Thermoremanent Magnetization in Er_{1-x}Tm_xFe₂O₄

J. Kim, J. S. Ahn, and B. W. Lee*

Department of Physics, Hankuk University of Foreign Studies, Yongin Kyugki-do 449-791, Korea *Corresponding author: B. W. Lee, e-mail: bwlee@hufs.ac.kr

Rare-earth iron oxides of the form RFe₂O₄ (R=Y, Er, Tm, Yb and Lu) have a rhombohedral structure with the space group R3m [1]. It is composed of the alternate stacking of the hexagonal Fe₂O_{2.5} layer (W-layer) and the hexagonal RO_{1.5} layer (U-layer) along the c-axis. Strong magnetic interactions between the localized Fe moments give rise to magnetic ordering below 250 K [2-5]. The thermoremanent magnetization (TRM) is observed in R=Tm, Yb and Lu [4, 5]. In order to understand origin of TRM, we have studied the magnetic properties of Tm substituted ErFe₂O₄. We prepared stoichiometric polycrystalline samples by solid-state reaction. Figure 1 shows the temperature dependence of field-cooled (FC) and zero-field-cooled (ZFC) magnetization for ErFe₂O₄ and Er_{0.5}Tm_{0.1}Fe₂O₄ with an applied magnetic field of 100 Oe. The field-cooling effect is observed in Er_{0.9}Tm_{0.1}Fe₂O₄, but not in ErFe₂O₄.



Fig. 1. Temperature dependence of magnetization for ErFe₂O₄ and Er_{0.9}Tm_{0.1}Fe₂O₄.

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Influence of Heat Treatment on the Superconductivities of MgB₂ Bulk Sintered in Different Processing Atmosphere

Y. Zhang*, S. H. Zhou, and S. X. Dou

Institute for Superconducting and Electronic Materials, University of Wollongong, Northfields Ave. Wollongong, NSW 2522, Australia

* Corresponding author: Author 1 Zhang, e-mail: yz268@u ow.edu.au

The effect of processing atmosphere and heat treatment on critical current density, J_c , was studied for MgB₂ bulk samples made using the in situ technique under argon atmosphere with three different purities, ultra-high, high, and welding grade. The critical temperature, T_c , and the amount of MgO show little variation for the samples treated under the three different argon atmospheres. The J_c for the sample processed under the welding grade argon has much weaker field dependence in high field region than that for the sample processed under the welding grade argon possessed small grains, and XRD refinement analysis revealed larger crystalline strains, to all of which the improvement in high field J_c of this sample can be attributed, because they are believed to act as effective pinning centres. These results verify the dual reaction model where the MgB₂ formation and the reaction between oxygen and precursor take place simultaneously, resulting in an optimal doping effect.



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