

Synthesis of Steroidal Nitrosoureas as Antitumor Activity

Jack C. Kim, Soon-Kyu Choi and Sung-Hwan Moon

Department of Chemistry, Pusan National University, Puan 607, Korea

(Received October 16, 1986)

Abstract □ Steroidal nitrosoureas have been synthesized and their antitumor activity on L1210 cells was evaluated. N-(2-Chloroethyl)-N-nitrosocarbamoyl-3-aza-A-homo-5 α -cholestane (5a) showed significantly low ED₅₀ value of 1.6 μ g/ml whose activity is equivalent to that of methyl-CCNU (ED₅₀ = 1.7mg/ml).

Keywords □ Steroidal nitrosoureas, Antitumor activity on L1210 cells, Chloroethyl- and methylnitrosourea analogs of 3-aza-A-homo-5 α -cholestane, ED₅₀.

Several steroid derivatives with attached alkylating moieties have been found to be active against selected animal tumor systems. These include the cholesterol derivative, phenesterin (1) (NSC-104469) and two estradiol derivatives, estradiol mustard (2) (NSC-112259) and estracyt (3) (NSC-89199) which have been used in clinical trial. A steroid nitrosourea agent has also been synthesized and demonstrated to be active against the growth of the DMBA-induced transplantable rat mammary tumor 13762 (4).

Based on the antitumor activity of these compounds, we have synthesized chloroethyl- and methylnitrosourea analogs of 3-aza-A-homo-5 α -cholestane.

EXPERIMENTAL METHODS

All melting points were uncorrected. IR spectra were recorded on a Perkin-Elmer 710 B spectrophotometer. NMR Spectra were obtained with a Varian EM-360 A using CDCl₃ as a solvent unless specified otherwise and tetramethylsilane as internal standard. Thin layer chromatography plates (3 \times 9cm⁻¹) were made with slurry medium of 30g of silica gel G. Type 60 and 100 ml of CHCl₃:CH₃OH (2:1, v/v) and their chromatograms were developed with C₆H₆:CH₃OH (8:2, v/v).

3-Aza-A-homo-5 α -cholestan-3-one (2)

A mixture of 5 α -cholestan-3-one oxime (5g, 1,24 mmol) and triphenyl phosphine (7,25g, 10 mmol) in 100 ml of CCl₄ was refluxed for 4 h. The mixture was evaporated *in vacuo* and crystallized from MeOH (charcoal) to afford **2** (3,8g, 88%

yield) as a colorless crystal. mp 275-277°C, lit. (6) mp 277,5-277,6°C; IR (KBr) 1650cm⁻¹ (lactam C=O). Anal. (C₂₇H₄₇NO) C,H,N.

3-Aza-A-homo-5 α -cholestane (2)

A mixture of lactam **2** (3g, 7,3 mmol) and LiAlH₄ (3g, 8,0 mmol) in dry dioxane (450 ml) was refluxed for 48 h. The excess LiAlH₄ was decomposed with H₂O, and filtered. The filtrate was extracted with CHCl₃, and the extract was washed (water and brine), dried (MgSO₄), and evaporated to dryness. Crystallization from acetone furnished **3** as a colorless solid (2,2g, 78% yield); mp 132-134°C; NMR (CDCl₃, δ ppm) 4,3(m, 1H, NH), 2,6-3,5(m, 4H, C-2, C-3), 0,65-2,0(m, steroid-H), Anal. (C₂₇H₄₉N) C,H,N.

N-(2-Chloroethyl) carbamoyl-3-aza-A-homo-5 α -cholestane (4a)

To a solution of **3** (0,8g, 2 mmol) in dry CHCl₃ (15 ml) was added 2-chloroethyl isocyanate (0,4 ml, 5 mmol) for a period of 3 h, at room temperature. The solvent was evaporated *in vacuo* to an oily residue which was chromatographed on alumina to yield pure **4a** as an oily product (0,7g, 69% yield); IR (neat) 1665cm⁻¹ (urea C=O), NMR (CDCl₃, δ ppm) 4,9(br s, H, NH), 3,1-3,5(m, 4H, C-2, C-3), 3,5-3,8(m, 4H, ClCH₂CH₂N-), 0,65-2,0(m, steroid-H). Anal. (C₃₀H₅₃ClN₂O)C,H,N. **N-Methylcarbamoyl-3-aza-A-Homo-5 α -cholestane (4b)**

