

Tannins from *Rubus coreanum*

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Abstract—Tannins were isolated from the stems of *Rubus coreanum* and identified as (–)-epicatechin, (+)-catechin, procyanidin B-4 and sanguin H-4 by spectral analysis.

Keywords—*Rubus coreanum* · Rosaceae · tannin · flavan-3-ol · proanthocyanidin · Ellagitannin

Rubus coreanum (Rosaceae), which has been used in folk medicine as a tonic for aged people,^{1,2)} is a fruit drug growing throughout southern Korea.^{3,4)} From *Rubus coreanum* only the triterpenoids such as niga-ichigoside F₁, F₂, suavissimoside R₁, coreanoside F₁ are known.^{5)–7)} As our part of research to find the biologically active components from plant polyphenols, we first isolated two flavan-3-ols, (–)-epicatechin(1) and (+)-catechin(2), a proanthocyanidin, procyanidin B-4(3) and an ellagitannin, sanguin H-4(4) from the stems of this plant. This paper deals with the isolation and structural elucidation of these compounds.

Experimental

General Experimental Procedures - The ¹H- and ¹³C-NMR spectra were recorded at 200,500 MHz (¹H-NMR) and 50,100MHz (¹³C-NMR) with a Bruker AM-200 and Bruker AMX 500 spectrometer. Chemical shifts are given in δ (ppm) scale with TMS as internal standard. EIMS spectra were obtained with GC-MS/MS-DS, TSQ 700 (U.S.A) spectrometer. Negative FAB-MS were measured at 1.5 kv (accelerating voltage) with glycerol matrix. Chromatography was carried out on Amberlite XAD-2 (16-45mesh, Fluka), Sephadex LH-20 (25-100 μm,

Pharmacia), MCI-gel CHP-20P (75-150 μm, Mitsubishi).

TLC was carried out on silicagel 60 F₂₅₄ (Merck) and precoated cellulose F₂₅₄(Merck) in the solvent system benzene—ethyl formate—formic acid (1:7:1 v/v%) and 2% HOAc. Chromatogram was detected by illumination with U.V light and by use of four spray reagent : (i) FeCl₃, (ii) 10% - H₂SO₄, (iii) anisaldehyde - sulphuric acid, (iv) NaNO₂-HOAc.

Plant Material - The stems of *Rubus coreanum* were collected from the herbarium of Chung Ang University, on september 1993. A voucher specimen is deposited in the herbarium of college of pharmacy, Chung Ang University.

Extraction and Isolation - The fresh stems (4 kg) were extracted with 80% aq. Me₂CO at room temperature (3 times). The acetone was removed *in vacuo* and the aqueous layer was defatted with ether. The extract were chromatographed on Amberlite XAD-2 (150x520 mm) with increasing concentration of methanol in water to give 6 fractions. Repeated CC of fraction 5 on Sephadex LH-20 (78x300 mm) with 80% EtOH followed by MCI-gel CHP-20P with H₂O-MeOH gradient system gave (–)-epicatechin(850 mg), (+)-catechin(50 mg), procyanidin B-4(1.5 g) and sanguin H-4(60 mg).

Compound 1 - White needles, EIMS $[M]^+$ m/z 290, $[\alpha]_D^{20}$ -58.0° ($c = 0.1$, acetone), 1H -NMR($Me_2CO-d_6 + D_2O$) δ : 7.18 (1H, d, $J=2$ Hz, H-2'), 6.85 (1H, dd, $J=2, 8$ Hz, H-6'), 6.78 (1H, d, $J=8$ Hz, H-5'), 6.09 (1H, d, $J=2.3$ Hz, H-8), 5.92 (1H, d, $J=2.3$ Hz, H-6), 4.89 (1H, s, H-2), 4.21 (1H, m, H-3), 2.83 (1H, dd, $J=4, 16$ Hz, H-4), 2.49 (1H, dd, $J=5, 16$ Hz, H-4). ^{13}C -NMR($Me_2CO-d_6 + D_2O$): δ 79.3 (C-2), 66.8 (C-3), 28.9 (C-4), 157.5 (C-5), 95.6 (C-6), 157.5 (C-7), 96.1 (C-8), 157.0 (C-9), 99.7 (C-10), 132.2 (C-1'), 115.2 (C-2'), 145.2 (C-3'), 145.3 (C-4'), 115.5 (C-5'), 119.3 (C-6')

