



in reference, Ryu *et al.*, 1992).

### Extraction, separation, purification and synthesis of compounds

The half-dried and chopped whole plant (1.5 kg) was extracted with MeOH for 7 days at room temperature (2 times) and 5 hours in 50°C (1 time), then concentrated in vacuo. The MeOH extract (90 g) was suspended in H<sub>2</sub>O (800 ml) and successively partitioned with CH<sub>2</sub>Cl<sub>2</sub> (800 ml), EtOAc (800 ml) and *n*-BuOH (800 ml). The CH<sub>2</sub>Cl<sub>2</sub> phase was evaporated under vacuum to yield the CH<sub>2</sub>Cl<sub>2</sub> extract (40 g).

The CH<sub>2</sub>Cl<sub>2</sub> extract was chromatographed on silica gel with gradient *n*-Hexane-CH<sub>2</sub>Cl<sub>2</sub> (1:8~) and CH<sub>2</sub>Cl<sub>2</sub>-MeOH (~5:1) to give nine subfractions (PM1~9), which were assayed by SRB method. The PM7 fraction, which was exhibited cytotoxic activity, was rechromatographed with SiO<sub>2</sub> column using *n*-hexane:CH<sub>2</sub>Cl<sub>2</sub>:MeOH (15:30:1) and sephadex LH-20 column using acetone:MeOH (1:1) to give two subfraction. The first and second subfraction was further purified by preparative HPLC (Econosil C-18, 10×250 mm; 3.0 ml/min) with MeOH and MeCN:MeOH (1:1) to afford compound I (21 mg) and compound II (10 mg), respectively.

#### Compound I

white powder; mp. 280°C; IR  $\nu_{\max}$  cm<sup>-1</sup>: 3300~3500 (OH), 2900, 1700 (C=O), 1450, 1380, 1250; EIMS  $m/z$  (rel. int., %): 456 (M<sup>+</sup>, 3), 438<sup>a</sup>(6), 248 (100), 233 (14), 203 (87), 190 (67), 175 (32), 133 (41), 105 (28), 81 (26), 69 (40), 55 (46); <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 0.72, 0.80, 0.88, 0.90 ( $\times$  2), 0.92 (each 3H, s, CH<sub>3</sub> $\times$  6), 1.11 (3H, s, 29-CH<sub>3</sub>), 2.78 (1H, br.dd, H-18), 3.38 (1H, t, J=2.48 Hz, H-3), 5.25 (1H, t, J=3.40 Hz, H-12); <sup>13</sup>C-NMR (75.5 MHz, CDCl<sub>3</sub>): Table I

#### Compound II

white powder; mp. 154°C;  $\nu_{\max}$  cm<sup>-1</sup>: 2900, 1700 (C=O), 1550, 1450, 1390, 1250; EIMS  $m/z$  (rel. int., %): 454 (M<sup>+</sup>, 33), 408 (33), 248 (100), 233 (47), 203 (97), 189 (55), 133 (62), 105 (42), 95 (35), 55 (42); <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 0.81, 0.90, 0.93, 1.03, 1.05, 1.09 (each 3H, s, CH<sub>3</sub> $\times$  6), 1.14 (3H, s, 29-CH<sub>3</sub>), 2.40 (1H, m, H-2), 2.52 (1H, m, H-2), 2.84 (1H, br.dd, H-18), 5.31 (1H, br.s, H-12); <sup>13</sup>C-NMR (75.5 MHz, CDCl<sub>3</sub>): Table I

#### Synthesis of oxo-oleanolic acid (Cainelli, G. *et al.*, 1984; Huneck, S., 1953; Yagi, A. *et al.*, 1978)

To a solution of oleanolic acid (20 mg, 0.044 mmol) dissolved in benzene (7 ml), PCC (pyridinium chlorochromate, 30 mg, 0.14 mmol) was added and the reaction mixture was stirred for 8 hours at room tem-

