Isolation of an Unusual Aloenin-acetal from Aloe

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Abstract—A hitherto unknown substance was isolated from the commercial sample of *Aloe arborescens*, which was shown to be the 4′, 6′-O-ethylidene-aloenín, most likely an artifact caused by hot-air drying procedure for the product preparation.

Keywords-Aloe arborescens · Liliaceae · acetal · 4', 6'-ethylidene-aloenin

Aloe arborescens Miller has not only been used as an emmenagogue, a febrifuge in pleurisy and phthisis, and a remedy for gastrointestinal disorders, constipation, burns, insects bites^{1,2)}, but also been widely used as materials of cosmetics and health foods.

Because of the wide medicinal and cosmetic uses, the leaves of *Aloe* species have been the focus of chemical studies.

The leaves of A. arborescens is known to contain a number of anthracene and chromone derivatives such as chrysophanol, aloe-emodin, barbaloin, aloesin and its cinnamoyl esters, and a phenyl- α -pyrone, aloenin^{3~5}).

Aloenin, a major constituent, has been reported to exhibit inhibitory effects on gastric juice secretion of rats⁶, on rat mast cell degranulation⁷, on histamine release from rat mast cells induced by compound 48/80 or A23187⁸) and on carrageenin-induced rat-paw inflammatory edema³, and promotion effect on hair growth in depilated mice⁹.

It was, recently, observed that a single oral administration of the methanol soluble fraction from the leaves of *Aloe* species caused a significant decrease in alcohol dehydrogenase activity in rat liver cytosol¹⁰.

As a part of our interest in alcohol metabolism modifiers, a systematic chemical investigation of the plants has been carried out. In the course of phytochemical examinations of *Aloe* species, a new compound, tentatively named aloenin X, was isolated from the ethylacetate soluble fracton of the methanol extract prepared from the commercial sample of *A. arborescens* leaves in addition to known compounds, aloe-emodin and aloenin.

The IR spectrum of aloenin X exhibited peaks at 3414(OH), 1682(conjugated ester), 1626, 1597, 1564(aromatic ring) and $1100{\sim}1000 \, {\rm cm}^{-1}$ (glycoside bond). The UV spectrum showed absorption maxima at $219.4 \, {\rm nm}(\log \, \epsilon \, 4.36)$ and $301.6 \, {\rm nm}(\log \, \epsilon \, 4.27)$, and in an alkaline solution showed a bathochromic shift by $40.2 \, {\rm nm}$. These spectra of aloenin X revealed strong similarities with those of 4-methoxy-2-pyrones¹¹⁾ and of aloenin⁶⁾.

The ¹H-NMR spectrum in DMSO-d₆ showed the presence of a methyl(δ 1.24, d, J=5.01 Hz), an aromatic methyl(δ 2.13, s), a methoxyl (δ 3.84, s), two pairs of meta-coupled aromatic

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protons(δ 5.59 and 6.17, d each, J=2.03 Hz and δ 6.40 and 6.51, d each, J=1.69 Hz), an anomeric proton(δ 4.96, d, J=7.69Hz), seven oxymethins(δ 3.19 \sim 4.71), two alcoholic OHs (δ 3.25 and 5.22, brs each) and a phenolic OH(δ 9.78, brs).

The 13 C-NMR spectrum displayed 21 carbon resonances. Two methyl carbon resonances were located at δ 19.84 and 20.39, one methoxyl carbon resonance at δ 56.13, hexose carbon resonances at δ 101.14, 79.79, 74.20, 72.94, 67.39, 65.88, respectively. Judging from the DEPT experiment, it is clear that the remaining carbon resonances are due to 5 methines and 7 quarternary carbons.

The mass spectrum of aloenin X exhibited a molecular ion peak at m/z 436 in accord with the formula $C_{21}H_{24}O_{10}$ and a fragment ion peak at m/z 248 corresponding to the aglucone.

The anomeric proton was observed as a doublet at δ 4.96 and an anomeric carbon, δ 101.14, indicating an O-glycoside.

Hydrolysis of aloenin X with 3% hydrochloric acid afforded glucose as a sugar moiety and the genin, which was found to be identical with that of aloenin by direct comparison (mmp, co-TLC, EIMS) with an authentic sample.

Appearance of a three proton doublet at δ 1.24 and a quartet at δ 4.71 assignable to ethylidene group in ¹H NMR together with its molecular weight suggested that aloenin X be an ethylidene derivative of aloenin.

