

단 신

치환페놀류로부터 2-아릴벤조[b]푸란 유도체의 합성

崔洪大* · 徐翊子 · 孫炳華†
 동의대학교 화학과 · 기초과학연구소
 †부경대학교 자연과학대학 화학과
 (1999. 7. 13 접수)

Synthesis of 2-Arylbenzo[b]furan Derivatives from Substituted Phenols

Hong-Dae Choi*, Pil-Ja Seo and Byeng-Wha Son†

Department of Chemistry and Research Institute of Basic Science, Dongeui University, Pusan 614-714, Korea

†Department of Chemistry, Pukyong National University, Pusan 608-737, Korea

(Received July 13, 1999)

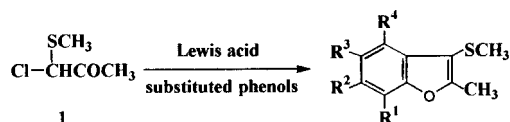
As a part of the constituents isolated from *Krameria ramosissima*,¹ *Machilus glaucescens*,² *Myroxylon balsamum*,³ *Sophora tomentosa*,⁴ *Styrax obussia*,⁵ and *Zanthoxylum ailanthoides*,⁶ there are many well known natural products possessing a benzo[b]furan skeleton which has the various aryl groups in the 2-position.

Recently, we showed an excellent method for preparation of 2-methylbenzo[b]furans from the one step reaction of substituted phenols with 1-chloro-1-(methylthio)acetone (**1**) in the presence of Lewis acid⁷ (Scheme 1). For our basic research programs toward the synthesis of natural products containing 2-arylbenzo[b]furan ring, we employed 2-chloro-2-(methylthio)acetophenone (**2**) instead of **1**. Thus it was easily confirmed that 2-phenylbenzo[b]furans ring could be prepared by the reaction of phenol and cresol isomer with **2** in the presence of zinc chloride⁸.

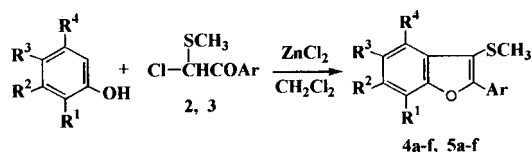
As shown in Scheme 2, the present paper describes a general synthetic method for 2-arylbenzo[b]furan derivatives (**6,7**) by the reaction of substituted phenols with **2** as well as 2-chloro-2-methylthio-(3',4'-methylenedioxy)acetophenone (**3**) and successive desulfurization of the resulting 2-aryl-3-methylthiobenzo[b]furans (**4,5**). The treatment of substituted phenols and **2** with zinc chloride in methylene chloride at 0°C gave the corresponding 3-methylthio-2-phenylbenzo[b]furans (**4**) in high yields. The compounds **4** were easily desulfurized by heating

with Raney nickel in ethanol to give 2-phenylbenzo[b]furans (**6**) in good yields.

Next we examined the reaction of substituted phenols with **3** under the same conditions as the preparation of **4**. The 2-(3',4'-methylenedioxyphenyl)-3-(methylthio)benzo[b]furans (**5**) were obtained from substituted phenols and

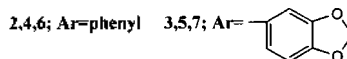
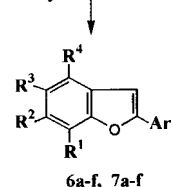


Scheme 1.



- a: R¹=R²=R³=H, R⁴=CH₃
- b: R¹=R²=R³=H, R⁴=C₂H₅
- c: R¹=R²=R³=H, R⁴=*i*-C₄H₉
- d: R¹=R²=R³=H, R⁴=*t*-C₄H₉
- e: R¹=R³=CH₃, R²=R⁴=H
- f: R¹=R³=H, R²=R⁴=CH₃

Raney-Ni EtOH



Scheme 2.

3. The desulfurization of **5** with Raney nickel in boiling ethanol afforded the corresponding 2-(3',4'-methylenedioxyphenyl)benzo[b]furans (**7**).

Of the several methods available for the preparation of 2-arylbenzo[b]furan ring, the route⁹ through the reaction of an *o*-halogenophenol with a copper(I) arylacetylide seems to be an useful procedure for 2-arylbenzo[b]furan construction. However, this route requires longer reaction time and the yield is very low.

