

Charge/Discharge Properties of Carbon Added LiFePO₄

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Abstract : Phospho-olivine LiFePO₄ cathode materials were prepared by hydrothermal reaction. Carbon black was added to enhance the electrical conductivity of LiFePO₄. The structural and morphological performance of LiFePO₄ and LiFePO₄-C powders were characterized by X-ray diffraction (XRD) and FE-SEM. LiFePO₄/SPE/Li and LiFePO₄-C/SPE/Li cells were characterized electrochemically by charge/discharge experiments. The results showed that the discharge capacity of LiFePO₄/SPE/Li cell was 103 mAh/g at the first cycle. The discharge capacity of LiFePO₄-C/SPE/Li cell with 5 wt% carbon black was the largest among LiFePO₄-C/SPE/Li cells, 126 mAh/g at the first cycle and 123 mAh/g after 30 cycles, respectively. It was demonstrated that cycling performance of LiFePO₄-C/SPE/Li cell with 5 wt% carbon black was better than that of LiFePO₄/SPE/Li cell.

Key Words : LiFePO₄, Hydrothermal, Carbon, Properties

1. Introduction

Recently, lithium secondary batteries have been widely used in devices such as electrical appliances due to high energy density, perfect cyclic performance and excellent capacity retention. The active cathode material, such as Li_xCoO₂ or LiMnO₂ which provide high potentials and high capacities over 150mAh/g. However, such as Li_xCoO₂ was a high cost material and not compatible with environment [1]. Since the pioneering works of Padhi et al., mixed ortho- phosphates LiMPO₄(M=Co, Ni, Mn, Fe) isostructural to olivine have been intensively studied as lithium insertion cathode materials for the next generation of lithium secondary batteries[2-3]. The choice of the cathode material has high capacity that can be retained for up to long cycles and also low cost and eco-friendly materials. Among LiMPO₄ series of materials, LiFePO₄ is a low cost material and highly compatible to the environment. However, it is difficult to attain its full capacity(170mAh g⁻¹), because its electronic conductivity is very low and diffusion of Li⁺ ion in the olivine structure is low [4-5].

In this paper, phospho-olivine LiFePO₄ was prepared by hydrothermal reaction and carbon nanotube added for enhance the electrical conductivity of LiFePO₄.

2. Experimental

LiFePO₄ was prepared with the starting materials of Lithium hydroxide monohydrate (LiOH · H₂O, 99.95%), Iron(II) sulfate heptahydrate (FeSO₄ · 7H₂O, 99.99+%), Phosphoric acid crystals (H₃PO₄, 99.999+%) and L-Ascorbic acid (C₆H₈O₆, 99+%) in a molar ratio for Li:Fe:P=3:1:1. Addition of ascorbic acid as a reducing agent to the precursor was useful in prohibiting the conversion of Fe²⁺ to Fe³⁺ during the hydrothermal reaction and generation of α-Fe₂O₃ during the annealing process. The mixed solution was heated at 150°C for 3h. After being cooled to room temperature, the solution was filtered to

separate the precipitate powder, and the powder was washed with ultrapure water and were parched for the vacuum drying at 100°C for 1h. The obtained powder was dried and the dried powders were further dried at 400°C in nitrogen atmosphere for 1h. In order to improve low electron conduction of LiFePO₄, carbon black(several weight percent) was added. The crystalline phases were identified with X-ray diffraction (XRD, Dmax/1200, Rigaku). The X-ray diffraction pattern was collected by a step-scanning mode in the range of 10-80° with a step time of 5°/min. The powder morphologies of electrodes before and after cycling were examined with a Hitachi S-4700 FE-SEM which had an accelerating voltage of 15 kV.

A composite electrode was prepared by mixing LiFePO₄, Sp270 and polyvinylidene fluoride (PVDF) in a weight ratio of 70:25:5. This mixture was coated onto an Al-foil, cut into 2cm² sections and heated at 110°C for 24h under vacuum. A lithium foil was used as an anode. The 25PVDF/LiClO₄EC₁₀PC₁₀ as SPE prepared by mixing of LiClO₄ and propylene carbonate (PC), ethylene carbonate (EC) in a mole ratio of 1:10:10, and polyvinylidene fluoride-hexafluoro propylene (kynar2801) was added. The coin type cell was fabricated for the electrochemical tests in a potential range of 2.5-4.3 V at a constant current density of 0.1 mA/cm².

3. Results and discussion

XRD patterns of LiFePO₄-C powders are shown in Fig. 1. All the patterns can be indexed to a single-phase material having an orthorhombic olivine-type structure (space group Pmnb), which are the same as the theoretical value. No impurity is found in all the LiFePO₄-C powders.