Compound 2 - Off-white needles, EIMS $[M]^+$ m/z 290, $[\alpha]_D^{20}$ $+12.0^\circ$ ($c=0.1$, acetone), 1H -NMR($Me_2CO-d_6 + D_2O$) δ : 6.89 (1H, m, H-2'), 6.78 (1H, m, H-5'), 6.76 (1H, m, H-6'), 6.02 (1H, d, $J=2.3$ Hz, H-8), 5.87 (1H, d, $J=2.3$ Hz, H-6), 4.55 (1H, d, $J=7.3$ Hz, H-2), 3.99 (1H, m, H-3), 2.52-2.89 (2H in total, m, H-4). ^{13}C -NMR($Me_2CO-d_6 + D_2O$): δ 82.9 (C-2), 68.6 (C-3), 29.1 (C-4), 157.5 (C-5), 96.6 (C-6), 158.0 (C-7), 95.7 (C-8), 156.0 (C-9), 99.7 (C-10), 132.3 (C-1'), 115.7 (C-2'), 146.0 (C-3'), 145.9 (C-4'), 116.1 (C-5'), 120.3 (C-6')

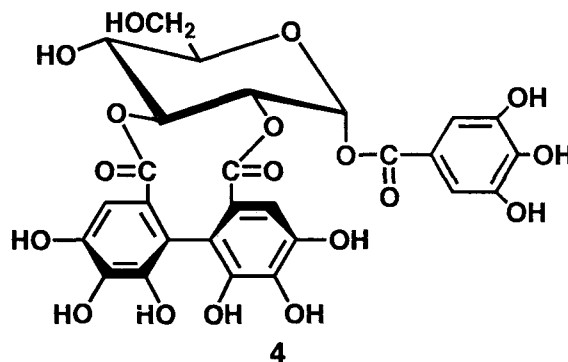
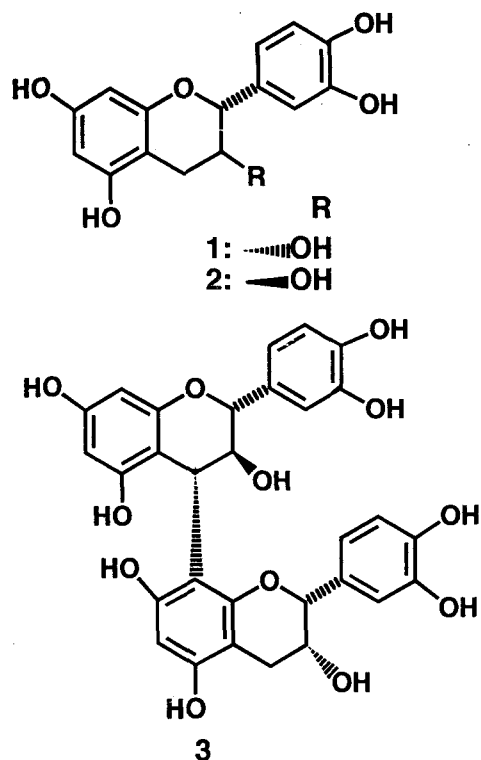
Compound 3 - Brown amorphous powder, Negative FAB-MS $[M-H]^-$ m/z 577, $[\alpha]_D^{20}$ -177.0° ($c=0.1$, acetone), 1H -NMR($Me_2CO-d_6 + D_2O$) δ : 7.27 (1H, d, $J=1.9$ Hz, H-2' (u or t)), 7.03 (1H, d, $J=1.9$ Hz, H-2' (u or t)), 6.90 (1H, dd, $J=1.9, 8.0$ Hz, H-6' (u or t)), 6.88 (1H, dd, $J=1.9, 7.9$ Hz, H-6' (u or t)), 6.84 (1H, d, $J=7.9$ Hz, H-5' (u or t)), 6.82 (1H, d, $J=8.0$ Hz, H-5' (u or t)), 6.80 (d, $J=1.9$ Hz, H-2' (conformer)), 6.76 (d, $J=1.9$ Hz, H-2' (conformer)), 6.73 (d, $J=8.1$ Hz, H-5' (conformer)), 6.69 (d, $J=8.1$ Hz, H-5' (conformer)), 6.54 (dd, $J=1.9, 8.1$ Hz, H-6' (conformer)), 6.41 (dd, $J=1.9, 8.1$ Hz, H-6' (conformer)), 6.21 (s, H-6 (conformer, t)), 6.08 (1H, s, H-6(t)), 5.97 (d, $J=2.4$ Hz, H-8 (conformer u)), 5.96 (d, $J=2.4$ Hz, H-6 (conformer u)), 5.89 (1H, d, $J=2.4$ Hz, H-8 (u)), 5.87 (1H, dd, $J=2.4$ Hz, H-6(u)), 4.99 (1H, s, H-

