

# A New Phenolic Aldehyde from the Seeds of *Phytolacca americana*

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商陸種子의 新揮 醜성 알데하이드

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Caffeic aldehyde was isolated from the seeds of *Phytolacca americana*. This is the first reported occurrence of the compound in nature.

## Introduction

The genus *Phytolacca* has been extensively investigated<sup>1,2)</sup> and previous works on the seeds of *Phytolacca americana* have yielded 3-acetylaleuritic acid<sup>3)</sup> and a new lignan, americanin A<sup>4)</sup>. We now wish to report the isolation of a new naturally occurring phenolic aldehyde from the seeds of *Phytolacca americana*.

The compound, C<sub>9</sub>H<sub>8</sub>O<sub>3</sub>, mp 203~4°, M<sup>+</sup>(m/e) 164, gave a green colouration in FeCl<sub>3</sub> reaction and its IR spectrum showed the presence of OH (3240cm<sup>-1</sup>), α,β-unsaturated CO(1655cm<sup>-1</sup>), trans-double bond (1615 and 980cm<sup>-1</sup>) and an aromatic system (1595, 1530 and 1449cm<sup>-1</sup>). The formation of a di-O-methyl ether, mp 81~3° and a diacetate, mp 115° indicated the presence of two phenolic groups. The NMR spectrum of the diacetate showed a singlet (6H) at δ 2.31, a double doublets centered at 6.63(1H, J=8 and 16Hz, a doublet at 7.4 (1H, J=16 Hz), a doublet at 9.64 (1H J=8Hz) and signals due to three aromatic protons at from 7.20 to 7.39.

These evidences clearly suggested that this compound is 3,4-dihydroxycinnamic aldehyde. This suggestion was confirmed by the formation of 3,4-dimethoxybenzoic acid and 3,4-dimethoxycinnamic acid(methylester) from the dimethyl ether by oxidation with KMnO<sub>4</sub> and Ag<sub>2</sub>O, respectively.

To our knowledge this compound has not been previously reported in the literature.

## Experimental

**Isolation of caffeic aldehyde** The acetone extract of the seeds of *P. americana* was extracted with CHCl<sub>3</sub> followed by H<sub>2</sub>O. The water soluble fraction was chromatographed over Silica gel using 3% MeOH-CHCl<sub>3</sub> as an eluent and crystallized from MeOH to give brown powdery crystals of caffeic aldehyde, mp 203~4°, UV<sub>EtOH</sub><sup>max</sup>nm(log ε): 226(4.05), 242(3.97), 255(4.00), 350(4.34), with AlCl<sub>3</sub> 226(4.04), 387(4.22), with AlCl<sub>3</sub>+HCl 226(4.09), 240(4.04), 255(4.01), 350(4.27); with H<sub>3</sub>BO<sub>3</sub>/NaAc 267(4.06), 321~31(3.99), 376(4.42). MS m/e(%): 164(M<sup>+</sup>, 61),

147(22), 136(28), 118(22), 110(51), 89(100), 77(87), 63(96), 53(93), 39(87), 29(32). Anal. Calcd for  $C_9H_8O_3$ : C, 65.85; H, 4.91. Found; C, 65.82; H, 4.70.

**Acetylation of caffeic aldehyde** Caffeic aldehyde (100mg) was acetylated with  $Ac_2O$  (2ml) and pyridine (1ml) in the usual way.

The reaction product was crystallized from benzene-ether (4:1) to give yellowish prisms, mp  $115^\circ$ ,  $IR_{\nu_{max}}$ : 1757 and  $1200\text{cm}^{-1}$  (acetate), no OH band.

Anal. Calcd for  $C_{13}H_{12}O_5$ : C, 62.92; H, 4.85. Found: C, 62.88; H, 4.84.

**Methylation of caffeic aldehyde** To a solution of caffeic aldehyde (30mg) in acetone (20ml) was added 1ml of  $(CH_3)_2SO_4$  and 0.5g of  $K_2CO_3$ . The mixture was refluxed on the steam bath for 3hr and filtered. The filtrate was evaporated, and extracted with ether. The ether layer was washed with  $NaHCO_3$  and then  $H_2O$ , dried and evaporated. The residue was crystallized from MeOH to give dimethylether, mp  $81-3^\circ$ . (Lit.<sup>5)</sup>, mp  $82-83^\circ$ ). NMR ( $CDCl_3$ )  $\delta$ : 3.96(s,  $2 \times Meo$ ), 6.63(dd, 1H,  $J=8$  and  $17\text{Hz}$ ), 6.8~7.25 (3H, aromatic H), 7.46(d, 1H,  $J=17\text{Hz}$ ) 9.77(d, 1H,  $J=H_z$ ).

**$KMnO_4$ -oxidation of dimethylether** To a solution of dimethylether (50mg) in acetone (10ml) was added 1ml of 10% KOH and then an acetone solution of  $KMnO_4$  dropwise at room temperature until colour of  $KMnO_4$  did not disappear. After degrading an excess of

$KMnO_4$  by addition of EtOH, the reaction mixture was filtered and chromatographed over Silica gel using benzene-ether (4:1) as an eluent to give 3,4-dimethoxybenzoic acid, mp  $174-5^\circ$ , which was identical with an authentic sample by direct comparison (TLC, mmp).

**$Ag_2O$ -oxidation of dimethylether** To a suspension of  $Ag_2O$  (1g) in 10% NaOH (20ml) was added a sample (100mg) and stirred for 10min at  $60^\circ$ , and filtered. The filtrate was acidified with HCl, extracted with ether, and the ether solution was washed, dried and methylated with  $CH_2N_2$  in an usual way to give 3,4-dimethoxycinamic acid methylester, mp  $60^\circ$ , which was identical with an authentic sample.

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