17	1 <i>.</i> 57	1.63	- 0.06
	1.94	1.95	- 0.01
18	1.57	1.63	- 0.06
	1.94	1.95	- 0.01
19	3.93	4.05	- 0.12
20	4.98	5.01	R
21	1.62	1.56	+ 0.06

threo trans threo OR
$$(CH_2)_7$$
 $(CH_2)_7$ $(CH_2)_5$ $(CH_2)_5$

1 R=H, 1a R=TMSiOH, 1s R=(S)-MTPA, 1r R=(R)-MTPA

Preparation of TMSi-derivatives

Approximately 10 μ g of **1**, **2**, and **3** were separately treated with 0.2 μ l of pyridine and 2 μ l of *N*,*O-bis*-(trimethylsilyl) acetamide (BSA) for 5 h to give a **1a**, **2a** and **3a**, respectively (Laprevote et al., 1992).

Preparation of Mosher esters

1 mg of **1** in 0.5 mL of CH_2Cl_2 , were added sequentially 0.2 mL pyridine, 0.5 mg 4-(dimethylamino)-pyridine, and 12 mg of (R)-(-)- α -methoxy- α -(trifluoromethyl)-phenylacetyl (MTPA) chloride. The mixture was left at room temperature overnight and purified over a micro-column (0.6 × 6 cm) of silica gel (230-400 mesh) eluted with 3-4 mL of hexane- CH_2Cl_2 (1:2); the eluate was dried, CH_2Cl_2 (5 mL) was added, and the CH_2Cl_2 was washed using 1% NaHCO₃ (5 mL × 3) and H_2O (5 mL × 2); the washed eluate was dried *in vacuo* to give the S Mosher ester of **1**. Using (S)-(+)- α -methoxy- α -(trifluoromethyl)-phenylacetyl (MTPA) chloride afforded the R Mosher ester (Mosher et al., 1973; Rieser et al., 1992; Gu et al., 1994). Their pertinent 1H NMR chemical shifts are given in Table II.

Bioassay

The extracts, fractions, and isolated compounds were routinely evaluated for lethality to brine shrimp lethality test (BST) (Meyer et al., 1982; McLaughlin, 1991).

MTT cytotoxicity tests

Seven-day *in vitro* MTT cytotoxicity tests against human tumor cell lines were carried out at the Cell Culture Laboratory, Purdue Cancer Center, Purdue University, using standard protocols for A-549 (human lung carcinoma) (Giard *et al.*, 1973), MCF-7 (human breast carcinoma) (Soule *et al.*, 1973), HT-29 (human colon adenocarcinoma) (Fogh *et al.*, 1975), A-498 (human kidney carcinoma) (Giard *et al.*, 1973), PC-3 (human prostate adenocarcinoma) (Kaighn *et al.*, 1979), and MIA PaCa-2 (human pancreatic carcinoma) (Yunis *et al.*, 1977) with adriamycin as a positive control.

RESULTS AND DISCUSSION

Compound 1 was isolated as a white amorphous powder. The molecular weight of 1 was indicated by a peak at m/z 595 [MH]⁺ in the FABMS spectrum. The HRFABMS spectrum gave m/z 595.4574 for the [MH]+ (calcd 595.4580) corresponding to the molecular formula C₃₅H₆₃O₇. The presence of four hydroxyl groups was indicated by the losses of four molecules of H2O from [M+H]+ in the FABMS spectrum and by a broad absorption in the IR spectrum at 3467 cm⁻¹. This is confirmed by preparation of the tetra-TMSi derivative (1a). The EIMS of 1a showed successive losses (m/z 792, 702, 612, 522) of four TMSiOH (m/z 90) units from the M^+ (m/z 882) of the tetra-TMSi. The spectral data of 1 provided the characteristic features for the usual α , β unsaturated y-lactone moiety with a 4-OH, including an IR carbonyl absorption at 1734 cm⁻¹, a UV (MeOH) λ_{max} at 220 nm, six proton resonances at 8 7.19 (H-33), 5.06 (H-34), 3.84 (H-4), 2.52 (H-3b), 2.40 (H-3a), and 1.43 (H-35) in the ¹H NMR, and seven carbon resonances at δ 174.68 (C-1), 151.91 (C-33), 131.16 (C-2), 78.03 (C-34), 69.93 (C-4), 33.34 (C-3), and 19.11 (C-35) in the 13 C NMR spectra (Table I) (Rupprecht et al., 1990). The presence of isolated double bond in 1 was suggested by two-proton multiplet at δ 5.35-5.43 and two carbon resonances at δ 128.95 and 130.85 in the ¹H and ¹³C NMR spectra (Colman-Saizarbitoria et al., 1998). The position of the double bond was determined at C-23/24 from the double-relayed COSY spectrum, which showed correlation cross peaks between H-23 (δ 5.37) and H-21 $(\delta 1.48)$. The configuration of the double bond was assigned as cis by comparing the NMR spectra with other double bond containing acetogenins of known configuration (Woo et al., 1995; Hopp et al., 1997).