In summary, our route for the formation of 2-arylbenzo[b]furan derivatives (**6,7**) consists of two steps: i) a successive dehydrocyclization of Freidel-Crafts reaction intermediate which was prepared from the treatment of substituted phenols with the chlorides **2,3** in the presence of zinc chloride; ii) the reductive desulfurization of the resulting products **4,5**. Now the application for total synthesis of natural products involving the skeletal structure of various benzo[b]furan rings with 3,4-methylenedioxyphenyl and highly methoxylated phenyl groups in the 2-position is in progress.

EXPERIMENTAL

Melting point was determined on a Gallenkamp melting point apparatus and uncorrected. ¹H NMR spectra were recorded on a Hitachi R-1500 (FT, 60MHz) spectrometer using tetramethylsilane as an internal standard. IR spectra were recorded on a JASCO FT/IR-300E spectrophotometer. Mass spectra were measured with a Hewlett Packard 5970 GC/MS system. Silica gel 60 (70-230 mesh, E. Merck) was used for column chromatography.

Synthesis of 3-Methylthio-2-phenylbenzo[b]furans 4: General Procedure. ZnCl₂ (218 mg, 1.6 mmol) was added to a stirred solution of **2** (300 mg, 1.5 mmol) and a substituted phenol (1.5 mmol) in CH₂Cl₂ (5 mL) at 0°C under N₂ atmosphere, and stirring was continued at the same temperature for 1h. The reaction was quenched by the addition of water, then the mixture was extracted with CH₂Cl₂ (10 mL), and the extract was dried over anhydrous MgSO₄. The solvent was removed *in vacuo* and the residue was purified by column chromatography (hexane/ethyl acetate=15/1) to give **4**. **4a**: Yield 89%; mp. 67-68°C ¹H NMR (CDCl₃) δ2.37 (s, 3H), 2.48 (s,

3H), 6.82-8.37 (m, 8H); IR (KBr) 2917, 1472, 1443, 1255, 1203, 1084, 1066 cm⁻¹; MS: m/z 254 (M⁺). **4b**: Yield 90%; mp. 37-38°C; ¹H NMR (CDCl₃) δ1.31 (t, 3H, J=7.6Hz), 2.38 (s, 3H), 2.80 (q, 2H, J=7.6Hz), 7.08-8.37 (m, 8H); IR (KBr) 3020, 2959, 2863, 1469, 1339, 1256, 1066 cm⁻¹; MS: m/z 268 (M⁺). **4c**: liquid; Yield 92%; ¹H NMR (CDCl₃) δ1.33 (d, 6H, J=6.5Hz), 2.38 (s, 3H), 2.84-3.31 (m, 1H), 7.10-8.36 (m, 8H); IR (neat) 2958, 1734, 1684, 1558, 1457, 1066 cm⁻¹; MS: m/z 282 (M⁺). **4d**: liquid; Yield 91%; ¹H NMR (CDCl₃) δ1.43 (s, 9H), 2.39 (s, 3H), 7.26-8.38 (m, 8H); IR (neat) 3063, 2961, 2867, 1558, 1520, 1489 cm⁻¹; MS: m/z 296 (M⁺). **4e**: Yield 80%; mp. 116-117°C; ¹H NMR (CDCl₃) δ2.38 (s, 3H), 2.46 (s, 3H), 2.53 (s, 3H), 6.97-8.33 (m, 7H); IR (KBr) 2918, 1473, 1420, 1204, 1067, 1031 cm⁻¹; MS: m/z 268 (M⁺). **4f**: liquid; Yield 51%; ¹H NMR (CDCl₃) δ2.31 (s, 3H), 2.42 (s, 3H), 2.49 (s, 3H), 6.85-8.21 (m, 7H); IR (neat) 2917, 2857, 1621, 1486, 1330, 1092 cm⁻¹; MS: m/z 268 (M⁺).

