

포스터 발표 P-17

## 폴리 이미드 실록산과 PVP 혼합물로부터 유도된 고투과성 C-SiO<sub>2</sub> 막의 개발

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Development of high permeable C-SiO<sub>2</sub> membranes  
derived from poly (imide siloxane) / PVP blends

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### 1. Introduction

Carbon molecular sieve (CMS) membranes have superior gas permeation and separation performance compared with polymeric membranes<sup>1,3</sup>. Up to now, CMS membranes mostly have been mostly focused on the kinds of precursor and pyrolysis condition (pyrolysis temperature, heating rate, pyrolysis atmosphere). In this study, we prepared high permeable C-SiO<sub>2</sub> membranes derived from Si-polyimide/polyvinylpyrrolidone using polymer-blending method, and we will discuss the gas separation properties of the flat of composite type C-SiO<sub>2</sub> membranes.

### 2. Experimental Section

The chemical structure of poly(imide siloxane) (Si-PI) synthesized in this study is illustrated in Fig. 1. Polyvinylpyrrolidone (PVP, MW 10,000) was used as into pore forming agent..The thermal labile

polymer, PVP (2 and 5% by weight) was added to Si-poly(amic acid) solution, respectively.

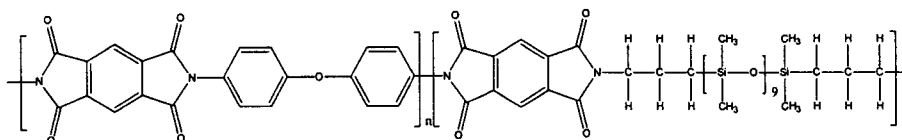


Fig 1. The structure of Si-polyimide

### 3. Results and Discussion

The Si-PI/PVP has distinguished thermal decomposition steps. The first decomposition was observed around 400–450 °C due to the thermal exclusion of PVP. The second decomposition began at 500 °C by the cleavage of imide ring in the siloxane domain. Finally, the degradation of imide ring was observed at 550 °C.

The gas permeation properties through the CMS membranes were investigated at 25 °C for single gas molecules [He (2.6 Å), CO<sub>2</sub> (3.3 Å), O<sub>2</sub> (3.46 Å), and N<sub>2</sub> (3.64 Å)].

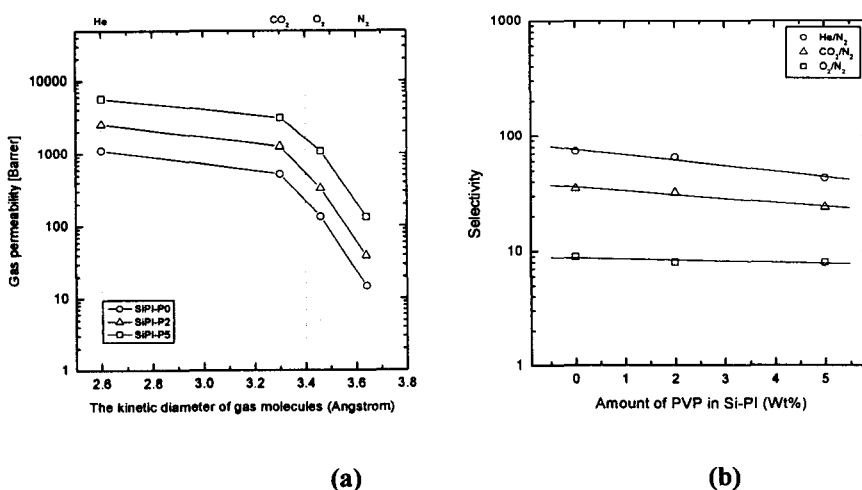


Fig. 2. (a) Gas permeabilities and (b) selectivities of C-SiO<sub>2</sub> membranes pyrolyzed at 550 °C as a function of the amount of PVP in Si-PI

As shown in Fig. 2, the C-SiO<sub>2</sub> membranes derived from Si-PI/PVP showed that the gas permeabilities increased with the amount of PVP and their selectivities showed tradeoff relations. Fig. 3 shows the gas permeabilities and selectivities of CMS membranes pyrolyzed with different pyrolysis temperatures (500, 550, 600, and 700 °C). As a result, their permeabilities and selectivities significantly affected by the pyrolysis temperature. Among them, CMS membranes pyrolyzed at 550 °C showed the highest permeabilities.