Methylisocyanate (3 ml, 5,1 mmol) was added to a solution of **3** (0,8g, 2 mmol) in dry CHCl₃ (15 ml) and reacted at room temperature for 3 h. The reaction mixture was evaporated to dryness

under reduced pressure at a temperature not exceeding 40°C, and the residue was then crystallized from CH₃OH to give **4b** as white solids (0.75g, 83% yield), mp 158–60°C; IR(KBr) 1620cm⁻¹ (urea C=O); NMR (CDCl₃, δ ppm) 2.8 (d, 3H, NMe), 4.2 (q, 1H, NH), 0.65–2.0 (m, steroid-H). Anal. (C₂₉H₅₃N₂O) C, H, N.

N-(2-Chloroethyl)-N-nitrosocarbamoyl-3-aza-A-homo-5α-cholestane (5a)

Sodium nitrite (0.5g, 7 mmol) was added slowly to an ice-cold solution (-5°C) of **4a** (0.5g, 1 mmol) in 20 ml of glacial acetic acid. The reaction mixture was stirred at 0°C for 3h, and it was then poured into ice-water and extracted with CHCl₃. The extract was washed (H₂O and brine), dried (MgSO₄), and evaporated to oily residues which were chromatographed on alumina to afford pure **5a** as an oily product. (0.48g, 70% yield). IR (neat) 1690cm⁻¹ (urea C=O); NMR (CDCl₃, δ ppm) 4.6–4.9 (t, 2H, CH₂Cl), 3.85–4.15 (t, 2H, -CH₂-N-NO), 0.65–2.0 (n, steroid-H). Anal. (C₃₀H₅₂ClN₃O₂) C, H, N.

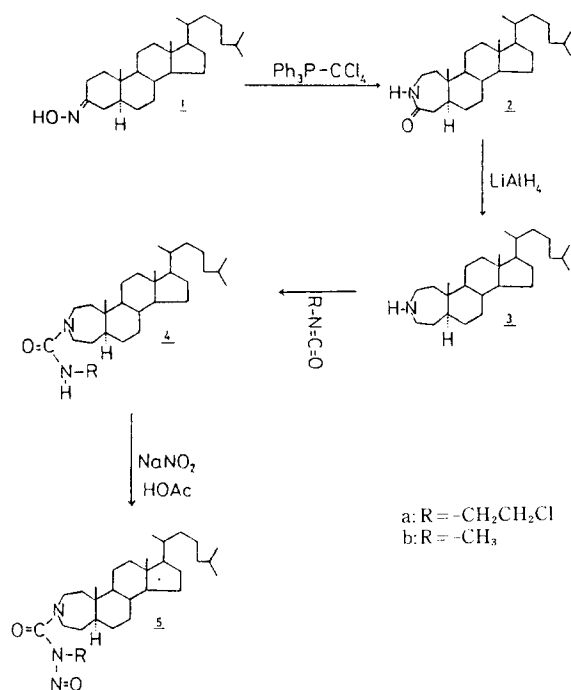
N-Methyl-N-nitrosocarbamoyl-3-aza-A-homo-5α-cholestane (5b)

Following the general procedure described above, reaction of sodium nitrite (0.5g, 7 mmol) and **4b** (0.5g, 1 mmol) in 20 ml of glacial acetic acid furnished **5b** (0.46g, 87% yield); mp 101–102°C; IR(KBr) 1700cm⁻¹ (urea C=O); NMR (CDCl₃, δ ppm) 3.1 (s, 3H, N-Me), 0.65–2.0 (m, steroid-H). Anal. (C₂₉H₅₁N₃O₂) C, H, N.

