

Preparation and Luminescence Properties of LaPO₄:Eu/LaPO₄ Core/Shell Nanoparticles

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Abstract

Due to an efficient red emission, Eu³⁺ ions have doped in various host materials. Eu-doped LaPO₄/LaPO₄ core/shell nanophosphor which emits red light in visible range, was synthesized by using a solution system. The mean size of the nanoparticle observed by TEM was about 7-8 nm. In this core/shell system, the thickness of LaPO₄ shell was optimized on the basis of relative PL emission intensity. In addition, the luminescent properties of LaPO₄:Eu/LaPO₄ core/shell particle have been compared with Eu-doped LaPO₄ bulk particle and nanoparticle.

1. Introduction

The LaPO₄ has been already employed to host of various luminescence materials such as laser materials and phosphors in lighting applications and in versatile information displays. It is known that the LaPO₄ has a monoclinic phase of monazite structure crystallographically, wherein La³⁺ ion is nine-coordinated to oxygen atoms and forming polyhedron described pentagonal interpenetrating in geometric[3]. The size of La³⁺ ion is similar to those of other lanthanide ions and La³⁺ ion site can be easily substituted by them. The substitutions of Eu³⁺ or Tb³⁺ for a portion of La³⁺ construct luminescent phosphors and the incorporations of Nd³⁺, Pr³⁺ or Er³⁺ result in optical amplifying materials. Several researchers have reported the syntheses of LaPO₄ particles via diverse methods such as solid-state reaction, spray pyrolysis and hydrothermal method. In recent, Haase group have successfully fabricated Eu-, Tb- and Er-doped LaPO₄ nanocrystals by the liquid phase synthesis using high-boiling coordinating solvent[4-7].

On the other hand, rare earth ions doped nano-sized luminescent materials have been attracting much attention interest due to their high potential of becoming a new class of luminescent materials such

as phosphors in information displays and fluorescence lamp, optical

amplifier by their optical properties. A crystalline mixed rare-earth phosphate of varying composition, have recently been proposed as primary hosts. For example, Eu³⁺-doped LaPO₄ phosphor can be applied as a red component for display[1]. On the other hand, because one of major attributes of the nanoparticles is their size-dependent physicochemical properties. These nano-sized materials take advantage of size-induced changes in structural, optical and electronic properties to enhanced luminescent materials. On the other hand, in the Eu³⁺ doped LaPO₄ nanoparticle system, Eu³⁺ ions on the surface were predicted defect. Therefore, Eu³⁺ doped LaPO₄ nanoparticle surface needs constructing inorganic shell on there.

In the present work, we have prepared Eu³⁺-doped LaPO₄/LaPO₄ core/shell nano-phosphor by liquid-phase reaction for application as display materials. The synthesized nano-powders were characterized by using a XRD, TEM, FT-IR. In addition, the optimal composition was investigated on the basis of relative PL emission intensity.

2. Experimental

LaCl₃·7H₂O (99.999%, Aldrich) together with EuCl₃ in the appropriate molar ratio were dissolved in methanol and tributyl phosphate, and the methanol was subsequently removed with a rotary evaporator. Then, diphenyl ether was added into the solution. Under vacuum condition of 80 Torr, the water ingredients contained in the solution were distilled off. A freshly prepared 2M solution of crystalline phosphoric acid was mixed with trihexylamine and the mixture was added into the metal solution and heated up to 200 °C under dry nitrogen. After 16 h the heating was stopped and the hot solution has been cooled to room temperature. The 3/4 of reaction mixture which contains LaPO₄:Eu nanoparticles was removed from

the flask. The remaining 1/4 of the reaction mixture was used for preparation of core/shell nanoparticles. The pretreatment route of LaPO_4 shell has been applied a close resemblance $\text{LaPO}_4:\text{Eu}$ nanoparticles. The prepared LaPO_4 shell solution was blended with reaction mixture. This solution was loaded into the metering pump and added drop by drop over 2 h to the reaction mixture of $\text{LaPO}_4:\text{Eu}$ core and heated up to 200 under dry nitrogen. After 16 h the heating was stopped and the hot solution has been cooled to room temperature. Reaction mixture was diluted with MeOH and was filled into a ultrafiltration cell equipped with membrane filter. The solution was moved through the membrane under a nitrogen pressure and the purification route was repeated 4 times. Finally, the core/shell nanoparticles were obtained from the reaction mixture by removing the methanol with rotary evaporator.