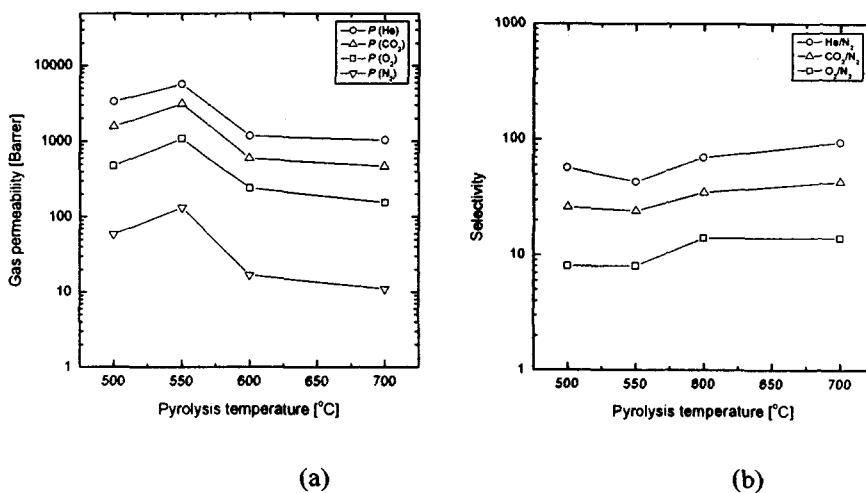


Fig. 3. (a)Gas permeabilities and (b)selectivities of C-SiO<sub>2</sub> membranes derived from Si-PI/PVP as a function of pyrolysis temperature

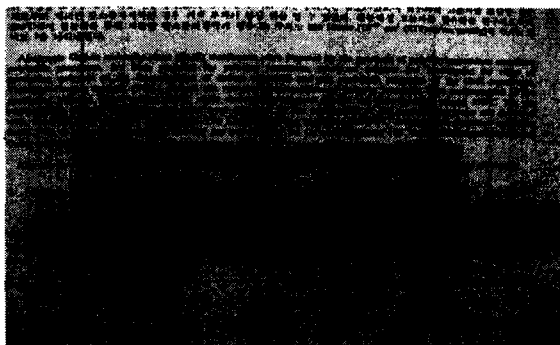


Fig 4. Dip-coated alumina tube and the pyrolyzed composite membrane.

Fig 4 show the dip-coated alumina tube. and the pyrolyzed composite membrane at 550 °C. Above all, to control the dip-coating time and speed to extract the alumina tube in the polymer solution is important to coat the alumina with the polymer solution.

#### 4. Conclusions

By the polymer-blending method, the gas permeabilities of C-SiO<sub>2</sub> membranes were improved because the thermal labile polymer acted as a pore forming agent during the pyrolysis. Up to now, the gas permeabilities and selectivities of CMS membranes mainly controlled by the pyrolysis conditions. However, the performances of CMS membranes should be improved by the polymer-blending method as confirmed in this study. Consequently, the C-SiO<sub>2</sub> membranes pyrolyzed at 550 °C showed O<sub>2</sub> permeability of 135 Barrers [ $10^{10}$  cm<sup>3</sup>(STP)cm/cm<sup>2</sup>·s·cmHg] and the O<sub>2</sub>/N<sub>2</sub> selectivity of 9 whereas the carbon-silica membranes prepared using the polymer-blending method showed O<sub>2</sub> permeability of 1080 Barrers and O<sub>2</sub>/N<sub>2</sub> selectivity of 8.

#### References

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