BULLETIN

OF THE KOREAN CHEMICAL SOCIETY

Volume 13, Number 3 June 20, 1992

BKCS 13(3) 219-344 ISSN 0253-2964

Communications

Micellization of Dipalmitoylphosphatidylcholine Vesicle by Apolipoprotein A-I Without the C-Terminal Segment

Yun Soo Bae† and Hyoungman Kim*

Department of Biological Science and Engineering, KAIST, Taejon 305-701

Received May 13, 1991

The micellization of phospholipid vesicles by a certain group of proteins is an intriguing aspect of bilayer-protein interactions and there has been an increased interest on this subject in recent years¹⁻³. In a previous communication, we compared the mode of interactions between human apolipoprotein A-I (apo A-I) and dipalmitoylphosphatidylcholine (DPPC) vesicles under the conditions of vesicular complex formation and micellar complex formation4. It was found that only the C-terminal stretch of apo A-I protein penetrates the bilayer when the lipid/protein ratio is 5000. This was the condition under which the vesicle morphology did not change upon binding with the protein. Time-dependent labeling of apo A-I protein with 3-(trifluoromethyl)-3-(m-[125I] iodophenyl) diazirine ([125I]TID) during the micellar complex formation followed by fragmentation into four segments indicated that the whole length of the polypeptide chain of apo A-I is involved in the interaction with the hydrophobic interior of the vesicle from the beginning. However, even when the differences in length of these fragments were taken into account, the C-terminal segment was more extensively labeled with [125I]TID than any other segment (Figure 6 of reference 4). These results suggest the importance of the Cterminal region for the micelle formation. If this is the case. there is a possibility that apo A-I protein from which the C-terminal segment is removed may show appreciably reduced capacity to form micellar complex. The present investi-

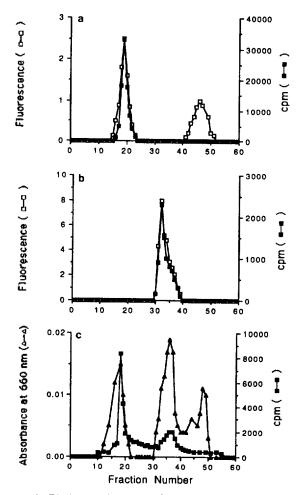


Figure 1. Elution profiles on a Sepharose CL-4B (1.8×40 cm) of DPPC/apo A-I reaction mixtures. a) DPPC/apo A-I-Dns, 5000: 1 (mol/mol), b) DPPC/apo A-I-Dns, 100:1 (mol/mol), c) DPPC/C⁻ apo A-I, 100:1 (mol/mol). Incubations were performed at 42°C for 24 h. The fraction volumes are 1.5 ml. Lipid fractions were assayed by obtaining ³H counts per min in 0.5 ml aliquots. Apo A-I-Dns fractions were monitored by means of the Dns probe fluorescence intensity. Excitation wavelength was 340 nm and emission was observed at 500 nm. C⁻ apo A-I was analyzed by Lowry method. Apo A-I-Dns: dansylated apo A-I protein.

[†] Laboratory of Immunochemistry, Genetic Engineering Research Institute, KIST, P.O.Box 17, Taeduck Science Town, Taejon 305-606.

^{*} To Whom correspondence should be addressed.

gation was undertaken to address this question.

Since the segment from Leu-189 to Arg-215 labeled with [1251]TID in vesicular complex4, our interest was to cleave off the C-terminal segment including this stretch from apo-A-I. For this purpose, apo A-I preparation was treated with 2 M hydroxylamine (1-5 mg/ml) in 6 M guanidine hydrochloride and the pH was adjusted to 9.6. This solution was incubated for 4 h at 45°C. Under this condition it is expected that the peptide bond between Asn-184 and Gly-185 is cleaved5. The reaction was terminated by introducing formic acid to bring the pH to 2-3. The C-terminal-depleted apo A-I (Capo A-I) was isolated using the gel elution method and its purity was confirmed by Laemmli SDS-PAGE⁶. Unilamellar ³H-DPPC vesicle was prepared by the reverse phase evaporation method7. The size range of the vesicle was reduced to 500-700 Å in diameter with a Heat System Sonifier cell disrupter.

Figure 1 shows the elution profiles of DPPC vesicle/apo A-I protein complexes at molar ratios of 5000 and 100, respectively, and that of DPPC vesicle/C⁻ apo A-I protein complex at the molar ratio of 100. These profiles were obtained by incubating vesicle/protein mixture for 24 h at 42°C and then passing through a Sepharose CL-4B column (1.4×40 cm). The lipid concentration was determined by liquid scintillation counting. The concentration of the apo A-I protein was monitored by measuring fluorescence intensity of dansylated protein⁸. The C⁻ apo A-I concentration was obtained by the Lowry method⁹.