3. Results and Discussion

We are preparing various nanoparticles that can be applied as novel materials for information displays. Here we report the synthesis of $\text{LaPO}_4:\text{Eu}/\text{LaPO}_4$ core/shell nanoparticles from solution system and their luminescence properties. The X-ray diffraction patterns of $\text{LaPO}_4:\text{Eu}/\text{LaPO}_4$ core/shell nanoparticles synthesized by liquid-phase reaction and bulk $\text{LaPO}_4:\text{Eu}$ particles calcined at 1100 for 12 h by solid state method are shown in Fig. 1. Although the XRD pattern of the core/shell nanoparticle is not clear, the XRD patterns are well matched with that of the $\text{LaPO}_4:\text{Eu}$ nanoparticles reported by Riwozki. The XRD patterns of the nanocrystal powder is similar to that of bulk $\text{LaPO}_4:\text{Eu}$ and is consistent with the monazite-structure of bulk LaPO_4 .

The TEM photograph of Fig. 2 shows morphology and size of the prepared LaPO_4 core/shell nanoparticle. It is confirmed that the size of the particles is about 7~8 nm with narrow size distribution. Also, the TEM image shows the $\text{LaPO}_4:\text{Eu}$ core/shell nanoparticles have well-dispersed without aggregation. The inset exhibits a higher magnification of a particle to illustrate the high crystallinity.

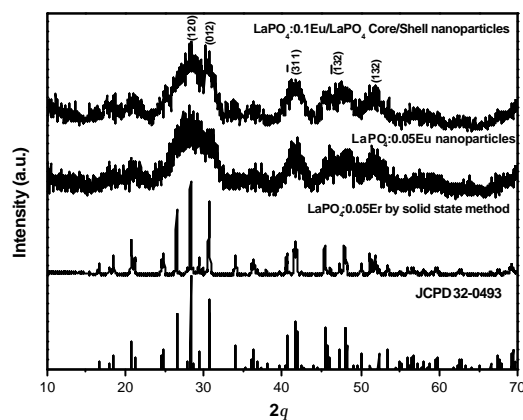


Fig. 1. XRD patterns of $\text{LaPO}_4:0.1\text{Eu}/\text{LaPO}_4$ Core/Shell nanoparticles, $\text{LaPO}_4:0.05\text{Eu}$ nanoparticles, $\text{LaPO}_4:0.05\text{Er}$ by solid state method and JCPDS file.

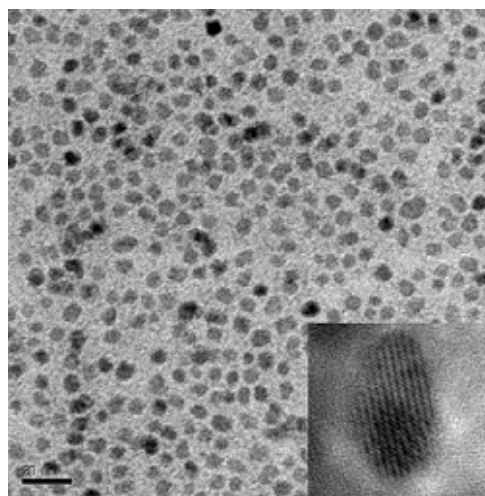


Fig. 2. TEM photograph of $\text{LaPO}_4:\text{Re}/\text{LaPO}_4$ core/shell nanoparticles.

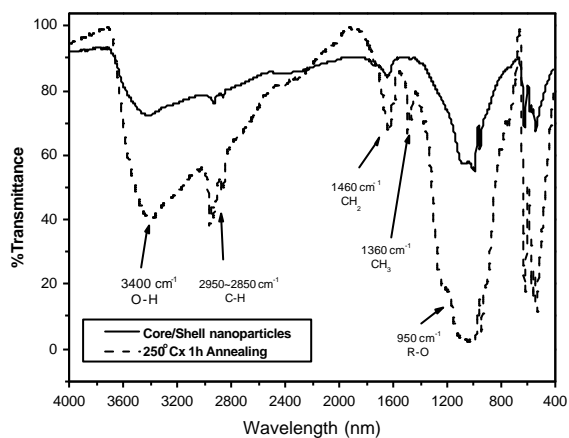


Fig. 3. FT-IR spectrum of $\text{LaPO}_4:\text{Re}$ core/shell nanoparticles

The composition of the surface of the prepared $\text{LaPO}_4:\text{Eu}/\text{LaPO}_4$ core/shell nanoparticles has been examined through FT-IR spectroscopy. As shown is given in Fig. 3. The broad absorption band at $3,400\text{ cm}^{-1}$ is assigned to O-H stretching vibration. Peaks at $2,950 \sim 2,850\text{ cm}^{-1}$ are due to C-H stretching vibration and at $1,460$ and $1,360\text{ cm}^{-1}$ are originated from bending vibrations of $-\text{CH}_2$ and $-\text{CH}_3$, respectively. In addition, some peaks between cm^{-1} and cm^{-1} are due to organic and inorganic PO_4^{3-} on the surface of the shell. This results could be considered that the organic molecules used in the reaction were adsorbed on the surface of the $\text{LaPO}_4:\text{Eu}/\text{LaPO}_4$ core/shell nanoparticles. Therefore, the surface of $\text{LaPO}_4:\text{Eu}/\text{LaPO}_4$ core/shell nanoparticles are capped by the used high b.p solvent molecules.

