

Synthesis of Sesquiterpene Derivatives as Potential Antitumor Agents; Elemene Derivatives

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Derivatives of elema-1,3-diene were synthesized in several steps as polar analogs of β -elemene, antitumor agent under clinical phase. The lactone ring of compound **1** was opened by LiAlH_4 to give diol **2** which was selectively protected by TBDPSCl. After acetylation of the secondary alcohol, the acetylated product was ozonolyzed and reduced to give elemene derivative **4** which was converted to diolefin **8** via selenides subsequent deprotection by tetrabutylammonium fluoride gave two compounds **9**, **10**.

Key words: β -Elemene, Elema-1,3-diene, Antitumor agent, α -Santonin

INTRODUCTION

β -Elemene (Fig. 1), a new antitumor agent, is one of sesquiterpenes isolated from *Rhizoma zedoariae*. It was synthesized and shown to have potent antitumor activity (Shi, 1981; Fu, 1994) and is under clinical trial. According to Wang (Wang, 1999), β -elemene has a different mechanism of action compared to the conventional chemotherapeutic agents, it was expected to have an excellent antitumor activity with less toxicity in treatment of cancer diseases (Xu, 1996; Zuo, 1998). But it showed the poor oral bioavailability possibly due to its low solubility (Qian, 1996). Recently, a few chemical modification of β -elemene was attempted to improve its physicochemical properties (Jia, 1991) but systematic studies were not achieved. Chemical synthesis of β -elemene has been reported by several authors (Patel, 1967; McMurray, 1985), including the method using α -santonin as starting material to establish the correct the

stereochemistry. It will be interesting to modify each of the three olefins of β -elemene to a polar functional group because increased polarity may lead to enhanced oral bioavailability and antitumor activity.

In the present study, we tried to synthesize polar analogs of elemene using α -santonin as starting material to obtain the improved antitumor agent.

MATERIALS AND METHODS

Melting points were determined on a Mel-temp II laboratory device and are uncorrected. $^1\text{H-NMR}$ spectra were recorded on Bruker Ac80 fourier transform spectrophotometer for 80 MHz or Bruker AM300 with Me_4Si as internal standard; chemical shifts (δ) are reported in parts per million, and signals are quoted as s (singlet), d (doublet), t (triplet), or m (multiplet). TLC was carried out using precoated plates with silica gel 60F 254 purchased from Merck or from Analtech Co.

5 α H,4,6,11 β H-Eudesm-2-en-6,12-olide (1)

Compound **1** was prepared by known method (Simonovic, 1963; Grieco, 1977)

Colorless crystal, m.p. 142° (lit. 142-143°C) (Grieco, 1977); IR (KBr) cm^{-1} 2966, 2937, 2876, 1768, 1456, 1207, 1151, 1020, 985, 688; $^1\text{H NMR}$ (CDCl_3) δ 0.95 (3 H, s, CH_3), 1.16 (3 H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$), 1.24 (3 H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$) 3.79 (1 H, t, $J=10$ Hz $-\text{CHO}-$), 5.46~5.56 (2 H, m, $-\text{CH}=\text{CH}-$).

5 α H,4,6,11 β H-Eudesm-2-en-6,12-diol (2)

The suspension of LiAlH_4 (0.78 g, 20.6 mmol) in ether

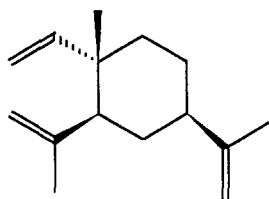


Fig. 1. β -Elemene

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(20 ml) was slowly added to the solution of lactone **1** (1.34 g, 5.7 mmol) in ether (40 ml) at 0°C and then refluxed for 4 h. The reaction was quenched by using EtOAc and ammonium chloride. The solution was extracted by ether several times and ethereal layer was dried and concentrated to give white residue which was separated by column chromatography using hexane/EtOAc (1:1) to give white crystal (1.15 g, 84.7%). m.p.: 162–164°C; IR (KBr) cm^{-1} : 3322, 2950, 2929, 1056, 1056, 694; $^1\text{H NMR}$ (CDCl_3): δ 0.85 (3 H, s, $-\text{CH}_3$), 0.9 (3H, d, $J=1.65$, $>\text{CH}-\text{CH}_3$), 1.22 (3 H, d, $J=1.65$, $>\text{CH}-\text{CH}_3$), 3.48–3.60 (3 H, m, CHOH & CH_2OH), 5.39–5.52 (2 H, m, $-\text{CH}=\text{CH}-$).