The presence of a mono-THF ring in **1**, with two OH groups adjacent to the ring, was suggested by 1 H NMR resonances at δ 3.43 (H-15/20) and 3.80 (H-16/19), and 13 C NMR peaks at δ 74.02 (C-15), 82.66 (C-16), 82.63 (C-19), and 73.58 (C-20); similar peaks are characteristic of other mono-THF acetogenins having two OH groups

adjacent to the ring, such as corrosolin (Corter et al., 1991) and xylomaticin (Colman-Saizarbitoria et al., 1994). Determination of the relative stereochemistry around the mono-THF moiety was done by Borns technique (Born et al., 1990) and by comparison with reported spectral data of a series of synthetic mono-THF diol compounds of known relative configuration (Harmange et al., 1992). The arrangement was assigned as threo between C-15/16 and C-19/20 because of the proton signals of compound 1 at δ 3.43 for H-15/20 and carbon signals at δ 74.02 for C-15 and 73.58 for C-20. The 1H NMR signals at δ 1.98 and 1.67 corresponding to H-17 and H-18 are typical methylene proton signals of a trans-THF ring, since the methylene proton signals for cis-THF rings are δ 1.94 and 1.82 (Gu et al., 1994b). The presence of one free hydroxyl along the hydrocarbon chain was suggested by a proton signal at δ 3.59 (H-10) and a carbon signal at δ 71.77 (C-10) (Jossang et al., 1990). By analysis of the EIMS fragmentations of 1a (Fig. 1), the placement of the mono-THF ring with its two flanking OH groups were determined to be at C-15 to C-20, and the free OH group was located along the hydrocarbon chain at C-10. The absolute stereochemistries of the carbinol stereocenters in 1 have been determined using advanced Mosher ester methodology which analyzes differences between the ¹H NMR chemical shifts of S- and R-MTPA [methoxy (trifluoromethyl) phenyl acetatel ester derivatives on both sides of the stereogenic carbinol centers (Mosher et al., 1973; Rieser et al., 1992; Gu et al., 1994). The ¹H NMR data for **1s** and 1r derivatives are summarized in Table II. Based on Moshers arguments, C-15 and C-20 were assigned to have the R absolute configuration, since the signs of $\Delta\delta$ sr were negative for H-16 and H-19 showing relatively less shielding for this side in the S-MTPA ester. As the relative stereocheimstry form C-15 to C-20 of compound 1 was known to be threo/trans/threo, the absolute configurations of C-15 R, C-16 R, C-19 R and C-20 R were readily concluded. Analysis of differences between the S- and R-Mosher derivatives (1s and 1r) around the γ -lactone ring moiety exhibited negative results for H-3, H-4, H-33, H-34, and H-35 and positive results for H-5, thus suggesting R configuration for C-4 (Table II). The absolute configuration at C-34 in the butenolide ring has been directly determined as S only for uvaricin (Jolad et al., 1985) and squamocin (Born et al., 1990; Araya et al., 1994; Sahai et al., 1994). However, all acetogenins are predicted as having the 34 S (or 36 S in the thirty-seven carbon acetogenins) configuration by comparisons of NMR data of the diagnostic protons and carbons with those of uvaricin and squamocin. Recently, a novel use of Mosher ester data for determining the relative configuration between C-4 and C-34 of the 4-hydroxylated annonaceous acetogenins has been described by Hoye et al. (Hoye, et al., 1994a, 1994b). Analysis of the ¹H NMR data of 1s and 1r, using Hoye's method, indicated that

Table III. Brine Shrimp Lethality for 1, 2 and 3, and Cytotoxicities in Human Solid Tumor Cell Lines for 1

	BST LC ₅₀ (µg/mL)	Human cancer cell line ED ₅₀ (μg/mL)					
Compound		A-549	MCF-7	HT-29	A-498	PC-3	MIA PaCa-2
1 Adriamycin	$1.62 \times 10^{-2} \text{NT}$	$4.41 \times 10^{-1} 2.08 \times 10^{-3}$	8.51×10^{-4} 1.91×10^{-2}	1.83 3.39×10^{-2}	$1.78 \times 10^{-6} $ 2.78×10^{-3}	$4.61 \times 10^{-2} 2.36 \times 10^{-2}$	$1.48 \times 10^{-1} \\ 1.51 \times 10^{-3}$

BST (brine shrimp lethality test): **2** (BST $LC_{50} = 1.02 \times 10^{-2} \,\mu\text{g/mL}$), **3** (BST $LC_{50} = 6.30 \times 10^{-4} \,\mu\text{g/mL}$), A-549 (human lung carcinoma), MCF-7 (human breast carcinoma), HT-29 (human colon adenocarcinoma), A-498 (human kidney carcinoma), PC-3 (human prostate adenocarcinoma), MIA PaCa-2 (human pancreatic carcinoma), adriamycin (positive control standard), NT (not tested).

the $|\Delta\delta|$ values for H-33 and H-34 were 0.24 ppm and 0.07 ppm, respectively (Table II), and strongly suggested "unlike" relative configuration between C-4 and C-34 for 1. Since C-4 in 1 has the *R* configuration, C-34 must possess the *S* configuration, as is usual. Compound 1 is a novel C₃₅ mono-THF acetogenin with double bond at C-23/24; it was named annomocherin.