2(t)), 4.88 (0.5H, s, H-2 (conformer u or t)), 4.70 (1H, d, $J=7.7$ Hz, H-2(u)), 4.60 (1H, dd, $J=7.7, 9.7$ Hz, H-3(u)), 4.51 (m, H-2 (conformer u or t)), 4.47 (1H, d, $J=9.7$ Hz, H-4(u)), 4.39 (m, H-3 and H-4 (conformer)), 4.28 (1H, m, H-3(t)), 4.13 (m, H-3 (conformer)) ^{13}C -NMR($Me_2CO-d_6 + D_2O$) δ : 79.8 (C-2t), 83.3 (C-2u), 73.1 (C-3u), 66.8 (C-3t), 38.1 (C-4u), 29.0 (C-4t), 154.7-158.2 (C-5u, 5t, 7u, 7t, 9u, 9t), 97.2 (C-6u), 96.4 (C-6t), 96.8 (C-8u), 107.7 (C-8t), 106.4 (C-10u), 101.3 (C-10t), 131.2-132.1 (C-1'u and t), 114.3-120.6 (C-5'u and t, C-6'u and t), 144.8-145.7 (C-4'u and t, C-3'u and t) u: upper unit t: terminal unit

Compound 4 - Brown amorphous powder, Negative FAB-MS $[M-H]^-$ m/z 633, $[\alpha]_D^{20}$ $+99.0^\circ$ ($c=0.1$, acetone), 1H -NMR($Me_2CO-d_6 + D_2O$) δ : 7.26 (2H, s, galloyl-H), 6.81 (1H, s, HHDP-H), 6.60 (1H, d, $J=4$ Hz, glc-1), 6.53 (1H, s, HHDP-H), 5.53 (1H, t, $J=9$ Hz, glc-3), 5.27 (1H, dd, $J=4, 9$ Hz, glc-2). ^{13}C -NMR($Me_2CO-d_6 + D_2O$): δ 119.9 (galloyl C-1), 110.4 (2C, galloyl C-2, 6), 145.9 (2C, galloyl C-3, 5), 139.8 (galloyl C-4), 165.9 (galloyl-COO-), 114.3, 114.5 (HHDP C-1, C-1'), 125.9, 126.4 (HHDP C-2, C-2'), 107.2, 107.5 (HHDP C-3, C-3'), 144.3, 144.4 (HHDP C-4, C-4'), 136.1, 136.4 (HHDP C-5, C-5'), 145.1, 145.2 (HHDP C-6, C-6'), 169.3, 170.2 (HHDP-C=O), 91.0 (glc-1), 73.7 (glc-2), 76.0 (glc-3), 66.8 (glc-4), 77.9 (glc-5), 60.8 (glc-6).

Results and Discussion

Compound 1 - White needles, $[\alpha]_D^{20}$ -58.0° ($c=0.1$, acetone), gave a blue color with ferric chloride reagent and an orange-red color with anisaldehyde-sulphuric acid reagent which supposed to have a flavan-3-ol unit. The 1H -NMR spectrum of **1** exhibited a ABX type of B-ring at δ 6.78 (1H, d, $J=8$ Hz, H-5'), 6.85 (1H, dd, $J=2, 8$ Hz, H-6') and 7.18 (1H, d, $J=2$ Hz, H-2') and two meta coupled doublet signals ($J=2.3$ Hz) at δ 5.92 (H-6) and 6.09 (H-8) suggesting 5,7-dihy-



droxylation pattern of A-ring. The $^1\text{H-NMR}$ spectrum of **1** further showed a singlet signal at δ 4.89 induced by H-2, a multiplet signal at δ 4.21 by H-3 and two double-doublet signals at δ 2.83 ($J=4$, 16Hz) and 2.49 ($J=5$, 16Hz) by H-4 indicated (-)cis catechin type.^{8),9)} The EIMS spectrum showed a molecular ion peak at m/z 290. From these results, **1** was characterized as (-)-epicatechin.