In comparison of the ¹H- and ¹³C-NMR of aloenin X with those of aloenin, the signals due to the aglucone moiety of both compounds appeared at almost same positions.

C-4' and C-6' signals of aloenin X were dramatically downfield of its position of aloenin and C-3' and C-5' signals were upfield, indicating acetalation of the C-4' and C-6' oxygens of the sugar. As expected, the signals due to the sugar moiety of aloenin X appeared at

almost the same positions as those of methyl 4,6-O-ethylidene- β -p-glucopyranoside¹²⁾. The β -configuration of p-glucosyl moiety of aloenin X, was inferred by the coupling constant value of 7.69 Hz for the anomeric proton.

Therefore, the structure of aloenin X has now been established as 4-methoxy-6- $\{2-(4', 6'-O-ethylidene-)-\beta-p-glucopyranosyloxy-4-hydroxy-6-methylphenyl\}-2-pyrone.$

Although there were a few instances of isolation of ketals such as 4', 6'-O-isopropylidene aloin B from an Aloe species¹³⁾, this is the first report of isolation of an acetal from plants. Nevertheless, it would be an artifact formed during hot-air drying procedure for the product preparation but the cause for its formation remains unclear.

EXPERIMENTAL

General experimental procedures—Melting points were measured on a Mitamura-Riken apparatus and are uncorrected. IR measurements were recorded in KBr pellets on a JASCO FT/IR-5300 spectrometer and EI-MS spectra were obtained on a Hewlett Packard Model 5985 B GC/MS System. A Bruker CXP-300 spectrometer was used to record ¹H(300 MHz) and ¹³C(75.08 MHz) NMR spectra in DMSO-d₆ with TMS as internal standard and chemical shifts are reported in δ (ppm) and coupling constants are in Hz.

Plant material — Aloe arborescens sample used in this investigation was a commercial product prepared in KJM Aloe Co., Ltd.

Isolation of aloenin X—The sample(100 g) was refluxed with MeOH. The MeOH extract (21 g) was partitioned with hexane(1.7 g), CHCl₃(1.3 g), EtOAc(2.4 g), successively. The EtOAc soluble portion was subjected to column chromatography over Si gel eluting with CHCl₃ and then CHCl₃ containing increasing amounts of MeOH. Separation was monitored by TLC

and fractions containing aloenin X as a major component were combined. A further purification by flash chromatography(eluent: CHCl₃-MeOH =9:1) afforded aloe-emodin(1.8 mg), aloenin X(47.4 mg) and aloenin(2.2 mg), which were found to be pure on TLC.

Structural elucidation of known compounds.—The structures of the known compounds, aloenin, its aglucone and aloe-emodin were each confirmed by comparing their MS and NMR data with those described in the literature 14,15).

Aloenin X—Needles from MeOH, mp 246 \sim 247°; UV, IR and 1 H-NMR: see text, 13 C-NMR: see Table I.

Aloe-emodin—Needles from MeOH, mp 216 \sim 217°; ¹H-NMR (DMSO-d₆) δ: 4.62(2 H, brs, 3-CH₂), 7.30(1 H, s, H-2), 7.37(1H, d, J=7.8 Hz, H-7), 7.66(1 H, s, H-4), 7.72(1 H, d, J=7.8 Hz, H-5), 7.79(1 H, t, J=7.8 Hz, H-6), 11.95(2 H, brs, 2 x OH); EIMS m/z (rel. int., %) 270(95), 241(100).

Aloenin—Needles from MeOH, mp 147~148°; UV, λ_{max} 230, 303; (+NaOH) 247, 346 nm; ¹H–NMR (DMSO-d₆) δ : 2.13(3 H, s, 12-Me), 3.04~3.69(6 H, m), 3.84(3 H, s, 4-OMe), 4.80(1 H, d, J=7.4 Hz, H-1'), 4.96(2 H, brs, OH), 5.59 (1 H, d, J=2.2 Hz, H-3), 6.22(1 H, d, J=2.2 Hz, H-5), 6.38(1 H, brs, H-11), 6.48(1 H, d, J=2.0 Hz, H-9), 9.81(1 H, brs, OH); EIMS m/z (rel. int., %) 262(C₁₄H₁₄O₅+, 19.3) 248 (100), 151(25.5), 150(24).

