

Effects of Nutritional Sources on Degradation of Polychlorinated Biphenyls (PCBs) by *Pseudomonas* sp. P2

Sang-Ki CHOI and Jung-Ho KIM

Dept. of Environmental Science, Kyungsan University, Kyungsan, Kyungpook, 712-240, Korea

(Manuscript received 26 January 1996)

The effects of nutritional sources on growth of *Pseudomonas* sp. P2 were investigated in medium containing biphenyl as a carbon source. To determine characterization of *Pseudomonas* sp. P2, the incubation time was determined to 100 h of the log phase in the growth curve. The optimal compositions for the growth of *Pseudomonas* sp. P2 degrading polychlorinated biphenyls (PCBs) were 1000 mg/L NH_4NO_3 , 1000 mg/L KH_2PO_4 , 100 mg/L $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 30 mg/L $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 200 mg/L NaCl, and 10 mg/L $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$. *Pseudomonas* sp. P2 showed the degradability of 59.3%, 57.6%, 51.4%, and 48.7% at 500 mg/L, 1000 mg/L, 1500 mg/L, and 2000 mg/L of the PCBs within insulating oil after 100 h incubation under the optimum conditions, respectively.

Key words : Polychlorinated biphenyls, *Pseudomonas* sp., Characterization, Degradation, Insulating oil.

1. INTRODUCTION

Polychlorinated biphenyls (PCBs) represents many isomers with different numbers of chlorine substituted in biphenyl. While there are 209 different PCBs isomers (Bedard et al., 1987), the commercial PCBs are the mixture of these isomers and were used as insulating oil in the electrical transformers and capacitors (Tanabe, 1988; May et al., 1992).

PCBs has been concerned for several decades because of persistence in environment, accumulation in animal tissue, and potential carcinogen. PCBs was usually distributed as a mixture of PCBs isomers in environment (May et al., 1992). Although PCBs was recalcitrant, recent studies have shown that PCBs has been extensively dechlorinated in some environment. The biodegradability of PCBs is important to minimize its impact on biota (Furukawa, 1978).

Many works have been done to isolate strains degrading PCBs. Furukawa et al. (1979) demonstrated the metabolic fates of PCBs by *Alcaligenes* sp. Y42 and *Acinetobacter* sp. P6. Furukawa et al. (1983) also studied the biodegradability of PCBs, and its metabolic products by *Acinetobacter* sp. P6. Shields et al. (1985) also investigated the mineralization of PCBs by *Alcaligenes* sp. and *Acinetobacter* sp. Kim et al. (1986) paid attention to the degra-

dation of PCBs by *Alcaligenes aquamarinus*. Kohler et al. (1988) studied the biodegradability of Aroclor 1254 by *Acinetobacter* sp. P6 and *Arthrobacter* sp. B1. Gibson et al. (1993) worked that *Pseudomonas* sp. LB400 was grown in medium containing biphenyl as a carbon source.

Some studies have also been conducted to understand the effects of environmental factors on microbial degradation of PCBs. Kim and Kim (1990) studied the effects of environmental factors on the degradation of Aroclor 1242 by *Pseudomonas* sp. Liu (1981) studied the PCBs biodegradation whose factors affecting degradation. Bedard et al. (1986) reported the characterization of microorganisms with regard to the ability to degrade PCBs.

If *Pseudomonas* sp. P2 is considered as strain biodegrading PCBs within insulating oil, it is essential that the optimum conditions for growth are determined. Investigation for nutritional sources could also provide useful information on the optimum conditions degrading PCBs within insulating oil by *Pseudomonas* sp. P2. Therefore, it was attempted to investigate the culture conditions of the *Pseudomonas* sp. P2 for further characterization.

2. Materials and Methods

2.1 Microorganism and Medium

The *Pseudomonas* sp. P2 isolated from the soil

on the Sinchun stream in Taegu was used (Choi, 1996). *Pseudomonas* sp. P2 was grown aerobically in 30 mL nutrient broth medium with 150 rpm shaking incubator for seed culture at 30°C 12 h. The cells were then harvested by centrifugation, and washed thoroughly by deionized water.

The compositions of basal medium were 1000 mg/L $(\text{NH}_4)_2\text{SO}_4$, 1000 mg/L KH_2PO_4 , 200 mg/L $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 100 mg/L NaCl, 10 mg/L $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$, and 20 mg/L $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$. Initial pH of medium was adjusted to 7.0 (Kim et al., 1986).