Figure 1c, when compared with Figure 1a and b, shows that C⁻ apo A-I is present in the vesicular complex as well as in the micellar complex. This may, in turn, mean that the C-terminal section of apo A-I is not indispensable for breaking down the vesicles. The reduced micellization capability of C⁻ apo A-I as compared to that of intact protein may be simply due to the decreased length of the polypeptide chain.

In view of an earlier obsevation that only the C-terminal segment of apo A-I protein interacts with the vesicles when the lipid/protein value is large⁴, the appreciable binding of C⁻ apo A-I to the vesicles as shown in Figure 1c is somewhat unexpected. The only explanation we have now is that parts other than the C-terinal region, which remains attached to the vesicles when digested with trypsin, also initially bind to the vesicle. Further digestion experiments with C⁻ apo A-I protein is required to shed light on this problem.

Acknowledgement. Supported in part by the Korea Science and Engineering Foundation which is gratefully appreciated.

References

- A. Jonas, S. M. Drengler, and B. W. Patterson, J. Biol. Chem. 255, 2183 (1980).
- A. Jonas and S. M. Drengler, J. Biol. Chem. 255, 2190 (1980).
- 3. J. W. Lee and H. Kim, EEBS Lett., 241, 181 (1988).
- 4. Y. S. Bae and H. Kim, J. Biochem., 106, 1019 (1989).
- P. Bornstein and G. Balian, Methods in Enzymology, 47, 132 (1977).
- 6. U. K. Laemmli, Nature 227, 680 (1970).
- 7. F. Szoka Jr. and M. Papahadjopoulos, Proc. Natl. Acad.

Sci. USA, 91, 4194 (1978).

- 8. A. Jonas, Biochim. Biophys. Acta, 393, 471 (1975).
- O. H. Lowry, N. J. Roseberg, A. L. Fart, and R. J. Randall, J. Biol. Chem., 193, 265 (1951).

Reaction of $arachno-S_2B_7H_8^-$ with $(CO)_5Cr^{\dagger}C$ $(OCH_3)R^{\dagger}$: Synthesis and Characterization of $arachno-4-RCH_2-6,8-S_2B_7H_8$ $(R=CH_3, IIa; C_6H_5, IIb)$.

Hee-Joo Jeon, Jae-Jung Ko[†], Si-Joong Kim[‡], Doo-Soon Shin[‡], and Sang Ook Kang^{*}

Department of Chemistry, College of Natural Sciences, Korea University, Chochiwon 339-700

[†]Department of Chemical Education, Korea National University of Education, Chung-buk 363-791

[‡]Department of Chemistry, Korea University, Seoul 136-701

Received November 18, 1991

Arachno-S₂B₇H₈⁻ has been shown to react with a variety of polarizable organic compounds¹ such as nitriles and ketones to generate the corresponding hypho-CH₃CNS₂B₇H₈⁻² and hypho-S₂B₆H₉, ⁻³ respectively.

$$S_2B_7H_8^- + CH_3CN \rightarrow hypho-CH_3CNS_2B_7H_8^-$$
 (1)

$$S_2B_7H_8^- + (CH_3)_2CO \rightarrow hypho-S_2B_6H_9^-$$
 (2)

The result of the reactions above suggests that the *arachno*- $S_2B_7H_8^-$ anion might also readily attack other polarized multiple bonds. We have found that arachno- $S_2B_7H_8^-$ anion readily reacts with Fisher-type carbene complexes at room temperature. In contrast to the reactions with nitriles and ketones, cage addition results in the production of new alkyl substituted thiaboranes, arachno-4-RCH₂-6,8- $S_2B_7H_8$ (R=CH₃ IIa: C₆H₅ IIb), in good yield.

In a typical experiment, a solution of Na⁺S₂B₇H₈⁻ was prepared by the reaction in vacuo of excess NaH(~0.1 g, 4.2 mmol) with arachno-6,8-S₂B₇H₉⁵ (0.45 g, 3 mmol) in tetrahydrofuran (~25 mL) at ~-20°C. To this solution 0.80 g (3.2 mmol) of (CO)₅Cr¹C(OCH₃)CH₃|⁶ in THF was added at ~78°C and allowed to warm slowly to room temperature and continued to stir overnight. The solution gradually turned dark green, suggesting the formation of a chromathiaborane complex. Protonation with HCl followed by extraction with hexane gave a reddish-yellow solid. Subsequent separation was performed by flash chromatography with hexane to give 0.18 g (1.01 mmol) of arachno-4-CH₃CH₂-6,8-S₂B₇H₈. IIa. This corresponds to a 34% yield based on consumed arachno-6,8-S₂B₇H₉.

In an analogous reaction, 0.45 g (3 mmol) of arachno-6,8- $S_2B_7H_9$, ~0.1 g (4.2 mmol) of NaH, and 1.0 g (3.2 mmol) of (CO)₅Cr † C(OCH₃)C₅H₅ † ? were reacted in ~30 mL of THF in vacuo. The reaction mixture was initially warmed to -20 $^{\circ}$ C whereupon the solution also gradually turned dark green.