

# Co-doped Gd<sub>2</sub>O<sub>3</sub> nanorods: Structure and magnetism

Sandeep K.S. Patel\*, Prasanta Dhak, Min-Kwan Kim,  
Jae-Hyeok Lee, Miyoung Kim, and Sang-Koog Kim

National Creative Research Initiative Center for Spin Dynamics and Spin-Wave Devices, Nanospinics Laboratory,  
Research Institute of Advanced Materials, Department of Materials Science and Engineering,  
Seoul National University, Seoul 151-744, South Korea

## 1. Introduction

Diluted magnetic semiconductors (DMSs) have attracted a considerable amount of attention owing to their promising applications in spintronics and magneto-optics [1,2]. Over the past decades, the enhancement of DMS magnetic properties such as Curie temperature has been the focus of intense research. Curiously, in this context, TM-ion-doped rare-earth (R<sup>3+</sup>) oxides have yet to be given significant consideration. Binary rare-earth oxides are the most stable rare-earth compounds, in which rare-earth ions typically hold the trivalent state. In TM ions, magnetic moment arises from the partially filled outermost 3d electrons, whereas in R<sup>3+</sup> ions, magnetic moment arises from the inner 4f incomplete sub-shell. Considering the importance of R<sup>3+</sup> rare-earth oxides, particularly the broad interest in Gd-based oxides, Gd<sub>2</sub>O<sub>3</sub> may be an advanced potential DMS candidate after doping with Co<sup>2+</sup> ions. The ferromagnetism of the Gd atom below 289 K drives the study of Gd<sub>2</sub>O<sub>3</sub>'s magnetic properties.

## 2. Experimental method and Results

We synthesized Gd<sub>2-x</sub>Co<sub>x</sub>O<sub>3</sub> ( $x = 0, 0.02, 0.04, 0.06$ ) nanorods via the following hydrothermal procedure. First, Gd<sub>2</sub>O<sub>3</sub> powders and stoichiometric amounts of Co(NO<sub>3</sub>)<sub>2</sub> · 6H<sub>2</sub>O were dissolved in diluted nitric acid solution. Then, 25% NH<sub>4</sub>OH solution was slowly added to the above solution under vigorous stirring, adjusting the pH value of the new colloidal solution between 10 and 12. The hydrothermal reaction was conducted at 190°C for 20 h in a Teflon-lined autoclave. The solid products were collected by filtration and washed with distilled water. Subsequent 3 h heat treatment at 550°C induced the formation of Co-doped Gd<sub>2</sub>O<sub>3</sub> nanorods. From FESEM and TEM images, the morphology is rod-shaped with less than 100 nm diameter and 200 nm–2 μm long. With increasing Co-doping concentration, the length and diameter of the nanorods decreases. EDS elemental mappings shows the uniform distribution of the Gd, Co, and O elements in the crystal.

## 3. Discussion and Conclusions

XRD, Raman, XPS, and TEM measurements showed the samples to have a single cubic phase structure of Gd<sub>2</sub>O<sub>3</sub> doped with Co<sup>2+</sup> cations, without any cobalt clusters. The XPS spectra revealed the Co ions to be in divalent Co<sup>2+</sup> states and in octahedral symmetry. All the samples exhibited mostly paramagnetism along with very weak antiferromagnetism; this was due to the non-interacting nature of the Gd<sup>3+</sup> ions which, incapable of undergoing strong ferromagnetic interaction with the neighboring ions. The samples also exhibited negative Curie–Weiss temperature due to the super-exchange interaction of Gd<sup>3+</sup> and Co<sup>2+</sup> ions via O<sup>2-</sup> ions and coupling between Co<sup>2+</sup>–Co<sup>2+</sup> pairs, indicating the presence of very weak antiferromagnetism in the Gd<sub>2-x</sub>Co<sub>x</sub>O<sub>3</sub> ( $x = 0, 0.02, 0.04, 0.06$ ) nanorods.

#### 4. References

- [1] H. Ohno, Science **281**, 951 (1998).
- [2] S. A. Wolf *et al.*, Science **294**, 1488 (2001).