2.2 Effects of Nutritional Factors

The effects of 1000 mg/L of $(\text{NH}_2)_2\text{CO}$, $(\text{NH}_4)_2\text{SO}_4$, NaNO_3 , NH_4Cl , NH_4NO_3 , and KNO_3 as nitrogen sources were compared one another on the growth of *Pseudomonas* sp. P2. The optimum concentration of selected nitrogen was monitored between 250 mg/L and 2000 mg/L. The effects of 1000 mg/L of KH_2PO_4 , K_2HPO_4 , NaH_2PO_4 , and H_3PO_4 as phosphorus sources were observed in the growth of *Pseudomonas* sp. P2. The optimum concentration of selected phosphorus was monitored between 250 mg/L and 2000 mg/L. Metal ions were selected to $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, NaCl, and $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$.

The 1000 mg/L biphenyl as a carbon source was added to 30 mL liquid medium. The initial optical density of *Pseudomonas* sp. P2 was adjusted to 0.25 at 660 nm. The *Pseudomonas* sp. P2 inoculated in liquid medium was incubated at 30°C for 160 h with 150 rpm shaking incubator. The growth was monitored by measuring optical density at 660 nm with a UV/Visible spectrophotometer (Varian, Model CARY-1).

2.3 Degradation of PCBs

The insulating oil contained PCBs was obtained directly from transformer with Aroclor 1242 (Lee et al., 1979; Kim and Park, 1989). To check the influences of PCBs concentration, *Pseudomonas* sp. P2 was cultivated aerobically at 30°C with 500 mg/L, 1000 mg/L, 1500 mg/L, and 2000 mg/L of PCBs within insulating oil. The initial cell mass was adjusted to 130 mg/L. The degradation of PCBs was ob-

tained after 100 h of the log phase in the growth curve.

2.4 Analysis of PCBs

An analysis of PCBs in medium was carried out by gas chromatography in terms of perchlorination meaning that PCBs were converted to decachlorinated biphenyl (DCB). PCBs extraction from medium was accomplished with n-hexane by magnetic stirrer for 1 h. PCBs extracted from medium was perchlorinated to DCB by SbCl_5 (Huckins et al., 1974; Trotter and Young, 1975; Kim and Moon, 1995).

The concentration of PCBs was analyzed by a Varian 3300 gas chromatography equipped with 63 Ni electron capture detector (ECD). The packing material of column (2 m long \times 0.6 cm i.d.) was packed with 3% OV-17 on 80/100 chrom. whp. The column temperature was 250°C while the detector and inlet temperature were 260°C. The flow rates of N_2 as carrier gas was 60 mL/min (Schutzmann et al., 1971; Armour, 1973; Moon et al., 1995).

3. Results and Discussion

3.1 Growth Curve

Fig. 1 showed the growth curve of *Pseudomonas* sp. P2 in medium containing 1000 mg/L biphenyl as a carbon source for 160 h incubation. After a lag phase for 20 h, *Pseudomonas* sp. P2 was increased rapidly until 100 h incubation. In the log phase,

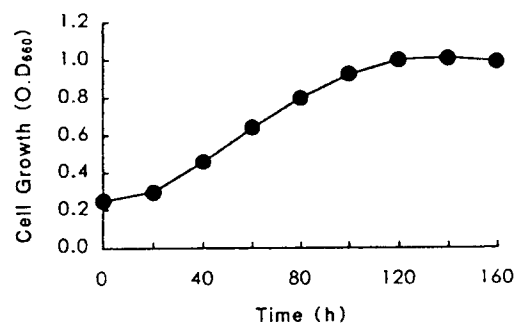


Fig. 1. Growth curve of *Pseudomonas* sp. P2 in medium containing 1000 mg/L biphenyl as a carbon source. Growth was observed by optical density at 660 nm by spectrophotometer.

Table 1. Effects of nitrogen sources on the growth of *Pseudomonas* sp. P2. Relative activity was calculated with O.D₆₆₀ after 100 h incubation

Nitrogen sources (1000 mg/L)	Relative activity (%)
NH ₄ NO ₃	100.0
(NH ₂) ₂ CO	82.1
(NH ₄) ₂ SO ₄	84.8
NaNO ₃	85.4
NH ₄ Cl	85.9
KNO ₃	80.4

the optical density of *Pseudomonas* sp. P2 was varied from 0.30 to 0.92. The maximum growth rate (μ_{max}) was calculated 0.33 day⁻¹ in the log phase. A maximum optical density 1.01 was obtained at 140 h, and a final optical density 0.99 was obtained at 160 h. For the study of optimum conditions, incubation time was determined at 100 h of the log phase in the growth curve.

