

### Machining property of Cobalt doped glass by using Nd:YAG(532nm) laser

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Suitable glasses for laser micromachining using a Nd:YAG laser have been developed. The composition of glasses is  $(70-x)\text{SiO}_2-x\text{B}_2\text{O}_3-10\text{Na}_2\text{O}-10\text{CaO}$  ( $x=0, 10, 15, 20, 25, 30$ ). These glasses contain the Co ions in glass matrix (adding to the mother glass). After machining with the 2nd harmonic beams of a pulsed Nd:YAG laser, the bump in the glass host, the pit at the bump in the glass host and the sole pit were(was) formed. This report was conducted for researching as for relationship between bonding intensity of glass and thermo-mechanical properties, and for finding out the possibility of the fabrication of micro lens array (MLA). The maximum height of bump in the glass was  $16\mu\text{m}$  and a diameter was  $262.3\mu\text{m}$ . We found that the Co ions acted as sensitizers in laser machining of the glasses and this process can be used for MLA fabrication process.

**Keywords:** Glass, Cobalt, Laser, Machining

### Nucleation and Crystallization of $\text{SiO}_2\text{-Na}_2\text{O-CaO-P}_2\text{O}_5$ biological glass by Differential Thermal Analysis

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Bioactive glasses of the system  $\text{SiO}_2\text{-Na}_2\text{O-CaO-P}_2\text{O}_5$  have been prepared by melting and annealing technique. The nucleation and crystallization mechanisms of a  $46\text{Na}_2\text{O-27CaO-25SiO}_2\text{-2P}_2\text{O}_5$ (mol%) glass was investigated by differential Thermal analysis (DTA) and X-ray diffraction analysis. The average activation energy for crystallization,  $E$ , was determined to be  $143\pm 10\text{KJ/mol}$  by the Kissinger method and  $141\pm 10\text{KJ/mol}$  by Augis-Bennett method. The Avrami constants were  $1.02\leq n\leq 1.2$ , and it appeared one direction growth. A nucleation-rate-type curve was determined by plotting either the reciprocal of the temperature corresponding to the crystallization peak maximum,  $1/T_p$ , or the height of the crystallization peak,  $(\delta T)_p$ , as a function of nucleation temperature,  $T_n$ . The temperature where nucleation can occur for this glass ranges from  $550$  to  $620^\circ\text{C}$  and the temperature of maximum nucleation rate is at  $600\pm 10^\circ\text{C}$ .

**Keywords:** Crystallization kinetics, Bioactive glass, Glass-ceramics, Surface nucleation