Acid hydrolysis of aloenin X—Aloenin X (10 mg) dissolved in 3% HCl(5 ml) had been heated under reflux for 2 hr, the mixture was

Table I. ¹³C-NMR spectral data of aloenin X and related compounds

Totaled compounds			
Carbon No.	1	2	3
2	164.28	163.78	
3	88.00	87.91	
4	171.11	170.79	
5	104.44	104.20	
6	157.90	157.44	
7	113.85	114.85	
8	156.33	156.40	
9	100.70	100.80	
10	159.57	160.96	
11	111.30	109.21	
12	139, 30	138.95	
12-CH ₃	19.84	19.66	
4 -OCH $_3$	56.13	56.13	
10-OCH ₃ (1'-OCH ₃)		55. 15	56.7
1'	101.14	99. 31	104.1
2'	74.20	73. 17	74.2
3'	72.94	76.72	72.9
4'	79.79	69.78	79.8
5′	65.88	77.11	65.9
6'	67.37	60.73	67.7
7′	98.76		99.4
7'-CH ₃	20.39		19.6

- 1: Aloenin X at 75.08 MHz in DMSO-d6.
- 2: Methylaloenin at 75.47 MHz in DMSO-d₆. 16)
- Methyl-4, 6-O-ethylidene-β-D-glucopyranoside at 15, 08 MHz in CDCl₃-CD₃OD=4: 1, ¹²⁾

diluted with water and extracted with ether. The ether extract gave the aglucone(4 mg). The aqueous layer was neutralized and the neutral solution on concentration under reduced pressure gave p-glucose, which was confirmed by TLC analysis.

Aglucone—Amorphous solid, mp 213 \sim 214°, ¹H-NMR (DMSO-d₆) δ: 2.08(3H, s, 12-Me), 3.82(3 H, s, 4-OMe), 5.59, 6.11(1 H, d, J= 2.2 Hz, H-5), 6.16(1 H, brs, H-11), 6.23(1H, d, J=2.0 Hz, H-9), 9.59(1 H, brs, OH), 9.65 (1 H, brs, OH); EIMS m/z(rel. int., %) 248 (100), 151(40), 150(40).

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LITERATURE CITED

- Namba, T.: Colored illustrations of Wakan-Yaku, Vol. II, Hoikusha Publishing Co., Osaka, pp. 218~221(1980).
- Perry, L.M.: Medicinal Plants of East and Southeast Asia, MIT Press, Cambridge, p. 234 (1980).
- Yamamoto, M., Masui, T., Sugiyama, K., Yokota, M., Nakagomi, K. and Nakazawa, H.: Agric. Biol. Chem. 55, 1627(1991).
- 4. Kodym, A.: Pharmazie 46, 217(1991).
- Yagi, A., Kambara, T. and Morinobu, N.: Planta Med. 53, 515(1987).
- Hirata, T. and Suga, T.: Bull. Chem. Soc. Jpn.
 842(1978).
- Nakagomi, K., Oka, S., Tomizuka, N., Yama-moto, M., Masui, T. and Nakazawa, H.: Rep. Ferm. Res. Inst. Jpn. 63, 23(1985).
- 8. Nakagomi, K., Yamamoto, M., Tanaka, H.,

- Tomizuka, N., Masui, T. and Nakazawa, H.: Agric. Biol. Chem. 51, 1723 (1987).
- 9. Yamamoto, M., Sugiyama, K., Yokota, M., Maeda, Y. and Inaoka, Y.: Jpn. J. Toxicol. Environ. Health 39, 409(1993).
- Shin, K.H. and Woo, W.S.: Int. Aloe Sci. Seminar, p. 20(1993).
- 11. Ganguly, A.K., Govindachari, T.R. and Mohamed, P.A.: *Tetrahedron* 21, 93(1965).
- Conway, E., Guthrie, R.D., Gero, S.D., Lukacs,
 G., Sepulchre, A.-M., Hagaman, E.W. and
 Wenkert, E.: Tetrahedron Lett. 4879(1972).
- Graf, E., Breitmaier, E. and Alexa, M.: *Planta Med.* 40, 197(1980).
- Conner, J. M., Gray, A.I., Reynolds, T. and Waterman, P.G.: Phytochemistry 26, 2995(1987).
- Conner, J.M., Gray, A.I., Waterman, P.G. and Reynolds, T.: J. Nat. Prod. 53, 1362(1990).
- Speranza, G., Dada, G., Lunazzi, L., Gramatica,
 P. and Manitto, P.: J. Nat. Prod. 49, 800 (1986).