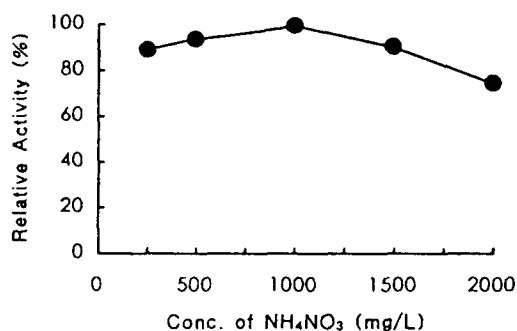


Fig. 2. Effects of NH₄NO₃ concentrations on the growth of *Pseudomonas* sp. P2. Relative activity was measured at O.D₆₆₀ after 100 h incubation.

3.2 Effects of Nitrogen Sources

The batch experiments were conducted to study the effects of nitrogen sources and nitrogen concentrations on the growth of *Pseudomonas* sp. P2. The effects of 1000 mg/L of NH₄NO₃, (NH₂)₂CO, (NH₄)₂SO₄, NaNO₃, NH₄Cl, and KNO₃ were shown in Table 1. The optimum nitrogen source was selected to be NH₄NO₃ shown maximum growth. Fig. 2 showed the effects of the concentrations of NH₄

Table 2. Effects of phosphorus sources on the growth of *Pseudomonas* sp. P2

Nitrogen sources (1000 mg/L)	Relative activity (%)
KH ₂ PO ₄	100.0
K ₂ HPO ₄	97.6
NaH ₂ PO ₄	88.9
H ₃ PO ₄	98.7

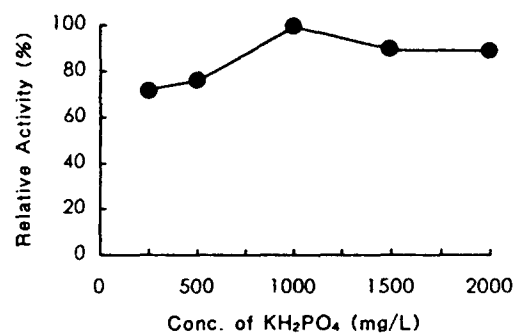


Fig. 3. Effects of KH₂PO₄ concentrations on the growth of *Pseudomonas* sp. P2.

NO₃ as the nitrogen source. The optimal NH₄NO₃ concentration was decided on 1000 mg/L on the growth of *Pseudomonas* sp. P2.

3.3. Effects of Phosphorus Sources

The effects of phosphorus sources and phosphorus concentrations on the growth of *Pseudomonas* sp. P2 were examined in the batch experiments. The effects of phosphorus sources were examined in 1000 mg/L of KH₂PO₄, K₂HPO₄, NaH₂PO₄, and H₃PO₄. As shown in Table 2, the optimum phosphorus source was revealed to be KH₂PO₄. Therefore the optimal KH₂PO₄ concentration was investigated between 250 mg/L and 2000 mg/L. In Fig. 3, the optimal KH₂PO₄ concentration was confirmed to be 1000 mg/L on the growth of *Pseudomonas* sp. P2.

3.4. Effects of Metal Ions

To improve the growth of *Pseudomonas* sp. P2, the effects of metal ions were examined out as MgSO₄ · 7H₂O, CaCl₂ · 2H₂O, NaCl, and FeSO₄ · 7H₂O. Fig. 4 showed the effects of MgSO₄ · 7H₂O. The relative activities of the growth at 0 mg/L and 250 mg/L of MgSO₄ · 7H₂O were 54.3% and 77.3%,

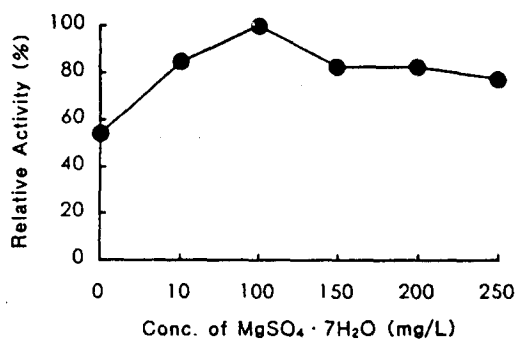


Fig. 4. Effects of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ concentrations on the growth of *Pseudomonas* sp. P2.

