The Distribution of ²²⁸Ra in, and the Eddy Diffusivity of, Surface Waters of the Coastal Southwestern East Sea

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Radium isotopes were measured in 11 surface water samples collected during April 26-29, 1994 in the coastal southwestern East Sea. Manganese fiber extraction techniques were used to concentrate Ra isotopes from large volume seawater samples. The activities of ²²⁸Ra in the coastal southwestern East Sea were in the range of 5.74 to 7.68 dpm/100L in eleven samples. The ²²⁸Ra activities generally decrease seaward with increasing distance from the coast. In order to estimate the horizontal eddy diffusion coefficients, we applied a simple one-dimensional diffusion model to their distribution. The horizontal eddy diffusivity in the coastal southwestern East Sea were estimated to be in the order of 10⁶ cm²/sec.

INTRODUCTION

Radium is an alkaline earth element and exists mostly in dissolved phase in sea water as Ra²⁺. Radium is classified as biointermediate element. Radium isotopes are continuously produced from the decay of thorium isotopes.

Most of ²²⁶Ra in the ocean water is produced by the decay of ²³⁰Th in bottom sediments. ²²⁶Ra is released to sediment pore water and subsequently diffuses upward into the bottom water (Cochran and Krishinaswami, 1980). Since Koczy (1957) suggested that ²²⁶Ra was a useful time tracer with its half-life of 1,622 years in studying oceanic mixing processes occurring on a time scale of a few thousand years, geochemical behaviour of Ra isotopes and its potential as time tracer were intensively studied during GEOSECS program of 1970's.

²²⁸Ra with a half-life of 5.75 years could be a useful tracer in studying mixing processes in the coastal sea or near-bottom waters (Yamada *et al.*, 1986; Somayajulu *et al.*, 1987; Kasemsupaya *et al.*, 1993). ²²⁸Ra is supplied to coastal sea water from the decays of its parent nuclide ²³²Th in shelf sediments and dispersed to the open sea by advection and diffusion of surface water. Thus the concentration of ²²⁸Ra decreases seaward with increasing distance from the coast. Kaufman *et al.* (1973), Knauss *et al.*

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(1978) and Moore et al. (1980) have studied horizontal movements within the surface water from the distribution of ²²⁸Ra by estimating horizontal eddy diffusion coefficients. The vertical distribution of ²²⁸Ra has been determined to estimate vertical eddy diffusion coefficients across and/or within the thermocline and near-bottom region (Trier et al., 1972; Moore, 1972; Okubo, 1980; and Somayajulu et al., 1987).

In this study, we have attempted to measure ²²⁶Ra and ²²⁸Ra in coastal waters of the East Sea (Sea of Japan). We determined the distribution of ²²⁸Ra activities in the coastal southwestern East Sea to estimate horizontal eddy diffusion coefficient of the surface waters.

MATERIALS AND METHODS

Study area

The East Sea is a marginal sea (area: 1,007,600 km², mean depth: 1,684 m, maximum depth: 4,049 m) surrounded by the Korean peninsula and Japanese islands and connected to neighbouring seas through four straits: Korea, Tsugaru, Soya and Tartarskiy Straits (Woo et al., 1995). The East Sea water below 200 m is occupied by a remarkably uniform water mass which is characterized by relatively low salinities and temperatures (0.1-0.3°C,

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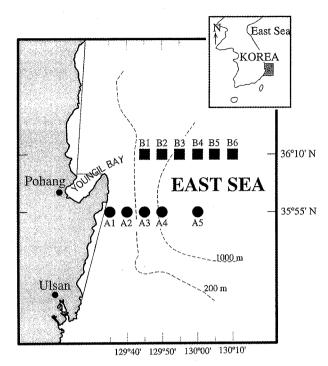


Fig. 1. A map showing the sampling stations. Circles represent the locations of surface water sampling on the line of 35° 55' N and squares on 35° 55' N.

34.0-34.1%) and it is called the East Sea Proper Water. The East Korea Warm Current branched from Tsushima Currents flows northward along the Korean coast (Lee, 1992).

Surface water sampling

Surface water samples were collected during R/V Tamyang cruise on April 26-29, 1994 at the coastal southwestern part of the East Sea from the 5 stations on the line of 35° 55' N and 6 stations on that of 36° 10' N (Fig. 1). The surface water samples were collected by utilizing ship's pre-installed pump system designed for obtaining clean sea water samples. The samples were divided into two parts: one part (about 240 liter) was transferred to three 80-liter plastic containers, and the other part (about 60 liter) was stored in polyethylene carboys for the Ba(Ra)SO₄ coprecipitation. Locations of each station were exactly recorded by a global positioning system (Furuno GP-500). Temperature and salinity were measured by a CTD (Sea-Bird Electronics 911 plus).

