

## Chemical Constituent of *Aloe capensis*

Man Ki Park<sup>1</sup>, Jeong Hill Park<sup>1</sup>, Young Geun Shin<sup>1</sup>, Yong Seok Choi<sup>1</sup>, Kyeong Ho Kim<sup>2</sup>, Tae Hyeong Cho<sup>3</sup> and Seung Ki Lee<sup>1</sup>

<sup>1</sup>College of Pharmacy, Seoul National University, Seoul 151-742, Korea, <sup>2</sup>College of Pharmacy, Kwangwon National University, Chuncheon 220-701, Korea, <sup>3</sup>Namyang Aloe Co., Chopyung, Jincheon, Chungbuk 365-850, Korea

(Received January 9, 1997)

A C-glycosyl chromone, named as 7-O-methylaloesinol, was newly isolated from the leaf exudate of *Aloe capensis* and identified as 8-C- $\beta$ -D-glucopyranosyl  $\beta$ -2-[2-(R)-hydroxypropyl]-7-methoxy-5-methyl-4H-1-benzopyran-4-one by chemical and spectral evidence.

**Key words** : *Aloe capensis*, Liliaceae, 8-C- $\beta$ -D-glucopyranosyl-2-[2-(R)-hydroxypropyl]-7-methoxy-5-methyl-4H-1-benzopyran-4-one, 7-O-methylaloesinol

### INTRODUCTION

Aloe is the dried latex of *Aloe ferox* Miller and its hybrids (Cape aloe) or *Aloe barbadensis* Miller (Curacao aloe) and it has long been used in folk medicine to treat constipation, burns and dermatitis. Several anthraquinones, anthrones, chromones, and their C-glycosyl derivatives were isolated from various species of aloe (Park *et al.*, 1995; Park *et al.*, 1996; Reynolds, 1985; Conner *et al.*, 1989; 1990, Speranza *et al.*, 1986; 1993). In the course of isolating chromone components from the leaf exudate of *Aloe capensis*, a C-glycosylchromone (Fig. 1) was newly isolated from the *n*-BuOH extract.

### MATERIALS AND METHODS

#### Materials

The dried leaf exudate of *Aloe capensis* was purchased from Wha-Il Pharmaceutical Co. Ltd., in Seoul.

#### Instruments

Melting point was recorded on a Gallenkamp melting point apparatus and was uncorrected. UV spectra were measured on a Shimadzu UV-2100 UV/VIS spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Jeol JNM-LA300 spectrometer. IR spectra were obtained on Perkin-Elmer 1710 spectrometer. Mass spectra were obtained using VG TRIO-II GC/MS system and Jeol AX505WA mass spectrometer. Silica gel 60 and TLC plates were purchased from Merck (Germany).

Correspondence to: Jeong Hill Park, College of Pharmacy, Seoul National University, Seoul 151-742, Korea

HPLC was carried out on Samsung SLC-100 system (Samsung, Korea) using  $\mu$ -Bondapak C<sub>18</sub> column (4 mm  $\times$  300 mm, 10  $\mu$ m, Waters, USA).

#### Extraction and isolation

Dried exudate (500 g) of *Aloe capensis* was dissolved in H<sub>2</sub>O (3L) and extracted with EtOAc (3L  $\times$  3) stirring vigorously for 3 hours at room temperature. The aqueous layer was further extracted with water-saturated *n*-BuOH (3L  $\times$  3). The *n*-BuOH layer was concentrated *in vacuo* and the residue (180 g) was subjected to column chromatography on silica gel (800 g, 230-400 mesh, column: 80 mm  $\times$  1 m) using CHCl<sub>3</sub>/MeOH (5/1, v/v) as eluent. Six fractions were obtained and Fr. III (9.5 g) was further chromatographed on silica gel (200 g, 230-400 mesh, column: 4  $\times$  50 cm) using EtOAc/MeOH/H<sub>2</sub>O (10/1/0.5, v/v/v) as eluent. Compound 1 (900 mg) was isolated from the 7th fraction in 0.18% yield.

