

# A novel sonochemical approach for the synthesis of core/shell $\text{Fe}_3\text{O}_4/\text{SiO}_2/\text{Ag}$ nanocubes and $\text{SiO}_2/\text{Ag}$ nanospheres with superior catalytic properties

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A novel sonochemical approach was developed for the synthesis of different core/shell structures of  $\text{Fe}_3\text{O}_4/\text{SiO}_2/\text{Ag}$  nanocubes and  $\text{SiO}_2/\text{Ag}$  nanospheres. The total reaction time of the three sonochemical steps for the synthesis of  $\text{Fe}_3\text{O}_4/\text{SiO}_2/\text{Ag}$  nanocubes is shorter than the previously reported methods. A proposed reaction mechanism for the sonochemical functionalization of the silica and the silver on the surface of magnetic nanocubes was discussed in details. Transmission electron microscopy revealed that small Ag nanoparticles of approximately 10-20 nm in size decorated on the surface of  $\text{Fe}_3\text{O}_4/\text{SiO}_2$  nanocubes, and the energy dispersive spectroscopy mapping analysis confirmed the morphology of the structure. Additionally, X-ray diffraction data were used to confirm the formation of both phases of a cubic inverse spinel structure for  $\text{Fe}_3\text{O}_4$  and bcc structures for Ag in the core/shell structure of the  $\text{Fe}_3\text{O}_4/\text{SiO}_2/\text{Ag}$  nanocubes. The as-synthesized  $\text{Fe}_3\text{O}_4/\text{SiO}_2/\text{Ag}$  nanocubes displayed a high efficiency in the catalytic reduction reaction of 4-nitroaniline to 4-phenylenediamine, and with better performance than both Ag and  $\text{SiO}_2/\text{Ag}$  nanoparticles. The grafted silver catalyst was recycled and reused at least fifteen times without a significant loss of catalytic efficiency.