Synthesis of 2-(3',4'-Methylenedioxyphenyl)-3-(methylthio)benzo[b]furans 5: General Procedure. By the same procedure as described above for the preparation of **4**, compounds (**5**) were obtained from **3** (228 mg, 0.93 mmol), a substituted phenol (0.93 mmol) and ZnCl₂ (150 mg, 1.1 mmol). The residue was purified by column chromatography (benzene). **5a**: Yield 76%; mp. 120-121°C; ¹H NMR (CDCl₃) δ2.36 (s, 3H), 2.47 (s, 3H), 6.02 (s, 2H), 6.83-7.89 (m, 6H); IR (KBr) 2979, 2779, 1498, 1360, 1267, 1227, 1105, 1039 cm⁻¹; MS: m/z 298 (M⁺). **5b**: Yield 88%; mp. 115-116°C; ¹H NMR (CDCl₃) δ1.31 (t, 3H, J=7.1Hz), 2.37 (s, 3H), 2.79 (q, 2H, J=7.1Hz), 6.03 (s, 2H), 6.84-7.90 (m, 6H); IR (KBr) 2954, 2904, 1498, 1441, 1281, 1037 cm⁻¹; MS: m/z 312 (M⁺). **5c**: Yield 93%; mp. 92-93°C; ¹H NMR (CDCl₃) δ1.33 (d, 6H, J=7.02Hz), 2.36 (s, 3H), 2.83-3.18 (m, 1H), 6.02 (s, 2H), 6.83-7.85 (m, 6H); IR (KBr) 2939, 2906, 1496, 1441, 1288, 1037 cm⁻¹; MS: m/z 326 (M⁺). **5d**: liquid; Yield 89%; ¹H NMR(CDCl₃) δ1.42 (s, 9H), 2.37 (s, 3H), 6.03 (s, 2H), 6.98-7.85 (m, 6H); IR (neat) 2966, 2900, 1504, 1475, 1363, 1267, 1063 cm⁻¹; MS: m/z 340 (M⁺). **5e**: Yield 82%; mp. 105-106°C; ¹H NMR (CDCl₃) δ2.35 (s, 3H), 2.44 (s, 3H), 2.50 (s, 3H), 6.02 (s, 2H), 6.84-7.85 (m, 5H); IR (KBr) 2914, 1471, 1250, 1228, 1115, 1036 cm⁻¹; MS: m/z 312 (M⁺). **5f**: Yield 50%; mp. 93-94°C; ¹H NMR (CDCl₃) δ2.31 (s, 3H), 2.41

(s, 3H), 2.83 (s, 3H), 6.01 (s, 2H), 6.82-7.83 (m, 5H); IR (KBr) 2947, 1502, 1444, 1255, 1234, 1037 cm^{-1} ; MS: m/z 312 (M^+).

Synthesis of 2-Phenylbenzo[b]furans 6: General Procedure. Raney nickel (W-2, 2-4 g) was added to a solution of **4** (200-400 mg) in EtOH (20 mL), and the mixture was heated at 60-70°C for 1h. The Raney nickel was filtered off and the solvent was evaporated off. The residue was purified by column chromatography (hexane/ethyl acetate=15/1) to give **6**. **6a**: Yield 86%; mp. 129-130°C; $^1\text{H NMR}$ (CDCl_3) δ 2.44 (s, 3H), 6.72-8.21 (m, 9H); IR (KBr) 3013, 2914, 1466, 1266, 1199, 1039 cm^{-1} ; MS: m/z 208 (M^+). **6b**: Yield 81%; mp. 76-77°C; $^1\text{H NMR}$ (CDCl_3) δ 1.28 (t, 3H, $J=7.6\text{Hz}$), 2.75 (q, 2H, $J=7.6\text{Hz}$), 6.95-7.87 (m, 9H); IR (KBr) 2959, 1560, 1467, 1267, 1038, cm^{-1} ; MS: m/z 224 (M^+). **6c**: Yield 81%; mp. 87-88°C; $^1\text{H NMR}$ (CDCl_3) δ 1.31 (d, 6H, $J=7.0\text{Hz}$), 2.78-3.25 (m, 1H), 6.97-7.94 (m, 9H); IR (KBr) 2957, 1468, 1259, 1160, 1020 cm^{-1} ; MS: m/z 236 (M^+). **6d**: Yield 82%; mp. 105-107°C; $^1\text{H NMR}$ (CDCl_3) δ 1.39 (s, 9H), 6.97-8.0 (m, 9H); IR (KBr) 2957, 1560, 1474, 1363, 1277, 1164, 1021 cm^{-1} ; MS: m/z 250 (M^+). **6e**: Yield 78%; mp. 69-70°C; $^1\text{H NMR}$ (CDCl_3) δ 2.40 (s, 3H), 2.54 (s, 3H), 6.92-7.82 (m, 8H); IR (KBr) 2913, 1460, 1205, 1039, 1020 cm^{-1} ; MS: m/z 222 (M^+). **6f**: liquid; Yield 74%; $^1\text{H NMR}$ (CDCl_3) δ 2.43 (s, 3H), 2.50 (s, 3H), 6.85-7.93 (m, 8H); IR (neat) 2917, 2858, 1619, 1446, 1295, 1128, 1038 cm^{-1} ; MS: m/z 222 (M^+).