6 α -Acetoxy-12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11 β H-eudesm-2-ene (3)

To the solution of diol **2** (1.07 g, 4.5 mmol) and imidazole (0.884 g, 13.0 mmol) in DMF (20 ml) was slowly added *t*-butyldiphenylsilyl chloride (1.3 ml, 4.96 mmol) in DMF (5 ml) at 0°C and was stirred at room temperature for 4 h. The reaction mixture was evaporated under vacuum pressure to give colorless oil which was separated by column chromatography using hexane/EtOAc (10 : 1) to yield colorless oil, 12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11H-eudesm-2-en-6 α -ol (2.04 g, 95.7%); IR (neat) cm^{-1} : 3481, 2928, 2361, 1653, 1456, 1388, 1109, 694; $^1\text{H NMR}$ (CDCl_3): δ 0.78 (3H, d, $>\text{CH}-\text{CH}_3$), 0.87 (3H, s, $-\text{CH}_3$), 1.06 (9H, s, $-\text{C}(\text{CH}_3)_3$), 3.35–3.65 (3H, m, $>\text{CH}-\text{OH}$ & $-\text{CH}_2-\text{OSi}$), 5.40–5.60 (2H, m, $-\text{CH}=\text{CH}-$), 7.25–7.74 (10H, m, aromatic).

A solution of above compound (0.3 g, 0.63 mmol) in Ac_2O /pyridine (10 ml) was reacted at 60°C overnight and concentrated under vacuum pressure. The crude yellowish oil was separated by column chromatography to give colorless oil; IR (neat) cm^{-1} 2931, 1731, 1471, 1427, 1373, 1240, 1110, 823, 738, 701, 609, 505; $^1\text{H NMR}$ (CDCl_3): δ 0.77 (3 H, d, $>\text{CH}-\text{CH}_3$), 0.88 (3 H, s, $-\text{CH}_3$), 1.02 (3 H, d, $>\text{CH}-\text{CH}_3$), 1.06 (9 H, s, $-\text{C}(\text{CH}_3)_3$), 2.06 (3 H, s, $-\text{COCH}_3$), 3.50 (2H, d, $J=7.0\text{Hz}$, $-\text{CH}_2-\text{OSi}$), 5.03 (1 H, t, $J=10$ Hz, $-\text{CHO}$), 5.40–5.49 (2 H, m, $-\text{CH}=\text{CH}$), 7.28–7.72 (10 H, m, aromatic).

2,3-Dihydroxy-6 α -acetoxy-12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11 β H-elemane (4)

A solution of the acetate **3** (0.2 g, 0.385 mmol) in MeOH (10 ml) cooled at -78°C was treated with a saturated ozone solution in CH_2Cl_2 (10 ml). After 20 min, NaBH_4 (16 mg) was added at -78°C and the same amount of NaBH_4 was added every 15 min for 45 min. The reaction mixture was warmed to room temp and the solvent was removed under reduced pressure and crude residue was dissolved to water. The solution was extracted with ether several times and ethereal layer was washed with saturated NH_4Cl solution and dried over

Na_2SO_4 . The solution was evaporated to give white powder which was crystallized to colorless foam (0.12 g, 56.2%); IR (KBr) cm^{-1} : 3331, 2957, 1730, 1429, 1385, 1232, 1113, 1035, 806; $^1\text{H NMR}$ (CDCl_3): δ 0.90 (6H, br d, $J=1.9$ Hz, $>\text{CH}-\text{CH}_3$ $\times 2$), 1.08 (9H, s, $-\text{C}(\text{CH}_3)_3$), 2.02 (3H, s, $-\text{COCH}_3$), 3.47 (2H, d, $J=7.0$ Hz, $-\text{CH}_2-\text{OSi}$), 4.2 (2H, br s, $-\text{OH} \times 2$), 5.06 (H, t, $J=10$ Hz, $-\text{CHO}$), 7.20–7.69 (10H, m, aromatic).

2-Hydroxy-3-*o*-nitrophenylseleno-6 α -acetoxy-12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11 β H-elemane (5)

To the solution of diol **4** (9.4 g, 16.9 mmol) and *o*-nitrophenylselenocyanate (7.68 g, 33.8 mmol) in THF (50 ml) and pyridine (50 ml), tri-*n*-butylphosphine (8.4 ml, 33.8 mmol) was added slowly and stirred at room temperature for 2.5 h. After removal of solvent at reduced pressure, the residue was treated with ether and washed with d-HCl, saturated NaHCO_3 and NaCl solution. The ethereal solution was dried over Na_2SO_4 and evaporated to yield crude brown oil which was separated by column chromatography to give yellowish oil **5** (10.5 g, 82.2%). $^1\text{H NMR}$ ($\text{DMSO}-d_6$): δ 0.83 (3H, s, $-\text{CH}_3$), 0.85 (3H, d, $>\text{CH}-\text{CH}_3$), 0.98 (3H, d, $>\text{CH}-\text{CH}_3$), 1.0 (9H, s, $-\text{C}(\text{CH}_3)_3$), 2.04 (3H, s, $-\text{COCH}_3$), 3.50 (2H, m, $-\text{CH}_2-\text{OH}$), 5.0 (1H, t, $J=10$ Hz), 7.20–8.36 (14H, m, aromatic).