Radium isotopes analysis

We prepared Mn-fiber following the procedures

described by Moore et al. (1976). Briefly, the 0.4 M KMnO₄ solution was heated to 80-85°C in 2-liter beakers. A batch of fluffed fiber was soaked in the hot KMnO₄ solution for an hour until the fiber turned black. Then, the fiber was drew from the beaker and stand for 1 day. It was washed with distilled deionized water (DDW) and stored in a vinyl bag.

The surface water samples were immediately passed through two Mn-fiber cartridges (30 cm length × 7 cm inner diameter) connected in a series with flow rates of 5-7 liter/min (registered with a flow meter). Then, the Mn-fiber was removed from the cartridge and stored in vinyl bags.

Mn-fiber method is very effective for extracting radium isotopes from sea water (Reid et al., 1979; Moore et al., 1985). Moore and Reid (1973) reported radium uptake efficiency of 95% ±5% for Mn-fiber. It is not necessary to know extraction efficiency of Mn-fiber for the determination of the activity ratios between ²²⁸Ra and ²²⁶Ra. If the extraction efficiency is precisely known, Mn-fiber can be used for determination of the specific activities of ²²⁶Ra and ²²⁸Ra in seawater (Reid et al., 1979).

If we assume no isotopic fractionation between ²²⁸Ra and ²²⁶Ra during the extraction onto the Mn-fiber, and the extraction efficiencies of the first and the second column of Mn-fiber are same, then the extraction efficiency (E) and the specific activity may also be calculated directly from the following equations (Lee, 1991).

$$E = 1 - (A_2/A_1)$$
 (1)

where A_1 and A_2 are measured radium activities of the first and the second columns, respectively.

Radium activity (A₀) of the sea water is:

$$A_0 = A_1/E \tag{2}$$

Figure 2 shows a leaching processes of Mn-fiber following the procedures described by Moore *et al.* (1985). The Mn-fiber was leached with hot (50°C) 1 M hydroxylamine hydrochloride solution, 30% H₂O₂ and concentrated HCl. The solution was filtered through Whatman NO. 2 filter paper and the fiber was rinsed with 1 M HCl. This washing procedure was repeated 3 times and the washing solution was filtered.

About 50 ml of concentrated sulfuric acid was added to the leached solution and stirred. About 2 g of BaCl₂ was dissolved in 200-300 ml of DDW. About 70 ml aliquot of BaCl₂ solution was added to the sample and stirred. Radium isotopes in solution

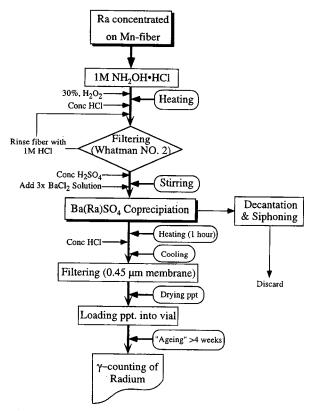


Fig. 2. A flow diagram showing the radium extraction procedures utilized in this study.

were associate with BaSO₄ and co-precipitated as Ba(Ra)SO₄.

Waiting for the settling of Ba(Ra)SO₄ precipitates, the precipitate was filtered with membrane filter (MPS, pore size: 0.45 µm). The Ba(Ra)SO₄ precipitates was dried at 60°C in a dry oven for 3 days and cooled in a desiccator. The precipitate was transferred into a pre-weighed plastic counting vial. The vial was weighed again to determine the net weight of Ba(Ra)SO₄ precipitates. Using this weight, the chemical yield was calculated by the following equation:

Chemical Yield (%)
$$= \frac{\text{weight of Ba (Ra) SO}_4 \text{ in vial (g)}}{\text{weight of dissolved BaCl}_2 \times 1.1209} \times 100 \quad (3)$$

where, 1.1209 is the weight conversion factor from BaCl₂ to BaSO₄.

The yield of radium recovery by BaSO₄ coprecipitation was estimated to be 100% within analytical error (Kim, 1992).