RESULTS AND DISCUSSION

N-(2-Chloroethyl)-N-nitrosocarbamoyl- and N-methyl-N-nitrosocarbamoyl groups substituted at the 3-aza-A-homo-5α-cholestane were prepared by a series of conversions starting from 5α-cholestan-3-one oxime (**1**) (Scheme 1). Beckmann rearrangement of the oxime **1** with a carbon tetrachloride-triphenyl phosphine combination gave 3-aza-A-homo-5α-cholestan-3-one (**2**) (**5**) in moderate yield. Reduction of the lactam **2** by LiAlH₄ afforded 3-aza-A-homo-5α-cholestan-3-amine (**3**) (**6**) which was reacted with 2-chloroethyl isocyanate and methyl isocyanate to yield N-(2-chloroethyl) carbamoyl-3-aza-A-homo-5α-cholestan-3-amine (**4a**) and N-methyl-carbamoyl-3-aza-A-homo-5α-cholestan-3-amine (**4b**), respectively.

Nitrosation of the unsymmetrical 1,3-disubstituted ureas, **4a** and **4b**, can theoretically give two isomeric nitrosoureas. However, sterically bulky 3-aza-A-homo-5α-cholestan-3-amine moiety made the nitrosation (under 99% HCOOH and dry NaNO₂) regioselective to yield exclusively the N



Scheme 1

-(2-chloroethyl)-N-nitrosocarbamoyl-3-aza-A-homo-5α-cholestan-3-amine (**5a**) and N-methyl-N-nitrosocarbamoyl-3-aza-A-homo-5α-cholestan-3-amine (**5b**). The purity of a nitrosourea has been shown by Montgomery, *et al.* (7) to be most clearly established by NMR spectroscopy. The spectral asymmetry of the -N(NO)-CONHCH₂CH₂Cl (A₂B₂X system) group due to the NH coupling of the adjacent methylene group can be clearly distinguished from the spectral symmetry of the -NHCON(NO)CH₂-CH₂Cl (A₂B₂ system) group.

Biological Data

Antitumor activity of compounds, **5a** and **5b**, was evaluated against the murine leukemic lymphoblast L1210 cell and as a positive control test methyl-CCNU was also applied to the present testing cells of L1210. The growth ratio for each dose of testing substance, Y was calculated by the following

$$\frac{T - C_0}{C - C_0} \times 100 = Y (\%)$$

where T = mean cell count for each dose of test substance after 48 hours incubation; C = mean cell count for control after 48 hours incubation; C₀ =

Table I. Antitumor Activity of Steroidal Nitrosoureas.

Comp. No.	<u>5a</u>	<u>5b</u>	methyl-CCNU
ED ₅₀ ($\mu\text{g/ml}$)	1.6	5.4	1.7

mean cell count at the start of incubation. When Y values were plotted against doses of methyl-CCNU semilogarithmically, a straight line could be obtained; a concentration of methyl-CCNU which could inhibit the growth of L1210 cells by 50% (ED₅₀) was estimated as 1.7 mg/ml. The synthesized compounds 5a and 5b, showed ED₅₀ values in Table I. The compound 5a showed significantly low ED₅₀ of 1.6 mg/ml.

LITERATURE CITED

1. Vollmer, E.P., Taylor, D.J., Masnyk, I.J., Cooney, D., Levine, B., Piczak, C. and

- Trench, L.: *Cancer Chemother. Rep.* part 3, **4**, 103(1973).
2. Wall, M.E., Abernethy, F.I., Carroll, D.J. and Taylor, D.J.: *J. Med. Chem.*, **2**, 810(1969).
3. Kirdani, R.Y., Muntzig, J., Varkarakis, M.J., Murphy, G.O. and Sandberg, A.A.: *Cancer Res.*, **34**, 1031(1974).
4. Lam, H.Y., Begleiter, A. and Goldenberg, G. J.: *J. Med. Chem.*, **22**, 200(1979).
5. Kim, J.C., Choi, S.K., Park, W.W. and Lee, Y. T.: *J. Korean Chem. Soc.*, **22**, 42(1979).
6. Shoppee, C.W., Kruger, G. and Mirrington, R. N.: *J. Chem. Soc.*, 1050(1962).
7. Johnston, J.P., McCaleb, G.S., Opliger P.S. and Montgomery, J.A.: *J. Med. Chem.*, **9**, 892(1966).
8. Klein, I.: *Tetrahedron Lett.*, 4307(1973).
9. Silverstein, R.M., Bossler, G.C., Morrill, T.C.: "Spectrometric Identification of Organic Compounds," John-wiley, New York, p.168, (1974).