**Table I.** <sup>13</sup>C-NMR and DEPT data of compounds I and II

C	Compound I		Compound II	
	<sup>13</sup> C ( $\delta$ )	DEPT	<sup>13</sup> C ( $\delta$ )	DEPT
1	32.48	CH <sub>2</sub>	39.15	CH <sub>2</sub>
2	25.18	CH <sub>2</sub>	34.16	CH <sub>2</sub>
3	76.18	CH	217.60	C
4	37.34	C	47.44	C
5	48.98	CH	55.37	CH
6	18.24	CH <sub>2</sub>	19.61	CH <sub>2</sub>
7	32.48	CH <sub>2</sub>	32.23	CH <sub>2</sub>
8	39.45	C	39.36	C
9	47.41	CH	46.64	CH
10	37.17	C	36.83	C
11	22.93	CH <sub>2</sub>	23.02	CH <sub>2</sub>
12	122.70	CH	122.37	CH
13	143.60	C	143.75	C
14	41.67	C	41.82	C
15	27.63	CH <sub>2</sub>	27.74	CH <sub>2</sub>
16	23.32	CH <sub>2</sub>	23.59	CH <sub>2</sub>
17	46.52	C	46.64	C
18	40.95	CH	41.16	CH
19	45.89	CH <sub>2</sub>	45.92	CH <sub>2</sub>
20	30.67	C	30.70	C
21	33.81	CH <sub>2</sub>	33.89	CH <sub>2</sub>
22	32.86	CH <sub>2</sub>	32.49	CH <sub>2</sub>
23	28.27	CH <sub>3</sub>	26.51	CH <sub>3</sub>
24	22.25	CH <sub>3</sub>	21.46	CH <sub>3</sub>
25	15.10	CH <sub>3</sub>	15.04	CH <sub>3</sub>
26	17.18	CH <sub>3</sub>	17.00	CH <sub>3</sub>
27	26.09	CH <sub>3</sub>	25.84	CH <sub>3</sub>
28	183.28	C	183.30	C
29	33.06	CH <sub>3</sub>	33.07	CH <sub>3</sub>
30	23.59	CH <sub>3</sub>	23.59	CH <sub>3</sub>

perature. The product was filtered and evaporated to dryness. The residue was chromatographed on silica gel with *n*-hexane:EtOAc (3:1). The main fraction was evaporated to dryness and was recrystallized from MeOH to give white powder (16 mg). The mp., TLC, <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra of this product were compared with those of compound II

#### Synthesis of epi-oleanolic acid (Huneck, S., 1953)

The synthetic oxo-oleanolic acid (10 mg, 0.022 mmol) and aluminiumisopropoxide (30 mg, 0.147 mmol) were dissolved in isopropanol (1 ml). The reaction mixture was stirred for overnight at 70°C. The reaction mixture was evaporated to dryness and 1N HCl (8 ml) was added. The reaction mixture was extracted three times with ether. The ether layer was dried over magnesium sulfate and concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel with *n*-Hexane:EtOAc (3:1) to give white powder (3 mg). The mp., TLC and <sup>1</sup>H-NMR spectrum of this product were compared with those of compound I

## RESULTS AND DISCUSSION

The CH<sub>2</sub>Cl<sub>2</sub> soluble fraction of *Pilea mongolica* has been found to exhibit the cytotoxic activity against several human cancer cell lines. Bioassay-guided fractionation led to isolation of two oleanane type triterpene as active principles.