The vials containing the Ba(Ra)SO₄ precipitates were tightly sealed with caps and epoxy glue to

keep the gaseous ²²²Rn from escaping. In order to secure the equilibrium between Ra and Rn, the samples were aged for 21 days. The activities of ²²⁸Ra and ²²⁶Ra were measured by non-destructive gammaray spectrometry using a well-type intrinsic germanium detector (Canberra Model GCW 2021) following the modified procedure of Kim and Burnett (1983). The activities of ²²⁸Ra were counted via ²²⁸Ac peaks at 338.4 keV, 911.1 keV and 969 keV. ²²⁶Ra activities were determined via ²¹⁴Pb photopeaks at 295.2 keV, 351.9 keV and ²¹⁴Bi at 609.3 keV.

RESULTS AND DISCUSSION

A seasonal thermocline was located at about 30~ 130 m depth (B1: 30-60 m, B2: 40-80 m, B4: 60-130 m) (Fig. 3.) and the permanent thermocline at about 60-180 m depth (B1: 60-80 m, B2: 80-100 m, B4: 130-160 m, B6: 130-180 m) (Shin, 1994). The depth of the thermocline became gradually deeper with the distance from the coast. The halocline was located at the same depth of thermocline (80-200 m). The vertical profiles of temperature and salinity are shown in Fig. 3. The surface water temperatures in the study area were between 15.11 and 15.64°C, decreasing seaward from the coast. The salinities of the surface waters above the permanent thermocline were in a ranged of 34.61 to 34.67%. The vertical pattern of salinity distribution in the surface layer was more complicated than that of temperature.

The horizontal distribution of ²²⁶Ra and ²²⁸Ra

The results of our determination of radium isotopes in the surface waters of the East Sea are presented in Table 1. The activities of ²²⁶Ra were in the range of 6.91-10.74 dpm/100L, increasing seaward with increasing distance from the coast. These ²²⁶Ra activities are generally in good agreement with the previously measure values (Okubo, 1980; Harada *et al.*, 1986; Yang *et al.*, 1992).

The activities of ²²⁸Ra were in the range of 5.74 to 7.68 dpm/100L in eleven samples and the mean value was 6.642±0.059 dpm/100L. The lowest ²²⁸Ra activity was found at B5 and the highest activity at A1. The activities of ²²⁸Ra at station A1 and A2 on continental shelf, was significantly higher than those of the other stations and decreased seaward with increasing distance from the coast (Table 1). This suggests that ²²⁸Ra is supplied to by shelf waters from shelf sediments, and disperse seaward

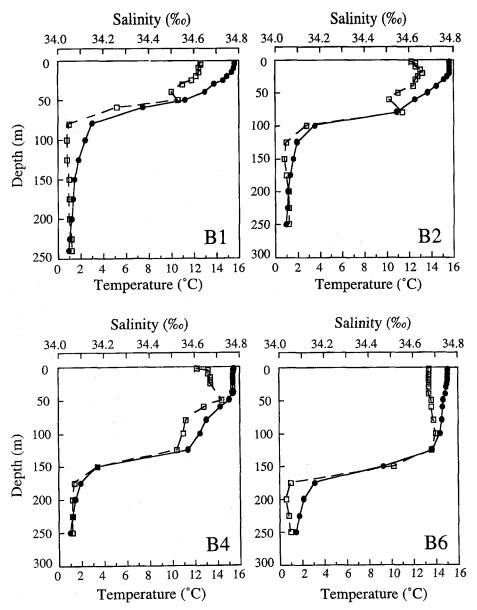


Fig. 3. Vertical profiles of salinity and temperature of St. B1, B2, B4 and B6 as recorded by a CTD in April 1994. Solid circles represent temperature and open squares salinity.

through surface waters by diffusive mixing processes (Fig. 4a.). The plot of ²²⁸Ra activity versus distance from the coastline is shown in Fig. 4b.

One-dimensional diffusion model

In this study one-dimensional, diffusion model is used for the estimation of horizontal eddy diffusion coefficients. This model has been used in previous studies of water mass mixing by ²²⁸Ra (Moore et al., 1972, 1980; Kaufman et al., 1973; Knauss et al., 1978; Yamada and Nozaki, 1986; Kasemsupaya et al., 1993). We applied the model equation to the hor-

izontal distribution of ²²⁸Ra activities in the surface waters of the East Sea.

If we assume that eddy diffusion plays a key role in dispersion of ²²⁸Ra in the surface layer and the system is in a steady state, then:

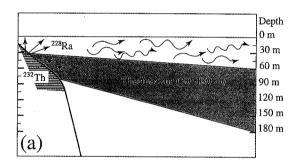
$$\frac{dC}{dt} = K_h \frac{\partial^2 C}{\partial x^2} - \lambda C = 0 \tag{4}$$

where, x is the distance in east-west direction from the coastline (cm), K_h the horizontal eddy diffusion coefficient (cm²/sec), λ decay constant (sec⁻¹), and C the activity of ²²⁸Ra (dpm/100L).

