# Study on the surface characteristics of parylene-C films in inductively coupled O<sub>2</sub>/CF<sub>4</sub> gas plasma

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#### **Abstract**

In this article, we reported the results of etching polymonochloro-para-xylylene (parylene-C) thin films using inductively coupled plasma and  $CF_4/O_2$  gas mixture. The  $CF_4$  gas fraction increased up to the approximately 16 %, the polymer etch rate increased in the range of 277 – 373 nm/min. It confirmed that the etch rate of the parylene-C mainly depended on the O radical density in the plasma. Using a contact angle measurement, the contact angle increased with increasing the  $CF_4$  fraction. Moreover, the contact angle was highly related a  $CF_x$  functional group on parylene films.

### 1. Introduction

Organic Thin Film Transistors (OTFTs) have been investigated for possible use in low-cost, large area, flexible electronic applications during many years.[1] Simultaneously, OTFT devices have been required to be scaled down.

Parvlene-C (poly-monochloro-para-xylylene) more suitable organic material for gate dielectric in OTFTs. A patterning process for the organic film is required in order to apply it to OTFTs. Specifically, the patterning of organic semiconductors is necessary for eliminating parasitic leakage, reducing cross talk in order to achieve high on/off ratios, and removing material from the optical path.[2] Patterning process of organic films has been widely carried out by O<sub>2</sub> plasma. However, it is well known that an oxygen plasma can cause damage the semiconductor and dielectric layer due to the interaction between the oxygen radials and polymer resulting in the formation of several functional groups on the polymer surface.[3] These functional groups can rotate into the bulk of the material and move away from the polymer

surface.[4] As a result, functional groups lead to degradation of transistor performance. Hence, preventing the functional groups is essential to improve the OTFTs' performance. Research in this field has been so far studied by many researching groups. However, various gas chemistries have not been widely examined to reduce the formation of the functional groups, even though the reaction of the polymer with the plasma-created species makes a large effect on the formation of the functional groups.

In this work, in order to obtain more information about the effect of  $CF_4$  gas addition to oxygen plasma, we investigated the plasma treated surface of parylene-C films and the plasma characteristics in inductively coupled  $O_2/CF_4$  mixing gas plasma.

## 2. Experimental

The 300-nm thick parylene-C films were deposited on Si (100) substrates by chemical vapor deposition. The wafers were then plasma-treated with a radio-frequency (RF) etching system.

Etching experiments were performed in a planar Inductively Coupled Plasma (ICP) reactor, which was described detail in our previous work. [5] A Quadrupole Mass Spectrometry (QMS) (HPR-30, Hiden Analytical) was performed in Residual Gas Analysis (RGA) mode, delivering data on the neutral species only. The etching conditions were: a total gas flow rate of 60 sccm; an operating pressure of 4 mTorr; an input ICP power of 300 W and; a bias power of 30 W. The gas mixing ratios were varied by adjusting the partial flow rate of the mixture components and were set in the range of 0 – 50 % CF<sub>4</sub> in the CF<sub>4</sub>/O<sub>2</sub> mixture. The parylene-C film samples used in this study were 2 × 2

cm² in area. In order to determine the etch rate, the etched depths of the parylene-C film were measured using a surface profiler (Alpha-step 500, Tencor). X-ray Photoelectron Spectra (XPS) were extracted using a VG Scientific ESCALAB 200R XPS system with Al (K $\alpha$ ) (1486.6 eV) radiation operating at 260 W. The binding energy was calibrated using C 1s peak at 284.5 eV. Static contact angles for water and diiodomethane were measured using a sessile drop method with a contact angle goniometer (SEO Phoenix 300) at room temperature.

#### 3. Results and discussion

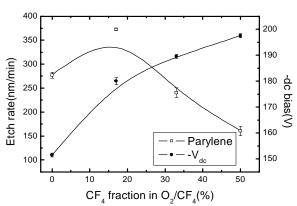


Fig. 1 Parylene-C etch rate and negative dc bias as functions of the  $CF_4$  gas fraction in the  $CF_4/O_2$  plasma

Figure 1 shows the parylene-C etch rate and negative dc bias as a function of the  $CF_4$  gas fraction in the  $CF_4/O_2$  plasma. From Fig. 1, it can be seen that, as the  $CF_4$  gas fraction increases up to the approximately 16 %, the polymer etch rate increases from 277 to 373 nm/min.

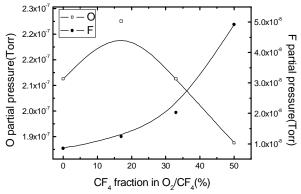


Fig. 2 O and F partial pressures as a function of the CF<sub>4</sub> gas fraction

As the CF<sub>4</sub> fraction of over 16%, the etch rate of the

parylene-C rapidly decreases. However, the negative dc bias increases monotonically up to 50% of  $CF_4$  fraction and was -197 V in 50%  $CF_4$  mixture.