Compound 2 - off white needles, $[\alpha]_D^{20} + 12.0^\circ$ ($c = 0.1$, acetone), gave positive results to FeCl_3 (dark green color) and anisaldehyde-sulphuric acid (pink) and showed almost same R_f value as (-)-epicatechin on TLC. The $^1\text{H-}^{13}\text{C-NMR}$ spectra were similar to those of **1** except one proton doublet signal at δ 4.55 ($J=7.3\text{Hz}$, H-2) indicating trans catechin.¹⁰⁾ The EIMS spectrum of **2** also showed its molecular ion peak at m/z 290. Thus **2** was elucidated as (+)-catechin.

Compound 3 - $[\alpha]_D^{20} - 177.0^\circ$ ($c=0.1$, acetone), was obtained as brown amorphous powder. This compound gave a blue color with ferric

chloride reagent and an orange-red color with anisaldehyde-sulphuric acid reagent and showed the low R_f value comparing with **1** or **2** on TLC which is typical of proanthocyanidin polymer. The negative FAB-MS spectrum of **3** exhibited $[\text{M-H}]^-$ peak at m/z 577, which suggested that this compound possesses a structural skeleton of a B-type proanthocyanidin dimer.¹¹⁾ The $^1\text{H-NMR}$ spectrum of **3** displayed complex patterns due to the existence of conformers generated by the restricted interflavan rotation.^{12),13),14)} Its distinctive H-2 terminal singlet signal at δ 4.99, a doublet signal at δ 4.70 [$J=7.8\text{Hz}$, H-2 (upper)], a double doublet signal at δ 4.60 [$J=7.8, 9.7\text{Hz}$, H-3 (upper)], and a doublet signal [$J=9.7\text{Hz}$, H-4(u)] at δ 4.47 revealed the presence of [4 \rightarrow 8]-linked upper catechin and terminal epicatechin.¹⁵⁾ And a quite similar C-2 upper signal (little effect on the C-2 resonance) at δ 83.3 compared with catechin (δ 82.9) also indicated C-4 as a trans orientation in the $^{13}\text{C-NMR}$ spectrum of **3**.¹⁵⁾

Therefore **3** was characterized as procyanidin B-4.^{14),15)} The chemical shifts of NMR data from the reference were coincident with **3**.¹⁶⁾ Direct comparison with authentic standard supported this conclusion(CO-TLC).¹⁷⁾

Compound 4 - $[\alpha]_D^{20} +99.0^\circ$ (c=0.1, acetone), was characterized as an ellagitannin by its blue colouration with ferric chloride and reddish-brown colouration with sodium nitrate acetic acid.¹⁸⁾ The ¹H-NMR spectrum of **4** showed a galloyl group at δ 7.26 (2H, s) and HHDP-H(each 1H, s) at δ 6.81 and δ 6.53. ¹H-NMR spectrum of **4** showed an anomeric proton doublet at δ 6.60 and its coupling constant ($J=4$ Hz) suggested to be α -configuration.¹⁹⁾ The ¹³C-NMR spectrum of **4** also showed one glucose moiety at δ 91.0, 77.9, 76.0, 73.7, 66.8, 60.8. The ¹H-NMR spectrum of **4** also showed 2, 3-HHDP glucose moiety from the low field shifted a double-doublet signal ($J=4, 9$ Hz, glc-2) at δ 5.27 and a triplet signal($J=9$ Hz, glc-3) at δ 5.53.^{20),21)} The negative FAB-MS spectrum showed a molecular ion peak at m/z 633. From these results, **4** was identified as sanguin H-4 unambiguously.²²⁾

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