Benzyl-2,6-dimethoxy and benzyl-2,3,5,6tetramethoxybenzoates from *Blainvillea latifolia* Linn.

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Abstract – Aromatic esters named benzyl-2,6-dimethoxy and benzyl-2,3,5,6-tetramethoxybenzoates along with previously reported compounds were isolated from aerial parts of *Blainvillea latifolia*.

Key words – *Blainvillea latifolia*. Compositae. sesquiterpene lactones. acyclic diterpene. benzyl-2,6-dimethoxybenzoate and benzyl-2,3,5,6-tetramethoxybenzoate.

Introduction

Blainvillea latifolia Linn., an erect annual herb, belongs to the family-Compositae, tribe-Heliantheae and subtribe-Ecliptineae (Stuessy, 1977 and Robinson, 1981). So far three species of this genus namely Blainvillea dichotoma (Bohlmann et al., 1981), B. acmella (Singh et al., 1985) and B. latifolia (Singh et al., 1988, Sawaikar et al. 1994, 1997) have been investigated chemically. In pursuing our interest in the constituents of this plant we now report the isolation and characterization of benzyl-2,6-dimethoxybenzoate 1 and benzyl-2,3,5,6-tetramethoxybenzoate 2 from its aerial parts in addition to previously reported subacaulin 3, 5-desoxy pumilin 4, zoapatanolide-A 5, zoapatanolide-B 6 and 18acetoxy-12,19-dihydroxy geranyl nerol 7.

Experimental

General-IR – Perkin Elmer 577, Magna-550 FTIR Nicolet spectrometers. UV: Varian Carry 118. ¹H NMR: Bruker WH 400 MHz, JEOL FX 90Q. EIMS: Varian Mat 711, Varian CH

7, Hitachi model RMU 6E mass spectrometers. HPLC: Knaufer instrument. [α]_D: Perkin

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Elmer polarimeter. CC: Over silica gel (BDH, 60-120 mesh). Prep. TLC: Over Kiesel gel PF₂₅₄, ⁶⁰F₂₅₄ (E. Merck) plates. Melting points were recorded in soft glass capillaries in an electrothermal m.p. apparatus and are uncorrected. Chemical shifts are reported in ppm.

Plant materials – The aerial parts of Blainvillea latifolia were collected from M/s United Chemicals and Allied Products, Calcutta and voucher specimen deposited at RUBL Herbarium, Jaipur.

Extraction and Isolation - The air-dried coarsely powdered aerial parts (2 kg) were extracted with Et₂O-petrol-MeOH (1:1:1) at room temperature for 24 hours. Evaporation of the solvent in vacuo gave a greenish semisolid mass, it was defatted by dissolving in 200 ml MeOH and leaving over night at 2°. After filtration precipitate was rejected and filtrate was column chromatographed over silica gel and gave fr.1 (petrol), fr.2 (petrol- Et_2O , 4:1), fr.3 (petrol- Et_2O , 1:1), fr.4 (Et_2O) and fr.5 (Et₂O-MeOH, 9:1). Preparative TLC of fr.3 (silica gel PF₂₅₄) yielded phytol 40 mg, β-sitosterol 50 mg, stigmasterol 20 mg, benzyl-2,6-dimethoxy benzoate 1 30 mg and benzyl-2,3,5,6-tetramethoxybenzoate 2 40 mg. Fraction 4 on prep. TLC, Et₂O-MeOH (9:1) afforded subacaulin 3, 15 mg (R_f 0.4), 5-desoxy pumilin 4, 12 mg (R_f 0.35), zoapatanolide-A 5, $20 \text{ mg} (R_f 0.2)$ and zoapatanolide-B 6, 18mg (R_f 0.15). Fraction 5 on repeated prep TLC gave impure product which was further separated by HPLC (MeOH-H₂O, 7:3, analytical) RP 18 column (Rt 10.8 min). giving a further amount of 5,6 and 18-acetoxy-12-19dihydroxy geranyl nerol, 7, 30 mg.

Benzyl-2,6-dimethoxybenzoate, 1 − Colorless oil, IR v_{max} (CCl₄): 3000-2800, 1750 (COOR), 1610, 1590 (Ph), 1480, 1400, 1310, 1265, 1125, 1080 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ: 7.44 (2H, dd, J=7.5, 2 Hz, 2 x Ar-H), 7.36 (3H, m, 3 x Ar-H), 7.26 (1H, d, J=7.5 Hz, 1 x Ar-H), 6.55 (2H, d, J=7.5 Hz, 2 x Ar-H), 5.37 (2H, s, O-CH₂-Ph), 3.8 (6H, s, 2 x OMe). MS m/z (rel. int.): 272 [M]⁺ (30) (Calc.

for $C_{16}H_{16}O_4$: 272), 181 [M-CH₂Ph]⁺ (5), 165 [C₆H₃(OMe)₂CO]⁺ (100), 149 [181-MeOH]⁺ (20), 91 [-CH₂C₆H₅]⁺ (50).

Benzyl-2,3,5,6-tetramethoxybenzoate, 2-Colorless oil, IR v_{max} (CCl₄): 2900-2800, 1735 (COOR), 1590 (Ph), 1380, 1280, 1130, 1100, 1080 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ : 7.47 (2H, brd, J=7.5 Hz, 2×Ar-H), 7.35 (3H, m, 3 x Ar-H), 6.57 (1H, s, 1 x Ar-H), 5.39 (2H, s, -O-CH₂-Ph), 3.85 (6H, s, 2 x OMe), 3.74 (6H, s, 2 x OMe). MS m/z (rel. int.): 332.126 [M]⁺ (90) (Calc. for $C_{18}H_{20}O_6$: 332.126), 317 [M-Me]⁺ (24), 289 [317-CO]⁺, 225 [C_6H_1 (OMe)₄CO]⁺ (18), 197 [225-CO]⁺ (18), 91 [-CH₂C₆H₅]⁺ (100).

Results and Discussion

Compounds 1 and 2 were isolated as colorless oil. The presence of aromatic ester functions was ascertained by the appearance of strong obsorption bands at 1750, 1610 cm⁻¹ and 1735, 1590 cm⁻¹ in their IR spectra, respectively. High resolution mass spectrometry has established their molecular formulae. The high field ¹H NMR spectrum of 1 showed the presence of a sharp singlet at δ 5.37 corresponding to benzylic protons. A singlet at δ 3.8 integrated for six protons was assigned to two methoxyl groups. Eight aromatic protons displayed signals in the region of δ 6.55 to 7.44. In the 'H NMR spectrum of 2 benzylic protons appeared at δ 5.39 as singlet and four methoxy groups exhibited a pair of singlets at δ 3.85 and δ 3.74 each integrated for six protons. Six aromatic protons gave signals in the region δ 6.57 to 7.47. Both compounds showed similar fragmentation behaviour. In the mass spectrum of 1 highly abundant molecular ion peak was observed at m/z 272 along with some important fragments at m/z 181 $[M-CH_2Ph]^+$, 165 $[C_6H_3]$ (OMe)₂CO]⁺ and 91[C₆H₅CH₂]⁺ while in the mass spectrum of 2 besides a molecular ion peak at m/z 332, important fragments were observed at m/z 317 [M-Me]+, 225 [M-C₆H₁ $(OMe)_4CO]^+$, $197[225-CO]^+$, and $91[C_6H_5CH_2]^+$.

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The above spectral data were in close agreement with those reported earlier for 1 by Joshi et al. 1983 and for 2 by Bohlmann et al. 1980.

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