

A Furan Derivative from *Cornus officinalis*

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Dimethyltetrahydrofuran *cis*-2,5-dicarboxylate a furan derivative has been isolated from the fruits of *Cornus officinalis*, and it was isolated for the first time in the nature. The structure was elucidated by NMR spectroscopic techniques.

Key words : *Cornus officinalis*, Cornaceae, Dimethyltetrahydrofuran *cis*-2,5-dicarboxylate, Kaempferol, Quercetin

INTRODUCTION

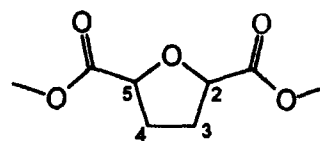
Fruits of *Cornus officinalis* Sieb. et Zucc. (Cornaceae) is a well-known traditional medicine for its tonic, analgesic and diuretic properties in Korea, Japan and China, and this plant is widely distributed in Korea (Yook, 1981; Lee, 1989).

Previous phytochemical studies on fruits of this plants have afforded iridoids (Endo *et al.*, 1973), tannins (Okuda *et al.*, 1984; Hatano *et al.*, 1989; Lee *et al.*, 1989), volatile flavor components (Miyazawa *et al.*, 1989), and an ursolic acid (Yamahara *et al.*, 1981; Kim *et al.*, 1996). Further investigation of fruits of this plant afforded a furan derivative, dimethyltetrahydrofuran *cis*-2,5-dicarboxylate, together with two common flavonoids, kaempferol and quercetin (Fig. 1). This paper reports their isolation and structural elucidation.

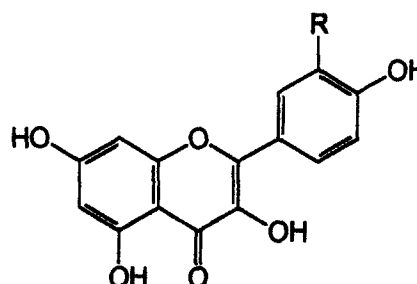
MATERIALS AND METHODS

Melting points were obtained on Gallenkamp melting point apparatus (uncorr.). IR spectra were recorded on a Nicolet model 205 FT-IR spectrophotometer. UV spectra were obtained on a Shimadzu UV₂₄₀ UV-Visible recording spectrophotometer. ¹H- and ¹³C-NMR spectra were determined on a Bruker AMX 400 spectrophotometer. MS was recorded on a VG70-VSEQ instrument. Analytical TLC was carried out on Merck aluminium plates precoated with Si gel 60 F₂₅₄. Chromatography was performed on Merck Si gel 60 (230~400 mesh) and Sephadex LH 20. LPLC was carried out on Duramat 80 equipped with a Merck Lichroprep Si 60 (240×10 mm) column.

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Compound I



Compound II R=H
Compound III R=OH

Fig. 1. Compounds isolated from *Cornus officinalis*.

Plant materials

The fruits of *Cornus officinalis* were collected at Suwon, Kyungkido, Korea in 1994. A voucher specimen is deposited in the herbarium of College of Pharmacy, Woosuk University (WSU-94-006).

Extraction and isolation

Fresh fruits (4 kg) of *Cornus officinalis* were extracted twice with MeOH at room temperature. The resulting MeOH extract (257 g) was subjected to successive solvent partitioning to give CH₂Cl₂ (35 g), EtOAc (30 g), n-BuOH (65 g) and water (130 g) soluble fractions. The EtOAc extract was applied over silica gel using gradient solvent system of EtOAc:MeOH (1:0 → 0:1)

as an eluent to give five sub-fractions (E1~E5). The first sub-fraction (20 g) was chromatographed on silica gel eluted with n-hexane:EtOAc:MeOH (10:10:1) to give five fractions. Fraction 4 (300 mg) was rechromatographed on LPLC (Chloroform:MeOH, 4.5:1) to yield 100 mg of compound **I**. Fractions 3 (30 mg) and 4 (20 mg) were applied over Sephadex LH 20 column chromatography (MeOH) to yield compounds **II** (15 mg) and **III** (10 mg).

Compound I (dimethyltetrahydrofuran *cis*-2,5-dicarboxylate): colorless needle in MeOH, mp. 76~78°C, $[\alpha]_D^{25}$; 0.0° (c 0.15, MeOH), IR $_{\text{max}}^{\text{KBr}}$ cm⁻¹ 1736 (C=O), 1228, GCMS:m/z (rel. int.); 188 (M⁺, 3), 59 (12), 71 (95), 89 (85), 102 (10), ¹H-NMR (CD₃OD, δ): 4.44 (2H, dd, *J*=7.1, 4.5 Hz, H-2,5), 3.69 (6H, s, 2×OCH₃), 2.73 (2H, dd, *J*=16.1, 4.5 Hz, H-3β, 4β), 2.59 (2H, dd, *J*=16.1, 7.1 Hz, H-3α, 4α), ¹³C-NMR (CD₃OD, δ): 175.1 (C=O), 174.0 (C=O), 66.7 (C-2, 5), 52.6 (2×OCH₃), 39.9 (C-3, 4).

