

Photocatalytic Activity of Hierarchical N doped TiO₂ Nanostructures

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Hierarchical N doped TiO₂ nanostructured catalyst with micro, meso and macro porosity have been synthesized by a facile self-formation route using ammonia and titanium isopropoxide precursor. The samples were calcined in different calcination temperature ranging from 300°C to 800°C at slow heating rate (5°C/min) and designated as NHPT-300 to NHPT-800. TiO₂ nanostructured catalyst have been characterized by physico-chemical and spectroscopy methods to explore the structural, electronic and optical properties. UV-Vis diffuse reflectance spectra confirmed the red shift and band gap narrowing due to the doping of N species in TiO₂ nanoporous catalyst. Hierarchical macro porosity with fibrous channel patterning was observed (confirmed from FESEM) and well preserved even after calcination at 800°C, indicating the thermal stability. BET results showed that micro and mesoporosity was lost after 500°C calcination. The photocatalytic activity has been evaluated for methanol oxidation to formaldehyde in visible light. The enhanced photocatalytic activity is attributed to combined synergetic effect of N doping for visible light absorption, micro and mesoporosity for increase of effective surface area and light harvestation, and hierarchical macroporous fibrous structure for multiple reflection and effective charge transfer.

Keywords: Photocatalysis, Hierarchical porous material, Visible light, Nitrogen-doped TiO₂