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Station No. —	Location		_ Distance* from	Activities (dpm/100L)		A ativity matic
	Latitude	Longitude	the coast (km)	Ra-226	Ra-228	 Activity ratio
A 1	35°55' N	129° 35' E	2.3	6.91±0.16	7.68 ± 0.13	1.111 ± 0.031
A2		129° 40' E	11.5	6.93 ± 0.15	7.41 ± 0.13	1.069 ± 0.030
A3		129° 45' E	20.8	7.92 ± 0.14	7.07 ± 0.19	0.893 ± 0.028
A 4		129° 50' E	30.0	8.22 ± 0.19	6.95 ± 0.25	0.846 ± 0.035
A5		130° 00' E	48.6	9.58 ± 0.24	5.99 ± 0.14	0.625 ± 0.022
B1	36° 10' N	129° 45' E	33.0	7.63 ± 0.16	6.76 ± 0.16	0.886 ± 0.028
B2		129° 50' E	42.3	7.79 ± 0.18	6.74 ± 0.32	0.866 ± 0.046
В3		129° 55' E	51.6	8.00 ± 0.16	6.19 ± 0.22	0.774 ± 0.032
B 4		130° 00' E	60.8	8.51 ± 0.19	5.89 ± 0.20	0.692 ± 0.028
B5		130° 05' E	70.1	9.45 ± 0.23	5.67 ± 0.18	0.601 ± 0.024
B6		129° 10' E	79.3	10.74 ± 0.25	5.74 ± 0.15	0.534 ± 0.019
Mean				8.223 ± 0.057	6.642 ± 0.059	0.8296 ± 0.009

^{*}Distance measured along the east-west direction.



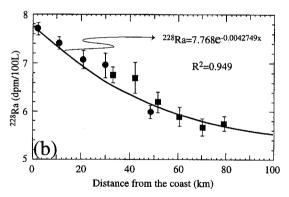


Fig. 4. (a) A conceptual sketch showing the diffusive pathway of ²²⁸Ra in the ocean. Short arrows denote diffusion of ²²⁸Ra across the sediment-water interface, (b) Plots of ²²⁸Ra activity as a function of distance from the coastline in the southwestern East sea. Circles indicate stations on the line of 35° 55' N and squares on 36° 10' N.

The boundary conditions are,

1)
$$C = C_0$$
 at $x = 0$

2) C
$$\rightarrow$$
 0 at $x\rightarrow\infty$.

A general solution to the equation (4) is as follows:

$$C = C_0 \exp[-(\lambda/K_h)^{1/2} \cdot x]$$
 (5)

We estimated the value of K_h by fitting equation (5) to the plot of ²²⁸Ra activity versus distance from the coastline (Fig. 4b), yielding the following equation:

²²⁸Ra =
$$(7.768 \pm 0.110)$$
 exp $[-(0.004275 \pm 0.000330) \cdot x]$
(R² = 0.949)

Thus the K_h was estimated to be $(1.79 \pm 0.28) \times 10^6$ cm²/sec (=10^(6.25 \pm 0.07)).

Our estimate is in the same order as the value from the Seto Inland Sea (Kasemsupaya et al., 1993: $1.1-1.4\times10^6$ cm²/sec) and Long Island Sound (Kaufman et al., 1973: $0.5-5\pm10^6$ cm²/sec). On the other hand, Kaufman et al. (1973) and Moore (1980) reported the K_h values ranging from 10^7 to 10^8 cm²/sec in the surface waters of the Australian Pacific and the North Atlantic Ocean.

CONCLUSIONS

In order to estimate the horizontal eddy diffusion coefficients in the coastal surface waters of the East Sea, we measured the ²²⁸Ra and ²²⁶Ra activities and applied a simple "one-dimensional" diffusion model to their distribution. The activities of ²²⁸Ra were in the range of 5.74 to 7.68 dpm/100L and the mean activity was 6.642 ± 0.059 dpm/100L. From the surface water distribution of ²²⁸Ra activities, the horizontal eddy diffusion coefficients in the coastal waters of the southwestern East Sea were estimated to be in the order of 10^6 cm²/sec. This result may provide useful information concerning the diffusion of coastal waters and dissolved pollutants to the East Sea.

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