Compound 2 was a white amorphous powder. The FABMS gave $[MH]^+$ at m/z 597 indicating a $[M]^+$ of 596. The presence of four OH moieties was suggested by successive losses of four H_2O molecules (m/z 18) in the FABMS as well as by the formation of a tetra-TMSi derivative, 2a. The IR carbonyl absorption band in 2 at 1751 cm⁻¹, the UV absorption λmax (MeOH) at 205 nm (log ε 4.0), the proton resonances at δ 7.18 (H-33), 5.06 (H-34), 3.80 (H-4), 2.39 (H-3a), 2.52 (H-3b), and 1.43 (H-35), and the carbon signals at δ 174.62 (C-1), 151.85 (C-33), 130.16 (C-2), 77.94 (C-34), 69.66 (C-4), and 18.99 (C-35) provided characteristic spectral features for an α , β -unsaturated γ -lactone fragment with a 4-OH. The existence of a mono-THF ring with two flanking hydroxyls in 2 was indicated by the ^{1}H NMR signals at δ 3.40 (H-15/20) and 3.80 (H-16/19) and 13 C signals at δ 73.98 (C-15/20) and 82.64 (C-16/19). The presence of a free hydroxyl along the hydrocarbon chain was suggested by a proton signal at δ 3.59 (H-10) and a carbon signal at δ 71.45 (C-10). The position of free hydroxyl group, as well as the total structure of the molecule, was established by close examination of the EIMS spectral fragmentation pattern of 2a (Fig. 2). The peaks at m/z 613, 543, 341, and 271 in the EIMS spectrum of 2a allowed placement of a mono-THF ring with two flanking hydroxyls between C-15 and C-20, whereas the ion fragments at m/z 385 and 295 suggested that the free hydroxyl group was on C-10. The relative stereochemistry C-15/16 and C-19/20 was defined by comparing the ¹³C NMR signals of the hydroxylated carbons at C-15/20 (δ 73.98) and C-16/19 (δ 82.64) and the ¹H NMR signals of H-15/20 (δ 3.40) and H-16/19 (δ 3.80) in **2**, with those of model compounds of known relative stereochemistry (Born et al., 1990). These data suggested that the relative configurations between C-15/16 and C-19/20 were both threo. Mono-THF ring of 2 possesses a trans configuration based on the observation that the methylene proton signals resonated at δ 1.98 (H-17b/18b) and 1.67 (H-17a/18a) (Harmange et al., 1992). Thus, the relative configuration for these four chiral centers in **2** was assigned as *threo/trans/threo*. So, this compound was identified as anno-nacin (Alkofahi et al., 1988).

Compound 3 was isolated as a white powder. The FABMS of 3 gave an [MH]⁺ at 625. Sequential losses of four molecules of H₂O from the [MH]⁺ in the FABMS as well as the formation of the tetra-TMSi derivative 3a confirmed the existence of four hydroxyl groups in 3. The presence of an α,β -unsaturated γ -lactone with a hydroxyl group at C-4 in 3 was suggested by the ¹H NMR resonances at δ 7.18 (H-35), 5.06 (H-36), 3.80 (H-4), 2.39 (H-3a), 2.53 (H-3a), and 1.43 (H-37) corresponding in the 13 C NMR spectrum to the resonances at δ 174.58 (C-1), 151.81 (C-35), 131.18 (C-2), 77.97 (C-36), 69.91 (C-4), 33.48 (C-3), and 19.10 (C-37) (Table I). The THF ring was placed between C-17 and C-22 based on peaks at m/z 543 and 613 in the EIMS fragmentation of 3a (Fig. 3). A free hydroxyl attached somewhere along the aliphatic chain could be identified by a multiplet at δ 3.58 in the ¹H NMR and a signal at δ 71.86 in the ¹³C NMR spectra. The peaks in the EIMS fragment of 3a at m/z 385 indicated that this hydroxyl was located along the hydro-carbon chain at C-10. Resonances at δ 3.40 (H-17/22) and 3.80 (H-18/21) in the ¹H NMR spectrum and δ 74.02 (C-17/22), and 82.62 (C-18/21) in the ¹³C NMR spectrum indicated the presence of a mono-THF ring and two flanking hydroxyls with a threo/ trans/threo relative stereocheimstry. Finally this compound was identified as annomontacin with C₃₇ (Jossang et al., 1991).

Table III summarizes the bioactivities for compounds 1-3. 1 was very toxic to the brine shrimp lethality test (BST) and showed potent and selective cytotoxicities to the human solid tumor cell lines, breast carcinoma (MCF-7) and kidney carcinoma (A-498). 1 exhibited potent and selective cytotoxicities against breast carcinoma (MCF-7) and kidney carcinoma (A-498) cell lines with 100 1,000 times the potency of adriamycin. Compounds 2 and 3 are very active in the brine shrimp lethality test. The acetogenins exert their biological effects, at least in part, through inhibition of mitochondrial electron transport (complex I) and the inhibition of the plasma membrane

NADH oxidase of tumor cells (Zeng et al., 1996; Alali et al., 1999).

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