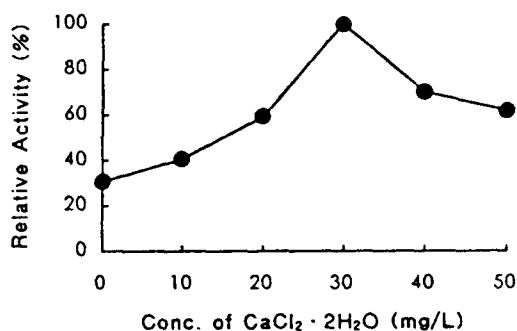


Fig. 5. Effects of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ concentrations on the growth of *Pseudomonas* sp. P2.

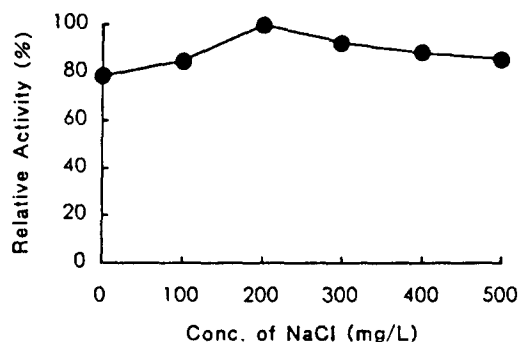


Fig. 6. Effects of NaCl concentrations on the growth of *Pseudomonas* sp. P2.

based on the criteria of 100 mg/L $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, respectively. The relative activities of the growth in case of 0 mg/L and 50 mg/L of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ were 30.6% and 61.8%, based on the criteria of 30 mg/L $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, respectively (Fig. 5). The optimum concentrations of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ and $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$

were found to be at 100 mg/L and 30 mg/L, respectively.

The relative activities of the growth at 0 mg/L and 500 mg/L of NaCl were 78.6% and 85.7%, based on the criteria of 200 mg/L NaCl, respectively. The optimum concentration of NaCl was 200 mg/L (Fig. 6). The relative activities of the growth at 0 mg/L and 25 mg/L of $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ were 37.5% and 82.5%, based on the criteria of 10 mg/L $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$, respectively. The optimum concentration of $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ was 10 mg/L (Fig. 7).

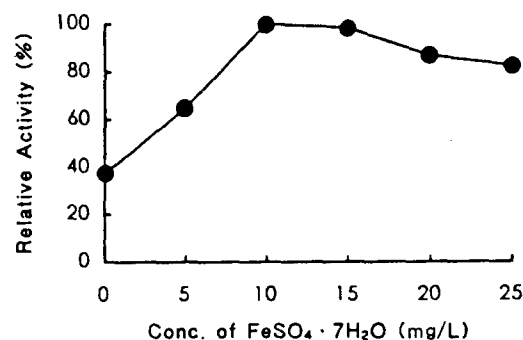


Fig. 7. Effects of $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ concentrations on the growth of *Pseudomonas* sp. P2.

Table 3. Optimal compositions of medium to degrade polychlorinated biphenyls (PCBs) by *Pseudomonas* sp. P2

Components	Concentrations (mg/L)
PCBs	1000
NH_4NO_3	1000
KH_2PO_4	1000
$\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$	100
NaCl	200
$\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$	10
$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$	30
Emulsifier ¹⁾	200
pH	7.0

1) Alkyl aryl ethoxylated phosphate as surfactant 101

3.5 Degradation of PCBs

To examine the effects of PCBs concentrations within insulating oil, *Pseudomonas* sp. P2 was cultivated in 500 mg/L, 1000 mg/L, 1500 mg/L, and 2000

mg/L of PCBs under the optimal medium compositions as indicated in Table 3.

To confirm degradability of PCBs by *Pseudomonas* sp. P2, the residual PCBs in medium was measured by gas chromatography after 100 h of the log phase in the growth curve. The chromatogram shown in Fig. 8 (A) was demonstrated by the standard decachlorinated biphenyl (DCB) with 6.5 min retention time. The DCB peaks of Fig. 8 (B) and (C) after perchlorination of samples were completely agreed with the peak of standard DCB. Com-

paring with Fig. 8 (B), the 57.6% degradation obtained at 1000 mg/L PCBs was confirmed to decrease peak height of DCB on Fig. 8 (C).