FE-SEM of LiFePO₄ and LiFePO₄ with carbon black 5wt% are shown in Fig. 2. As can be seen from Fig. 2(a), the average particle size of LiFePO₄ is around 200nm. It is shown from Fig. 2(b) the LiFePO₄ with carbon black 5wt% is around 150nm. The

smaller particles, which shorten the lithium ions diffusion distance between the surfaces and center during lithium intercalation and de-intercalation, are expected to contribute the enhanced electrochemical performance of the carbon black added LiFePO₄.

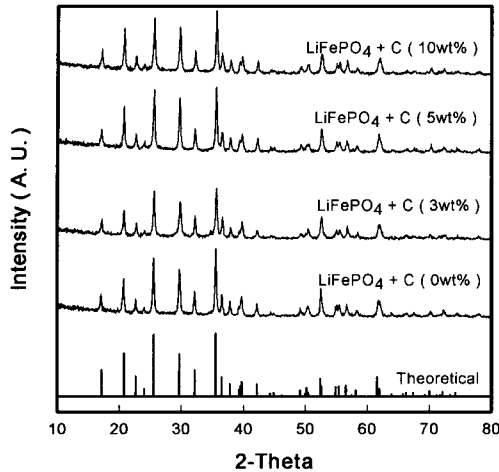


Fig. 1. XRD patterns of LiFePO₄ and LiFePO₄ with carbon black.

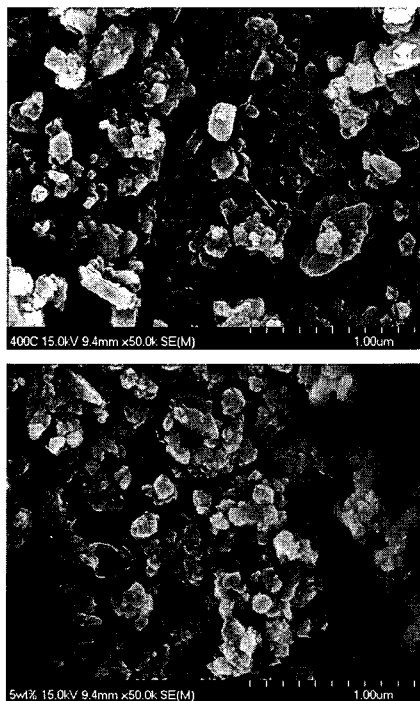


Fig. 2 FE-SEM images of (a) LiFePO₄ and (b) LiFePO₄ with carbon black 5wt%.

Cycling performance of LiFePO₄/SPE/Li and LiFePO₄-C(5wt%)/SPE/Li cells is shown in Fig. 3. As can be seen from Fig. 3, the discharge capacity of LiFePO₄/SPE/Li cell is 105 mAh/g at the first cycle. The initial discharge capacity of LiFePO₄ with 3 wt%, 5 wt% and 10 wt% carbon black is 126, 114 and 102 mAh/g, respectively. It is demonstrated that cycling performance of LiFePO₄-C/Li cell is better than that of LiFePO₄/Li cell and the discharge capacity is the largest when carbon content is 5 wt%.

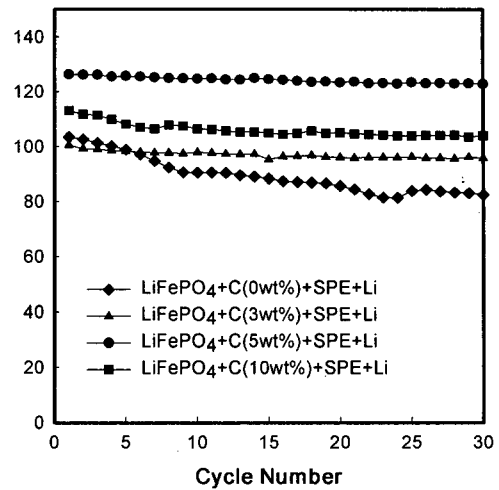


Fig.3. Cycling performance of LiFePO₄ added different amounts of carbon black.

4. Conclusions

Phospho-olivine LiFePO₄ cathode materials were prepared by hydrothermal reaction. Carbon black was added to enhance the electrical conductivity of LiFePO₄. The discharge capacity of LiFePO₄/SPE/Li cell is 105 mAh/g at the first cycle and with the carbon black 5wt% cell is the large among added other different amounts of carbon black, such as 3wt% and 10wt%. The discharge capacity of LiFePO₄-C(5wt%)/SPE/Li cell is 126 mAh/g at the first cycle and 123 mAh/g after 30 cycles, respectively. It is demonstrated that cycling performance of LiFePO₄-C/Li cell is better than that of LiFePO₄/Li cell and with 5wt% carbon black is best.

Acknowledgments

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