Fig. 4 shows the emission spectra in the range of $560\text{--}720\text{ nm}$ for $\text{LaPO}_4:\text{Eu}$ core/shell nanoparticles, $\text{LaPO}_4:\text{Eu}$ nanoparticles and $\text{LaPO}_4:0.1\text{Eu}$ bulk. The typical emission peaks due to the transition from $^5\text{D}_0$ to $^7\text{F}_j$ are observed. It was confirmed that the emission intensity of this core/shell nano-phosphor is almost similar to that of nano-phosphor. However, that is a little lower than the emission intensity of bulk phosphor.

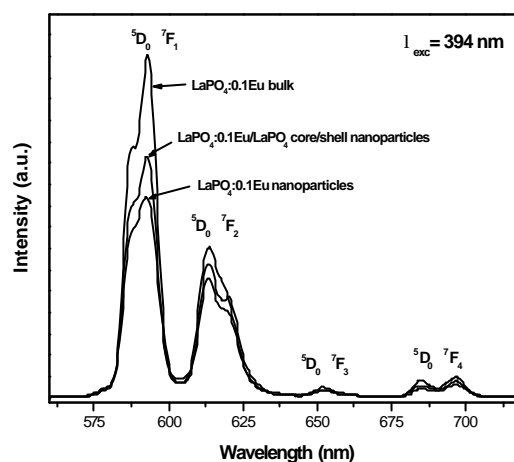


Fig. 4. Emission spectra of $\text{LaPO}_4:0.1\text{Eu}$ nanoparticles, $\text{LaPO}_4:0.1\text{Eu}/\text{LaPO}_4$ core/shell and $\text{LaPO}_4:0.1\text{Eu}$ bulk

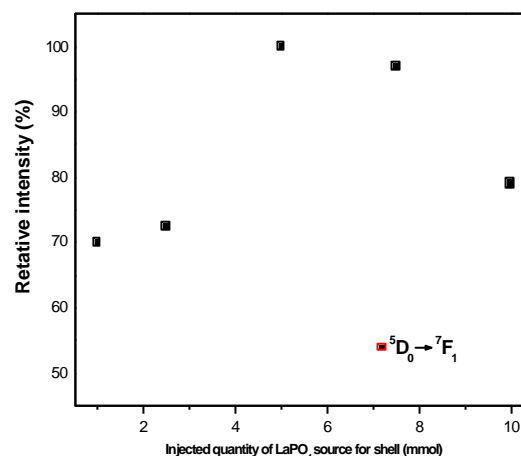


Fig. 5. Dependence of emission intensities on the contents of LaPO_4 shell in the $\text{LaPO}_4:0.1\text{Eu}/\text{LaPO}_4$ nanoparticles

The dependence of the relative emission intensities on the contents of LaPO_4 shell in Eu^{3+} doped $\text{LaPO}_4/\text{LaPO}_4$ core/shell nanoparticles is presented in Fig. 5. As can be seen, the emission intensity of the core/shell nanoparticles is increased with the

increasing contents of LaPO₄ shell. At the composition of LaPO₄:0.1Eu/LaPO₄ (5 mmol), a maximum emission intensity is achieved and then is decreased with increasing LaPO₄ shell contents. This is explained by the facts that the decreasing defect between Eu³⁺ ions concentration in the LaPO₄:Eu/LaPO₄ system.

4. Conclusions

In the present work, Eu-doped LaPO₄ core/shell nanoparticles by the designed liquid-phase reaction. The prepared core/shell nanoparticles were confirmed to be a single phase of monazite structure and to have the 7-8 nm of particles size with narrow size distribution. On the other hand, it was confirmed by FT-IR that the surface of nanoparticles are capped with the solvent molecules, which will possibly keep from the agglomerating between the nanoparticles. The optimum contents of LaPO₄ shell exhibiting the maximum emission intensity in the nanoparticles was about 5 mmol.

5. References

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6. Acknowledgement

This work was supported from Information Display R & D Center, one of the 21th Century frontier R & D Program funded by the Ministry of Commerce, Industry and Energy of Korea.