**Compound 1**: amorphous powder (EtOAc/MeOH),

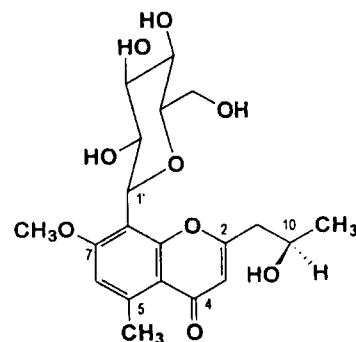


Fig. 1. The structure of compound 1.

mp: 140-142°,  $[\alpha]_D^{24}$ : -11.5° (MeOH, c 0.2),  $R_f$ : 0.22 (CHCl<sub>3</sub>/MeOH=5/1; Kieselgel 60F<sub>254</sub>), UV  $\lambda_{max}$  (log  $\epsilon$ ) MeOH: 226 (4.27), 242 (4.20), 250 (4.17), 292 (4.00), IR  $\nu_{max}$  (KBr): 3401, 2361, 1651, 1598, 1385 cm<sup>-1</sup>, <sup>1</sup>H-NMR (300 MHz, DMSO-*d*<sub>6</sub>,  $\delta$  ppm): 1.16 (3H, d,  $J$ =6.0 Hz, 11-H), 2.57 (2H, brd,  $J$ =6.0 Hz, 9-Hz), 2.74 (3H, s, 12-H), 3.2-4.0 (5H, m, sugar-H), 3.88 (3H, s, 7-OCH<sub>3</sub>), 4.13 (1H, m, 10-H), 4.73 (1H, d,  $J$ =9.6 Hz, 1'-H), 6.03 (1H, s, 3-H), 6.93 (1H, s, 6-H), <sup>13</sup>C-NMR (75 MHz, DMSO-*d*<sub>6</sub>,  $\delta$  ppm): 165.3 (C-2), 111.6 (C-3), 179.2 (C-4), 116.3 (C-4a), 141.5 (C-5), 111.9 (C-6), 160.4 (C-7), 113.2 (C-8), 157.6 (C-1a), 43.6 (C-9), 64.3 (C-10), 24.2 (C-11), 23.3 (C-5-CH<sub>3</sub>), 56.7 (C-7-OCH<sub>3</sub>), 73.4 (C-1'), 71.4 (C-2'), 79.1 (C-3'), 71.2 (C-4'), 82.2 (C-5'), 62.2 (C-6'), Mass [EI+,  $m/z$ ] (rel.int. %): 410[M<sup>+</sup>] (11), 392 (10), 366 (14), 277 (19), 259 (81), 233 (100), 217 (49), 193 (94), 121 (63), positive FAB-MS: 411[M+H]<sup>+</sup>.

### Synthesis of compound 1

Aloesin (100 mg), which was isolated in our laboratory, was methylated with CH<sub>2</sub>N<sub>2</sub> in diethylether to give 7-O-methylaloesin and NaBH<sub>4</sub> (200 mg) was added to a solution of 7-O-methylaloesin in MeOH (5 ml). The solution was stirred for 3.5 hrs at room temperature and acidified to pH 3 with 1M-HCl. The solution was extracted with *n*-BuOH and *n*-BuOH soluble part was evaporated *in vacuo* to give a mixture of diastereomers (60 mg). HPLC of the product showed two peaks; (10,*R*)-form at 7.5 min and (10,*S*)-form at 8.3 min. HPLC conditions; column: Waters  $\mu$ -Bondapak C<sub>18</sub> (4 mm $\times$ 300 mm, 10  $\mu$ m), mobile phase: linear gradient from 10% CH<sub>3</sub>CN in H<sub>2</sub>O to 20% CH<sub>3</sub>CN for 10 min, flow rate: 1.0 ml/min, detection: UV 293 nm. Compound 1 was separated from the product mixture by semi-prep HPLC; column: Lichrosorb C<sub>18</sub> (7  $\mu$ m, 10 mm $\times$ 250 mm), mobile phase: linear gradient from 10% MeOH to 40% MeOH in H<sub>2</sub>O for 30 min, flow rate: 2.5 ml/min, detection: UV 293 nm.

### RESULTS AND DISCUSSION

Compound 1, C<sub>20</sub>H<sub>26</sub>O<sub>9</sub>, gave blue fluorescence under long wavelength UV (365 nm). Positive FAB-MS spectrum showed [M+H]<sup>+</sup> ion peak at  $m/z$  411. The EI-MS spectrum showed fragmentation pattern very similar to that of aloeresin D (Speranza *et al.*, 1986), except that it lacked a fragment arising from p-coumaroyl moiety ( $m/z$  164). The <sup>1</sup>H NMR spectrum of 1 showed two methyl proton signals at  $\delta$  1.16 and  $\delta$  2.74, a methoxy proton signal at  $\delta$  3.88, an olefinic proton signal at  $\delta$  6.03 and an aromatic proton signal at  $\delta$  6.93. The anomeric proton signal of compound 1 appeared at  $\delta$  4.73 (1H, d,  $J$ =9.6 Hz) and its coupling

