Sterols and Triterpenoids from Codonopsis lanceolata

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Abstract—The roots of Codonopsis lanceolata contained α -spinasterol, Δ^7 -stigmasterol, oleanolic acid, echinocystic acid, and an unidentified triterpene acid, mp 249°.

Codonopsis lanceolata BENTH et HOOKER (Campanulaceae) is a perennial herb, roots of which are known as both edible and medicinal plant, being used as an expectorant. Although anti-inflammatory activity of this plant was reported¹⁾, these active principles have not been known yet. In this paper, isolation and idenification of sterols and triterpenes in this plant are described.

Sterols—Phytosterol mixture, mp 136°, was isolated from unsaponifiable fraction of the ether soluble extract. It showed positive Lieberman-Burchard reaction (very rapid color change from pink to blue green). Its IR absorption spectrum shows OH band at 3400cm^{-1} and absorption bands due to double bond at 1650, 970 (trans-disubstituted double bond) and $825 \sim 850 \text{cm}^{-1}$ (trisubstituted double bond). Although this substance showed single spot on silica gel plates with or without AgNO₃ and the spot was superimposible on that of the authentic sample of α -spinasterol, its mass spectrum exhibits another parent peak at m/e 414 besides peaks which are characteristic of α -spinasterol². GLC analysis conformed that this substance was a mixture of α -spinasterol and Δ -stigmastenol (27:73).

Triterpenoids—Three substances(compound A~C) were obtained from the hydrolysate of the crude saponin.

Compound A, mp 310°, was identified to be oleanolic acid by direct comparison with authentic sample (mixed mp, TLC, IR, and MS), and preparation of methyloleanolate(methylation with CH₂N₂) and 3-acetyloleanolic acid (acetylation with Ac₂O and pyridine).

Compound B(I), C₃₀H₄₈O₄·H₂O, mp 303~308°, showed positive Lieberman-Burchard reaction and tetranitromethane test. Its IR spectrum shows OH band at 3560cm⁻¹, carboxyl

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band at 1700 cm⁻¹, and trisubstituted double bond at 810~830 cm⁻¹. Acetylation of I with acetic anhydride-pyridine gave a diacetate(II), mp 271~275°. I was esterified on treatment with ethereal diazomethane to give a methyl ester(III), mp 206~208°, which was hydrolysed by 4% KOH to give I(30%).

Methylation of II or acetylation of III gave equally a methyl diacetate(IV), mp 197 \sim 199°, which consumed one mole of perbenzoic acid. Oxidation of IV with CrO₃-HOAc afforded an α,β -unsaturated ketone(V), mp 220 \sim 223° and treatment of III with Br₂-HOAc gave a bromo- γ -lactone(VI), mp 141 \sim 145°. On the basis of the above experiments, this compound is supposed to be a dihydroxy-olean-12-en-28-oic acid.

Mass spectra of I, II, and III show typical retro-Diels-Alder fragmentation and the rest of the peaks due to further cracking are similar to the usual cascading patterns expected for olean-12-ene series³⁾. Mass spectra analysis together with the saponification rate* of carbomethoxyl group of III suggest that one of hydroxyl groups should be located either at C-16 or C-22.

The IR spectrum of III exhibits strong bands at $1730 \,\mathrm{cm^{-1}}$, showing no intramolecular hydrogen bonding between hydroxyl and carbonyl groups^{4,5)}, thus suggesting the presence of an axial hydroxyl group at C-16 in I. The presence of a second hydroxyl group at C-3 is more probable from biogenetic ground. Its equatorial(β) oriention is supported by the presence of a triplet-like multiplet(1H) centered at δ 4.50 in the NMR spectrum of IV with splitting pattern characteristic of the proton α to the 3β -acetoxyl group⁶⁻⁹³. Observed values in PPM(0.72, 0.86×2, 0.94, 0.98×2, and 1.25) of the chemical shifts for the C-methyl groups of IV are in excellent agreement with the values (0.69, 0.87×2, 0.95, 0.98, 0.99, and 1.26 for Me 26, 23, 24, 25, 29, 30, and 27, respectively) calculated for 3β , 16α -diacetoxyolean-12-en-28-oic acid methyl ester based on the data presented by Tursch, et al. 10 and Ito, et al. 111 The value of the width at half-height of signal at 5.63 is ca 6 Hz, being in agreement with the assigned stereochemistry of an acetoxyl group as 16α .