2-Hydroxy-6 α -acetoxy-12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11 β H-elema-2-ene (6)

To the solution of monoselenide **5** (10.5 g, 13.9 mmol) in THF (130 ml) at 0°C was slowly added 50% aqueous hydrogen peroxide solution (6.4 ml). After adding, the reaction mixture was stirred at room temperature for 3h and diluted with water (200 ml), and concentrated to remove the organic solvent. The aqueous solution was extracted with ether 3 times and ethereal layer was dried and evaporated to give crude residue which was separated by column chromatography using hexane/EtOAc (25 : 1) to give yellowish solid **6** (6.7g, 89.8%); IR (KBr) cm^{-1} : 3442, 2931, 1729, 1469, 1373, 1238, 1110, 910, 823, 701, 503; $^1\text{H NMR}$ (CDCl_3): δ 0.78 (3H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$), 0.87 (3H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$), 1.00 (3H, s, $-\text{CH}_3$), 1.06 (9H, s, $-\text{C}(\text{CH}_3)_3$), 2.04 (3H, s, $-\text{COCH}_3$), 4.90–5.90 (3H, m, $-\text{CH}=\text{CH}_2$), 7.30–7.80 (10H, m, aromatic).

2-*o*-Nitrophenylseleno-6 α -acetoxy-12-(*t*-butyldiphenylsilyloxy-5 α H,4,6,11 β H-elema-2-ene (7)

A solution of monoolefin **6** (5.7 g, 10.62 mmol) in THF (200 ml) was treated with 2-nitrophenylselenocyanate (4.81 g, 21.2 mmol) and tri-*n*-butylphosphine (5.3 ml, 21.2 mmol) was added slowly at room temperature and stirred for 4 h. After removal of solvent in vacuum pressure, the residue was extracted with ether, washed with water

and dried over Na_2SO_4 . The ethereal solution was concentrated to red residue which was separated by column chromatography to give yellowish oil **7** (4.8 g, 61.4%); IR(neat) cm^{-1} : 2929, 2352, 1729, 1589, 1515, 1332, 1236, 1097, 730, 703; ^1H NMR (CDCl_3): δ 0.85 (3 H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$) 0.97 (3H, d, $J=1.6$ Hz, $>\text{CH}-\text{CH}_3$), 1.07 (9 H, s, $-(\text{CH}_3)_3$), 2.31 (3H, s, $-\text{COCH}_3$), 2.93 (H, m, $J=6.9$ Hz), 3.50 (1H, d, $J=6.9$ Hz) 7.26~8.37 (14H, m, aromatic).

6 α -Acetoxy-12-(*t*-butyldiphenylsilyl)oxy-5 α H,4,6,11 β H-elema-2,3-diene(**8**)

To the solution of above monoselenide **7** (4.8 g, 6.52 mmol) in THF (100 ml) at 0°C was slowly added 50% H_2O_2 solution (3 ml) and stirred at room temperature for 4 h. The reaction mixture was concentrated to crude oil which was separated by column chromatography (hexane/EtOAc=10:1) to give yellowish oil **8** (1.5 g, 44.4%); IR (neat) cm^{-1} : 3072, 2932, 2858, 1734, 1637, 1473, 1429, 1371, 1242, 1113, 912; ^1H NMR ($\text{DMSO}-d_6$): δ 0.82 (3 H, d, $>\text{CH}-\text{CH}_3$), 1.06 (9H, s, $-(\text{CH}_3)_3$) 1.69 (3H, s, $-\text{CH}_3$), 1.96 (3 H, s, $-\text{COCH}_3$), 4.66 (1H, s), 5.75 (1H, dd, $J=9.0$ Hz, 12.0 Hz, $-\text{CH}=\text{CH}_2$), 4.87~4.92 (3H, m, $-\text{CH}=\text{CH}_2$), 5.12 (1H, t, $J=10\text{Hz}$), 7.36~7.65 (10H, m, aromatic).