The degradation at 500 mg/L, 1000 mg/L, 1500 mg/L and 2000 mg/L of PCBs were 59.3%, 57.6%, 51.4% and 48.7%, respectively (Table 4). Choi (1996) reported that 1000 mg/L PCBs was degraded 2.0% at 40 h, 49.0% at 80 h, 60.0% at 120 h, and 60.0% at 160 h after incubation of *Pseudomonas* sp. P2. Based on Choi's report (1996), this study showed that the degradation rate of PCBs was not inhibited from 500 mg/L to 2000 mg/L of PCBs (Table 4). Similarly, Wong and Kaiser (1975) reported that inhibition of bacterial growth was not observed up to 1000 mg/L PCBs. Kim et al. (1986) reported that inhibition of PCBs degradation by *Alcaligenes aquamarinus* was occurred at 5000 mg/L PCBs.

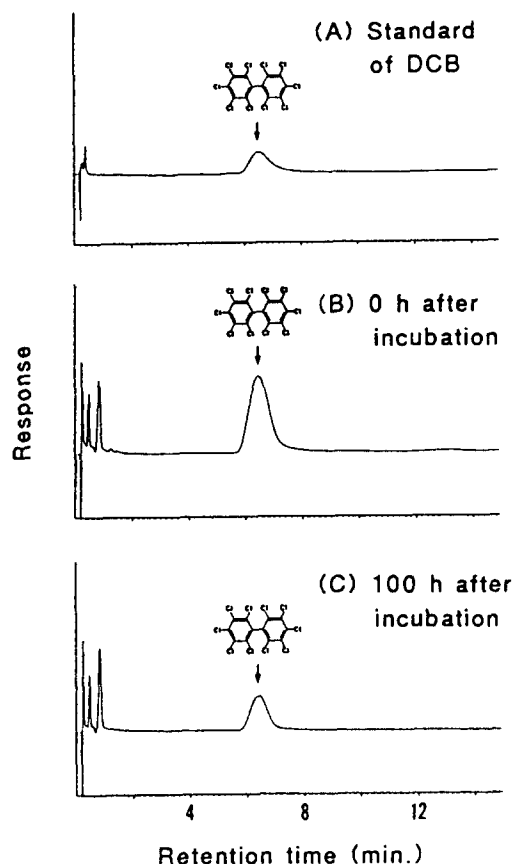


Fig. 8. GC-ECD chromatograms for analysis of polychlorinated biphenyls (PCBs). Standard decachlorinated biphenyl (DCB) in (A), perchlorinated medium after 0 h incubation in (B), and perchlorinated medium after 100 h incubation in (C).

Table 4. Degradation of polychlorinated biphenyls (PCBs) by *Pseudomonas* sp. P2 in medium containing PCBs as a carbon source after 100 h incubation

Concentrations (mg/L)	Degradation of PCBs (%)
500	59.3
1000	57.6
1500	51.4
2000	48.7

ACKNOWLEDGMENTS

This research was supported by a grant from the Korea Science and Engineering Foundation (No. 951-0606-035-2) in 1995~1996.

REFERENCES

- Armour, J.A., 1973, Quantitative perchlorination of polychlorinated biphenyls as a method for confirmatory residue measurement and identification, *J. Assoc. Off. Anal. Chem.*, 56, 987~993.
- Bedard, D.L., R.E. Wagner, M.J. Brennan, M.L. Haberl and J.F. Brown, Jr., 1987, Extensive degradation of Aroclors and environmentally