constant suggests a  $\beta$ -glucosidic linkage (Overend, 1972). These data indicated that compound 1 has an aloeresin D analogue lacking p-coumaroyl moiety. The <sup>13</sup>C NMR spectrum of 1 showed two methyl carbons at  $\delta$  23.3 and  $\delta$  24.2, one methoxy carbon at  $\delta$  56.7. By comparison of <sup>13</sup>C-NMR of C- or O-glycosyl compounds (Markham, 1982), hexose carbons of 1 appeared at  $\delta$  73.4 (anomeric carbon), 71.4, 79.1, 71.2, 82.2 and 62.2 suggesting the existence of C-glycosidic linkage and a glucose moiety in the molecule. All <sup>13</sup>C and <sup>1</sup>H NMR signals of compound 1 were assigned using DEPT, <sup>1</sup>H-<sup>1</sup>H COSY, <sup>13</sup>C-<sup>1</sup>H COSY and were confirmed by comparison of chemical shifts and coupling constants with those of aloesin analogues (Park *et al.*, 1995; Park *et al.*, 1996; Conner *et al.*, 1989; 1990; Speranza *et al.*, 1986, 1993). The configuration of C-10 in compound 1 was determined by Speranza's procedure (Speranza *et al.*, 1986) as follows. Aloesin was treated with CH<sub>2</sub>N<sub>2</sub> to give 7-O-methylaloesin, which was then reduced with NaBH<sub>4</sub>. In HPLC analysis, the product showed two peaks arising from (10,*R*) and (10,*S*) stereoisomers ( $R_f$ =7.5 and 8.3 min). Compound 1 coincided with former peak ( $R_f$ =7.5 min) which is (10,*R*)-form of 8-C-glucosyl-7-O-methylaloesol. Alkaline hydrolysate of authentic aloeresin D also coincided with former peak ( $R_f$ =7.5 min) which is (10,*R*)-form (Speranza *et al.*, 1986).

Thus, compound 1 was identified as 8-C- $\beta$ -D-glucopyranosyl-2-[2-(*R*)-hydroxypropyl]-7-methoxy-5-methyl-4H-1-benzopyran-4-one, and designated as 7-O-methylaloesinol. Although compound 1 and its stereoisomer were synthesized from aloesin to elucidate the structure of aloeresin D (Speranza *et al.*, 1986), compound 1 has not been reported in the nature so far.

### ACKNOWLEDGMENTS

We are grateful to Dr. Speranza for supplying the authentic sample of aloeresin D. This research was supported by the grant from the Ministry of Science and Technology, Korea.

### REFERENCES CITED

- Conner, J. M., Gray, A. I., Reynolds, T. and Waterman, P. G., Anthracene and chromone derivatives in the exudate of *Aloe rabaicensis*, *Phytochemistry*, 28, 3551-3553 (1989)
- Conner, J. M., Gray, A. I., Reynold, T. and Waterman, P. G., Anthrone and chromone components of *Aloe cremnophila* and *A. jacksonii* leaf exudates, *Phytochemistry*, 29, 941-944 (1990)
- Markham, K. R., *Techniques of flavonoid identification*, Academic Press, New York, 1982, p81.
- Overend, W. G., Glycosides, In Pigman, W. and Horton, D., (Eds.), *The carbohydrates: chemistry and bio-*

- chemistry*, Academic Press, New York, 1972, pp. 306-309.
- Park, M. K., Park, J. H., Kim, K. H., Shin, Y. G., Myoung, K. M. and Lee, J. H., Chemical constituents on *Aloe capensis*, *Saengyak Hakhoechi*, 26, 244-247 (1995)
- Park, M. K., Park, J. H., Shin, Y. G., Kim, W. Y., Lee, J. H. and Kim, K. H., Neoaloesin A : A new C-glucofuranosyl chromone from *Aloe barbadensis*, *Planta Med.*, 62, 363-365 (1996)
- Reynolds, T., Observations on the phytochemistry of the Aloe leaf-exudate compounds, *Bot. J. Linn. Soc.* 90, 179-199 (1985).
- Speranza, G., Dada, G., Lunazzi, L., Gramatica, P. and Manitto, P., A C-glucosylated 5-methylchromone from Kenya aloe, *Phytochemistry*, 25, 2219-2222 (1986)
- Speranza, G., Manitto, P., Cassara, P., Monti, D., Castri, D. D., Chialva, F., Studies on Aloe, 12. Furoaloesone, a new 5-methylchromone from Cape aloe, *J. Nat. Prod.*, 56, 1089-1094 (1993)