Although the direct comparison with authentic sample has been unavailable, the accumulated evidences for compound B clearly assure its identity with echinocystic acid.

Compound C, mp 249 \sim 251°, (M⁺, m/e 472), was assumed to be an isomer of compound B, however, further examination was not undertaken due to shortage of the material.

EXPERIMENTAL*

Extraction and Hydrolysis of Crude Saponin-MeOH extracts of the dried roots

^{*} It is known that the rate of saponification of the triterpene esters of the oleanane series having the carbomethoxyl at the C-17 position is very slow but the presence of the hydroxyl group β to the carbomethoxy group facilitates the hydrolysis¹²³.

^{**} The melting points were determined in open capilaries in sulfuric acid and are uncorrected. Optical rotations were measured in MeOH. IR spectra were recorded in KBr and UV spectra in EtOH. NMR spectra were taken in CDCl₃ using TMS as internal reference at 100 MHz. We are grateful to Dr. D.Y. Han, College of Pharmacy, Jung Ang University, for the measurements of the mass spectra.

(10 kg) were partitioned between Et₂O and water, and water layer was extracted with BuOH. After removal of solvent, the crude saponin was precipitated by addition of Et₂O into MeOH soln. of the residue(yield 15g). The crude saponin(3g) was hydrolyzed by refluxing in 5% MeOH-H₂SO₄ for 5hr. The aglycone fraction(1g) precipitated by addition of water was chromatographed on silica gel and elution with MeOH-CHCl₃ (1:100 to 1:20) gave 3 components, which have been designated as compound A, B, and C in order of increasing polarity on TLC (CHCl₃-MeOH-7% HOAc=5:1:1).

Compound A, mp 310°, IR 3400 cm⁻¹(OH), $1680 \text{ cm}^{-1}(\text{COOH})$, and $825 \text{ cm}^{-1}(\text{trisubstituted double bond})$. MS m/e 456(M⁺ 1.7), 248(100), 207(40.7), 203(98), 189(25.9), and 133(37), was identified as oleanolic acid by direct comparison with the authentic specimen.

Methylation with CH₂N₂ and acetylation with Ac₂O and pyridine of compound A gave a methylester, mp 198~200° and an acetate, mp 290~300°, which did not depress the mps of authentic samples of methyl oleanolate and acetyl oleanolic acid, respectively.

Compound B(I), mp 303~308°, [a]_D=+15.7° (c, 0.22 in MeOH), $\lambda = 204$ nm (log ϵ , 3.80), IR 3560 cm⁻¹(OH), 1700 cm⁻¹(COOH), and 810~830 cm⁻¹ (trisubstituted double bond); NMR δ 0.77(s, 3H), 0.79(s, 3H), 0.88(s, 3H), 0.97(s, 9H), 1.38(s, 3H), 3.27~3.33 (m, 1H), and 5.5(m, 1H). MS m/e 472(M+25), 264(52), 246(100), 219(17), 207(52), 201(67), 189(48), 149(9.8), and 131(25).

Anal. Calcd for $C_{30}H_{48}O_4 \cdot H_2O$: C, 73. 47; H, 10. 20. Found: C, 73. 98; H, 10. 28. Compound C, mp $249\sim251^{\circ}$, IR $3540 \,\mathrm{cm}^{-1}$ (OH) and $1698 \,\mathrm{cm}^{-1}$ (COOH), MS m/ϵ 472 (M⁺).