Synthesis of 2-(3',4'-Methylenedioxyphenyl)benzo[b]furans 7: General Procedure. By the same procedure as described above for the preparation of **6**, compounds (**7**) were obtained from **5** (150-250 mg) and Raney nickel (1.5-3.0 g). The residue was purified by column chromatography (benzene). **7a**: Yield 71%; mp. 132-133°C; $^1\text{H NMR}$ (CDCl_3) δ 2.43 (s, 3H), 6.01 (s, 2H), 6.79-7.45 (m, 7H); IR (KBr) 2897, 1462, 1230, 1200, 1107, 1037 cm^{-1} ; MS: m/z 252 (M^+). **7b**: Yield 86%; mp. 99-100°C; $^1\text{H NMR}$ (CDCl_3) δ 1.27 (t, 3H, $J=7.6\text{Hz}$), 2.79 (q, 2H, $J=7.6\text{Hz}$), 6.00 (s, 2H), 6.79-7.45

(m, 7H); IR (KBr) 2958, 1500, 1466, 1284, 1234, 1103, 1037 cm^{-1} ; MS: m/z 266 (M^+). **7c**: Yield 90%; mp. 113-114°C; $^1\text{H NMR}$ (CDCl_3) δ 1.30 (d, 6H, $J=7.0\text{Hz}$), 2.77-3.12 (m, 1H), 6.0 (s, 2H), 6.80-7.31 (m, 7H); IR (KBr) 2954, 2889, 1489, 1362, 1234, 1161, 1041 cm^{-1} ; MS: (m/z) 280(M^+). **7d**: Yield 84%; mp. 127-128°C; $^1\text{H NMR}$ (CDCl_3) δ 1.38 (s, 9H), 6.01 (s, 2H), 6.82-7.55 (m, 7H); IR (KBr) 2958, 2900, 1473, 1362, 1234, 1041 cm^{-1} ; MS: m/z 294 (M^+). **7e**: Yield 97%; mp. 121-122°C; $^1\text{H NMR}$ (CDCl_3) δ 2.39 (s, 3H), 2.51 (s, 3H), 6.00 (s, 2H), 6.77-7.32 (m, 6H); IR (KBr) 2910, 1564, 1495, 1284, 1234, 1045 cm^{-1} ; MS: m/z 266 (M^+). **7f**: Yield 93%; mp. 107-108°C; $^1\text{H NMR}$ (CDCl_3) δ 2.42 (s, 3H), 2.46 (s, 3H), 5.98 (s, 2H), 6.81-7.42 (m, 6H); IR (KBr) 2883, 1500, 1481, 1458, 1252, 1107 cm^{-1} ; MS: m/z 266 (M^+).

Acknowledgement. This work supported by the Research Foundation of Dongeui University, 1999 and gratefully acknowledged.

REFERENCES

1. Achenbach, H.; Grob, J.; Dominguez, X. A.; Star, J. V.; Salgado, F. *Phytochemistry* **1987**, *26*, 2041.
2. Talapatra, B.; Ray, T.; Talapatra, S. K. *J. Indian Chem. Soc.* **1978**, *55*, 204.
3. Oliveira, A.; Iracema, M.; Madruga, L. M.; Gottlieb, O. *Phytochemistry* **1978**, *17*, 593.
4. Komatsu, M.; Yokoe, I.; Shirataki, Y. *Chem. Pharm. Bull.* **1978**, *26*, 1274.
5. Takanashi, M.; Takizawa, Y.; Mitsunashi, T. *Chemistry Letters* **1974**, 869.
6. Sheen, W. S.; Tsai, I. L.; Teng, C. M.; Chen, I. S. *Phytochemistry* **1994**, *36*, 213.
7. (a) Choi, H. D.; Seo, P. J.; Son, B. W. *J. Korean Chem. Soc.* **1998**, *42*, 719. (b) Choi, H. D.; Seo, P. J.; Son, B. W. *J. Korean Chem. Soc.* **1999**, *43*, 237.
8. Choi, H. D.; Seo, P. J.; Son, B. W. *J. Korean Chem. Soc.* **1999**, in press.
9. (a) Schreiber, F.; Stevenson, R. *J. Chem. Soc., Perkin Trans I* **1976**, 1514. (b) Schneider, G.; Stevenson, R. *J. Org. Chem.* **1979**, *44*, 4710.