Compound **I** was obtained as colorless amorphous powder. EIMS and DEPT data established molecular formula of C<sub>30</sub>H<sub>48</sub>O<sub>3</sub>. In the mass spectrum the base peak at *m/z* 248 (C<sub>16</sub>H<sub>24</sub>O<sub>2</sub>) besides other peaks at *m/z* 203 (C<sub>15</sub>H<sub>23</sub>) and 133 (C<sub>10</sub>H<sub>13</sub>) were characteristic of a Δ<sup>12</sup>-amyrin skeleton with a carboxyl group in ring D/E and hydroxy groups in ring A and/or B. (Budzikiewicz, H. *et al.*, 1963; Akhtar, N. *et al.*, 1993) The presence of only tertiary methyl groups indicated the olean-12-ene skeleton (Budzikiewicz, H. *et al.*, 1963; Seo, S. *et al.*, 1981). The IR spectrum exhibited hydroxyl (3300~3500 cm<sup>-1</sup>) and carboxyl group (1700 cm<sup>-1</sup>) absorption bands. The <sup>1</sup>H-NMR spectrum clearly showed an olefinic proton at δ 5.25 (1H, t, J=3.40 Hz, H-12), a methine proton at δ 2.78 (1H, br.dd, H-18), an α-carbinol proton at 3.38 (1H, t, J=2.48 Hz, H-3), and seven methyl functionalities all attached to saturated carbons in the δ 0.72-1.11. Its broad-band-decoupled <sup>13</sup>C-NMR and DEPT spectrum exhibited the presence of 30 carbon signals (CH<sub>3</sub>×7, CH<sub>2</sub>×10, CH×5, C×8). These data showed two olefinic carbon signal at 122.7 (CH) and 143.6 ppm (C), and a carbonyl carbon at 183.3 ppm (C). These data were consistent with having a OH group at C-3 of oleanane type triterpene. <sup>1</sup>H and <sup>13</sup>C-NMR data of oleanolic acid were similar with compound **I**; the major difference between oleanolic acid and compound **I** was in the <sup>13</sup>C-NMR chemical shift of the C-24 carbon. C-24 methyl carbon (22.25 ppm) in compound **I** showed at downfield to 7 ppm than C-24 peak (15.6 ppm) of oleanolic acid. Also, the stereochemistry of the α-OH at C-3 was deduced from the comparison of the coupling constant with those reported ones (Akhtar, N. *et al.*, 1993; Chem, T. K. *et al.*, 1983; Hylands, P. J. *et al.*, 1980; Mahato, S. B. *et al.*, 1994). Consequently, we suggested that compound **I** was epi-oleanolic acid. The structure was further confirmed by comparison of <sup>1</sup>H-NMR data of epi-oleanolic acid methyl ester in previous papers (Ikuta, A., 1989; Konishi, T. *et al.*, 1981) and synthesis of epi-oleanolic acid from oxo-oleanolic acid.

Compound **II** was obtained as colorless amorphous powder and showed many spectral features in common with compound **I**. EIMS and DEPT data established molecular formula of C<sub>30</sub>H<sub>46</sub>O<sub>3</sub>. The IR spectrum exhibited carboxyl group (1700 cm<sup>-1</sup>) absorption band. The <sup>1</sup>H-NMR spectrum is consistent with the triterpene structure and clearly showed an olefinic proton at δ 5.31 (1H, br.s, H-12), a methine

**Table II.** Cytotoxicity of compounds **I** and **II**

ED <sub>50</sub> values					
cancer cell line	A549	SK-OV-3	SK-MEL-2	XF498	HCT15
compound <b>I</b>	3.2523	5.3536	8.1379	3.6646	3.8027
compound <b>II</b>	2.3578	6.5829	6.8531	4.1675	0.7156

ED<sub>50</sub> value of compound against each cancer cell line, which was defined as a concentration (μg/ml) that caused 50% inhibition of cell growth *in vitro*

proton at δ 2.84 (1H, br.dd, H-18), and seven methyl functionalities all attached to saturated carbons in the δ 0.81-1.14. <sup>13</sup>C-NMR and DEPT spectra of **II** were very similar to compound **I**, but the difference was that a new peak appeared at 217.60 ppm (C) in compound **II** instead of peak at 76.18 ppm (CH) in **I**. Consequently, we supposed that a hydroxyl group of **I** was reduced to ketone in to give **II**. On the basis of above mentioned evidence and previous papers, (Seo, S. *et al.*, 1975) compound **I** was suggested as oxo-oleanolic acid. Its structure was further confirmed by synthesis of oxo-oleanolic acid from oleanolic acid.

As shown in Table **II**, compound **I** and **II** showed significant cytotoxicity against cancer cells. Especially, compound **II** showed strong cytotoxic activity against HCT15 (colon cancer cells) (ED<sub>50</sub>: 0.7156 μg/ml).

## ACKNOWLEDGEMENT

This work was supported in part by the research grant from Korea Science & Engineering foundation (KOSEF; 93-0400-07). We wish to thank Dr. Jong Hwan Kwak, KIST for the collection of plant.

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