- transformed polychlorinated biphenyls by *Alcaligenes eutrophus* H850, *Appl. Environ. Microbiol.*, 53 (5), 1094~1102.
- Choi, S.-K., 1996, Chapter 1. Isolation and characterization of *Pseudomonas* sp. P2 degrading polychlorinated biphenyls (PCBs). In *Biodegradation of polychlorinated biphenyls (PCBs) by Pseudomonas* sp. P2. Master Thesis. Kyungsan Univ., 1~17pp.
- Furukawa, K., K. Tomomura and A. Kamibayashi, 1978, Effect of chlorine substitution on the biodegradability of polychlorinated biphenyls, *Appl. Environ. Microbiol.*, 35 (2), 223~227.
- Furukawa, K., N. Tomizuka and A. Kamibayashi, 1979, Effect of chlorine substitution on the bacterial metabolism of various polychlorinated biphenyls, *Appl. Environ. Microbiol.*, 38 (2), 301~310.
- Furukawa, K., N. Tomizuka and A. Kamibayashi, 1983, Metabolic breakdown of Kaneclors (polychlorobiphenyls) and their products by *Acinetobacter* sp., *Appl. Environ. Microbiol.*, 46 (1), 140~145.
- Gibson, D.T., D.L. Cruden, J.D. Haddock, G.J. Zylstra and J.M. Brand, 1993, Oxidation of polychlorinated biphenyls by *Pseudomonas* sp. strain LB400 and *Pseudomonas pseudoalcaligenes* KF707., *J. Bacteriol.*, 175 (14), 4561~4564.
- Huckins, J.N., J.E. Swanson and D.L. Stalling, 1974, Perchlorination of polychlorinated biphenyls, *J. Assoc. Off. Anal. Chem.*, 57 (2), 416~417.
- Kim, C.-K. and M.-S. Kim, 1990, Effects of environmental factors on degradation of Aroclors by gram negative bacteria, *Korean. J. Microbiol.*, 28 (2), 145~150.
- Kim, J.-H. and C.-H. Moon, 1995, Residual polychlorinated biphenyls (PCBs) in the sediment of the Kumho river, *Korean J. Environ. Agric.*, 14 (3), 272~281.
- Kohler, H.-P.E., D. Kohler-Staub and D.D. Focht, 1988, Cometabolism of polychlorinated biphenyls: Enhanced transformation of Aroclor 1254 by growing bacterial cells, *Appl. Environ. Microbiol.*, 54 (8), 1940~1945.
- May, H.D., A.W. Boyle, W. Allen Price II and C. K. Blake, 1992, Subculturing of a polychlorinated biphenyl dechlorinating anaerobic enrichment on solid media, *Appl. Environ. Microbiol.*, 58 (12), 4051~4054.
- Moon, C.-H., S.-K. Choi and J.-H. Kim, 1995, Determination of polychlorinated biphenyls in the soil by perchlorination, *J. Korean Environ. Sci.*, 4 (3), 249~258.
- Schutzmann, R.L., D.W. Woodham and C.W. Collier, 1971, Removal of sulfur in environmental samples prior to gas chromatographic analysis for pesticide residues, *J. Assoc. Off. Anal. Chem.*, 54 (5), 1117~1119.
- Shields, M.S., S.W. Hooper and G.S. Saylor, 1985, Plasmid mediated mineralization of 4-Chlorobiphenyl, *J. Bacteriol.*, 163 (3), 882~889.
- Tanabe, S., 1988, PCB problems in the future: Foresight from current knowledge, *Environ. pollut.*, 50, 5~28.
- Trotter, W.J. and S.J.V. Young, 1975, Limitation on the use of antimony pentachloride for perchlorination of polychlorinated biphenyls, *J. Assoc. Off. Anal. Chem.*, 58 (3), 466~468.
- Wong, P.T.S. and K.L.E. Kaiser, 1975, Bacterial degradation of polychlorinated biphenyls II- Rate studies, *Bull. Environ. Contamin. Toxicol.*, 13 (2), 249~255.

***Pseudomonas* sp. P2에 의한 Polychlorinated Biphenyls (PCBs) 분해에 대한 영양원의 影響**

崔尚基 · 金政鎬
慶山大學校 環境科學科
(1996년 1월 26일 접수)

유일한 炭素原으로서 Biphenyl을 包含하고 있는 液體培地에서 *Pseudomonas* sp. P2의 生育에 미치는 營養原의 效果에 關係 研究하였다. *Pseudomonas* sp. P2의 特性을 調査하기 爲한, 培養時間은 菌의 生育 曲線에서 對數成長期 末인 100時間으로 하였다. PCBs를 分解하는 *Pseudomonas* sp. P2의 生育을 爲한 最適 組成은 1000 mg/L NH_4NO_3 , 1000 mg/L KH_2PO_4 , 100 mg/L $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 30 mg/L $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 200 mg/L NaCl, 10 mg/L $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ 였다. *Pseudomonas* sp. P2에 依한 100時間 培養後 絶緣油 内에 있는 PCBs의 分解率은 500 mg/L에서 59.3%, 1000 mg/L에서 57.6%, 1500 mg/L에서 51.4%, 2000 mg/L에서는 48.7%였다.