Acetylation of I—A sample of I (30 mg) was acetylated by boiling with Ac₂O (2 ml) and pyridine (4 ml). The mixture was worked up in the usual manner and after recrystallization from MeOH yielded II, mp 271~275°, IR 1740 and 1250 cm⁻¹ (acetate); NMR § 0.73(s, 3H), 0.85(s, 6H), 0.93(s, 6H), 0.99(s, 3H), 1.27(s, 3H), 2.05(s, 3H), 2.09(s, 3H),

4.5(t like, 1 H), 5.69(t like, 1 H), and 5.42(t like, 1 H). MS m/e 556(M⁺ t), 249(13), 246(100), 201(33), 191(20), 189(67), and 131(46).

Methyl Ester(III)—A sample of I (600 mg) was treated with diazomethane in the usual manner. After working up in the usual way, III was crystallized from M \sim OH as fine needles, mp 206 \sim 208°, [α]²⁰₀=+26.6° (c, 0.52 in MeOH), IR 1730 cm⁻¹(ester), MS m/e 486 (M $^+$ 25), 278(48), 260(100), 149(20), 207(13), 201(33), 189(67), and 131(46). I was recovered in 30% yield by saponification of III with 4% MeOH-KOH for 5 hr.

Methyl Diacetate(IV)—Methylation of II, or acetylation of III under similar condition as above gave IV, which was crystallized from MeOH to give fine needles, mp $197\sim199^\circ$, IR $1750 \,\mathrm{cm^{-1}}$ (ester) and $1240 \,\mathrm{cm^{-1}}$ (acetate), NMR δ 0.72(s, 3H), 0.86(s, 6H), 0.94 (s, 3H), 0.98(s, 6H), 1.25(s, 3H), 2.04(s, 3H), 2.08(s, 3H), 3.62(s, 3H), 4.50(t like 1H), 5.65(t like 1H), and 5.40(t like 1H).

CrO₃-Oxidation of the Methyl Diacetate(IV)—To a boiling solution of IV(70 mg) in HOAc(20ml), CrO₃(100 mg) in 85% HOAc(10ml) was added dropwise over a period of 1 hr. Refluxing was continued further for an additional 90 min and then the mixture was poured onto ice. The solids were filtered and chromatographed on silica gel. Elution with CHCl₃ gave V as needles, mp 220~223°, λ_{BSF}^{BOF} 250 nm (log ε , 4,00).

Bromolactone(VI) of II—To a solution of II(35 mg) and NaOAc(300 mg) in HOAc (10 ml) was added dropwise a solution of bromine in HOAc(3%, 2 ml). It was kept at room temperature for 2hr and mixture then poured into water(50 ml) containing Na₂S₂O₃ (1 g) to discharge excess bromine. The precipitate was filtered, washed thoroughly with water and dried. Crystallization from MeOH gave needles, mp 141~145°, IR 1780 cm⁻¹ (γ-lactone).

Isolation of Sterol Mixture—The Et₂O soluble fraction was saponified with 5% MeOH-KOH for 2hr. After filtering filtrate was diluted with water and extracted with Et₂O. The repeated recrystallization of the unsaponifiable matter from MeOH gave phytosterol as colorless plates, mp 136°, IR 3450 cm⁻¹ (OH), 970 cm^{-1} (trans-disubstituted double bond), and 825 cm^{-1} (trisubstituted double bond), MS m/e 414 (M⁺ 41.5 for Δ -stigmasterol), 412 (M⁺ 48.1 for α -spinasterol), 399(15), 397(21.6), 369(26.6), 351(23.2), 300(25), 273(56.4), 271 (100), 255(93.8), 246(44.8), and 231(39); NMR δ 0.54(s, 3 H), 0.79(s, 3 H), 1.02(d, 3 H), 3.50(b, 1 H), and 5.0~5.3(m, 2 H).

Gas-Liquid Chromatography of Sterol Mixture—The chromatographic column was 100 cm×3mm glass and contained 3% OV-17 on Chromosorb W(60~80 mesh). Column temp., 260°; injection temp. and detector temp., 275°; carrier gas, N₂(40 ml/min). Cholesterol was run with the samples. RRT, Cholesterol, 1;α-spinasterol, 1.72; Δ'-stigmasterol, 1.95.

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