Compound II (kaempferol): yellowish needles in MeOH, Mg-HCl, Zn-HCl test, FeCl₃ positive, mp. 274~277°C, IR $_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3500 (OH), 1660 (C=O), 1620, 1510, 1260, UV $_{\text{max}}^{\text{MeOH}}$ nm 261, 320 (sh), 370, EIMS:m/z (rel. int.); 286 (M⁺, 100), 258 (21), 229 (25), 153 (19), ¹H-NMR (DMSO-*d*₆, δ): 12.60 (1H, s, 5-OH), 8.02 (2H, d, *J*=8.8 Hz, H-2', 6'), 6.86 (2H, d, *J*=8.8 Hz, H-3', 5'), 6.43 (1H, d, *J*=2.0 Hz, H-8), 6.21 (1H, d, *J*=2.0 Hz, H-6), ¹³C-NMR (DMSO-*d*₆, δ): 175.9 (C-4), 163.9 (C-7), 160.7 (C-5), 156.4 (C-9), 159.2 (C-4'), 146.8 (C-2), 135.6 (C-3), 121.8 (C-1'), 128.6 (C-2', 6'), 115.0 (C-3', 5'), 103.2 (C-10), 98.2 (C-6), 93.5 (C-8).

Compound III (quercetin): yellowish powder in MeOH, Mg-HCl, Zn-HCl test, FeCl₃ positive, mp. 309~312°C, IR $_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3390 (OH), 1670 (C=O), 1620, 1520, 1430, UV $_{\text{max}}^{\text{MeOH}}$ nm 254, 292 (sh), 370, EIMS:m/z (rel. int.); 302 (M⁺, 100), 273 (12), 245 (35), 229 (18), 153 (50), ¹H-NMR (DMSO-*d*₆, δ): 12.50 (1H, s, 5-OH), 7.68 (1H, d, *J*=2.1 Hz, H-2'), 7.54 (1H, dd, *J*=8.5, 2.1 Hz, H-6'), 6.89 (1H, d, *J*=8.5 Hz, H-5'), 6.40 (1H, d, *J*=1.8 Hz, H-8), 6.18 (1H, d, *J*=1.8 Hz, H-6), ¹³C-NMR (DMSO-*d*₆, δ): 175.7 (C-4), 163.9 (C-7), 160.8 (C-5), 152.4 (C-9), 147.5 (C-4'), 146.6 (C-2), 145.0 (C-3'), 135.8 (C-3), 122.0 (C-1'), 120.1 (C-6'), 115.6 (C-5'), 115.0 (C-2'), 103.2 (C-10), 98.2 (C-6), 93.4 (C-8).

RESULTS AND DISCUSSION

The furan derivative, compound **I** was obtained as colorless needles (MeOH). Its IR spectrum showed the presence of an ester (1736 cm⁻¹) group, and molecular formula was established as C₈H₁₂O₅ by ¹H-NMR (12H), ¹³C-NMR (8C) and mass (m/z 188) spectra. The formula showed three unsaturated-bond equivalents. ¹³C-NMR (DEPT) spectra of compound **I** exhibited the two carbonyl (δ175.1 and 174.0 ppm) signals, hence the remaining one unsaturated-bond equivalent should be

a cyclic. The remaining carbon signals were revealed a methine bearing oxygen (δ66.7 ppm), a methoxy (δ 52.6 ppm) and a methylene (δ40.0), thus the structure of compound **I** was supposed to have a symmetric skeleton. From the values of chemical shifts and coupling patterns of ¹H-NMR (¹H-¹H COSY), AMX spin system (δ4.44, 2.73 and 2.59 ppm), and ¹³C-NMR spectra, a remaining unsaturation was accounted for one ring suggestive of a 2,5-disubstituted tetrahydrofuran compound. Finally, compound **I** was characterized as dimethyltetrahydrofuran *cis*-2,5-dicarboxylate. The structure and stereochemistry of compound **I** was identified by comparison with the ¹H-NMR spectrum data of the synthesized compounds (*cis* δ4.59 and *trans* δ4.71 ppm, H-2,5 in CCl₄), those reported in the literature (Moore *et al.*, 1972).

Compounds **II** and **III** have very similar patterns in their UV, IR and NMR spectra. Two compounds showed positive results in FeCl₃, Zn/HCl, Mg/HCl tests, and UV spectra exhibited characteristic absorptions for flavonols (Mabry *et al.*, 1970). Compounds **II** and **III** were identified as well-known flavonols, kaempferol and quercetin, respectively, by comparison of IR, NMR and UV data in literature (Kim *et al.*, 1995; Do *et al.*, 1992), and finally established by comparison with authentic standards.

ACKNOWLEDGEMENT

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