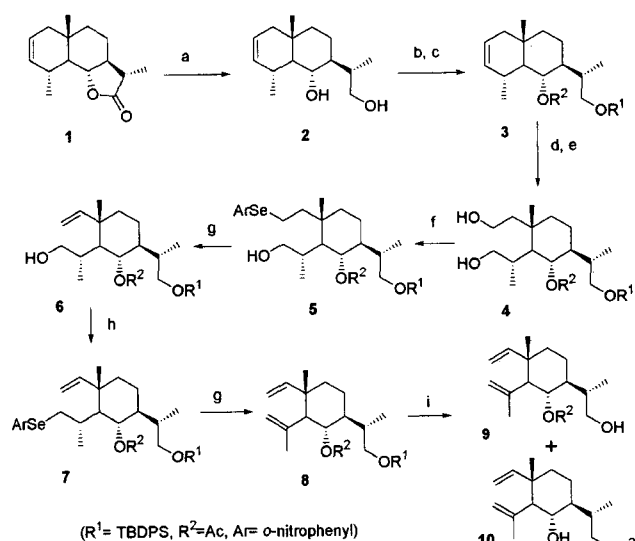
6 α -Acetoxy-5 α H,4,6,11 β H-elema-2,3-diene-12-ol(**9**) and 12-acetoxy-5 α H,4,6,11 β H-elema-2,3-diene-6 α -ol (**10**)

To the solution of the diolefin **8** (1.5 g, 2.89 mmol) was added 1M-tetrabutylammonium fluoride solution (3.8 ml, 3.8 mmol) at 0°C and stirred at room temperature overnight. The solution was concentrated and separated by column chromatography to give two alcohols **9** (350 mg, 43.2%) and **10** (370 mg, 45.7%) Compound **9**, IR (neat) cm^{-1} : 3480, 3076, 2934, 2856, 1740, 1637, 1367, 1240, 1035, 910 ^1H NMR (CDCl_3): δ 0.90 (3H, d, $J=9.0$ Hz, $>\text{CH}-\text{CH}_3$) 1.06 (3H, s, $-\text{CH}_3$), 1.69 (3H, s, $-\text{CH}_3$), 3.50 (2H, d, $J=6.0$ Hz, $-\text{CH}_2\text{OH}$), 4.65 (1H, s, $-\text{COCH}_3$), 4.87~4.93 (3H, m, $-\text{CH}=\text{CH}_2$), 5.15 (1H, t, 10Hz, $-\text{CHO}$), 5.75 (1H, m).

Compound **10**, IR (neat) cm^{-1} : 3420, 3082, 2936, 2878. 2631, 1734, 1373, 1028, 910; ^1H NMR ($\text{DMSO}-d_6$): δ 0.90 (3 H, d, $J=6.0$ Hz, $>\text{CH}-\text{CH}_3$) 1.00 (3 H, s, $-\text{CH}_3$), 1.78 (3 H, s, $-\text{CH}_3$), 2.06 (1 H, s) 3.69 (1 H, t, $J=9.0$ Hz, $>\text{CHOH}$), 4.78 (1 H, s, $-\text{CH}=\text{CH}_2$), 4.87~4.92 (2 H, m, $-\text{CH}=\text{CH}_2$). 5.15 (1 H, s), 5.73 (1 H, dd, $J=9.0$ Hz, 12.0 Hz).

RESULTS AND DISCUSSION

The starting material **1** was prepared by catalytic hydrogenation, epimerization and Shapiro reaction from α -santonin according to the published methods (Simonic, 1963; Grieco, 1977). The diol **2** was easily obtain-



Reagents: (a) LiAlH_4 , ether; (b) TBDPSCl, imidazole, DMF; (c) Ac_2O , Py; (d) O_3 , CH_2Cl_2 , MeOH; (e) NaBH_4 , MeOH; (f) ArSeCN , Bu_3P , THF, Py; (g) 50% H_2O_2 , THF; (h) ArSeCN , Bu_3P , THF; (i) TBAF, THF

Scheme 1. Synthesis of Elemane derivatives

ed in high yield from lactone **1** by treatment with LiAlH_4 /ether (Cardona, 1992) (Scheme 1). By taking advantage of the different reactivity of primary and secondary alcohol, selective protection of primary alcohol using *t*-butyldiphenylsilyl chloride in the presence of imidazole in DMF was achieved in high yield and subsequent acetylation of the secondary alcohol using Ac_2O /pyridine gave compound **3** in high yield. This olefin was treated with ozone at 78°C followed by reduction with NaBH_4 , gave the diol **4**. We attempted to synthesize bis-*o*-nitrophenyl selenide from **4** to afford divinyl compound directly via oxidation but we could not obtain bis-selenide. Conversion of **5** to divinyl compound was accomplished in four steps. The mono-selenide was obtained from diol in high yield using *o*-nitrophenylselenocyanate and tri-*n*-butylphosphine. Oxidative elimination of the monoselenide was carried out with hydrogen peroxide to give the olefinic alcohol, **6** which was subjected to another selenation followed by oxidative elimination to yield diolefin **8**. The compound **8** was deprotected by tetra-butylammonium fluoride to give separable mixture of two compounds **9** and **10**.

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