Figure 2 shows the variation of partial pressure for each neutral species of O and F in the  $CF_4/O_2$  plasmas which were obtained by the QMS. From Fig. 2, we can observe that the change of O radical density shows a very similar trend as that of polymer etch rates in the  $CF_4/O_2$  plasma. Accordingly, the polymer etch rate is closely dependent on the O radical density in the  $CF_4/O_2$  plasma.

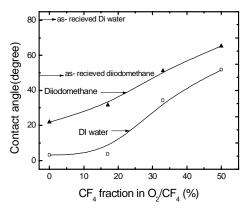


Fig. 3 Contact angles as a function of the CF<sub>4</sub> gas fraction

Figure 3 shows the contact angle and the surface energy with increasing  $CF_4$  fraction. In Fig. 3, we can confirm that the contact angle increases with  $CF_4$  fraction. For D.I. water, the contact angle increases from 3.1 to 51.8° with  $CF_4$  fraction in the range of 0 to 50%, respectively. At the same time, for diiodomethane solution, the contact angle increases from 21.9 to 65.3°, respectively.

Figure 4 represents the deconvolution of C 1s spectrum. As shown in Fig. 4(a), we can observe that the carbon peak is composed of three different carbon species in pure  $O_2$  plasma treatment. Each peak basically corresponds to C-C/H, C=O (or C-O), and O-C=O bonds which are observed at 284.5, 286.3, and 288.2 eV, respectively. [6] At the  $O_2/CF_4$  plasma, new peak was observed at the binding energy of approximately 291 eV in Fig. 4(b). In generally, the fluorine atom was formed to  $CF_x$  group on organic materials. So, peak of 291 eV is surely related to  $CF_4$  gas addition to the  $O_2$  plasma which corresponds to  $CF_x$  group.

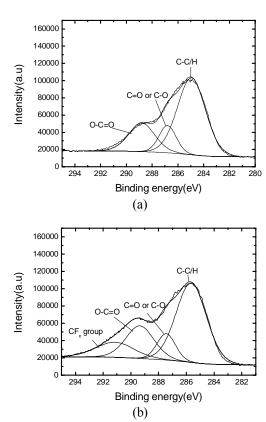


Fig. 4 Deconvolution of a C 1s spectrum at (a) pure-O<sub>2</sub> and (b) 50% CF<sub>4</sub> fraction

It is also known that the contact angle for  $CF_x$  groups is higher than that of the functional groups such as O-C=O and C=O (or C-O) bonds. Therefore, we can found that the formation of  $CF_x$  functional group is surely influenced to the contact angle in Fig. 3.

## 4. Summary

In this article, the results were obtained from the etch of poly-monochloro-para-xylylene (parylene-C) thin film, which was carried out in ICP etch system using  $CF_4/O_2$  gas mixture. The effects of  $CF_4$  addition to  $O_2$  plasma on etch rates were investigated. The  $CF_4$  gas fraction increased up to the approximately 16 %, the polymer etch rate increased in the range of 277 – 373 nm/sec. In contrast, the etch rate rapidly decreased with increasing the  $CF_4$  fraction over 16 %. Based on the QMS data, it was confirmed that the etch rate of the parylene-C mainly depended on the O radical density in  $CF_4/O_2$  plasma.

By a contact angle measurement, the contact angle increased with increasing the  $CF_4$  fraction. It is related to the generation of a  $CF_x$  functional group on parylene film after  $CF_4$  addition in  $CF_4/O_2$  plasma.

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#### 5. References

- 1. M. E. Roberts, S. C. B. Mannsfeld, R. M. Stoltenberg, and Z. Bao, Org. Electronics. 10, 377 (2009).
- 2. I. Kyissis, C. D. Dimitrakopoulos, and S. J. Purushothaman, J. Vac. Sci. Technol. B. 20, 956 (2002).
- 3. C. M. Chan, T.M. Ko, and H. Hiraoka, Surf. Sci. Rep. 24, 1 (1996).
- 4. W. J. Brennan, W.J. Feast, H.S. Munro and S.A. Walker, Polymer. 32, 1527 (1991).
- 5. M. Kim, N.-K. Min, S. J. Yun, H. W. Lee, A. Efremov, and K.-H. Kwon, Microelectron. Eng. 85, 348 (2008).
- D. A. Shutov, S.-Y. Kang, K.-H. Baek, K. S. Suh, and K.-H. Kwon, Jpn. J. Appl. Phys. 47, 6970 (2008).